

# TECHNOLOGIE CCS aneb S OXIDEM UHLIČITÝM POD ZEM

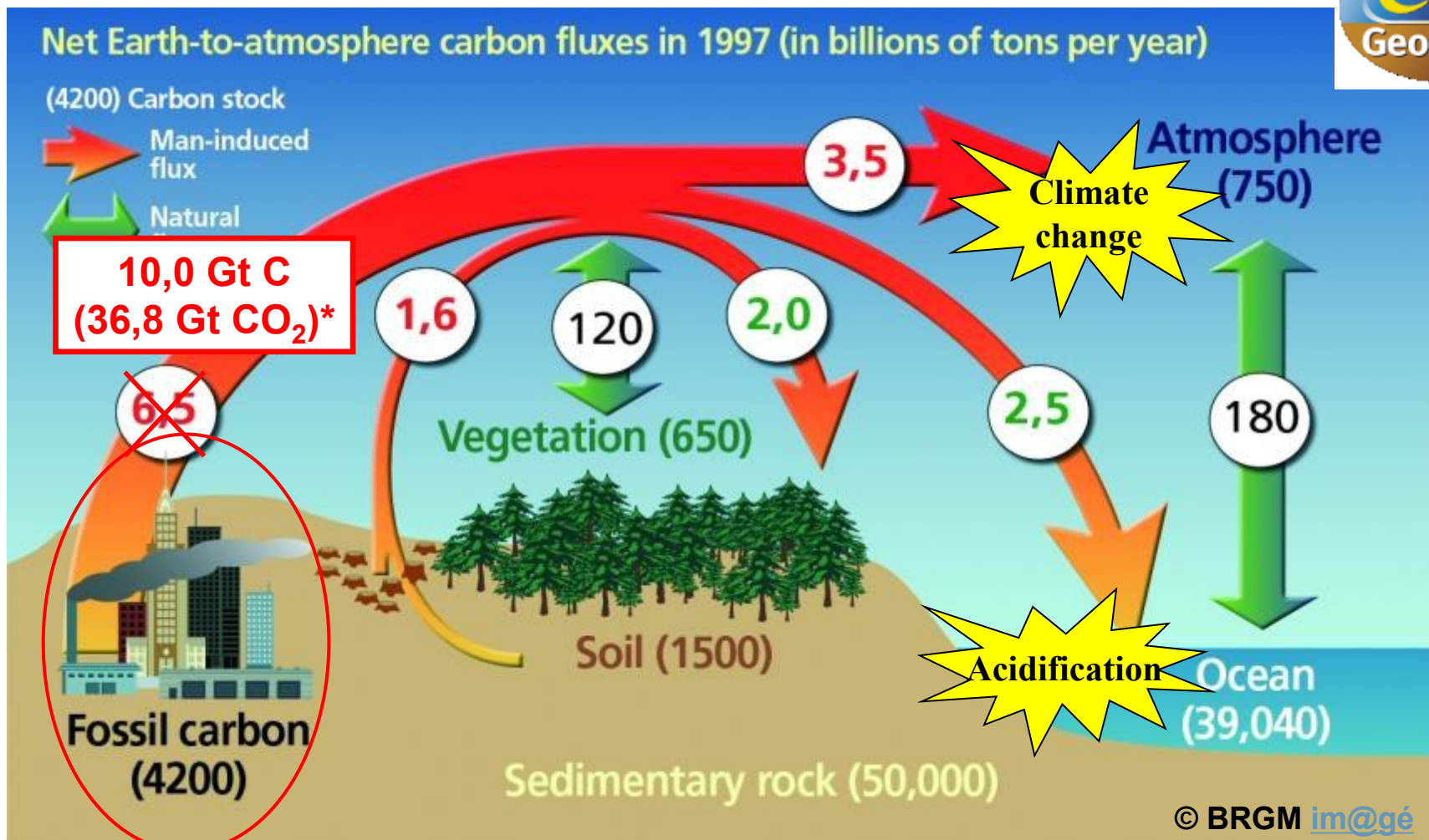
Vít Hladík (Česká geologická služba)

# Obsah přednášky

- Role CO<sub>2</sub> ve změně klimatu
- CCS jako mitigační technologie
- Princip technologie CCS – zachytávání, doprava, ukládání
- CCS ve světě
- Situace v ČR
- Ekonomické faktory
- Shrnutí

# Koloběh uhlíku mezi Zemí a atmosférou

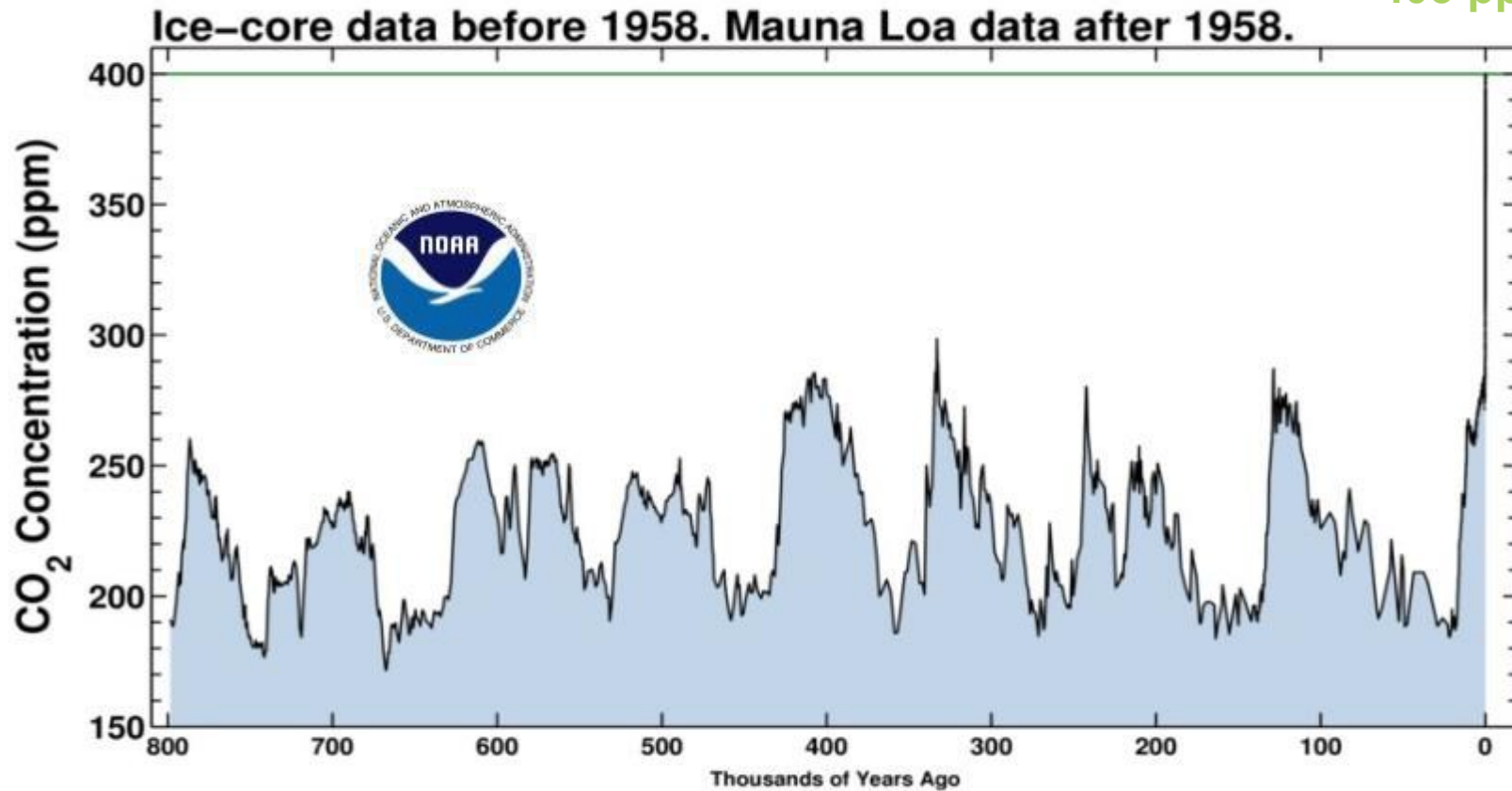
(v miliardách tun za rok)



\*odhad na rok 2017 podle Global Carbon Budget 2017

# Obsah CO<sub>2</sub> v atmosféře (ppm<sub>v</sub>)

403 ppm (2016)



# Pliocén (před 5.3 – 2.6 mil. lety)



- průměrná globální teplota o 3-4°C vyšší
- teplota na pólech o 10°C vyšší
- hladina moře o 5-40 m výše
- kontinentální drift 250 → 70 km od dnešní polohy
- vzniká Panamská šíje a Středozemní moře
- objevuje se australopithecus



The cumulative contributions to the global carbon budget from 1870  
 The carbon imbalance represents the gap in our current understanding of sources and sinks

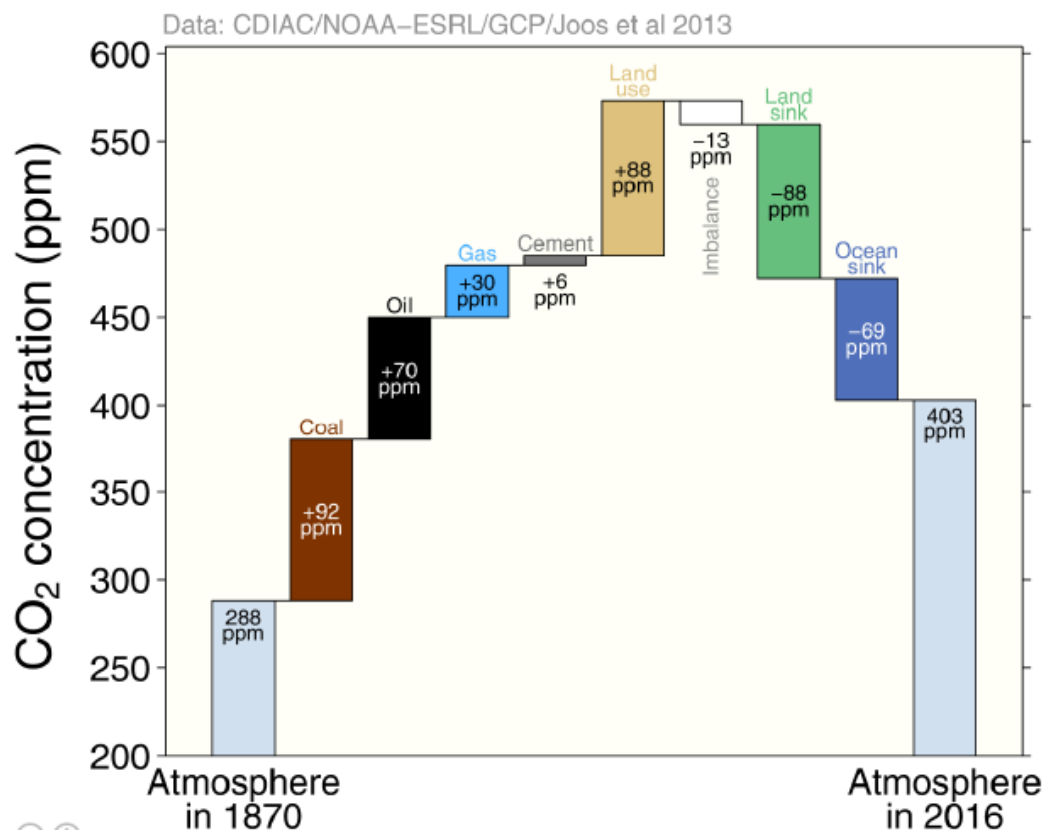


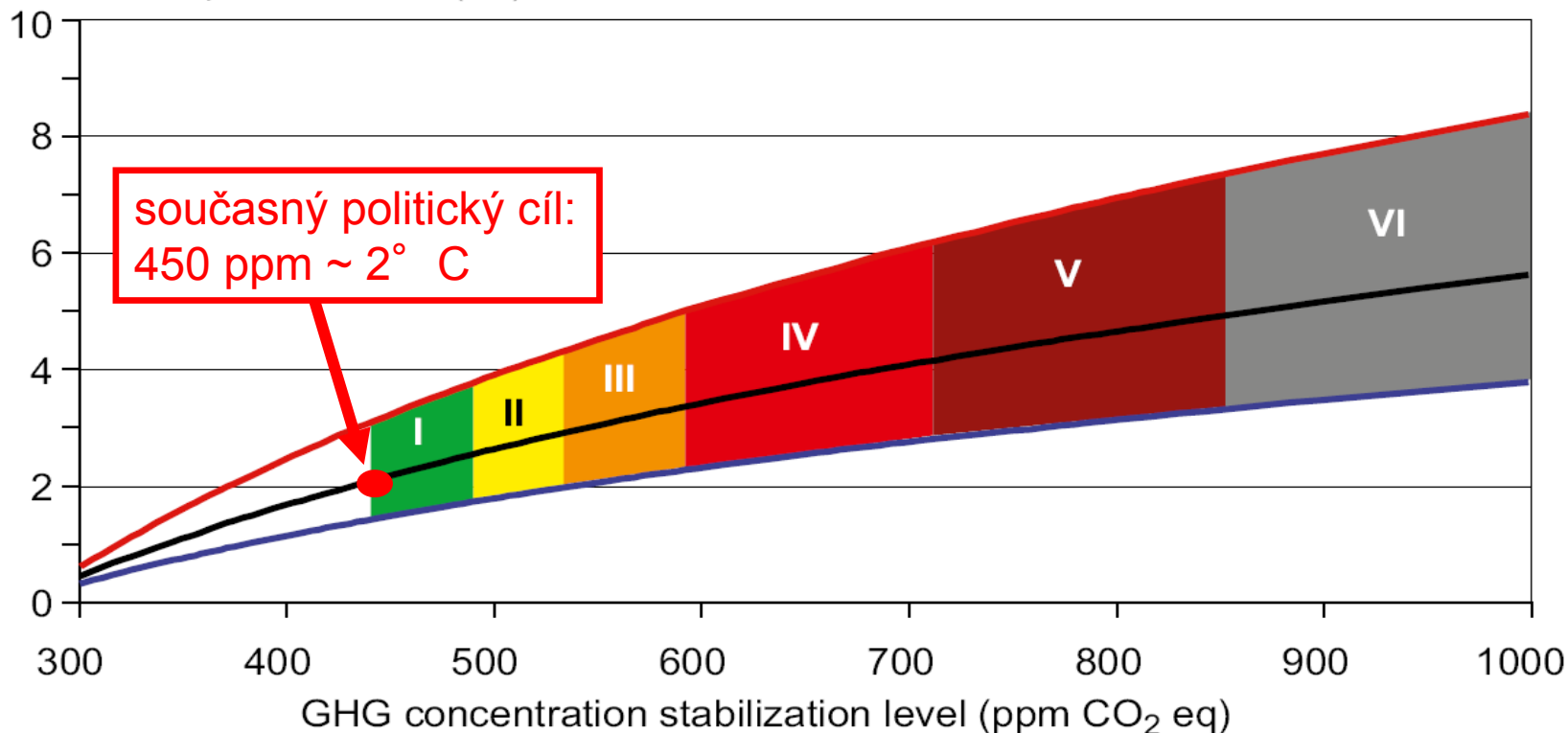
Figure concept from [Shrink That Footprint](#)

Source: [CDIAC](#); [NOAA-ESRL](#); [Houghton and Nassikas 2017](#); [Hansis et al 2015](#); [Joos et al 2013](#); [Khaliwala et al. 2013](#); [DeVries 2014](#); [Le Quéré et al 2017](#); [Global Carbon Budget 2016](#)

# Klimatické modelování – vztah mezi teplotou a obsahem CO<sub>2</sub> v atmosféře

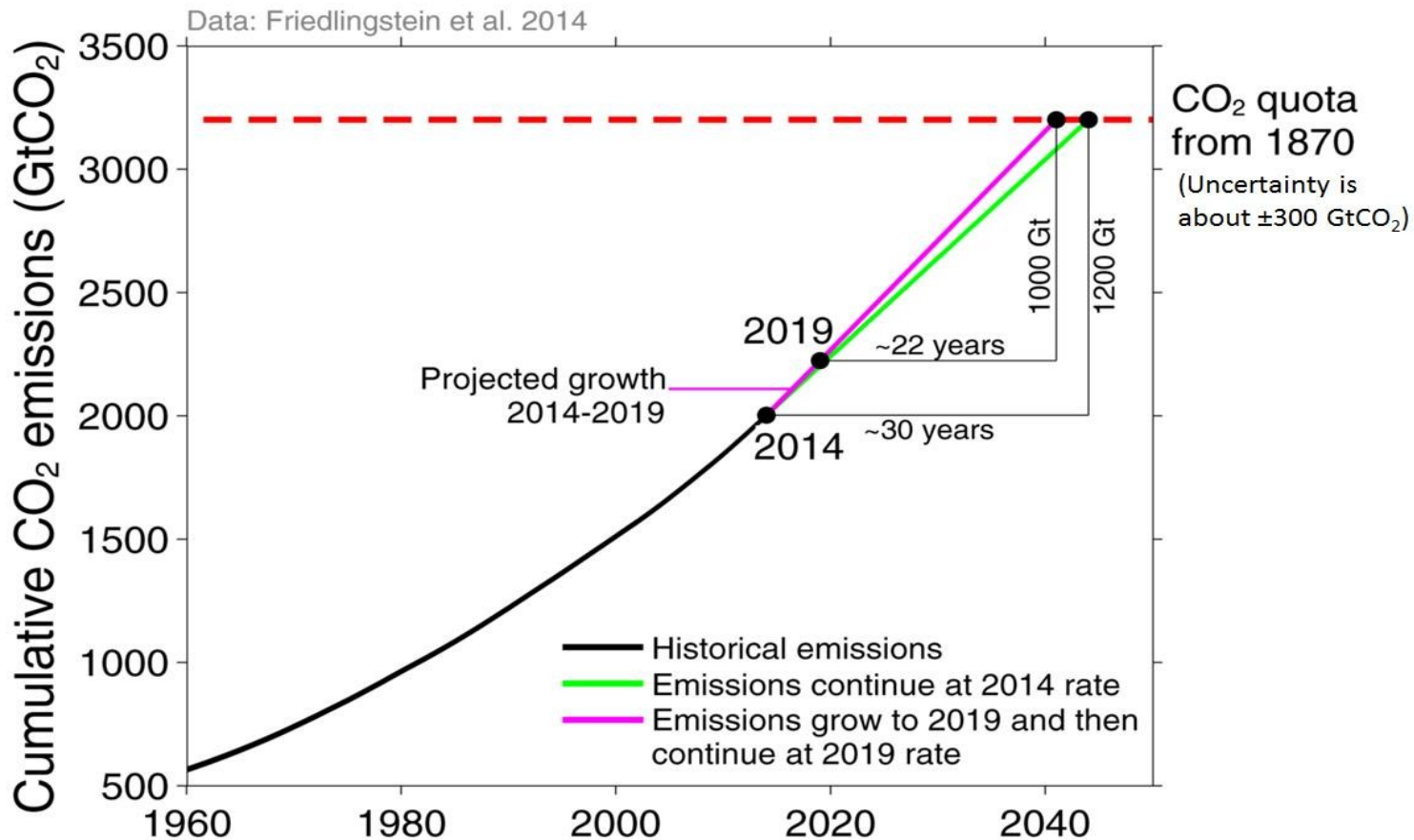
Equilibrium global mean temperature increase above pre-industrial (°C)

Source: IPCC 4th Assessment Report 2007



# Remaining emissions quota

Cumulative CO<sub>2</sub> emissions should remain below about 3200 Gt for a 66% chance of staying below 2°C  
 At present emissions rates the remaining budget would be used up in about 30 years

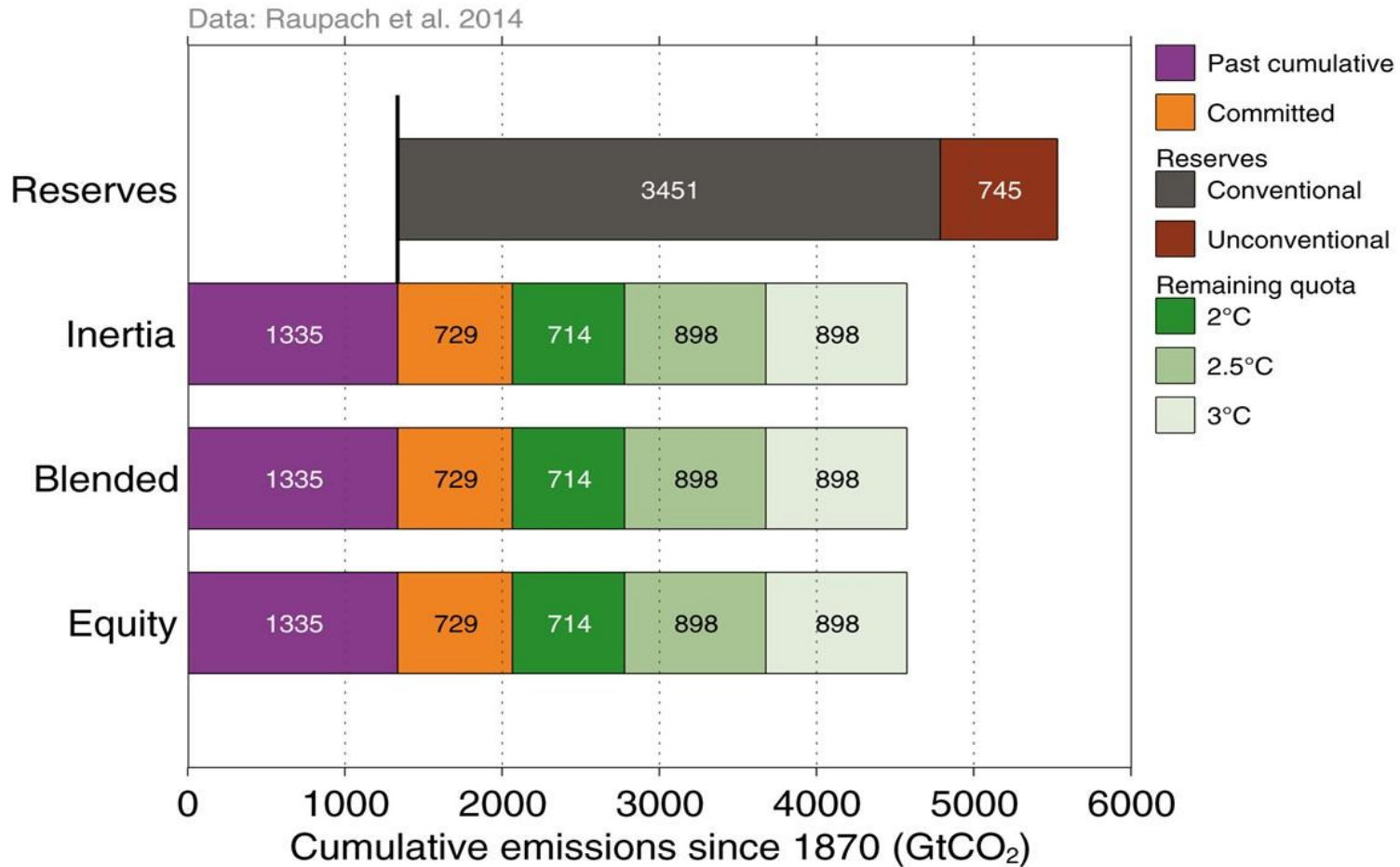


If emissions continue to grow as projected to 2019 and then continue at the 2019 rate, the remaining budget would be used up about 22 years from 2019



# Global Quotas, Committed Emissions, Fossil-Fuel Reserves

To keep temperatures below 2°C requires two-thirds of fossil fuels to remain in the ground\*  
 Committed emissions in existing infrastructure represents 50% of the remaining quota\*



\*Assuming a 50% chance to stay below 2°C and no carbon-capture and storage

Source: [Raupach et al 2014](#)

# Pařížská dohoda 2015



# Pařížská dohoda 2015

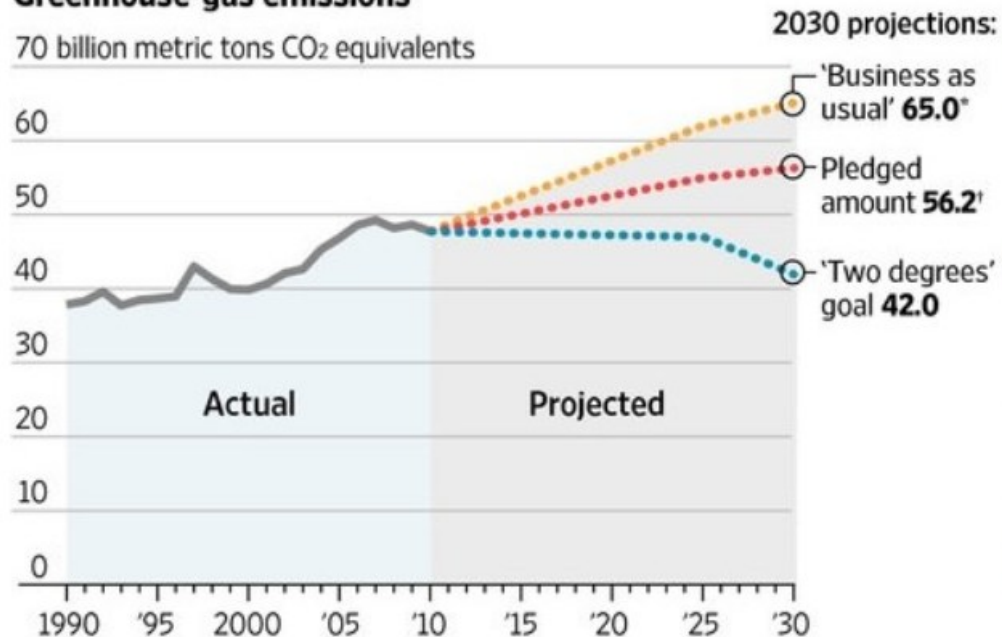


## Promising Start

Pledges made at the summit may not be enough to limit global warming to less than 2 degrees Celsius above pre-industrial levels but could spur a significant cut in emissions compared with existing policies.

### Greenhouse-gas emissions






70 billion metric tons CO<sub>2</sub> equivalents



\*Assumes no new national energy policies from 2010 onwards

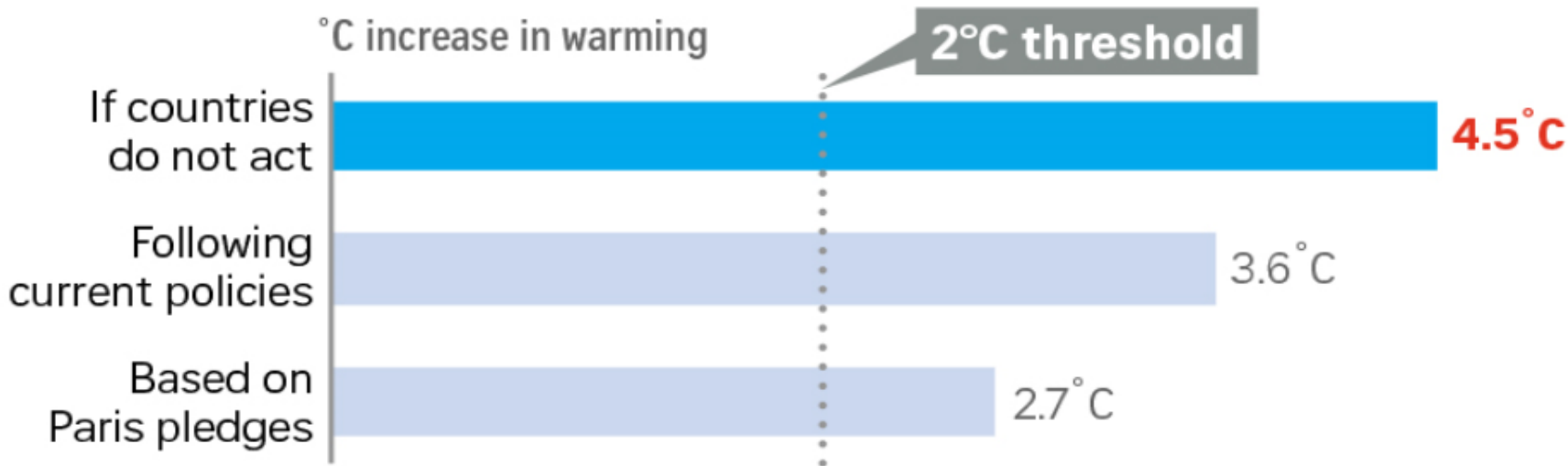
†Based on unconditional INDCs of 102 countries Source: Michel den Elzen, PBL Netherlands Environmental Assessment Agency

### Key points of the agreement

-  Aims to bring growth of greenhouse gases in the atmosphere to zero by the second half of this century
-  Requires for first time that all countries curb emissions, not just developed ones
-  Targets no more than 2° C (3.6° F) rise in temperature, but aspires to go further
-  Richer countries take lead in mobilizing more than \$100 billion annually to help developing countries meet agreement
-  Establishes a process to review, assess and update emissions-reduction plans

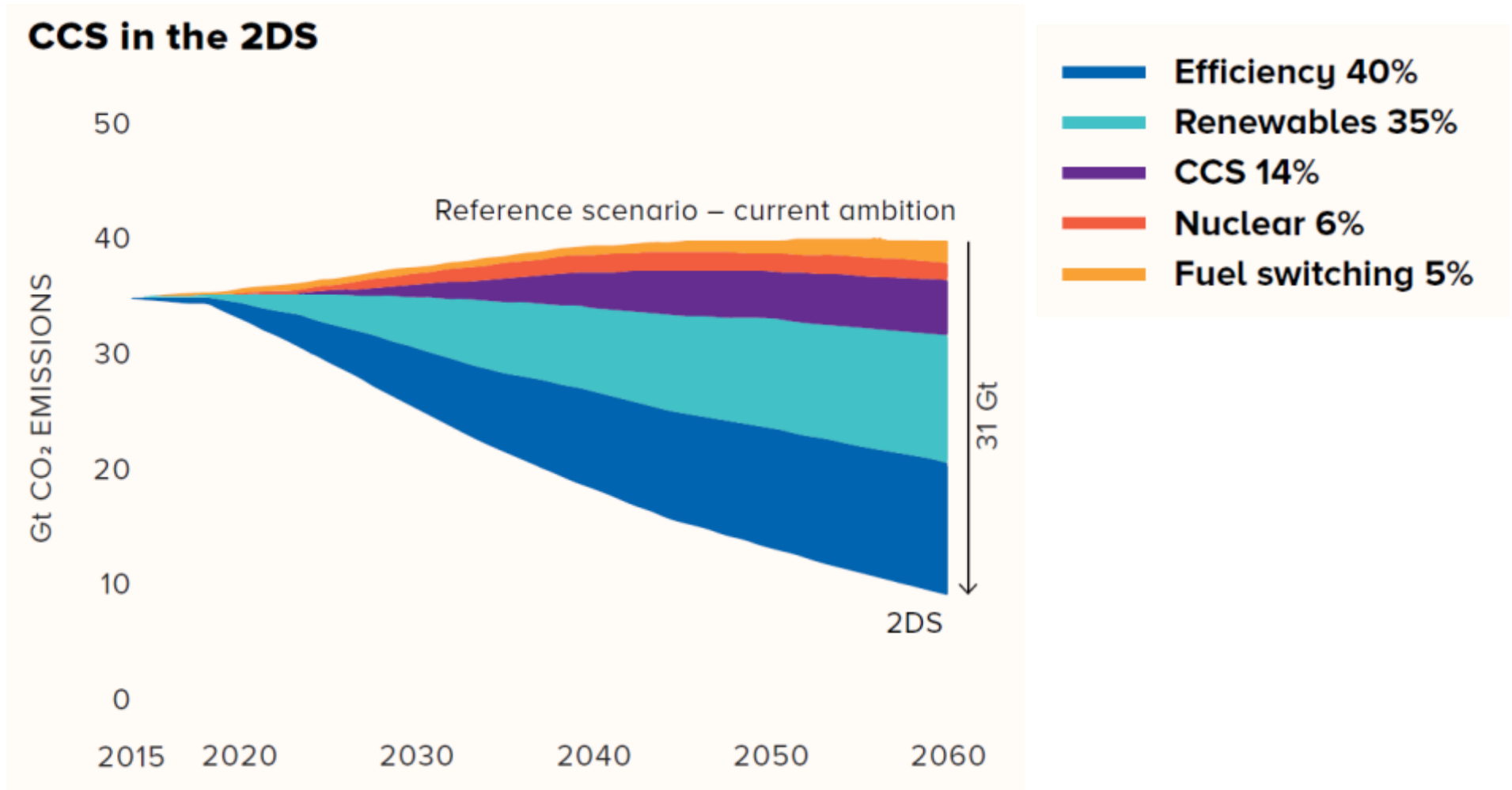
# Scénáře globálního oteplování

## Average warming projected by 2100



Sources: CLIMATE ACTION TRACKER, DATA COMPILED BY CLIMATE ANALYTICS, ECOFYS, NEW CLIMATE INSTITUTE AND POTSDAM INSTITUTE FOR CLIMATE IMPACT RESEARCH ST GRAPHICS ADAPTED FROM BBC

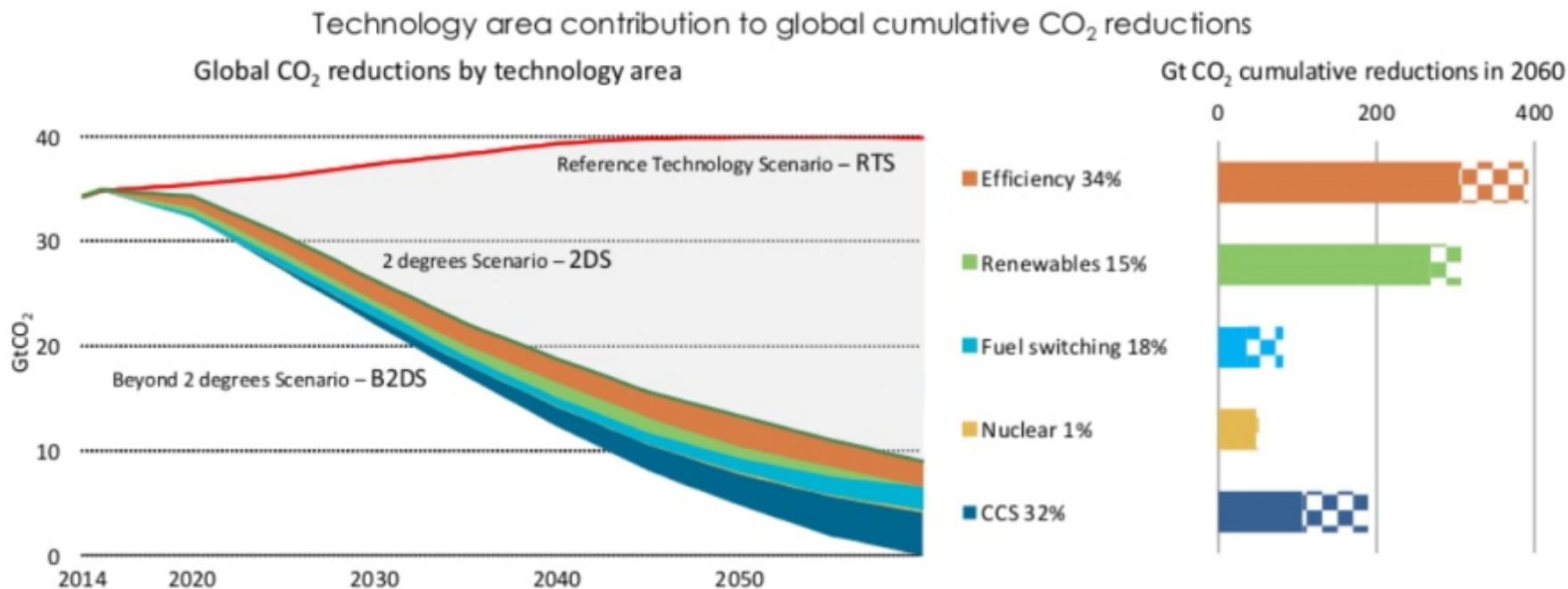
# Technologie pro redukci emisí CO<sub>2</sub>



Source: IEA Energy Technology Perspectives 2017

# Technologie pro redukci emisí CO<sub>2</sub>

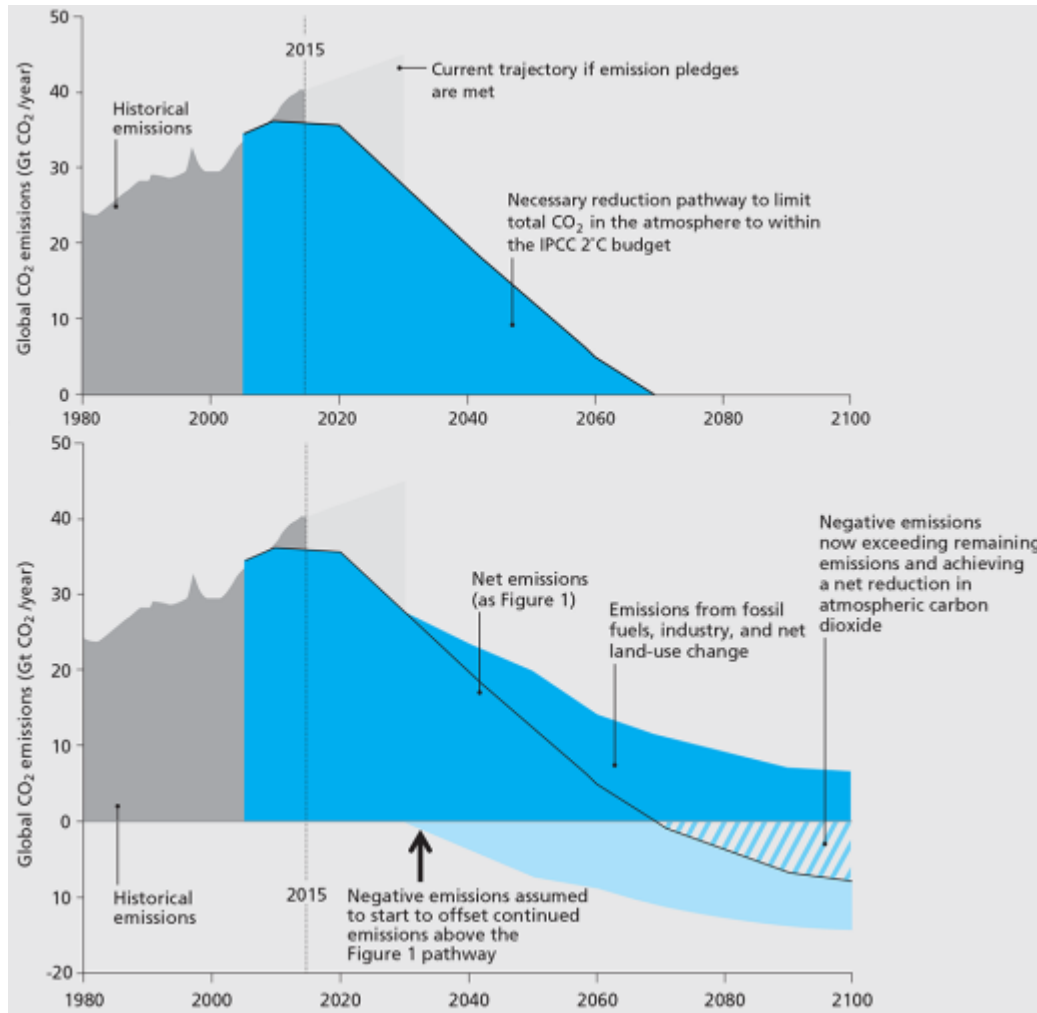
## How far can technology take us?



Pushing energy technology to achieve carbon neutrality by 2060 could meet the mid-point of the range of ambitions expressed in Paris.

Source: IEA Energy Technology Perspectives 2017

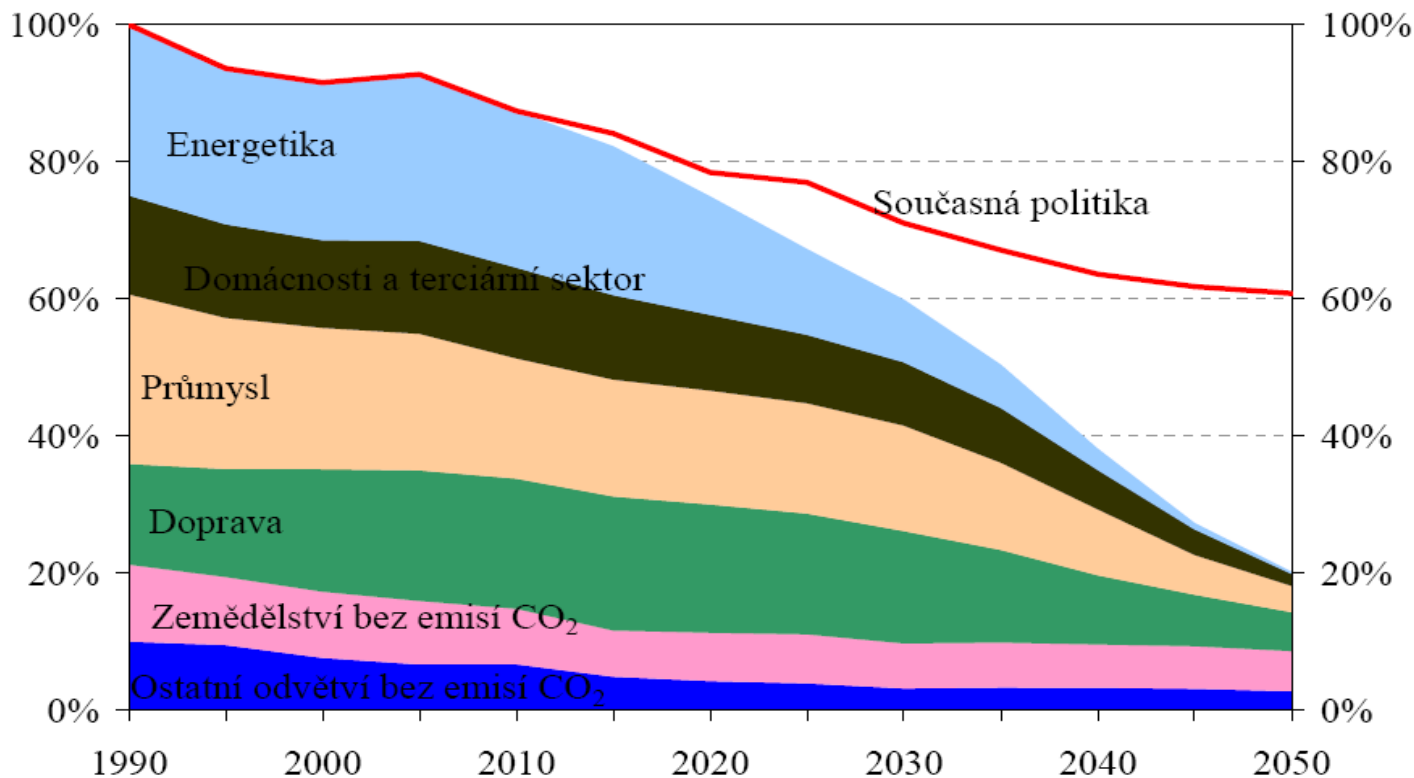
# Technologie pro redukci emisí CO<sub>2</sub>



Source: European Academies Science Advisory Council (EASAC) report 2018

# EU - Plán přechodu na konkurenceschopné nízkouhlíkové hospodářství do roku 2050 (Sdělení EK z 8.3.2011)

Obrázek 1: Projekce 80% snížení domácích emisí skleníkových plynů v EU (100 % = 1990)





# EU - Plán přechodu na konkurenceschopné nízkouhlíkové hospodářství do roku 2050 (Sdělení EK z 8.3.2011)

**Tabulka č. 1: Snížení emisí podle odvětví**

Snížení emisí skleníkových plynů oproti roku 1990	2005	2030	2050
Celkem	-7 %	-40 až -44 %	-79 až -82 %
Odvětví			
Energetika (CO <sub>2</sub> )	-7 %	-54 až -68 %	-93 až -99 %
Průmysl (CO <sub>2</sub> )	-20 %	-34 až -40 %	-83 až -87 %
Doprava (včetně emisí CO <sub>2</sub> z letecké dopravy; vyjma námořní)	+30 %	+20 až -9 %	-54 až -67 %
Domácnosti a služby (CO <sub>2</sub> )	-12 %	-37 až -53 %	-88 až -91 %
Zemědělství (bez CO <sub>2</sub> )	-20 %	-36 až -37 %	-42 až -49 %
Ostatní emise bez CO <sub>2</sub>	-30 %	-72 až -73 %	-70 až -78 %

# Technologie CCS

## = zachytávání a ukládání CO<sub>2</sub>

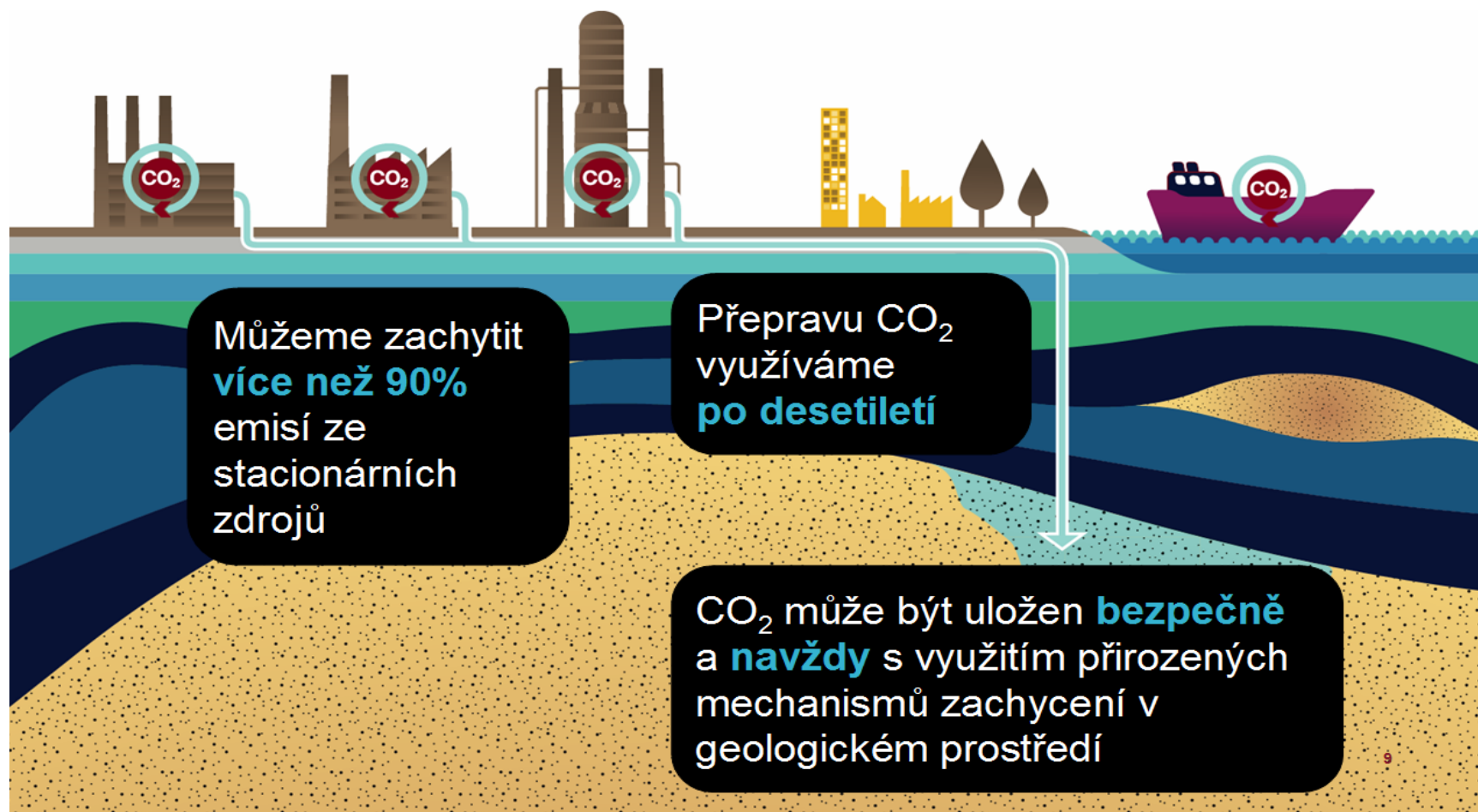
**CCS** (Carbon /Dioxide/ Capture and Storage) je tzv. mitigační opatření - cílem je snížení emisí CO<sub>2</sub> jako skleníkového plynu

**Princip:** zachycení emisí přímo u velkého stacionárního zdroje (elektrárna, ocelárna, cementárna, chemický provoz, úpravna plynu aj.), jeho přeprava a bezpečné uložení do vhodné geologické struktury

**Výchozí předpoklady:**

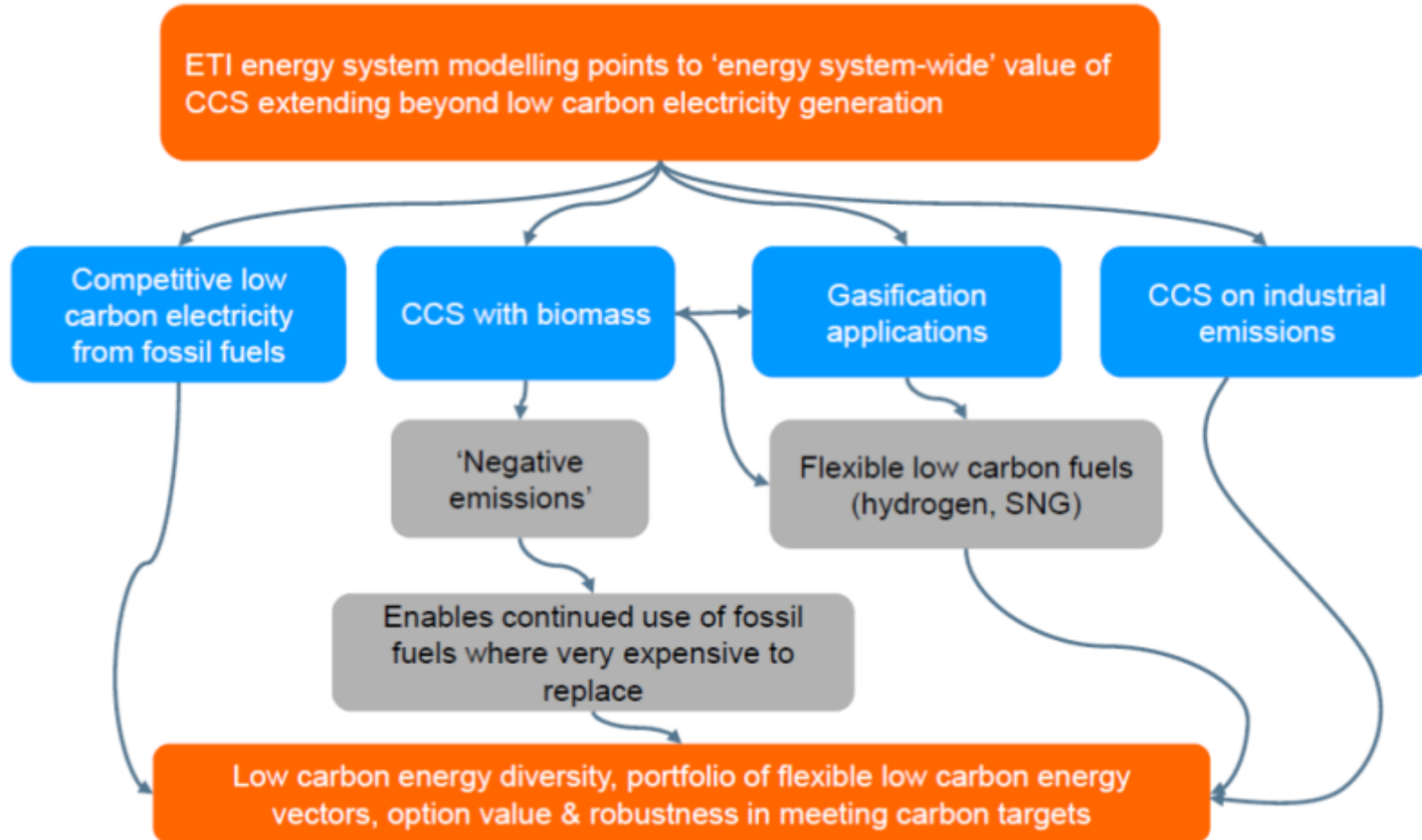
- dochází ke změně klimatu
- antropogenní emise CO<sub>2</sub> jsou významným hybatelem klimatické změny
- chceme s tím něco dělat

# Princip CCS





# CCS maintains multiple options and delivers significant value at a whole-system level



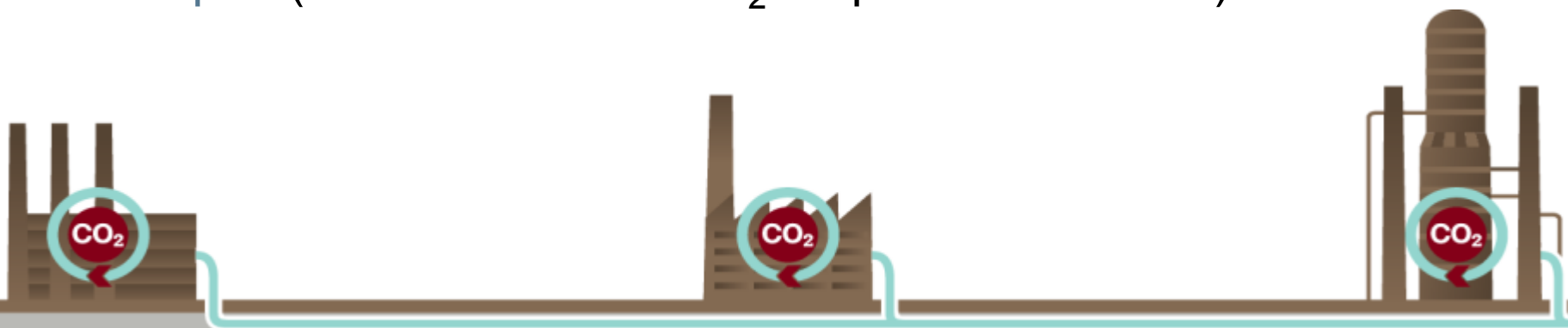
# Zachytávání CO<sub>2</sub>

**Primární cíle** = velké stacionární zdroje emisí produkující statisíce až milióny tun CO<sub>2</sub> ročně (menší zdroje mohou být seskupeny do klastrů):

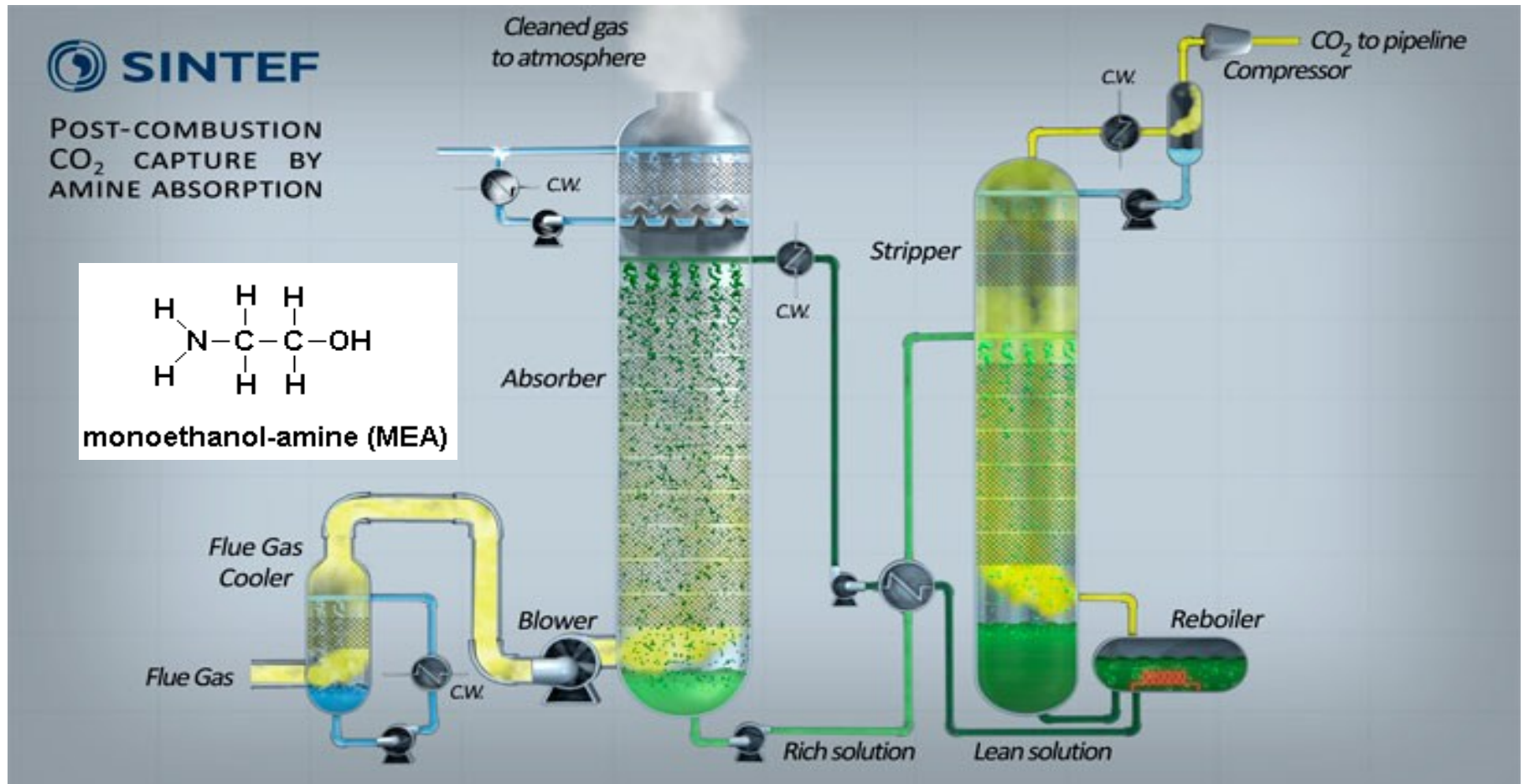
- elektrárny spalující fosilní paliva nebo biomasu
- zařízení na úpravu ropy a zemního plynu
- průmyslové provozy (výroba cementu, železa a oceli, papíru a celulózy)
- rafinerie a petrochemické provozy

V současnosti jsou komerčně dostupné **technologie 1. generace**:

- chemická absorpce (chemická vazba)
- fyzikální absorpce (rozpuštění v rozpouštědle – Selexol, Rectisol)
- adsorpce (adheze molekul CO<sub>2</sub> na povrch sorbentu)



# Princip chemické absorpce



# Princip adsorpce

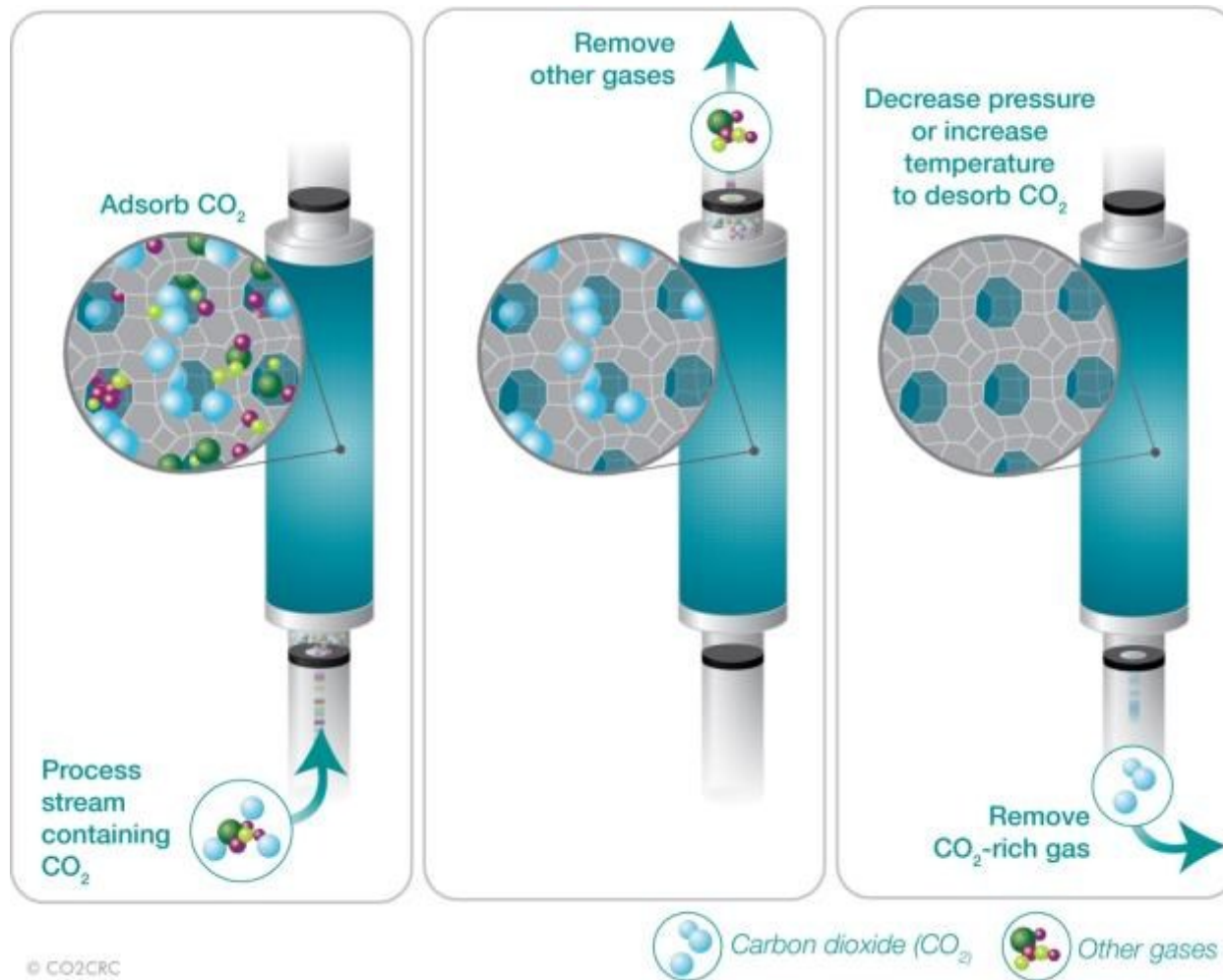


image courtesy CO2CRC

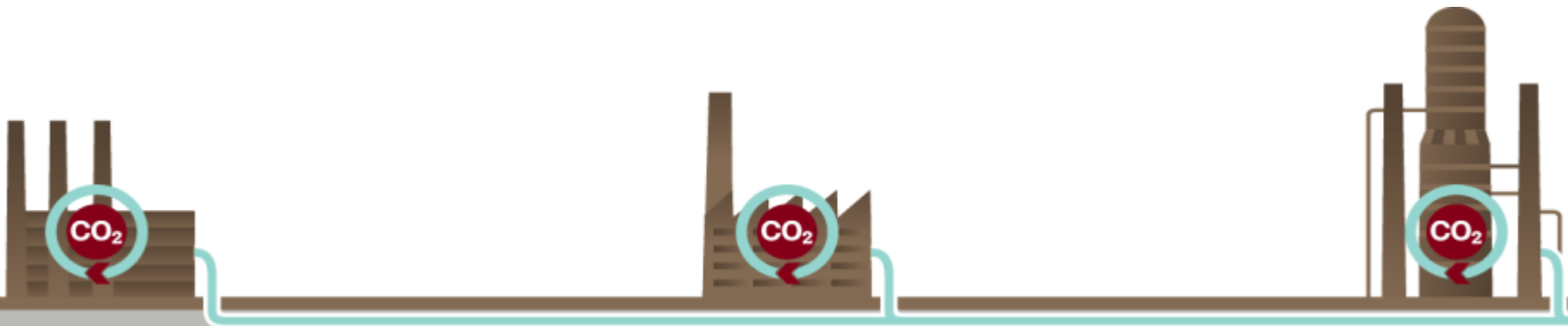
# Technologie dalších generací

2. generace – v současnosti v pilotním měřítku

3. generace – v současnosti v laboratorním měřítku

## Rozvíjené a zkoumané technologie

- nové absorbenty pro chemickou absorpci
- nové adsorbenty a adsorpční procesy
- membrány
- kryogenní destilace
- biogenní zachytávání (mikrobi, řasy)





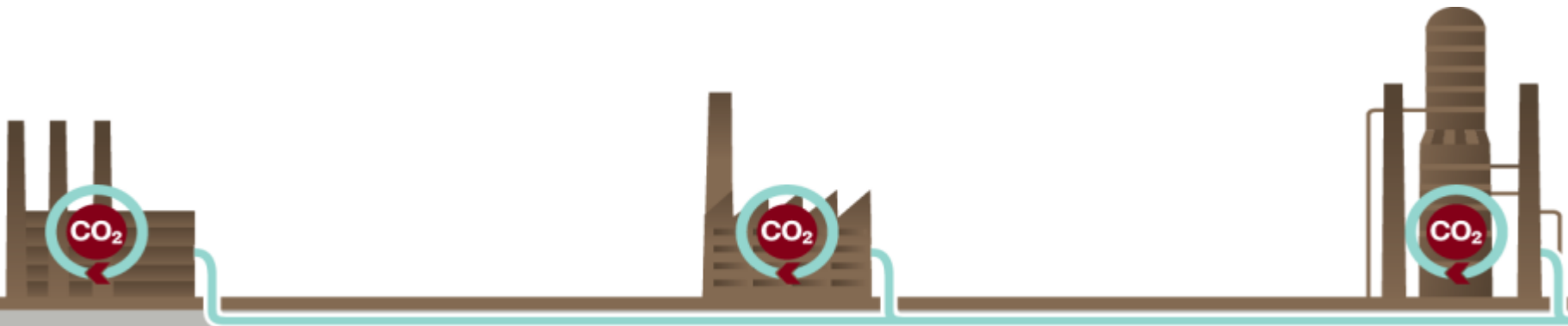
# Zachytávání CO<sub>2</sub> u spalovacích procesů

## Stávající technologie:

- Pre-combustion (před spalováním):  
CO<sub>2</sub> je zachytáván ještě před spálením paliva
- Oxy-fuel (spalování v kyslíkové atmosféře):  
CO<sub>2</sub> je separován během spalování
- Post-combustion (po spalování):  
CO<sub>2</sub> je zachytáván po spálení paliva  
(Tato technologie může být použita i pro stávající zařízení jako retrofit)

## Nově vyvíjené technologie:

- High temperature solid looping cycles (karbonátová a chemická smyčka)

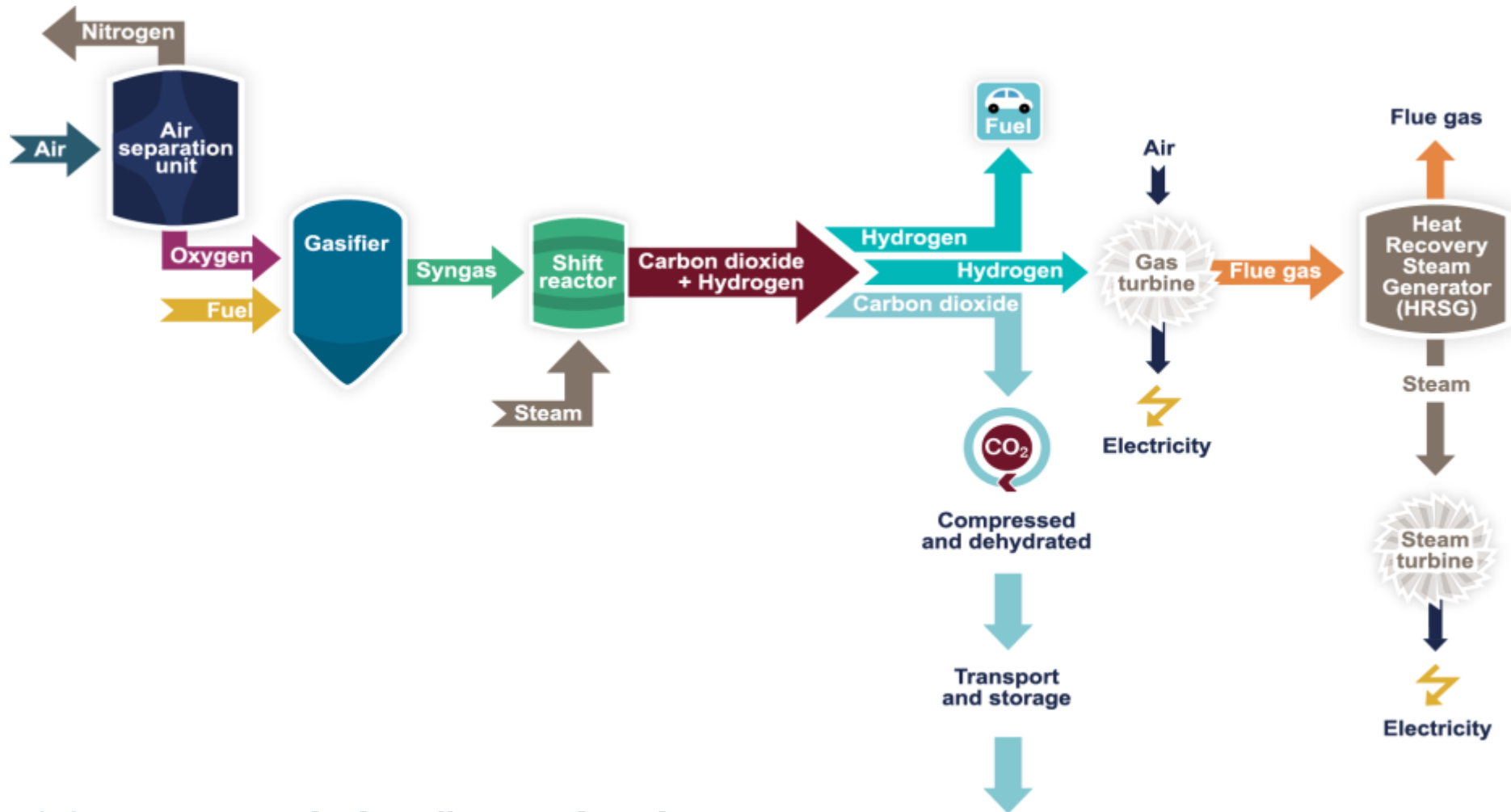


# Zachytávání po spalování



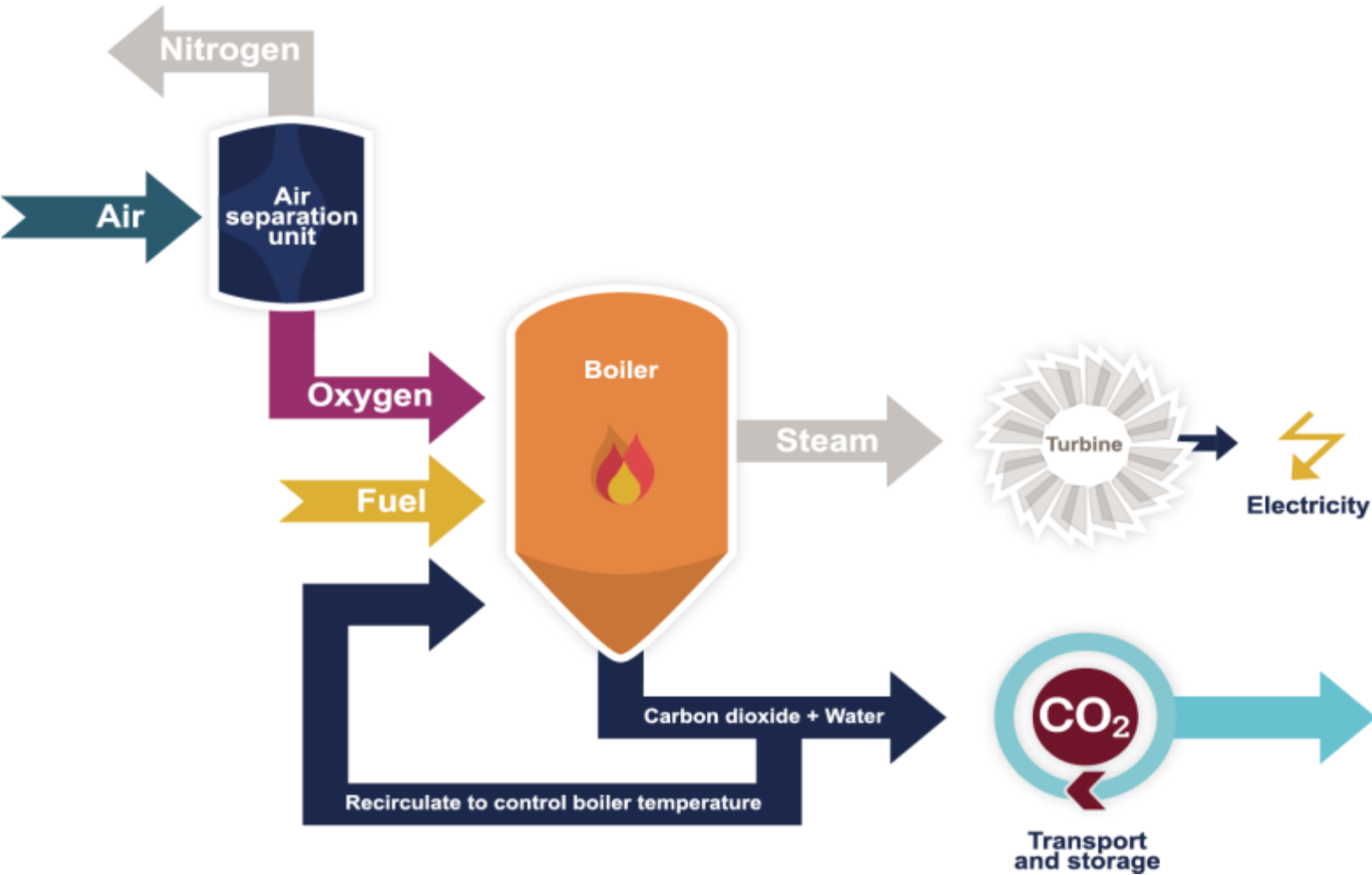
CO<sub>2</sub> je zachytáván po spálení paliva

# Pre-combustion



CO<sub>2</sub> je zachytáván před spálením paliva

# Technologie oxy-fuel

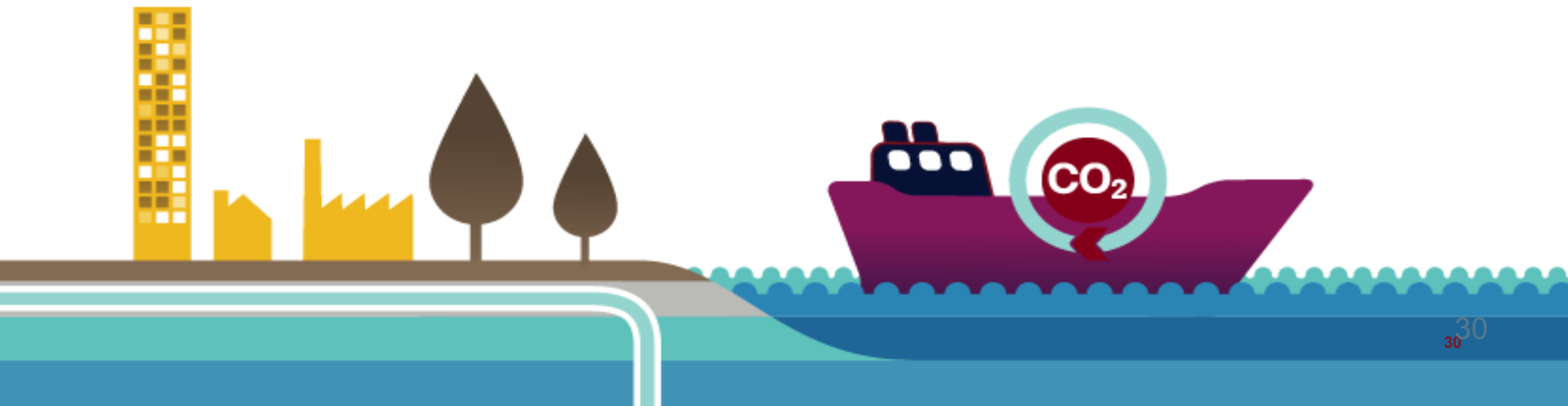


# Průmyslové procesy

- Úprava zemního plynu (absorpce / Sleipner, Snøhvit NOR)
- Zplynování uhlí / bitumenu (fyzikální absorpce)
  - výroba syngasu (Great Plains Synfuels Plant USA)
  - výroba vodíku (North West Sturgeon Refinery CAN)
- Výroba vodíku pomocí parního reformingu (absorpce n. adsorpce / Port Arthur Project USA, Shell Quest Project CAN)
- Chemická výroba:
  - výroba čpavku ( $\text{CO}_2$  je vedlejší produkt ve vysoké koncentraci)
  - výroba etanolu (ADM Illinois Industrial CCS Project USA – první aplikace BioCCS)
- Výroba oceli (chemická absorpce / ASI CCS Project UEA)
- Výroba cementu (zachytávání  $\text{CO}_2$  zatím v pilotním měřítku)

# Transport CO<sub>2</sub>

- Po zachycení je CO<sub>2</sub> stlačen do kapalného, popř. superkritického stavu, zbaven vody a dopravován na místo uložení.
- Pro větší množství CO<sub>2</sub> je nejvýhodnějším způsobem dopravy produktovod (v Sev. Americe provozováno téměř 5000 km) – typické podmínky 10-15 MPa, 15-30°C.
- Pro menší množství a velké vzdálenosti na moři lze využít tankery – dnes do 1800 t CO<sub>2</sub>, ~ -30 °C, ~ 1,6 Mpa; připravují se větší.



# Ukládání CO<sub>2</sub>

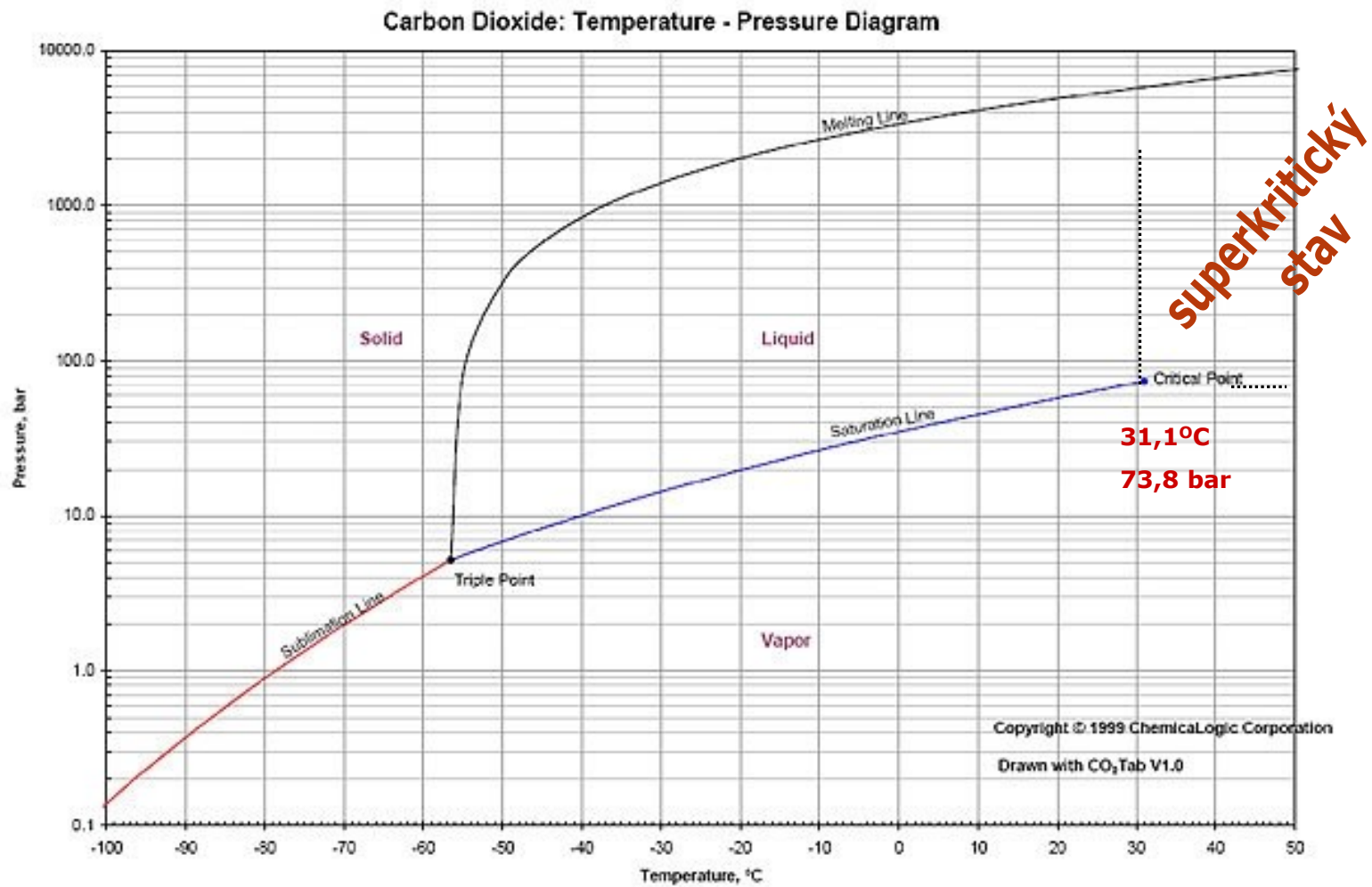


# Bezpečné ukládání CO<sub>2</sub>

- Využívá se **přírodních mechanismů** které v přírodě zadržují CO<sub>2</sub>, zemní plyn a ropu po milióny let
- Tekutý CO<sub>2</sub> se vtlačí hluboko do podloží do některého ze dvou základních typů úložišť:
  - hlubinné slané akvifery (hloubka 700 – 3000 m)
  - vytěžená ložiska ropy a plynu (hloubka až 5000 m)
- Oba typy úložišť mají vrstvu **porézní horniny**, která pohltí CO<sub>2</sub> a v nadloží nepropustnou vrstvu **těsnící horniny (cap rock)**, která úložiště utěsňuje



# Fázový diagram CO<sub>2</sub>

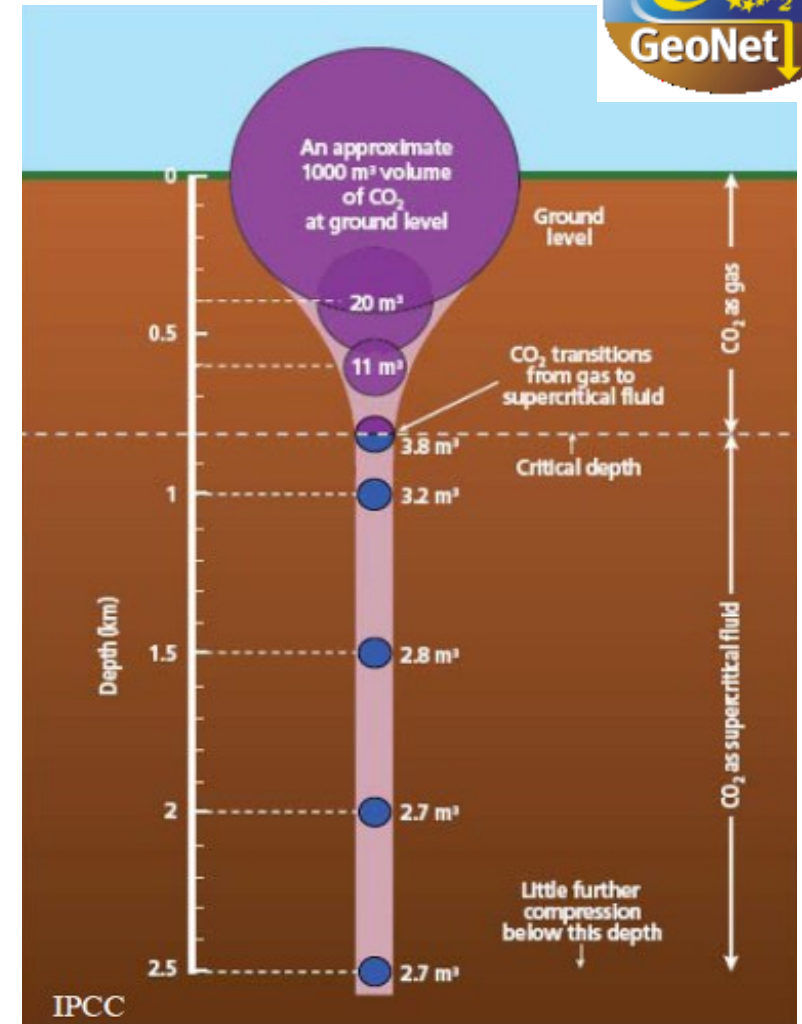


# Superkritický CO<sub>2</sub> – vlastnosti

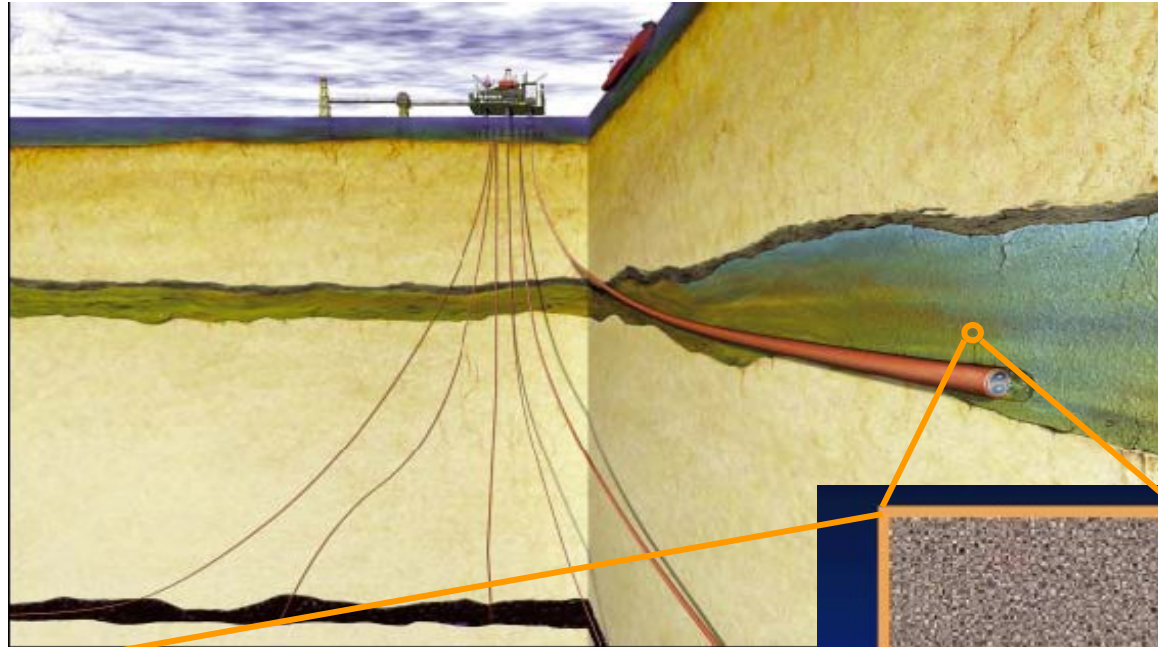


- plyn z hlediska viskozity
- kapalina z hlediska hustoty

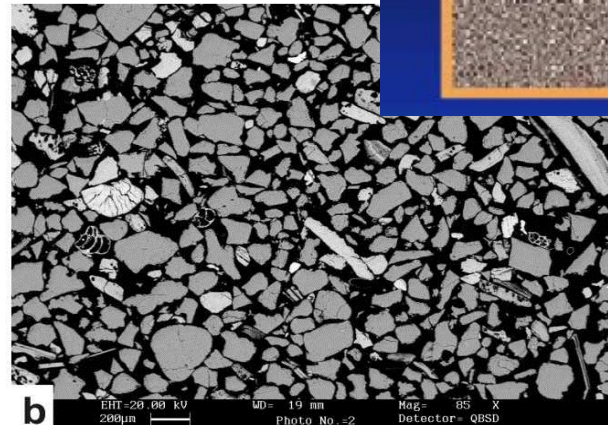
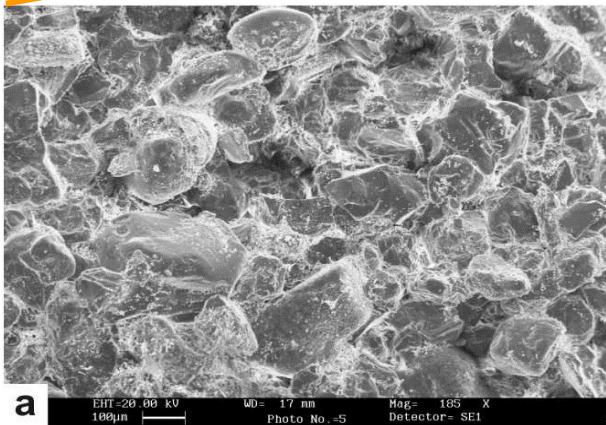
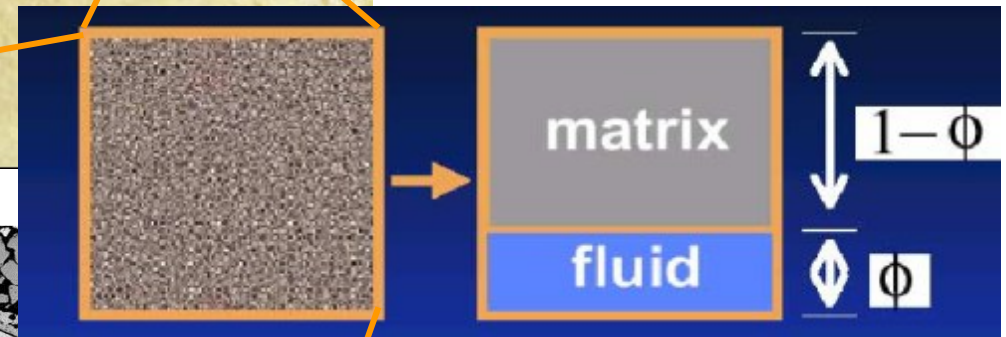
T=100° C, P=280bar (2800m)	hustota (kg/m <sup>3</sup> )	viskozita (cP)
superkritický CO <sub>2</sub>	615	0.05
voda	804	0.16
plyn (metan)	150	0.02



# Koncept vtláčení CO<sub>2</sub> do porézních hornin (akvifery, ložiska ropy a plynu)

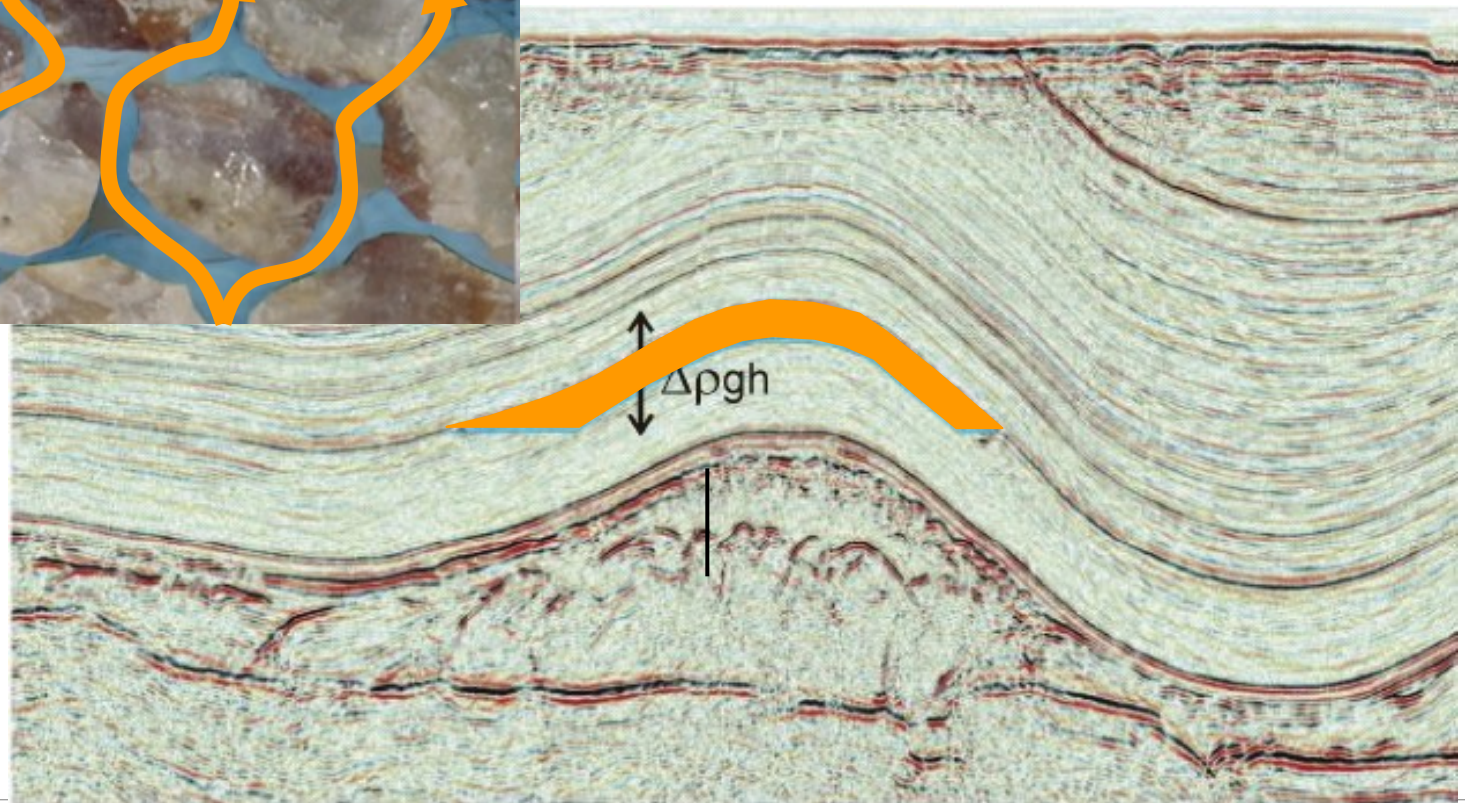


Courtesy CO2STORE



SEM images of the Utsira Sand  
a) Reflected light  
b) Transmitted light (pore-spaces are black)

# Strukturní trapping CO<sub>2</sub>



# Bezpečnost uložení CO<sub>2</sub> roste s časem

... díky 3 přirozeným mechanismům

## Reziduální trapping

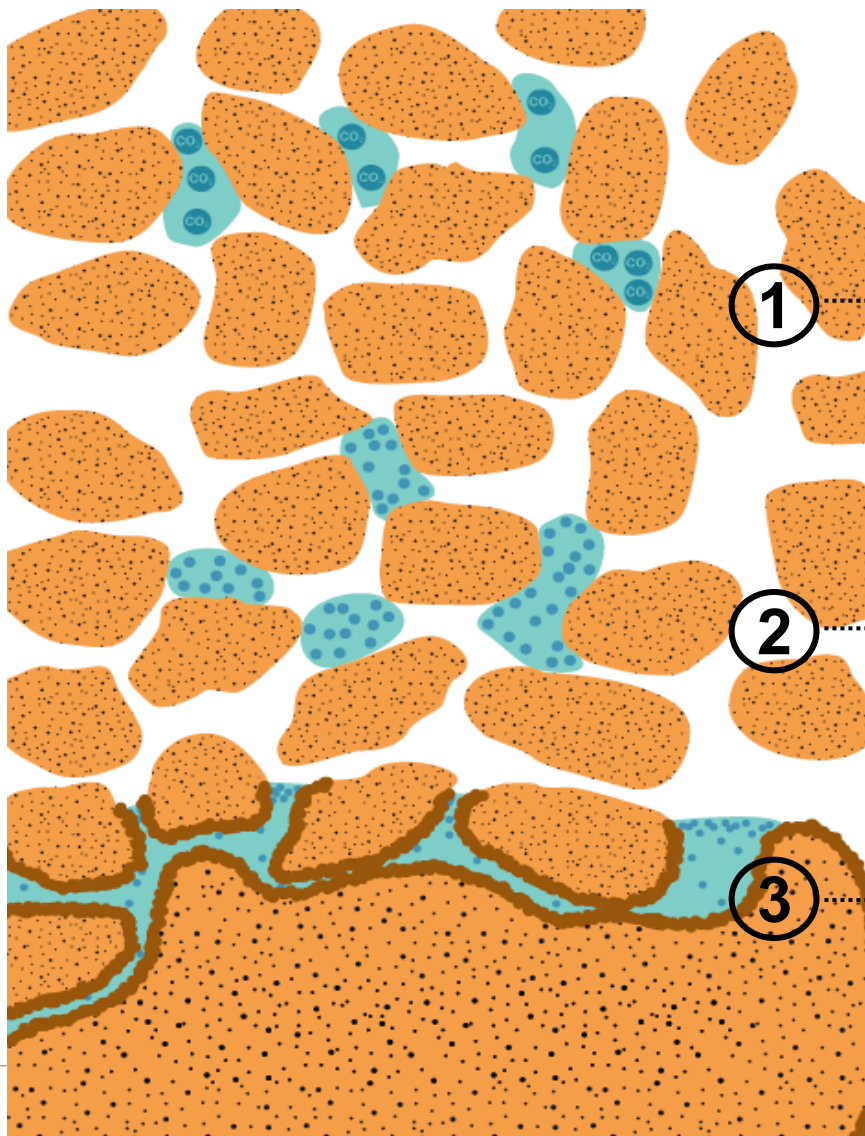
CO<sub>2</sub> je nevratně zachycen v mikropórech a nemůže se dále pohybovat

## Rozpouštění

CO<sub>2</sub> se rozpustí v okolní solance; ta pak klesá ke dnu rezervoáru

## Minerální trapping

CO<sub>2</sub> geochemicky reaguje s minerály a vytváří chemické vazby



# Požadované vlastnosti úložiště

## Geologické

- Kapacita umožňující uložit uvažované množství CO<sub>2</sub> (objem pórů, p-T podmínky)
- Injektivita zaručující dostatečnou rychlost vtlačení CO<sub>2</sub> do struktury (permeabilita, bariéry toku)
- Dostatečné těsnící vlastnosti nadloží, bránící úniku CO<sub>2</sub> z úložiště (zejména dostatečná mocnost a neporušenost těsnících hornin v nadloží)

## Ekonomické

- Vzdálenost od zdroje emisí
- Existující infrastruktura
- Dostupnost geologických a dalších údajů

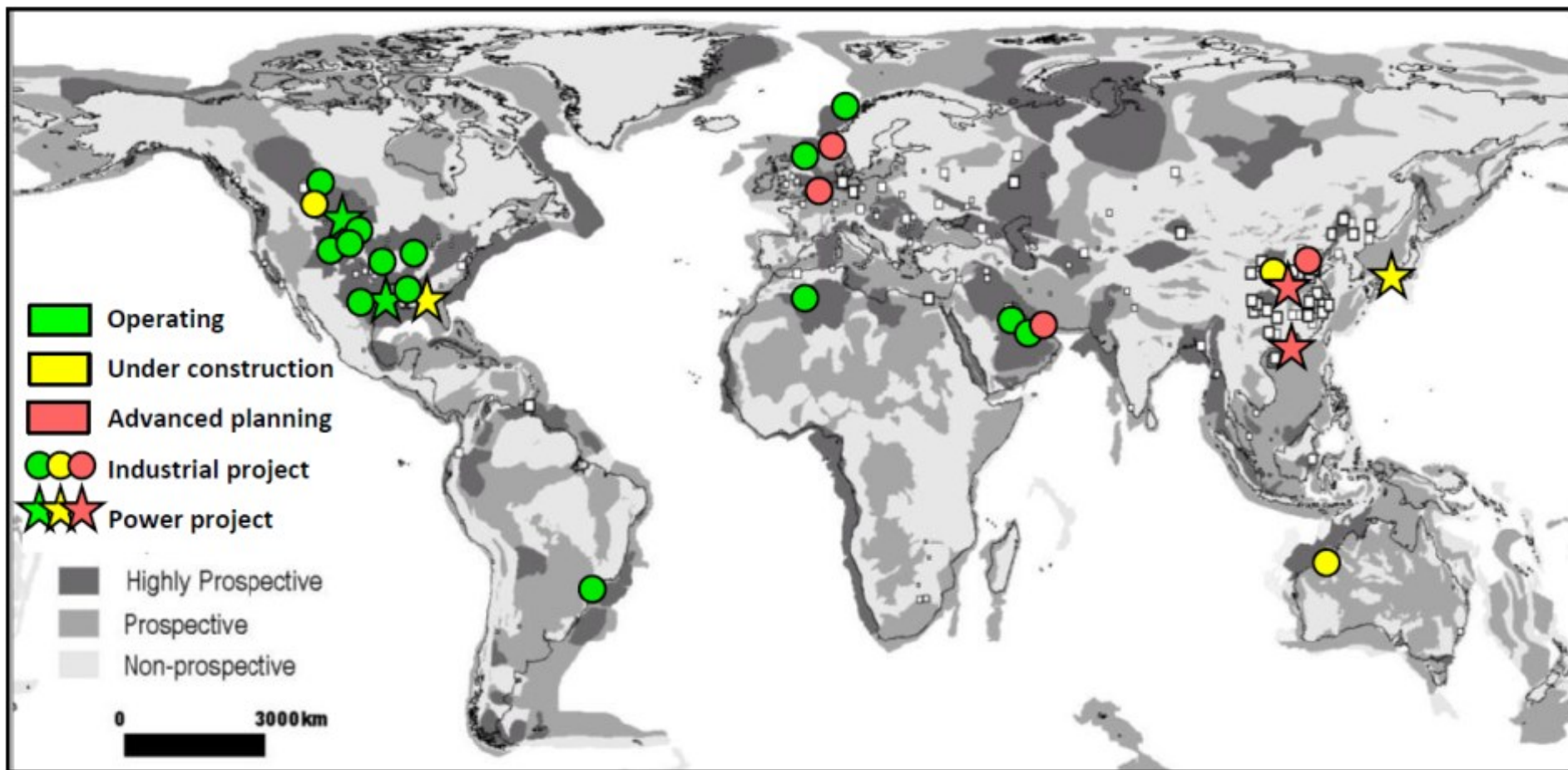
## Ostatní

- Konflikty zájmů
- Přijatelnost pro veřejnost

# CCS ve světě



# CCS ve světě

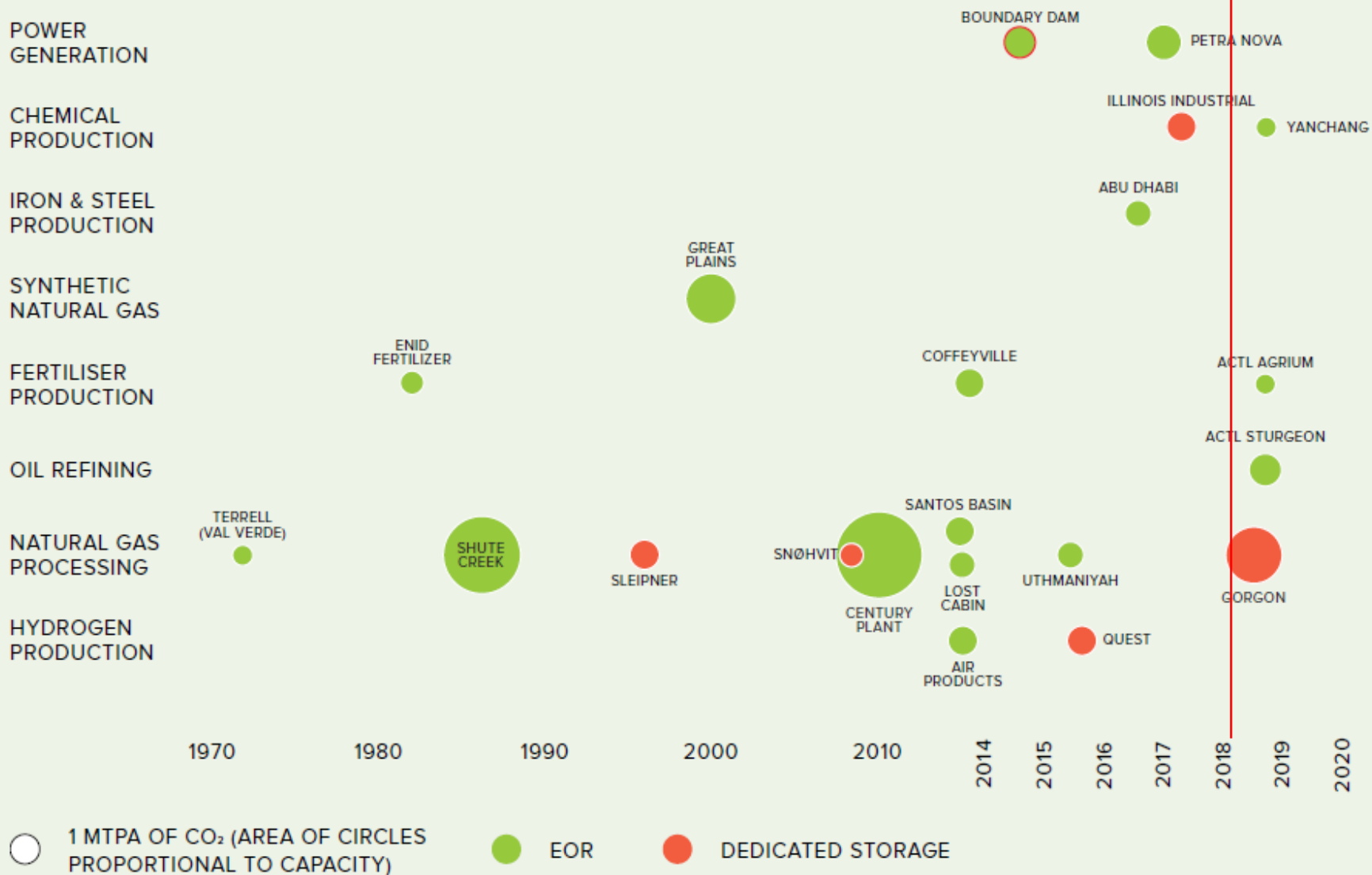


**17 operating plants, storing >22 Mtons CO<sub>2</sub> each year**

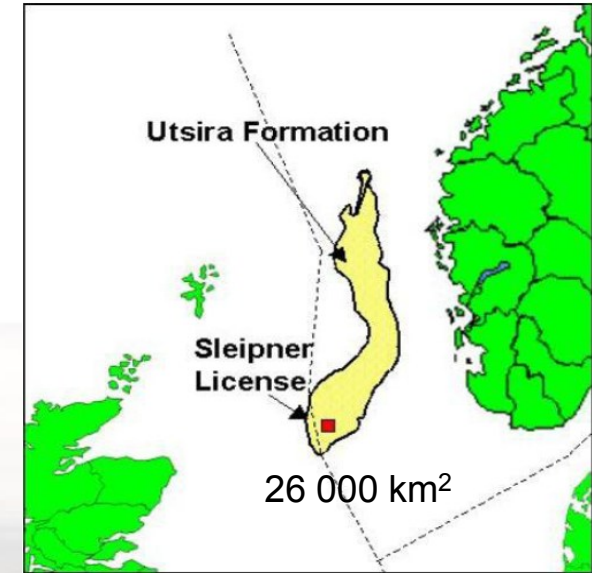


# CCS ve světě

**CCS large-scale facilities in operation and construction by industry and operations start date**



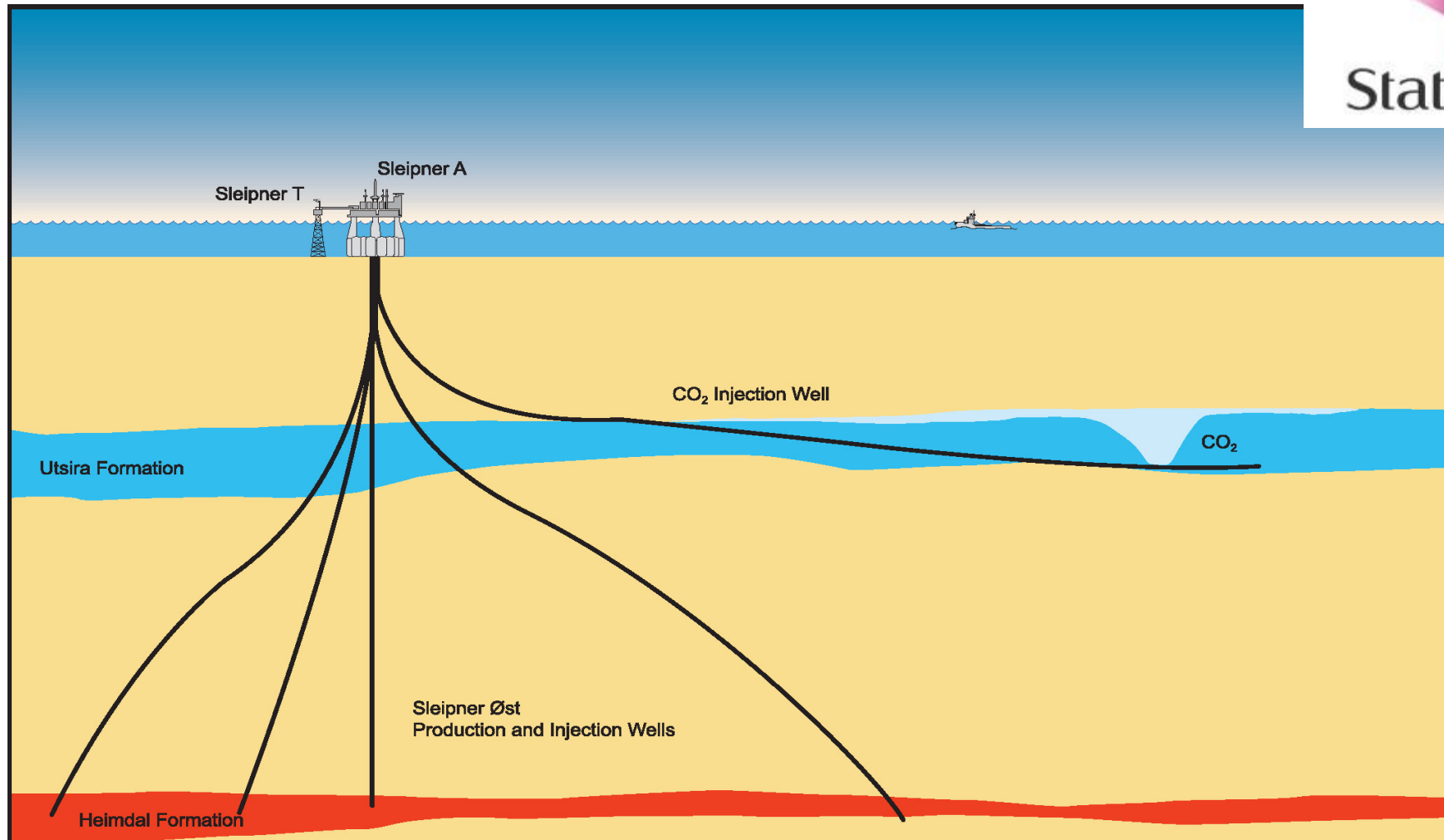
# Projekt Sleipner – 22 let ukládání CO<sub>2</sub> (od 1996)



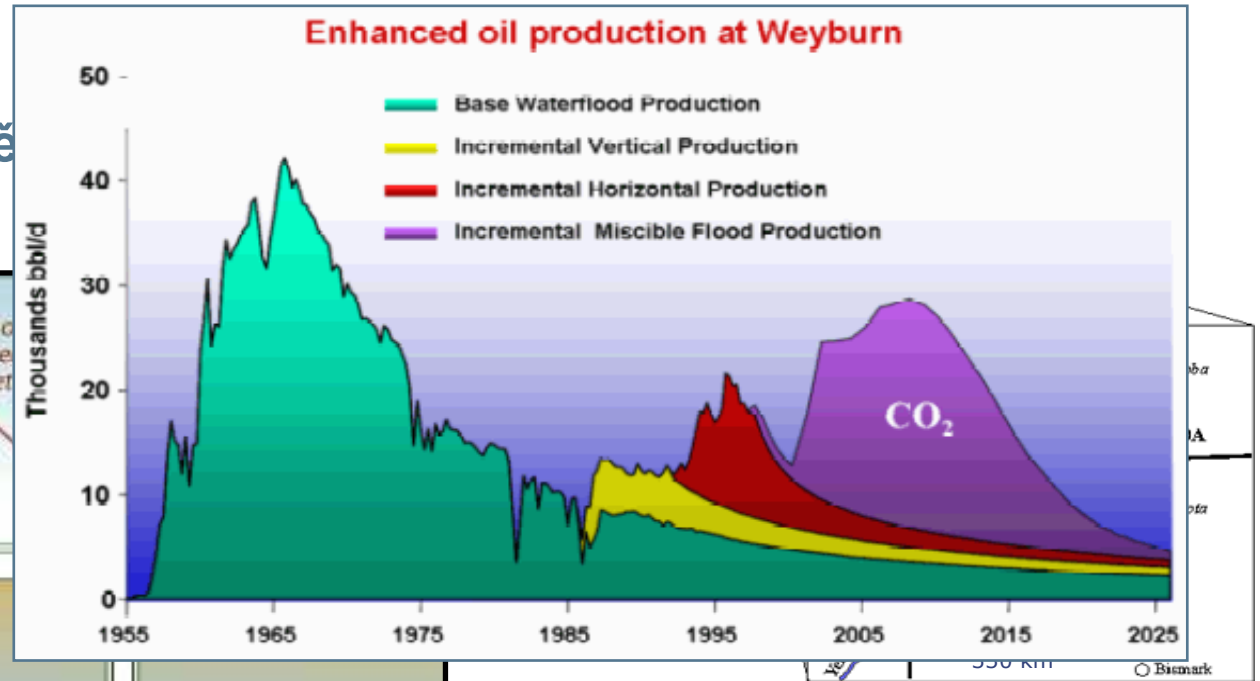
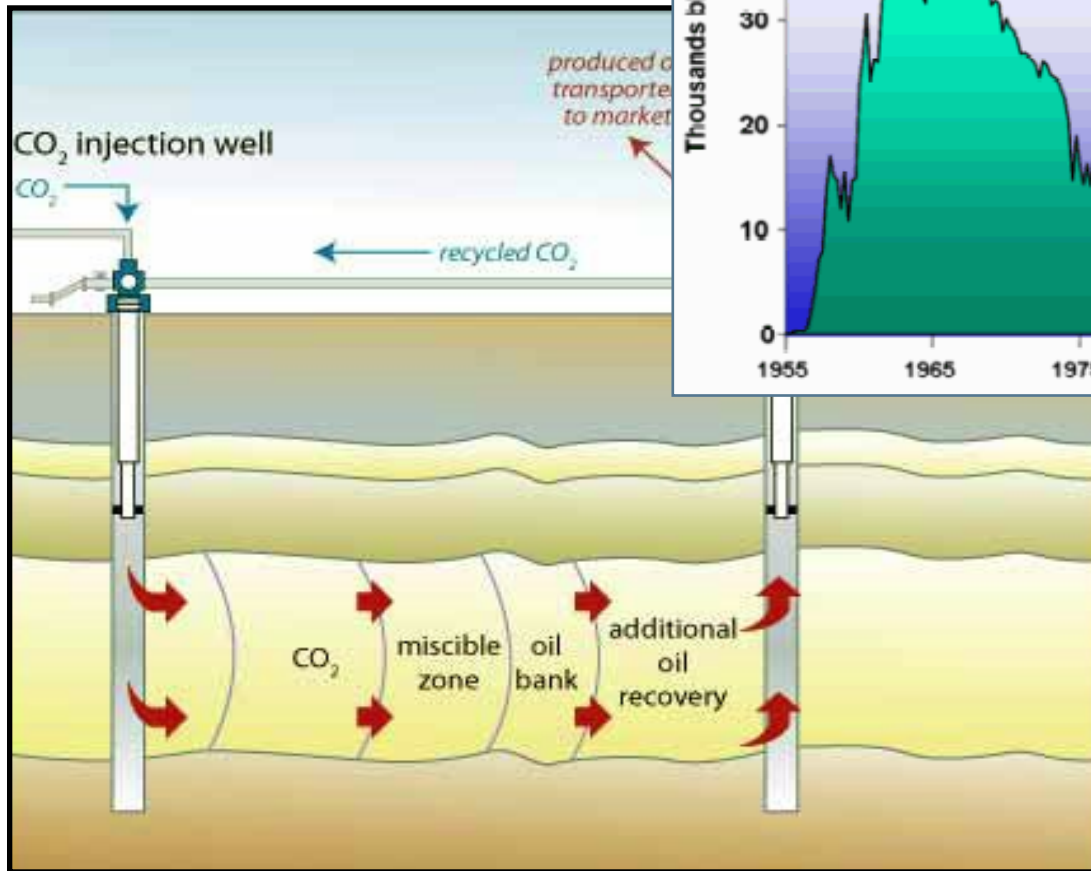
# Sleipner - schéma



Statoil



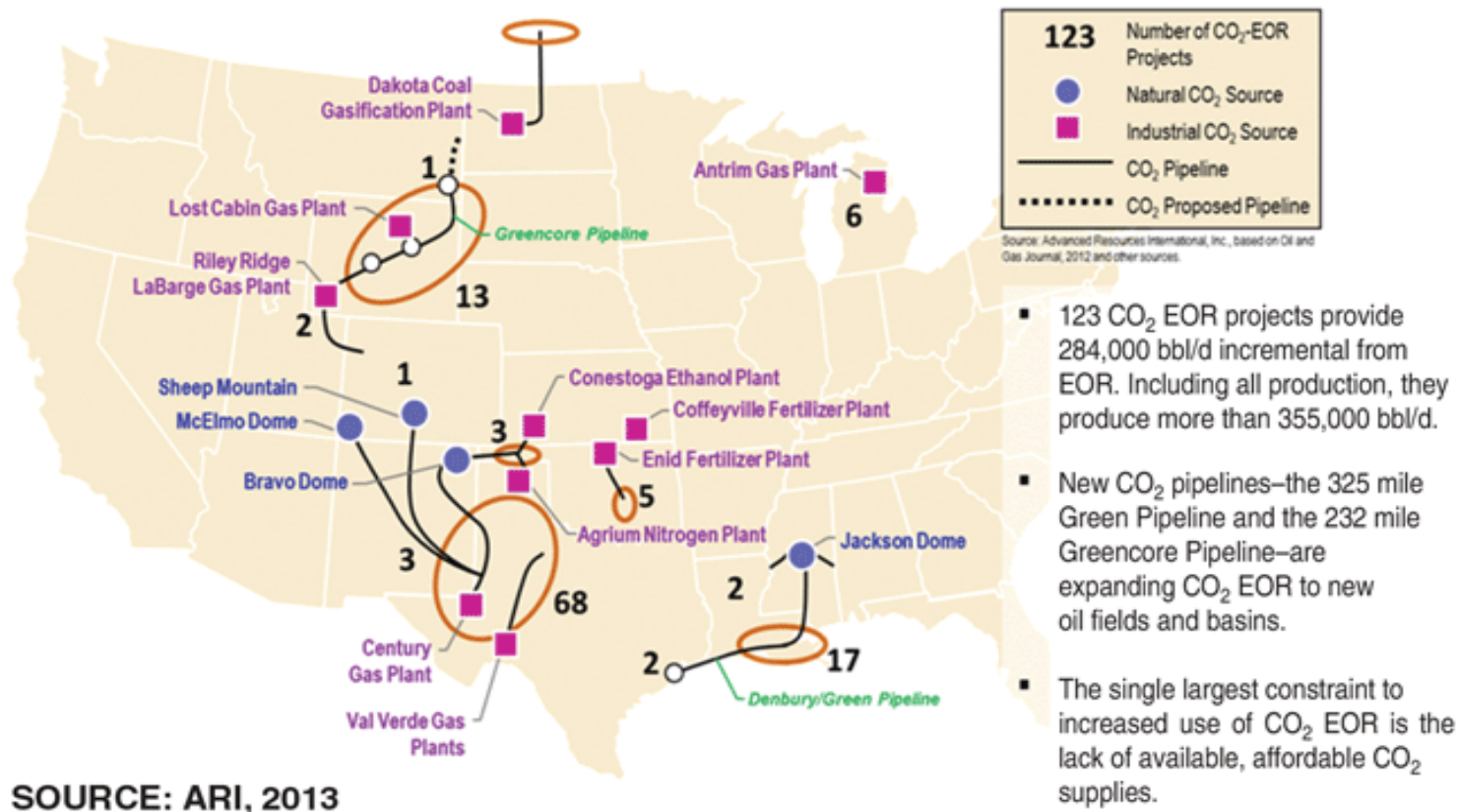
# Weyburn – EOR (druhotná intenzifikace tě



North Dakota coal gasification plant – 3,3 Mt CO<sub>2</sub> za rok

# CO<sub>2</sub>-EOR v USA

- projekty na komerční bázi od r. 1972
- v r. 2014 bylo v běhu 136 projektů EOR, 300 tis. barelů dodatečné ropy denně
- vybudováno přes 5 800 km produktovodů
- celkem injektováno přes 700 Mt CO<sub>2</sub>, 62 mil. tun v r. 2013 (z toho cca 20 % antropogenní)



# První elektrárny s CCS

## Boundary Dam

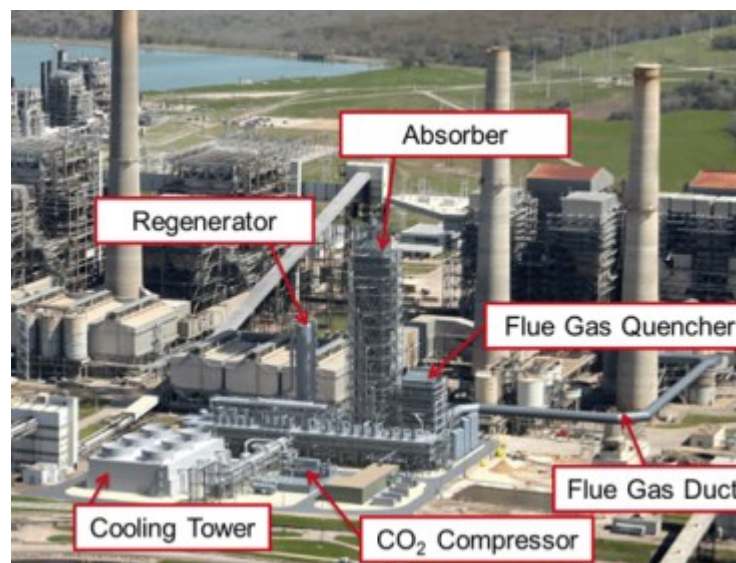
(Kanada, Saskatchewan)

- retrofit - blok 130 MW (HU)
- cca 1 mil. t CO<sub>2</sub> ročně
- post-combustion (Cansolv)
- EOR (66 km produktovod)
- zahájení 10/2014



## Petra Nova (USA, Texas)

- retrofit - blok 250 MW (HU)
- cca 1,5 mil. t CO<sub>2</sub> ročně
- post-combustion (MHI)
- EOR (130 km produktovod)
- zahájení 1/2017



# Situace v ČR

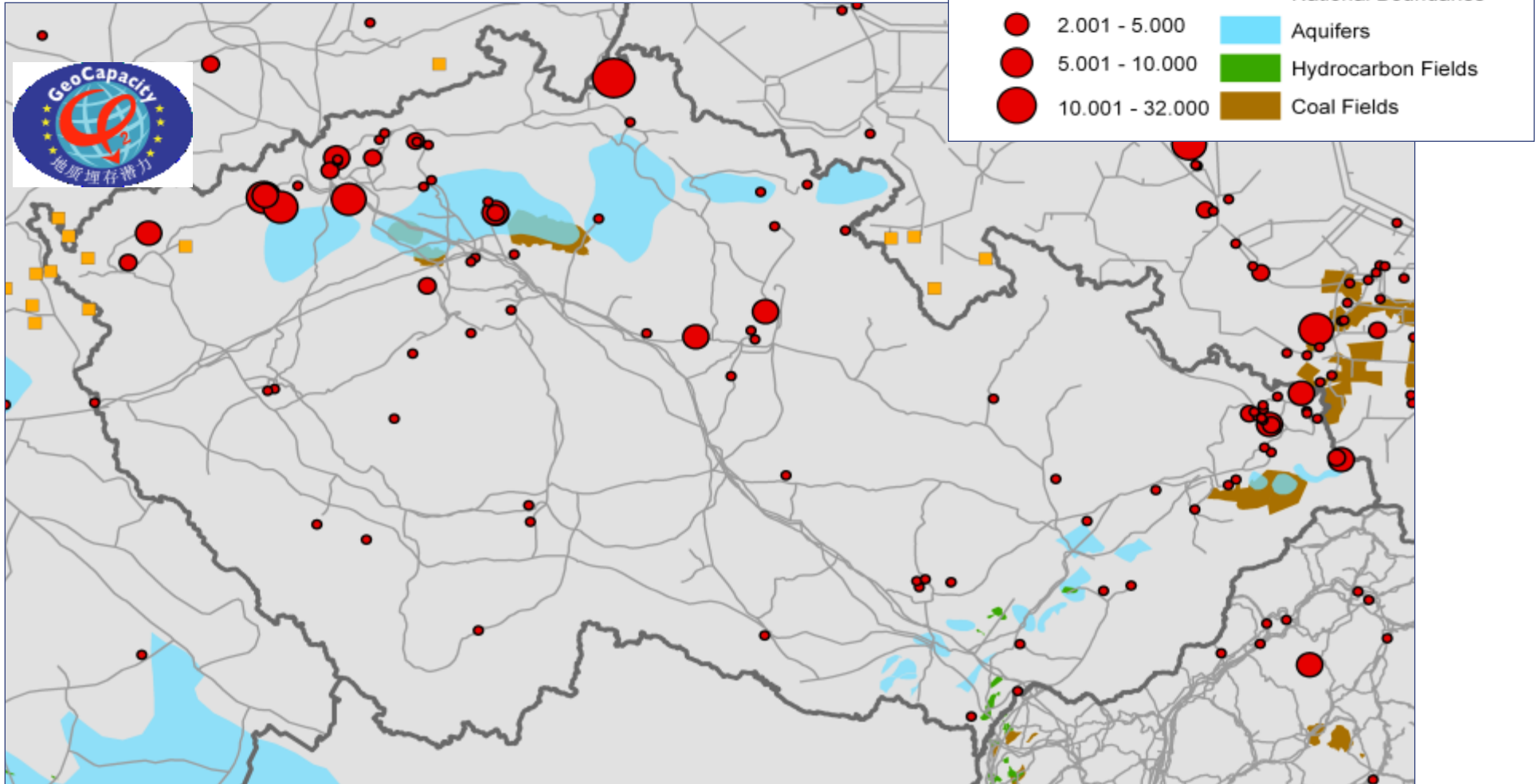
- CCS nepatří mezi priority energetické a klimatické politiky
- Aktualizace Státní energetické koncepce připouští možnou roli CCS po r. 2040 a doporučuje věnovat se výzkumu geologického ukládání CO<sub>2</sub>
- Politika ochrany klimatu 2017 – jeden ze scénářů založen na významném podílu CCS
- Zákon 85/2012 neumožňuje průmyslové využití CCS do r. 2020 a klade další překážky (max. 1 Mt na úložiště, finanční záruka)
- Podpora V&V – program MPO TIP, TAČR
- Nový impuls – Norské fondy 2009-2014 – Program CZ08 (CCS)

# Proč může být technologie CCS v budoucnu pro ČR důležitá?

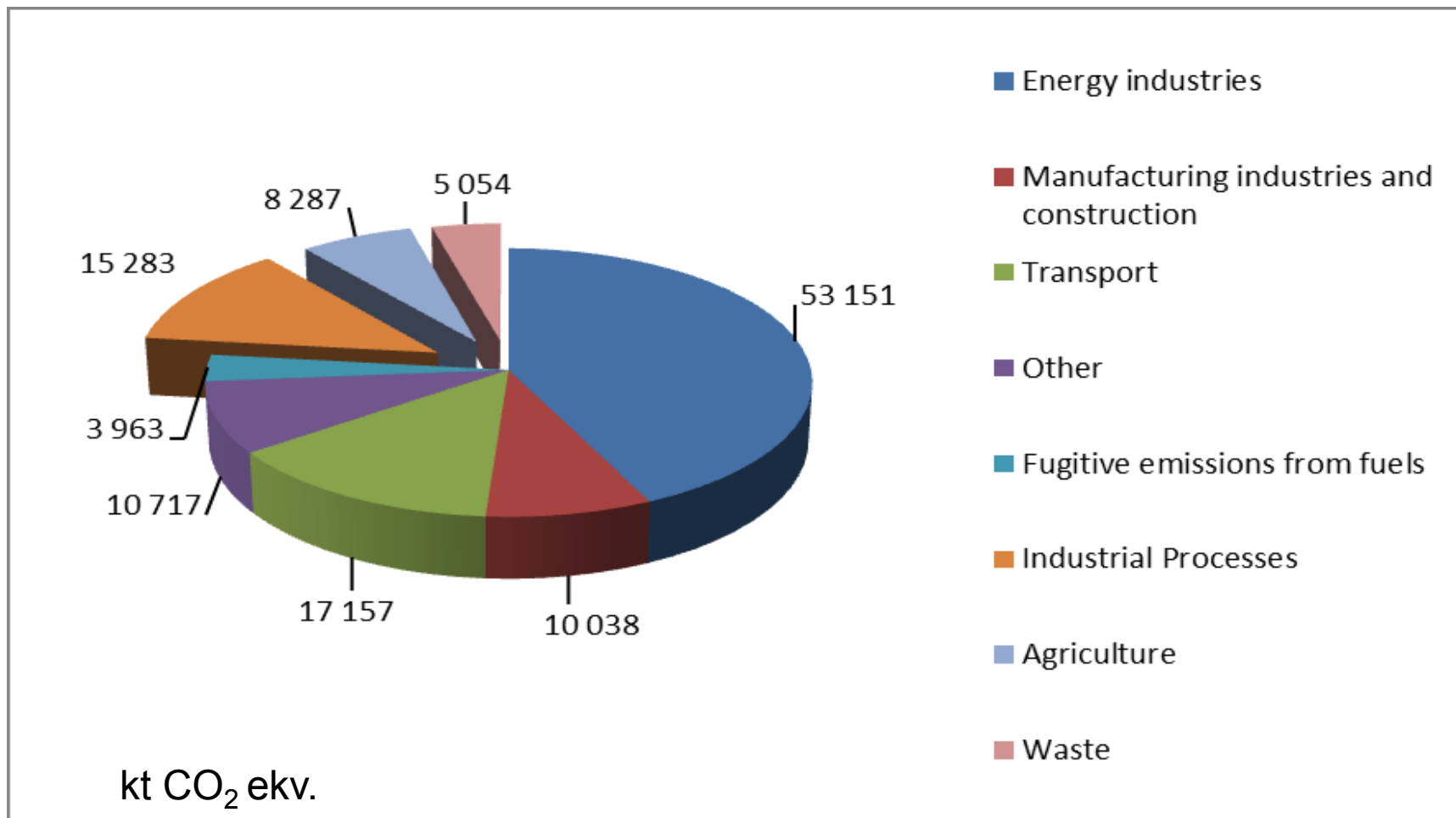
- Je to jediná metoda, která umožňuje snížit emise CO<sub>2</sub> ze spalování fosilních paliv až o 90 % (význam např. při zavedení EPS nebo pro udržení životnosti nových bloků)
- Pro některé průmyslové výroby (ocelárny, cementárny, některé chemické výroby) představuje jedinou alternativu snížení emisí
- Bez CCS bude dosažení cíle 80% snížení emisí v r. 2050 obtížné a drahé
- CCS může být v budoucnu levnější než nákup povolenek (nové technologie 2. generace)
- Kombinace spalování biomasy a CCS je uhlíkově negativní (zlepšení uhlíkové bilance)



# Potenciál pro CCS v ČR

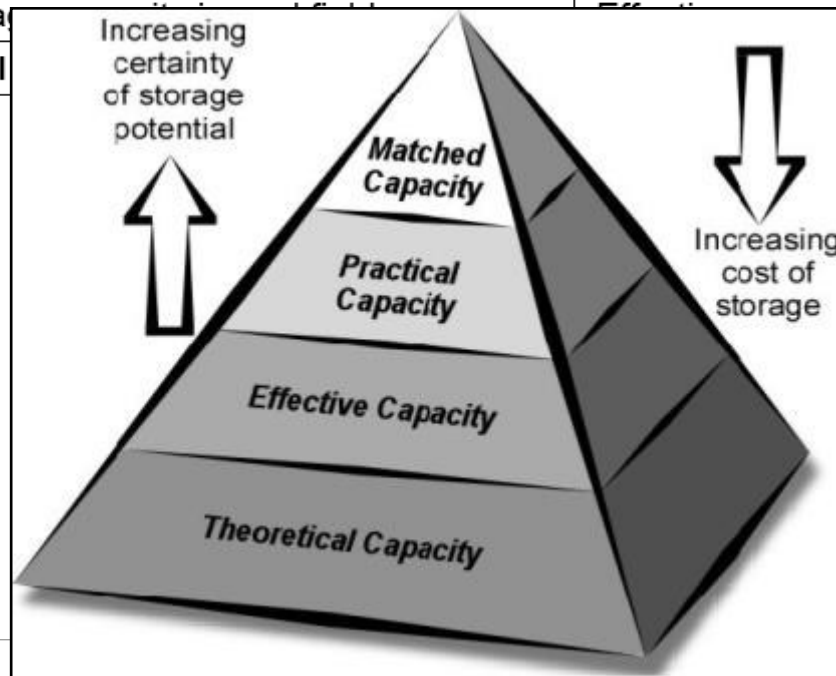


# Emise skleníkových plynů v ČR v r. 2014 podle sektorů



# Potenciál ukládání CO<sub>2</sub> v ČR

CO <sub>2</sub> emissions		Year(s)	Average CO <sub>2</sub> emissions (Mt)
CO <sub>2</sub> emissions from large point sources in database		2005	78
Total CO <sub>2</sub> emissions		2006	128
CO <sub>2</sub> storage capacity	Pyramid class	Conservative estimate (Mt)	Estimate in database (Mt)
Storage capacity in aquifers	Effective	766	2863
Storage capacity in hydrocarbon fields	N/A	33	33
Storage capacity in unconsolidated sediments	Effective	54	54
<b>Total</b>		<b>853</b>	<b>2950</b>



Značná nejistota u akviferů –  
nedostatek dat

# Technicko-ekonomické hodnocení integrace CCS technologií

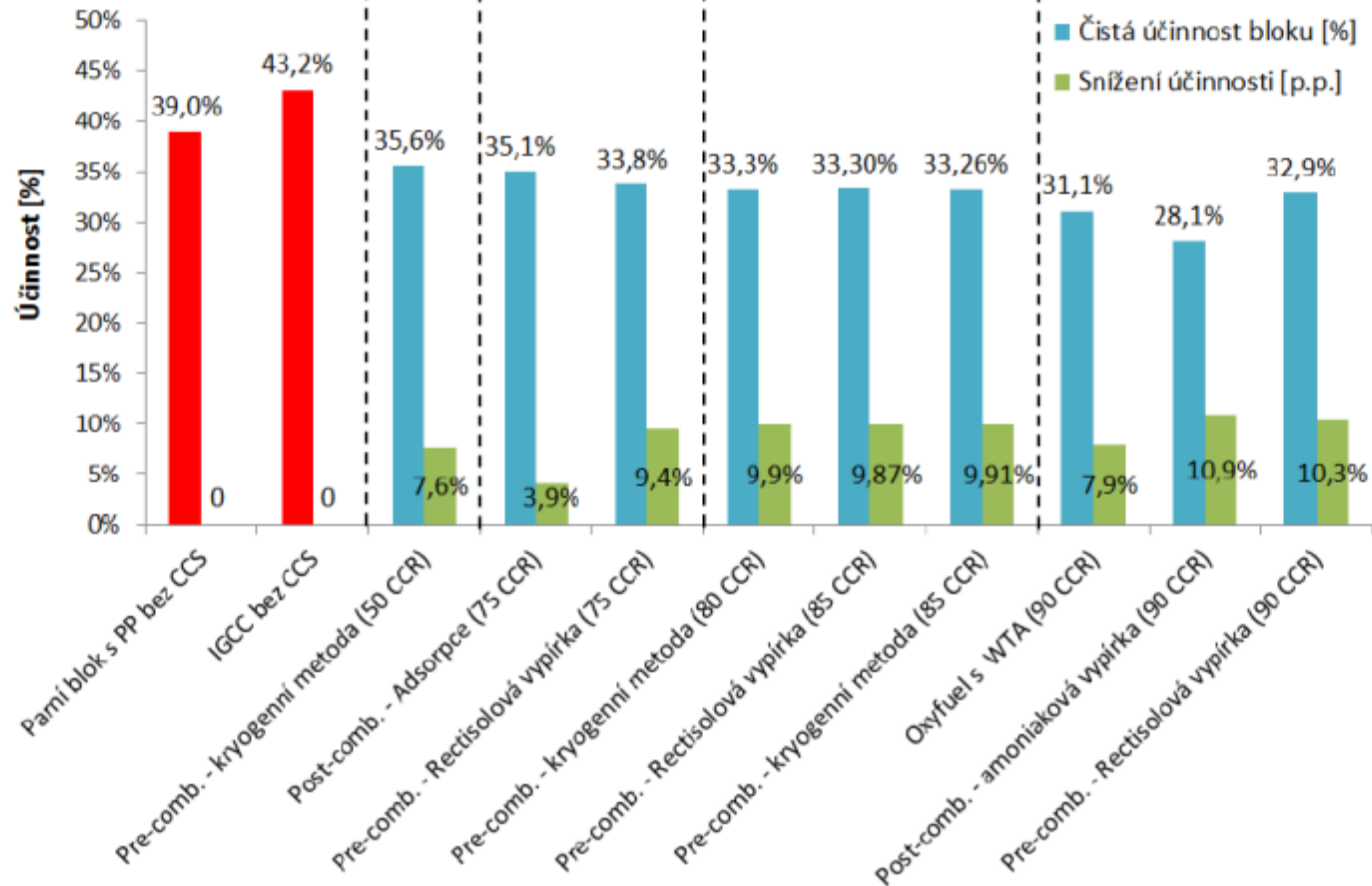


Parametry	FR-TI/379	TA02020205	NF-CZ08-OV-1-003
Elektrárna	Parní blok s podkritickými parametry		IGCC parní blok
Palivo (Lignit)	LHV = 9.75 MJ/kg, W=31%, A=30%, S=3%		LHV = 16.50 MJ/kg, W=31%, A=13%, S=1.3%
Návrhový výkon bloku	250 MWe		310 MWe
CCS technologie	Oxyfuel, post combustion – Amoniaková vypírka	Post-combustion - Adsorpce	Pre-combustion – Rectisolová vypírka, membrány, nízkoteplotní
CCR	90%	75%	85% (50 – 90%)
Období studií	2013	2015	2016

# Technicko-ekonomické hodnocení integrace CCS technologií



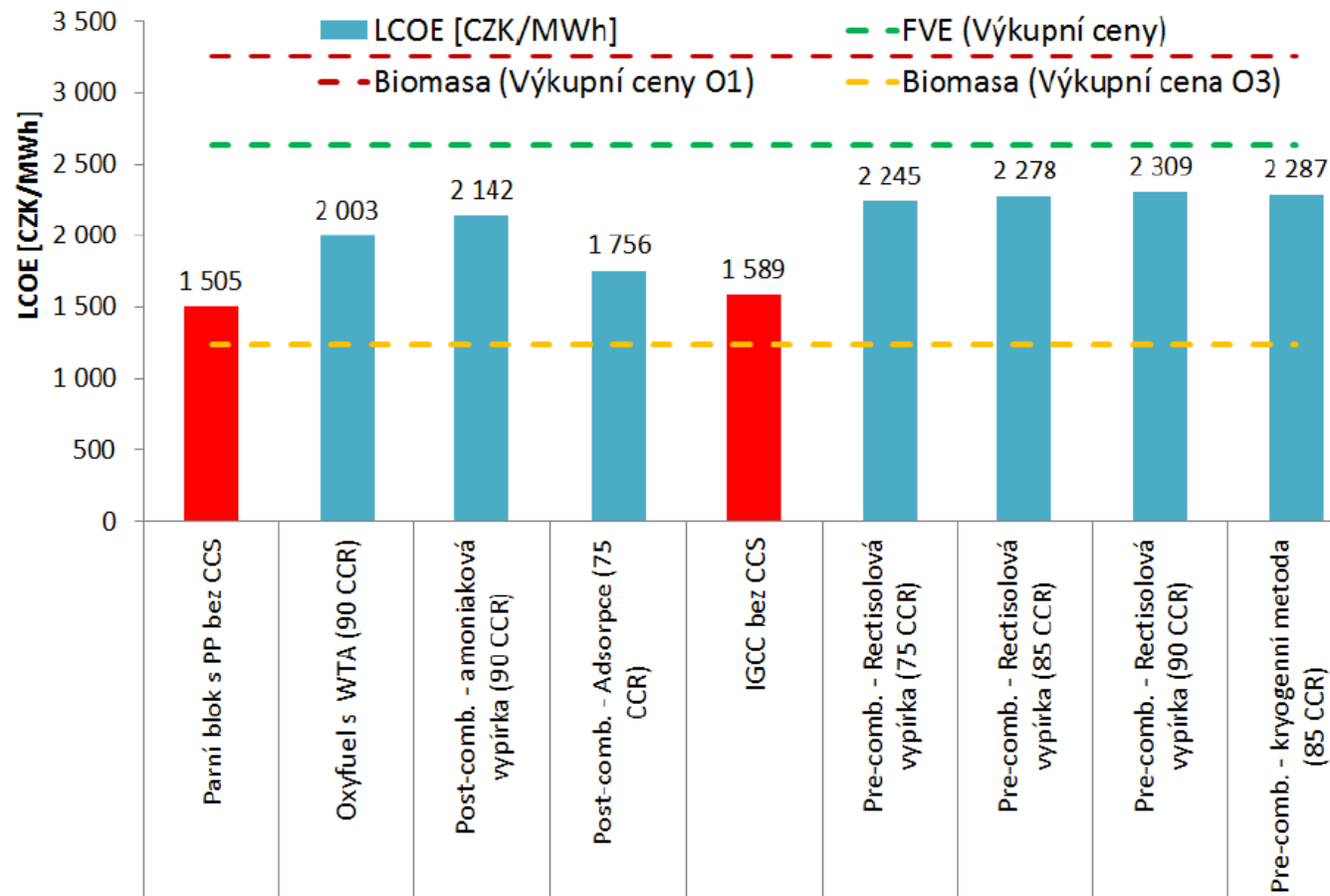
Norway  
grants



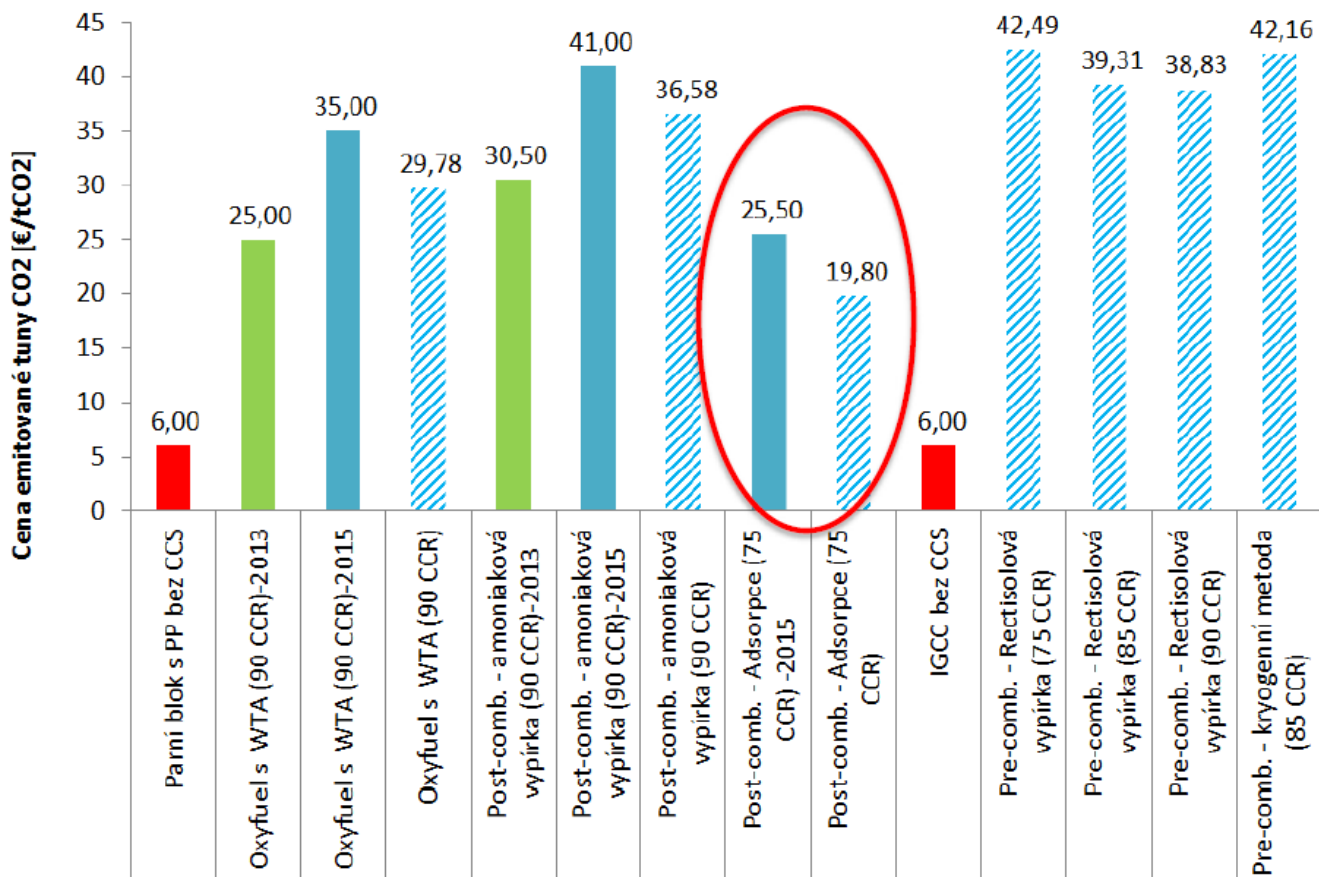
# Technicko-ekonomické hodnocení integrace CCS technologií



Norway  
grants



# Technicko-ekonomické hodnocení integrace CCS technologií



# Shrnutí

- Geologické ukládání CO<sub>2</sub> a technologie CCS představují nadějnou a potenciálně důležitou **součást opatření ke zmírnění změny klimatu**
- Technologie i úložná kapacita **jsou k dispozici**
- Rozsáhlejšímu zavedení technologie do praxe však brání **významné překážky**:
  - nedostatečná politická podpora
  - nejistá cena uhlíkových emisí v budoucnosti
  - nejasné financování demonstračních a prvních komerčních projektů
  - odpor veřejnosti k ukládání CO<sub>2</sub> na pevnině (v osídlených oblastech)

**Přesto – CCS bude potřeba k dosažení cílů snížení emisí; vývoj proto pokračuje a lze očekávat jeho zrychlení v budoucnosti.**



Tato prezentace je výstupem projektu ENOS

[www.enos-project.eu](http://www.enos-project.eu)



**ENOS**  
Enabling Onshore CO<sub>2</sub> Storage

Český informační portál pro technologie CCS najdete na

[www.geology.cz/ccs](http://www.geology.cz/ccs)



This project has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 653718

# BACKUP SLIDES

# introduction to the CCS process chain

- Definition:  
**CCS = Carbon dioxide Capture and Storage** or CO<sub>2</sub> capture and storage  
(also referred to, in short, as Carbon Capture and Storage)  
For Storage, also terms geological storage, sequestration or geosequestration are used.
- CCS is a **climate change mitigation technology** (see Lecture I).
- Main goal = **reduction of CO<sub>2</sub> emissions** to the atmosphere
- **Principle of CCS:** separation (capture) of CO<sub>2</sub> from waste gas streams at emission sources and its safe, long-term isolation from the atmosphere in deep subsurface
- CCS process chain has **three main components** (see next slide for illustration):
  - CO<sub>2</sub> capture – separating CO<sub>2</sub> from the waste gas stream
  - CO<sub>2</sub> transport – delivering the captured CO<sub>2</sub> to the storage site
  - CO<sub>2</sub> storage – injecting the captured CO<sub>2</sub> in a carefully chosen geological structure that will keep it permanently isolated from

# CO<sub>2</sub> capture

- **Purpose:** to produce a concentrated stream of CO<sub>2</sub> at high pressure that can readily be transported to a storage site
- Primary targets = **large-scale emission sources** emitting hundreds of thousands to millions tons of CO<sub>2</sub> annually (smaller sources can be clustered):
  - power plants combusting fossil fuels and/or biomass
  - oil and gas processing facilities
  - industrial plants (cement, iron and steel, pulp and paper production)
  - refineries and petrochemical plants
- Various **capture technologies** are used, based on the composition of the waste gas to be treated, concentration of CO<sub>2</sub> in the gas stream, pressure and temperature.

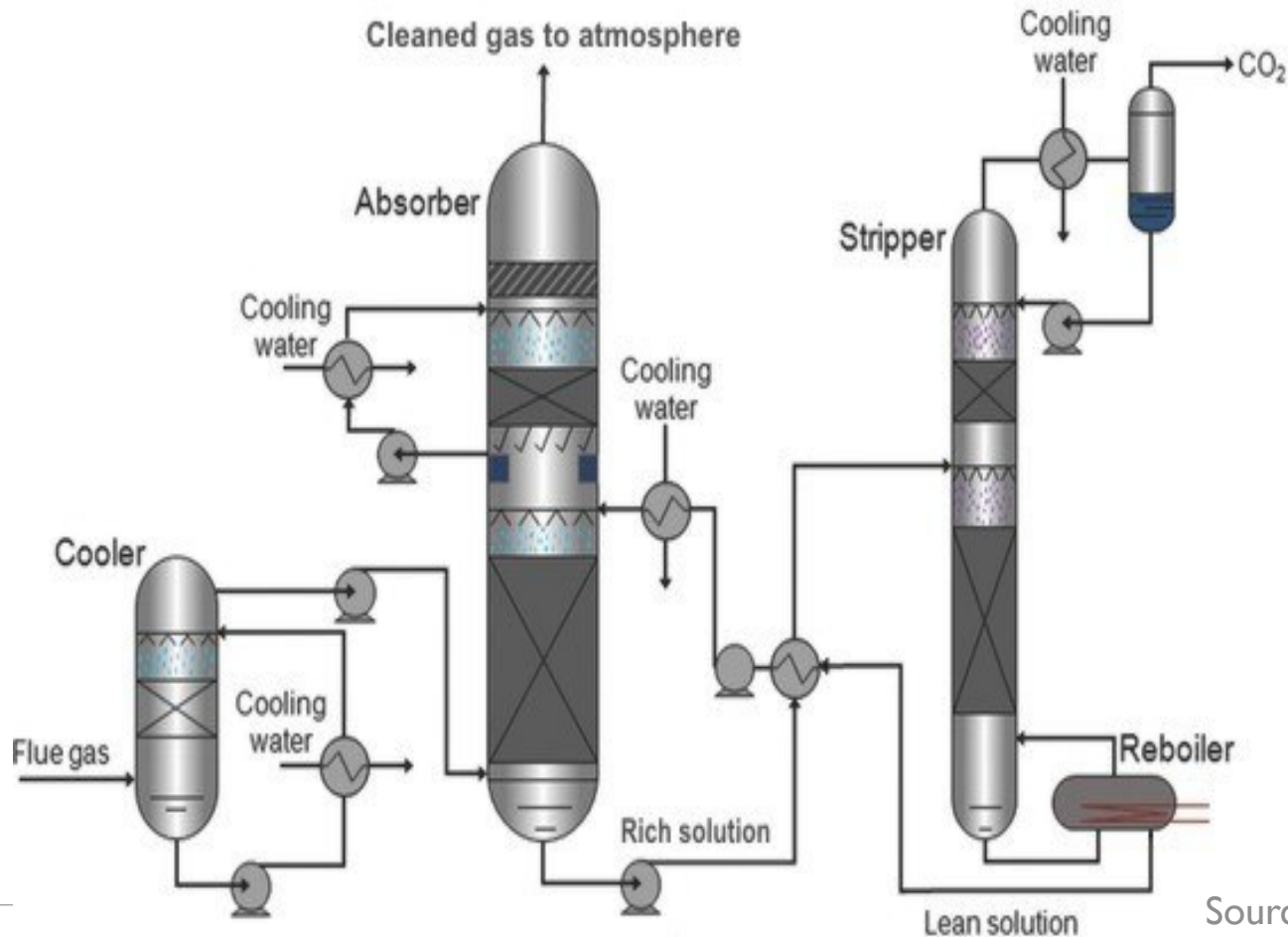
Technologies commercially available at present (1st generation technologies) are based on:

- chemical absorption
- physical absorption
- adsorption

# CO<sub>2</sub> capture by chemical absorption

- **Chemical absorption** is the most common technology used today.
- It utilises a reversible chemical reaction of CO<sub>2</sub> with an aqueous solvent, usually amine or ammonia.
- **The principle** is shown on the next slide:
  - The flue gas containing CO<sub>2</sub> (left) is cooled (if necessary) and transported (using a blower) to the Absorber. Here the CO<sub>2</sub> is brought into contact with the solvent and is chemically bound (typically at 40-60°C). The flue gas (almost CO<sub>2</sub>-free) is water-washed and leaves the Absorber (Cleaned gas to atmosphere).
  - The solvent with chemically bound CO<sub>2</sub> (Rich solution) is then pumped to the Stripper. Here the solvent is regenerated (CO<sub>2</sub> desorption = breaking the chemical bond between the solvent and CO<sub>2</sub>) at elevated temperature (100-140°C) and pressure. This causes a **thermal energy penalty**.
  - The regenerated solvent without CO<sub>2</sub> (Lean solution) is pumped back to the Absorber and the loose CO<sub>2</sub> leaves the Stripper for compression and storage.

# diagram of CO<sub>2</sub> capture process using chemical absorption



Source: Vega et al., 2014

# CO<sub>2</sub> capture by physical absorption

- **Physical absorption** is mostly used in cases where the gas stream has a high concentration of CO<sub>2</sub> at high pressure, or high CO<sub>2</sub> partial pressure (e.g. synthesis gas /syngas/).
- The process is similar to chemical absorption; the CO<sub>2</sub> is, however, captured by physical dissolution in the solvent.
- The regeneration of the solvent (stripping of CO<sub>2</sub> from the solution) is mainly achieved by reducing the pressure → lower energy penalty.
- Several well-established **commercial technologies** are available, e.g.:
  - Selexol (by UOP Ilc)
  - Rectisol (by Linde and Air Liquide)

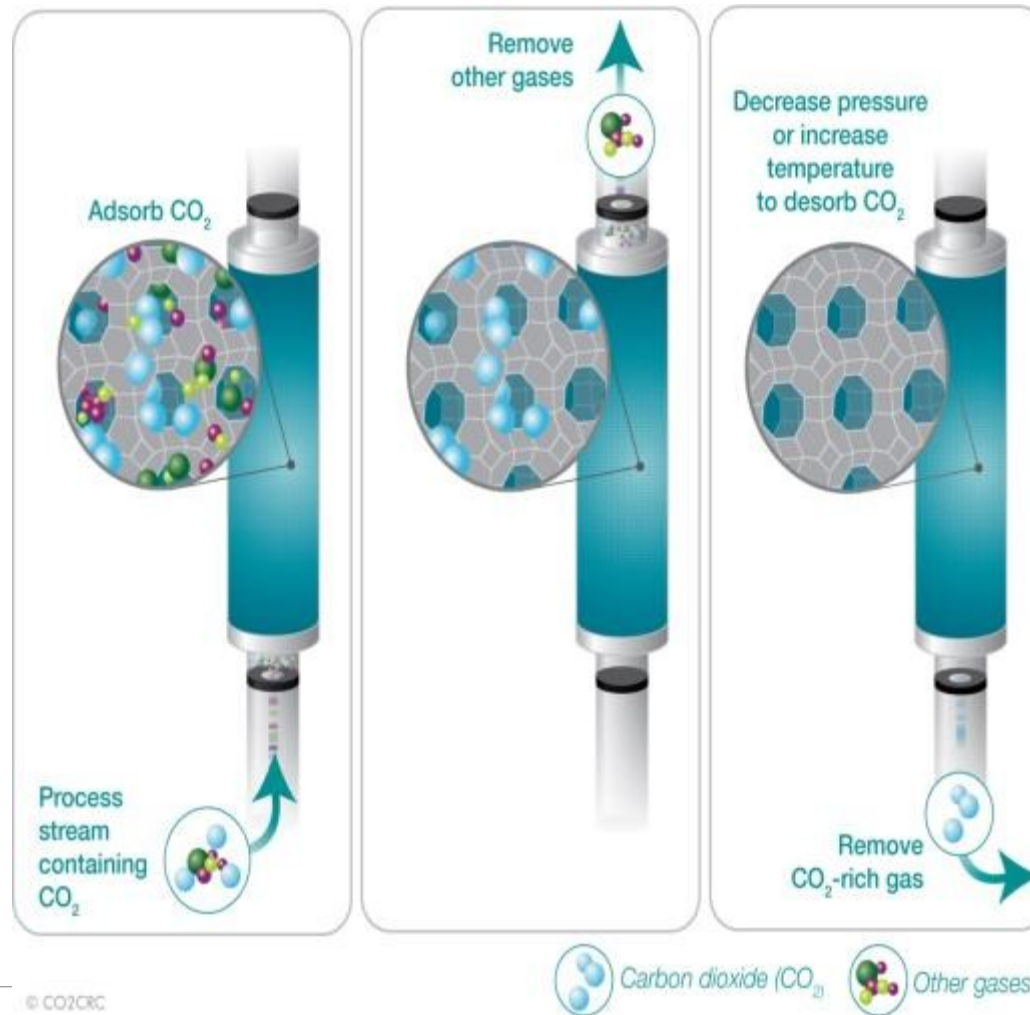
[IPCC,2005; ZEP, 2017]

# CO<sub>2</sub> capture by adsorption

- **Adsorption technology** is based on the adhesion of CO<sub>2</sub> molecules to the surface of the adsorbent where the molecules form a film.
- Adsorption may take place physically, involving van der Waals forces, chemically, involving covalent bonding, or due to electrostatic attraction.
- **The principle** is shown on the next slide:
  - CO<sub>2</sub> contained in the flue gas is adsorbed on the surface of an adsorbent (left)
  - other gases are removed (centre)
  - CO<sub>2</sub> is released (desorbed) by decreasing pressure or increasing temperature (right) and removed from the reactor for compression and storage
- Several examples of **commercial applications** exist, e.g. the Port Arthur Project (Texas, USA) – hydrogen plant; CO<sub>2</sub> is captured from syngas produced by steam methane reformers.



# principle of CO<sub>2</sub> capture by adsorption



© CO2CRC

# next generation technologies

- **2nd generation** capture technologies are currently available at pilot scale (10 or more times smaller than industrial scale); **3rd generation** technologies at laboratory scale.
- They include, e.g.:
  - enhanced solvents for chemical absorption
  - novel adsorbent materials and processes for adsorption
  - CO<sub>2</sub> capture by membranes
  - CO<sub>2</sub> separation by cryogenic distillation
  - microbial and algal systems (biogenic capture)

Next generation technologies are expected to deliver cost reduction, lower penalty in energy efficiency, operational flexibility, reduced health, safety & environment issues, etc.

[ZEP, 2017]

Further reading: ZEP report „Future CCS Technologies“

# CO<sub>2</sub> capture types at combustion processes

- At fossil-fuelled power plants and other combustion processes, three basic types of CO<sub>2</sub> capture are distinguished:
  - **post-combustion capture** – CO<sub>2</sub> is captured after the fuel is burned
  - **pre-combustion capture** – CO<sub>2</sub> is captured within the fuel treatment process
  - **oxy-fuel combustion** – fuel is burned in a special atmosphere (mixture of oxygen and CO<sub>2</sub>) instead of common air
- Emerging technologies – high temperature solid looping cycles (calcium and chemical looping)

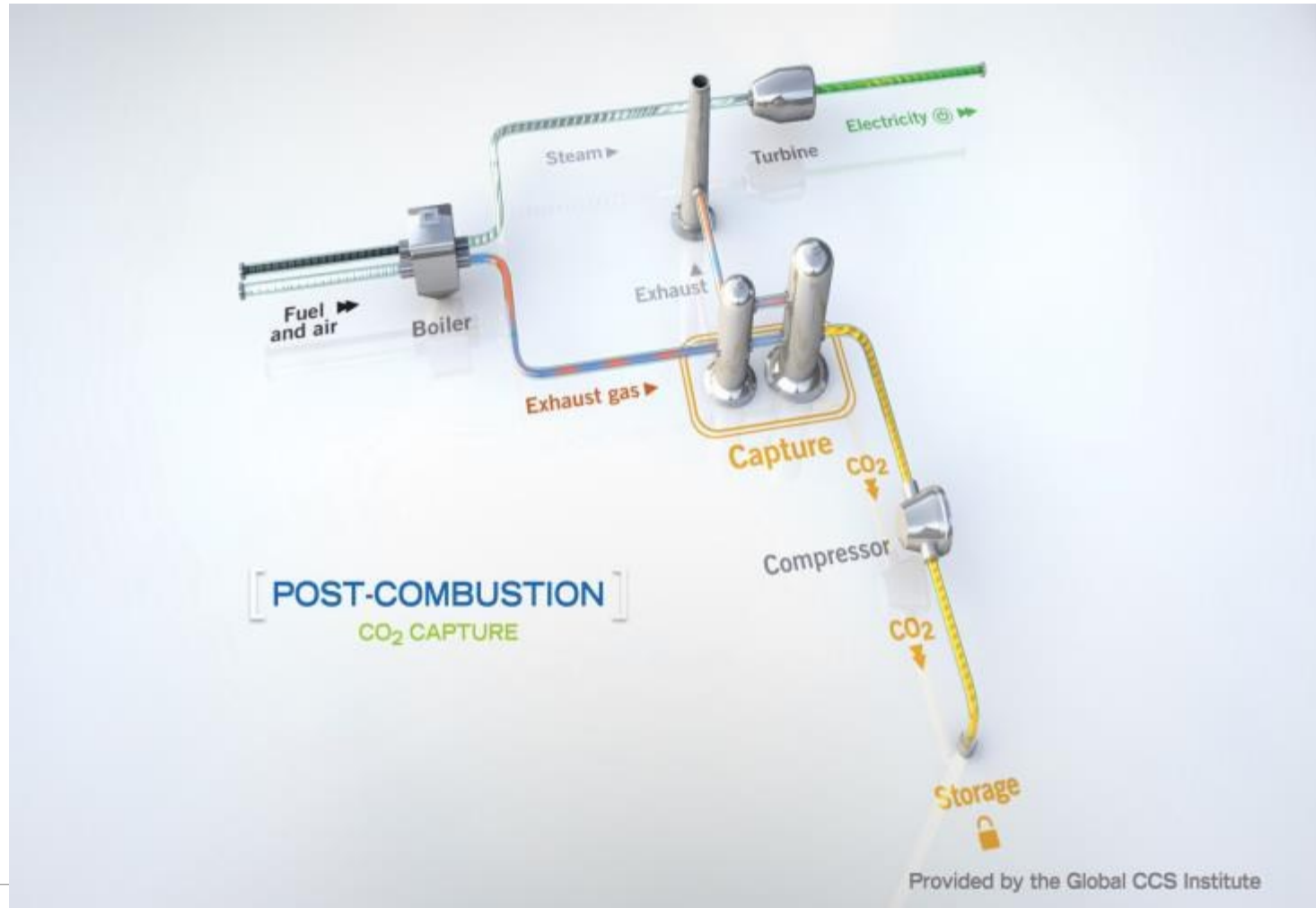
[IPCC,2005; ZEP, 2017]

# post-combustion capture

- The **principle** is shown on the next slide (power plant example):
  - fuel is burned in a boiler in air atmosphere, producing steam that runs a turbine producing electricity
  - flue gas is brought to a capture unit where CO<sub>2</sub> is captured (most commonly using chemical absorption)
  - captured CO<sub>2</sub> is sent for compression and storage while the remaining gas goes to exhaust
- The technology is used at the two currently operating power plants equipped with CCS – **Boundary Dam** (Canada) and **Petra Nova** (Texas, USA)

[IPCC,2005; MIT, 2016; ZEP, 2017]

# principle of post-combustion capture

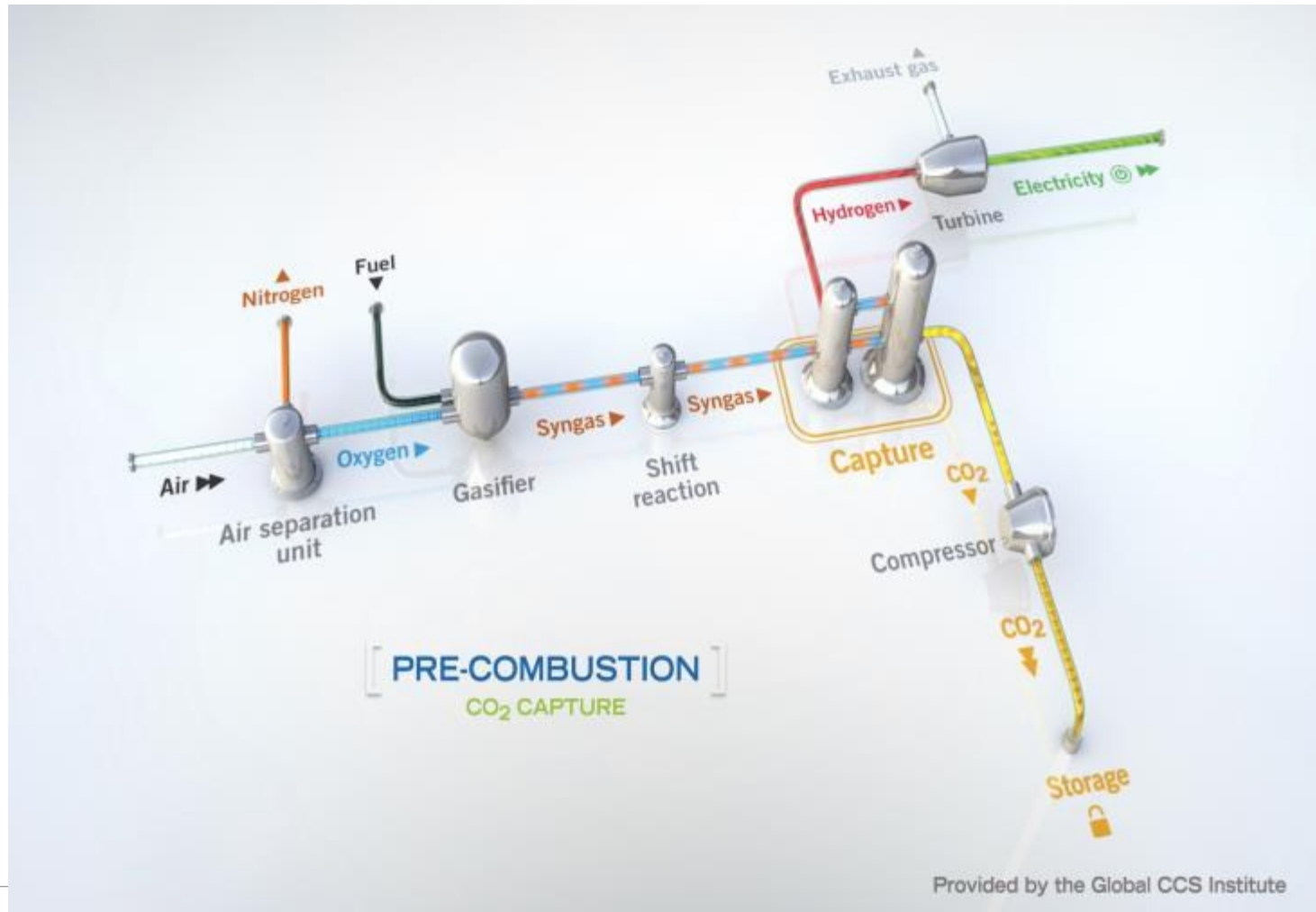


# pre-combustion capture

- The **principle** is shown on the next slide (power plant example):
  - air separation unit /ASU/ separates oxygen from air
  - fuel reacts with oxygen to create synthesis gas /syngas/ – mixture of carbon monoxide /CO/ and hydrogen /H<sub>2</sub>/
  - in the shift reactor syngas reacts with steam and a mixture of H<sub>2</sub> and CO<sub>2</sub> is created
  - the capture plant separates CO<sub>2</sub> (usually by chemical or physical absorption), which is sent to compression and storage
  - the remaining H<sub>2</sub>-rich gas is burned in a gas turbine to produce electricity
  - in addition, the remaining heat can produce steam, which can be used in a steam turbine to produce more electricity, then we have the Integrated Gasification Combined Cycle (**IGCC**)
- The technology has been demonstrated at pilot scale so far.

[IPCC,2005; ZEP,2017]

# principle of pre-combustion capture



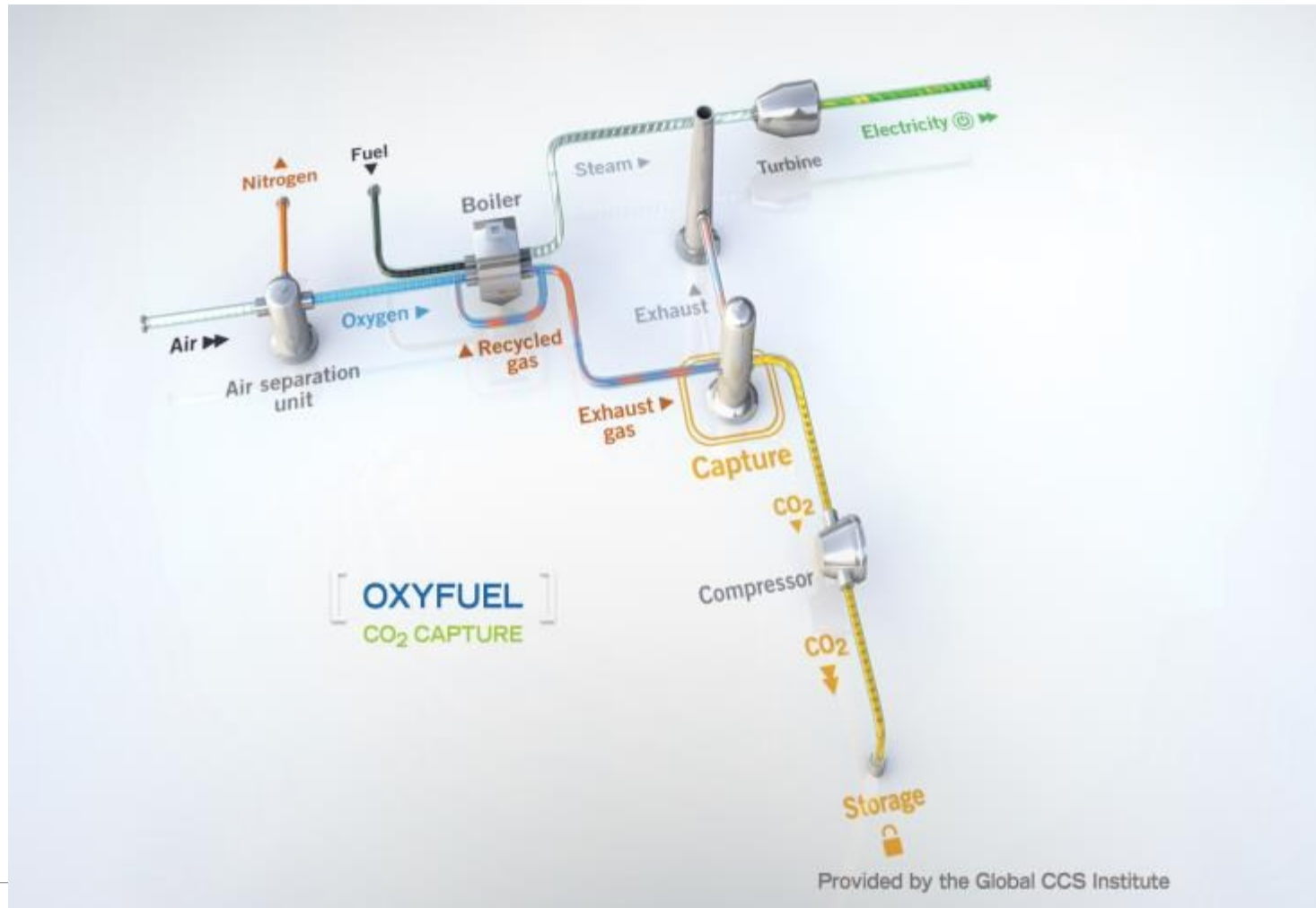
# oxy-fuel combustion capture

- The **principle** is shown on the next slide (power plant example):
  - air separation unit /ASU/ separates oxygen from air
  - fuel is burned in a special boiler in an atmosphere composed of oxygen and recycled flue gas
  - produced steam powers a steam turbine producing electricity
  - exhaust gas (mostly  $\text{CO}_2$  />80%/ and water vapour) is moved to the capture unit where  $\text{CO}_2$  is separated and sent to compression and storage
- The technology has been demonstrated at pilot scale so far.

[IPCC,2005; ZEP,2017]



# principle of oxy-fuel combustion capture



# industrial processes

Examples of processes with CO<sub>2</sub> capture potential:

- **natural gas sweetening** – CO<sub>2</sub> is separated (by chemical or physical absorption) from natural gas produced from the subsurface in case its content is above limit; this is the most frequent CCS application at present, incl. the two sole large-scale CCS projects in Europe – Sleipner and Snøhvit offshore Norway
- **coal/bitumen gasification** with CO<sub>2</sub> capture (mostly by physical absorption) in various modifications, demonstrated at industrial scale, e.g. at:
  - synthetic natural gas production (Great Plains Synfuels Plant, North Dakota, USA)
  - hydrogen production (North West Sturgeon Refinery, Alberta, Canada)
- hydrogen production by **steam methane reforming** with CO<sub>2</sub> capture (absorption or adsorption from CO<sub>2</sub>-H<sub>2</sub> mixture) – e.g. Port Arthur Project (Texas, USA), Shell Quest Project (Alberta, Canada)

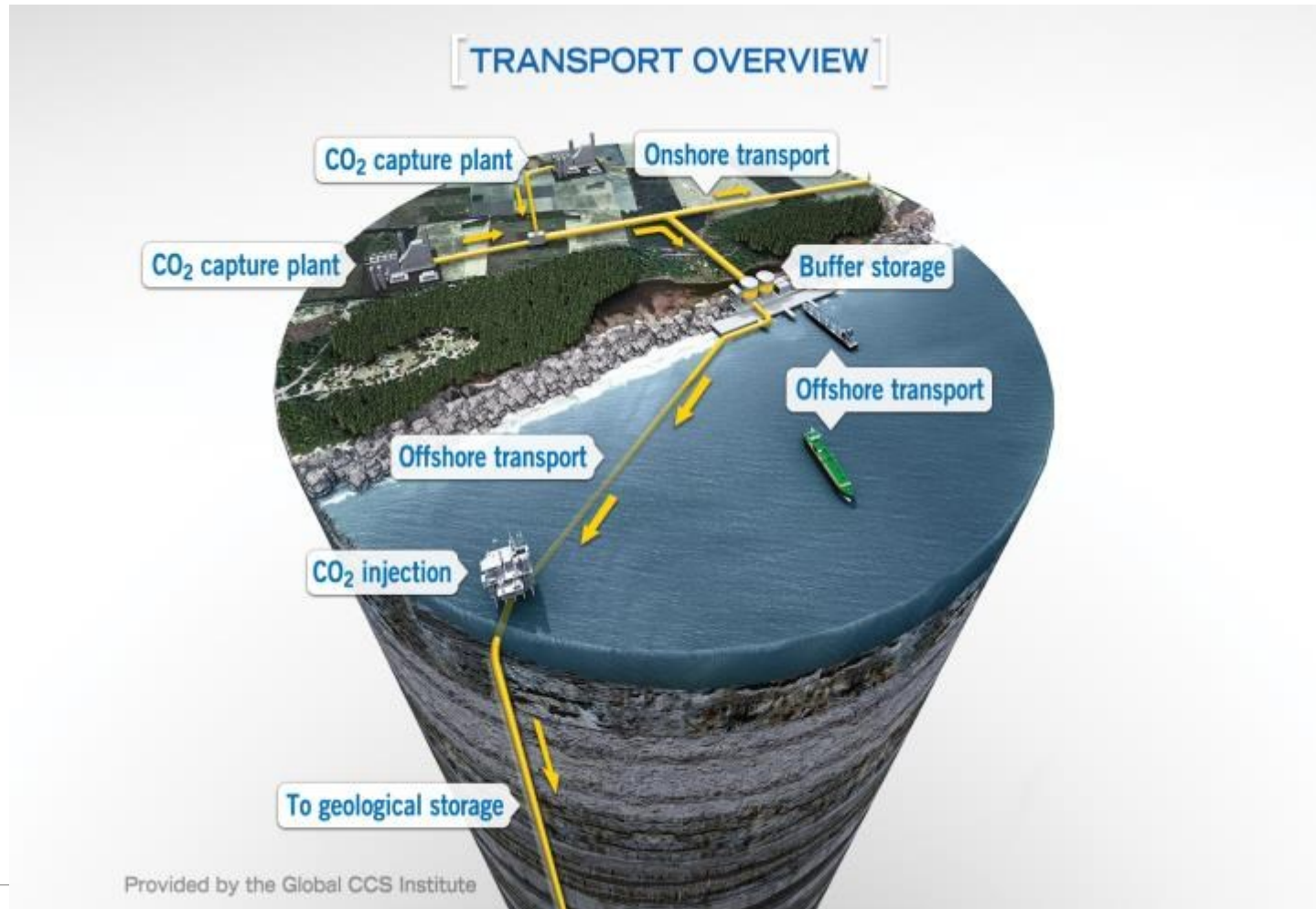
# industrial processes (cont.)

Examples of processes with CO<sub>2</sub> capture potential (cont.):

- chemical industry:
  - **ammonia production** – CO<sub>2</sub> is a by-product, usually in form of a highly concentrated stream → easier capture and purification
  - **ethanol production** - commercial-scale example – ADM Illinois Industrial CCS Project (USA) – the first large-scale BioCCS application; CO<sub>2</sub> is a >99% pure stream which is a by-product of ethanol production via anaerobic fermentation
- **steel production** – 1st demonstration project = ESI CCS Project (United Arab Emirates) – CO<sub>2</sub> is captured (chemical absorption) from waste gas produced by Direct Reduced Iron process (iron ore /Fe<sub>2</sub>O<sub>3</sub>/ is reduced to iron /Fe/ in a reactor, using hydrogen brought in syngas)
- **cement production** – CO<sub>2</sub> capture demonstrated at pilot scale

[Abu Zahra, 2015; IPCC, 2005; ZEP, 2017]

# CO<sub>2</sub> transport



# CO<sub>2</sub> transport

Three **main options** of CO<sub>2</sub> transport:

- pipelines – big-scale projects, both on- and offshore
- trucks & rail tankers – small onshore projects (typically pilot-scale)
- ships – small- and mid-scale offshore projects, especially with longer distance from capture plant to storage site

**CO<sub>2</sub> pipelines** are a well-established technology, especially in North America:

- almost 5000 km of CO<sub>2</sub> pipelines are operating in the USA and Canada
- typical operation conditions – pressure 10-15 MPa, temperature 15-30°C → CO<sub>2</sub> is transported in liquid phase, with high density
- compressors are used to pressurize the pipeline, booster stations are needed for longer distances
- additional processing of CO<sub>2</sub>-rich gas (after capture) is usually required to decrease the content of impurities, especially water vapour to limit corrosion risk

# CO<sub>2</sub> transport (cont.)

CO<sub>2</sub> transport by trucks and/or rail tankers is a well established industrial technology.

- Liquefaction of captured CO<sub>2</sub> is needed = compression, cooling, dehydration and distillation.
- Typical parameters for truck transport in Europe:
  - capacity of one truck - in the order of ~ 25 tons
  - pressure ~ 1.6 MPa
  - temperature ~ -30°C

## Ship transport:

- current ships (capacity  $\leq 1800$  t CO<sub>2</sub>) use similar pressures and temperatures to trucks
- bigger ships are needed for industrial CCS projects – lower temperatures (~ -52°C) and pressures (~ 0.65 MPa) are proposed

[Brownsort, 2015; IPCC,2005; ZEP,2017]

# CO<sub>2</sub> storage (more details in e-books 3-9)

„A geological formation shall only be selected as a storage site, if under the proposed conditions of use there is no significant risk of leakage, and if no significant environmental or health risks exist.“

European CCS directive [EU, 2009]

**Main types** of geological formations suitable for geological storage of CO<sub>2</sub> (see sketch diagram in the next slide):

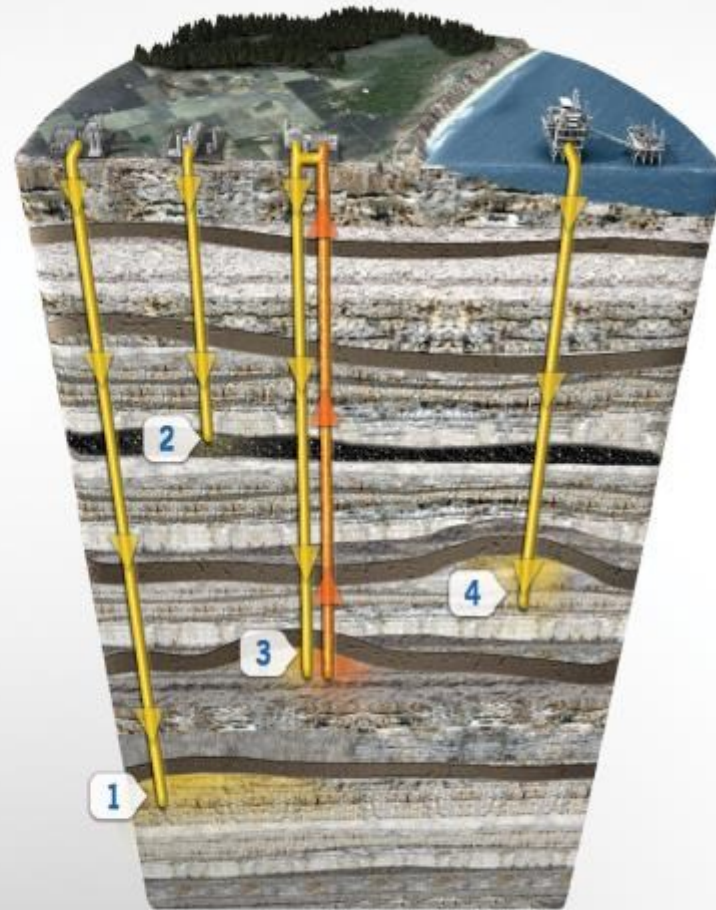
- Currently used options:
  - **deep saline aquifers** – more than 800 m below surface, rock pores contain brine (salt water) that is pushed back by injected CO<sub>2</sub>
  - **depleted hydrocarbon (oil and gas) fields** – CO<sub>2</sub> can fill the pore space made „empty“ by oil and gas production
- Emerging options:
  - unmineable coal layers (seams)
  - basalt formations

# CO<sub>2</sub> storage options

## [ STORAGE OVERVIEW ]

### SITE OPTIONS

- 1 Saline formations
- 2 Injection into deep unmineable coal seams or ECBM
- 3 Use of CO<sub>2</sub> in enhanced oil recovery
- 4 Depleted oil and gas reservoirs



Provided by the Global CCS Institute



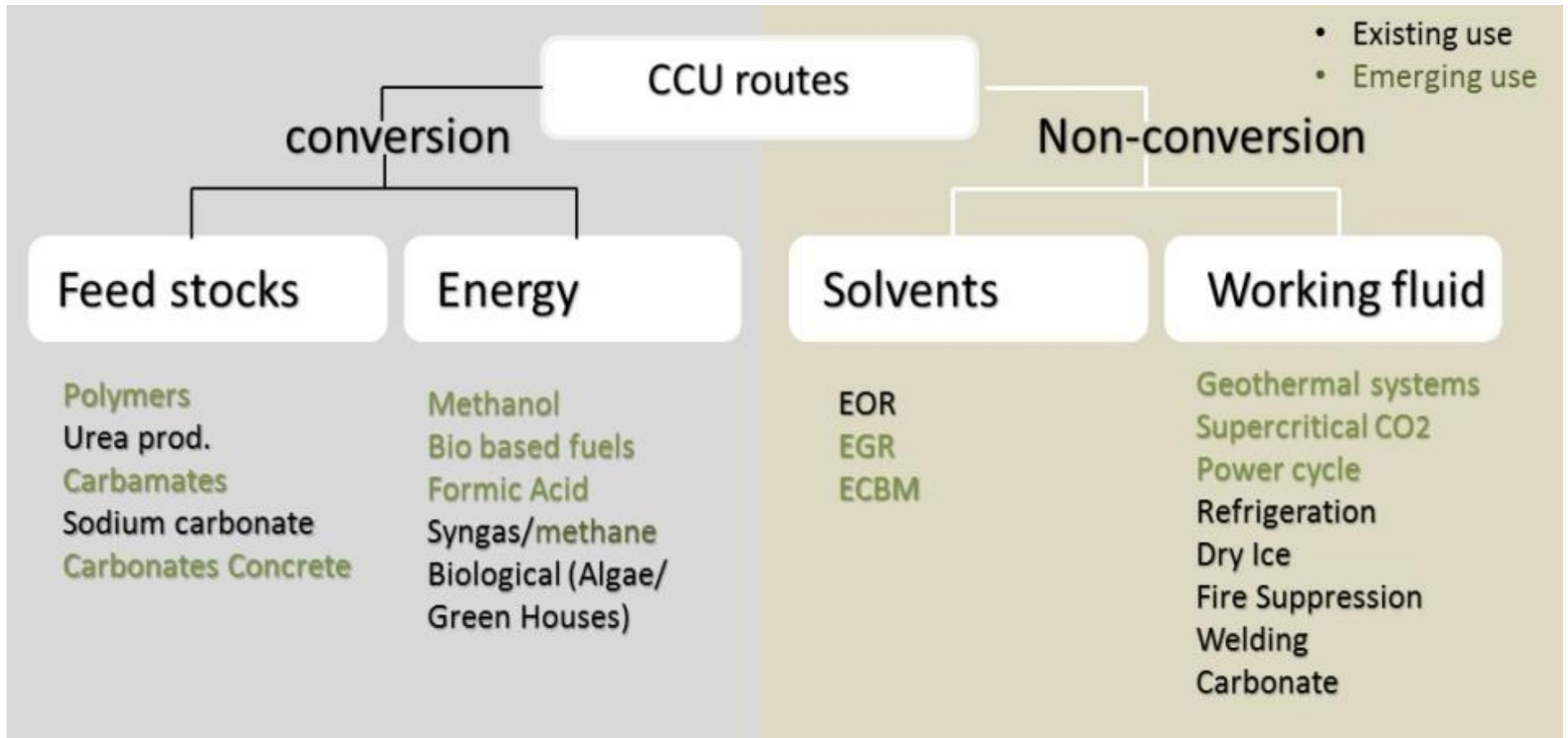
# CO<sub>2</sub> utilisation

**CO<sub>2</sub> capture and utilisation (CCU)**, sometimes also called CO<sub>2</sub> capture and use or re-use, handles CO<sub>2</sub> not as a waste but as a commercial product. If combined with permanent storage (incl. geological), it is abbreviated as CCUS.

**CO<sub>2</sub> utilisation routes** are shown in the next slide. Two main options are available:

- conversion – uses CO<sub>2</sub> as a source of carbon for various feed stocks or energy products, energy input is required for conversion; in many cases CCU can displace the use of fossil fuels
- non-conversion – CO<sub>2</sub> acts as a solvent, or working fluid  
This option includes various modifications of Enhanced Hydrocarbon Recovery where CO<sub>2</sub> is used to stimulate production of hydrocarbons from oil & gas fields and coal seams:
  - EOR = Enhanced Oil Recovery
  - EGR = Enhanced Gas Recovery
  - ECBM = Enhanced Coal-Bed Methane Recovery

# CO<sub>2</sub> utilisation routes



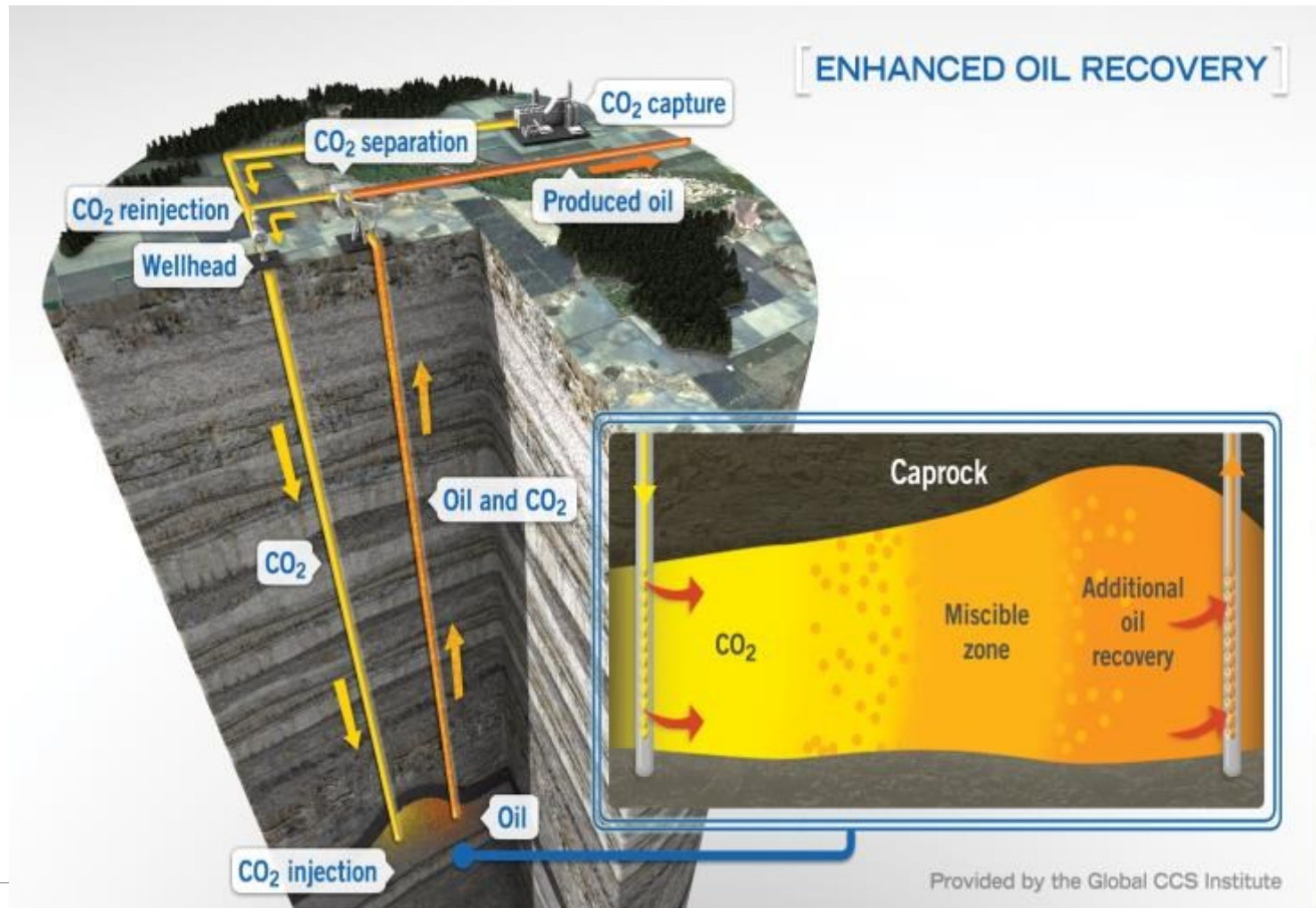
Source: ZEP, 2015

# Enhanced Oil Recovery (EOR)

CO<sub>2</sub>-driven Enhanced Oil Recovery (**CO<sub>2</sub>-EOR**) is a well-established CCUS industrial activity practiced from 1970s in the USA and Hungary.

- CO<sub>2</sub> acts as agent enabling to gain more oil from the field (see next slide) after primary production and (possibly) water-flooding effects are exhausted → **economic revenue**
- large part of the injected CO<sub>2</sub> stays in the reservoir; the rest is produced together with oil, separated and recycled
- originally the intention was to just produce more oil; nowadays the CO<sub>2</sub> storage element attracts more attention (so called „**EOR+**“ **concept** – injection scenarios are optimised to increase the volume of stored CO<sub>2</sub> + after production stops the wells are plugged and the field is turned to CO<sub>2</sub> storage)
- largely developed in the USA (> 120 projects, > 60 mil. tons of CO<sub>2</sub> used /~25% from anthropogenic sources/, > 280,000 barrels of oil produced per day, lack of CO<sub>2</sub> on the market), further projects in Canada, Brazil, Hungary, Turkey, Croatia

# principle of CO<sub>2</sub>-EOR



# limitations of CCU

- Some CCU technologies involve permanent storage of CO<sub>2</sub> (CCUS) but some do not – the storage is just temporary, or new products are delivered. It is always important to assess **if the technology contributes to climate change mitigation** = reduction of CO<sub>2</sub> emissions to the atmosphere. Carbon footprint and Lifecycle analyses are necessary for this assessment.
- The **amount of CO<sub>2</sub>** that various CCU(S) technologies can handle varies in the extent of several orders of magnitude. Some technologies can utilise only small amounts of CO<sub>2</sub>, without a real potential to increase → their mitigation potential is small.
- Some CCU products compete on the market with commercially well-established ones produced by „classical“ technologies. If the CCU products are more expensive, their **market potential is limited**. This can be only changed by:
  - incentives and subsidies to support CCU
  - regulation, e.g. ban of carbon-intensive products, carbon tax, etc.

# costs of CCS

The implementation of CCS technology **raises the investment and operational costs** for power and industrial projects. New power plants and industrial facilities can be designed to incorporate CCS from their inception, or the technology can be retrofitted to existing sources of CO<sub>2</sub> emissions. Overall, **the cost of each project can vary considerably**, depending especially on:

- the type of process generating CO<sub>2</sub> – e.g. power generation, cement, steel, hydrogen, ammonia production, etc.
- CO<sub>2</sub> content in the waste gas stream and its composition – generally higher CO<sub>2</sub> content and simpler gas composition mean lower capture costs
- the size of the project – big projects have normally lower per-unit cost due to economy of scale
- CO<sub>2</sub> capture technology used and its maturity – first-of-a-kind projects are generally very expensive, the following ones can benefit from their experience → „**learning curve**“
- distance to storage site

## costs of CCS (cont.)

- type of transport – pipelines are cost-effective for large volumes of CO<sub>2</sub> transported; ship transport can be beneficial for long distances and/or smaller volumes
- location of storage site – offshore storage is more expensive than onshore
- type of storage reservoir – depleted oil & gas fields are cheaper to develop than saline aquifers because of higher level of knowledge (abundant data are usually available from exploration and production of the field)

Other important observations:

- CO<sub>2</sub> capture usually represents the majority of CCS costs – most often reported as 75% or higher.
- CCS projects normally have high investment costs – the technology is not easily scalable, contrary to e.g. some renewable energy sources (wind, solar).
- CCS networks with transport and storage hubs can significantly decrease unit transport and storage costs.

# costs of CCS in power generation

CCS is a **cost-competitive low-carbon** power generation **technology** – its costs are comparable with other options (renewables, nuclear).

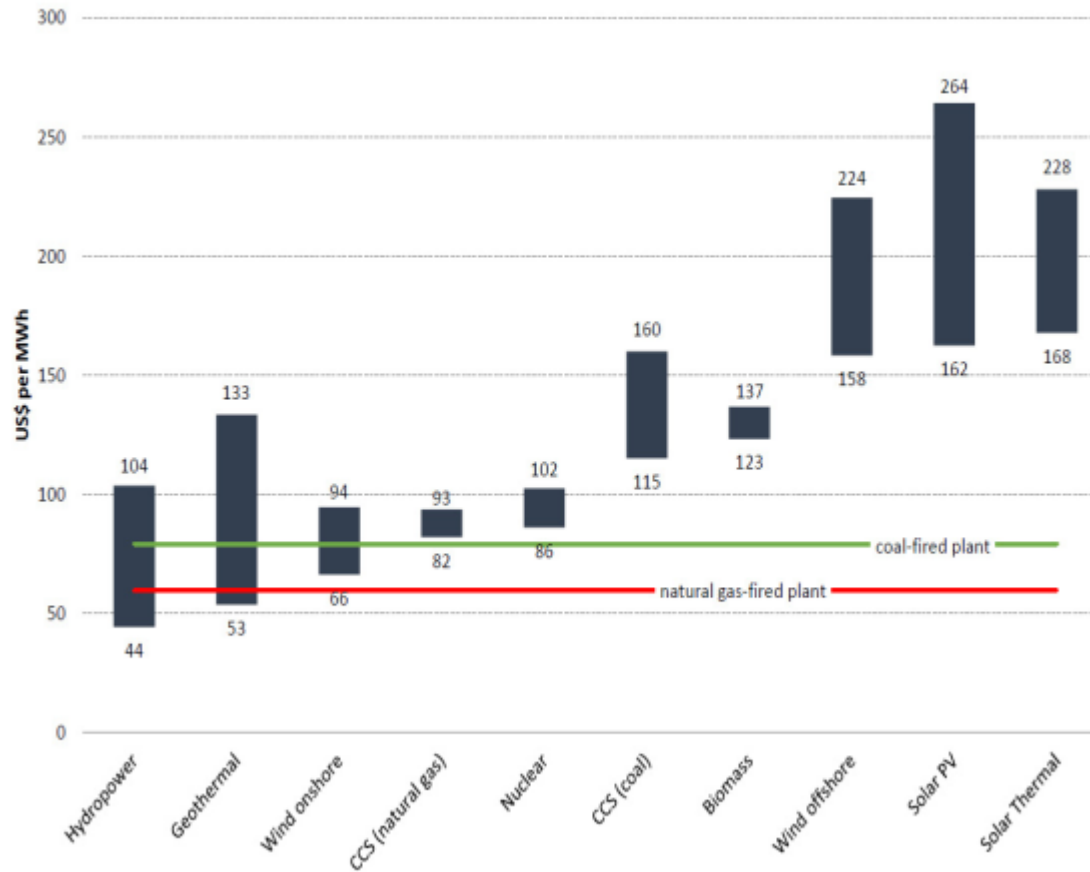
Exact cost comparison is difficult because various factors need to be taken into account, like e.g.:

- load factor (number of operation hours of the power plant per year)
- regime (baseload = continuous delivery of power vs. peak load = delivery on demand, when power supply is lacking)
- intermittency (continuous delivery of power vs. intermittent delivery depending on weather /wind energy/ or daytime /solar energy/)

**Levelised cost of electricity** (LCOE) is a frequently used measure to compare the costs (see next slide). It represents costs per unit of electricity generated over the life of a particular plant.



# levelised cost of electricity for low-carbon power generation technologies



# summary

- CO<sub>2</sub> capture, transport and storage are the three main parts of the CCS technological chain
- CO<sub>2</sub> capture can be applied at big point sources of CO<sub>2</sub> emissions - power plants combusting fossil fuels and/or biomass, oil and gas processing facilities, industrial plants (cement, iron and steel, pulp and paper production), refineries and petrochemical plants, etc.
- currently available CO<sub>2</sub> capture technologies are based on chemical or physical absorption and adsorption
- main CO<sub>2</sub> transport options include pipelines, ships and trucks
- favourable CO<sub>2</sub> storage structures are depleted hydrocarbon fields and deep saline formations (aquifers)
- CCU handles CO<sub>2</sub> not as a waste but as a commercial product that can be further utilised as feed stock, energy source, solvent or working fluid
- implementation of CCS technology raises the investment and operational costs for power and industrial projects

# **Budeme v ČR potřebovat CCS?**

# BUDEME!

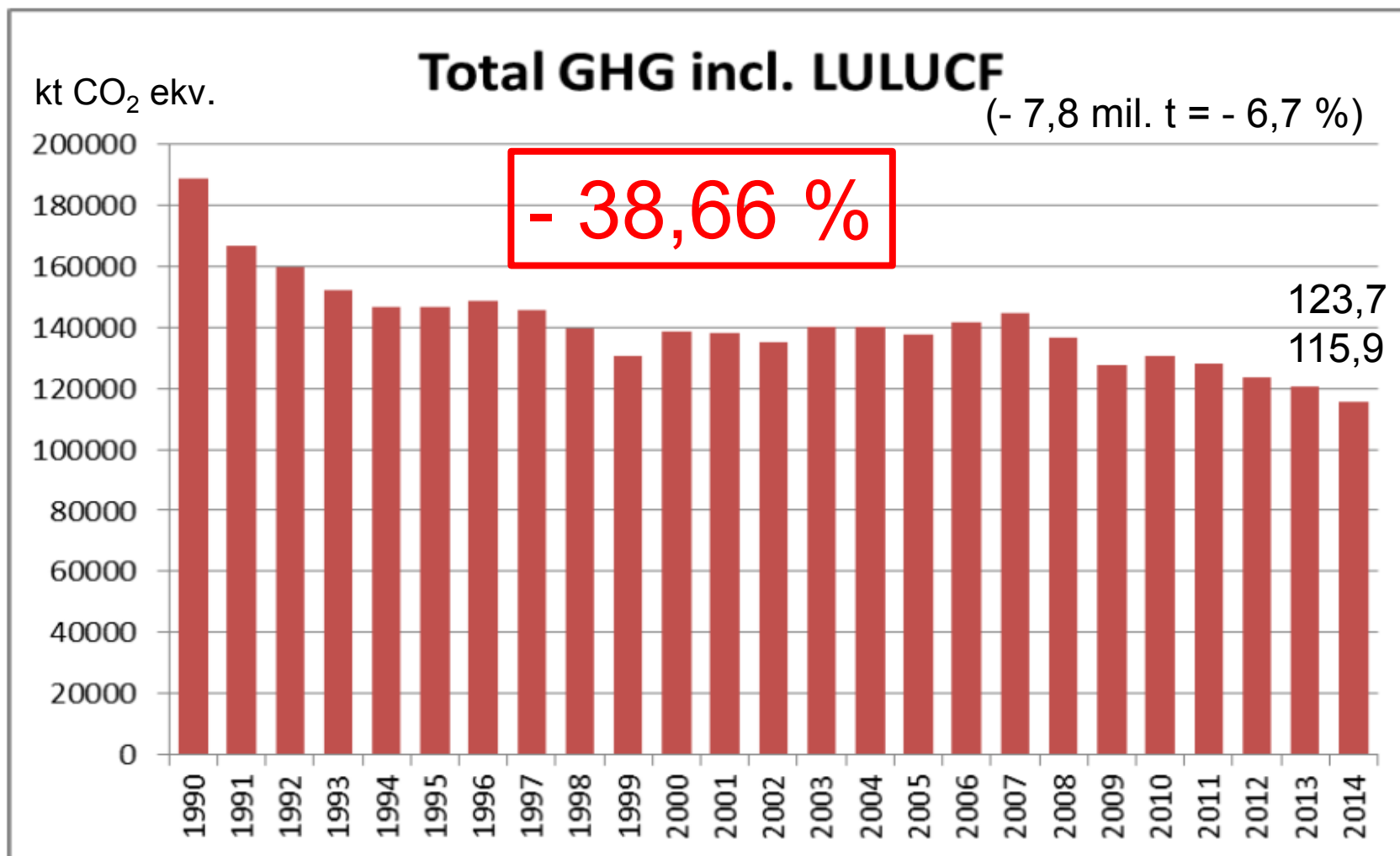
- Potřebujeme všechny nízkouhlíkové technologie
- S CCS to bude levnější než bez CCS
- CCS umožní spalovat fosilní paliva s nízkými emisemi CO<sub>2</sub>
- Je to jediná možnost dekarbonizace pro některé technologie
- Umožní negativní emise (bio-CCS)

# NEBUDEME!

- Je to příliš drahé
- Máme jiné, lepší technologie
- Je to nebezpečné
- Je to nevyzkoušené
- Obyvatelstvo to nedovolí
- Bude to zakázáno

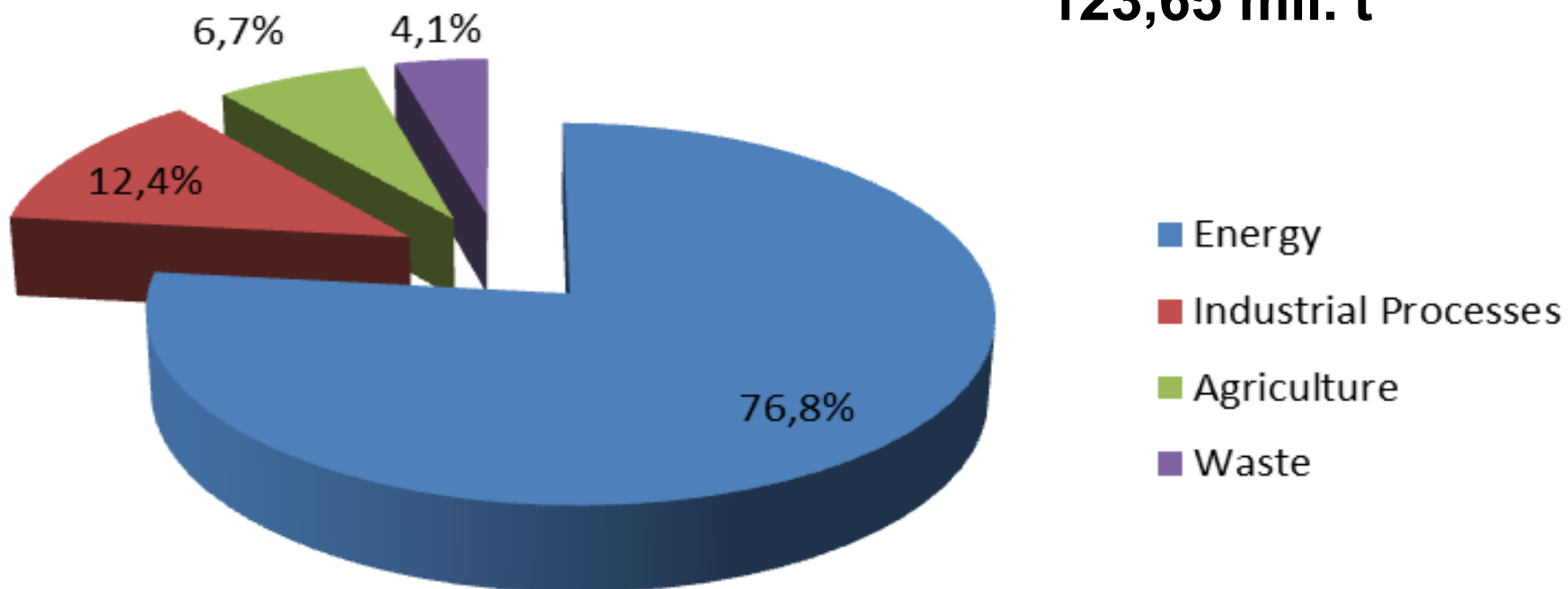


# Vývoj emisí skleníkových plynů v ČR 1990-2014

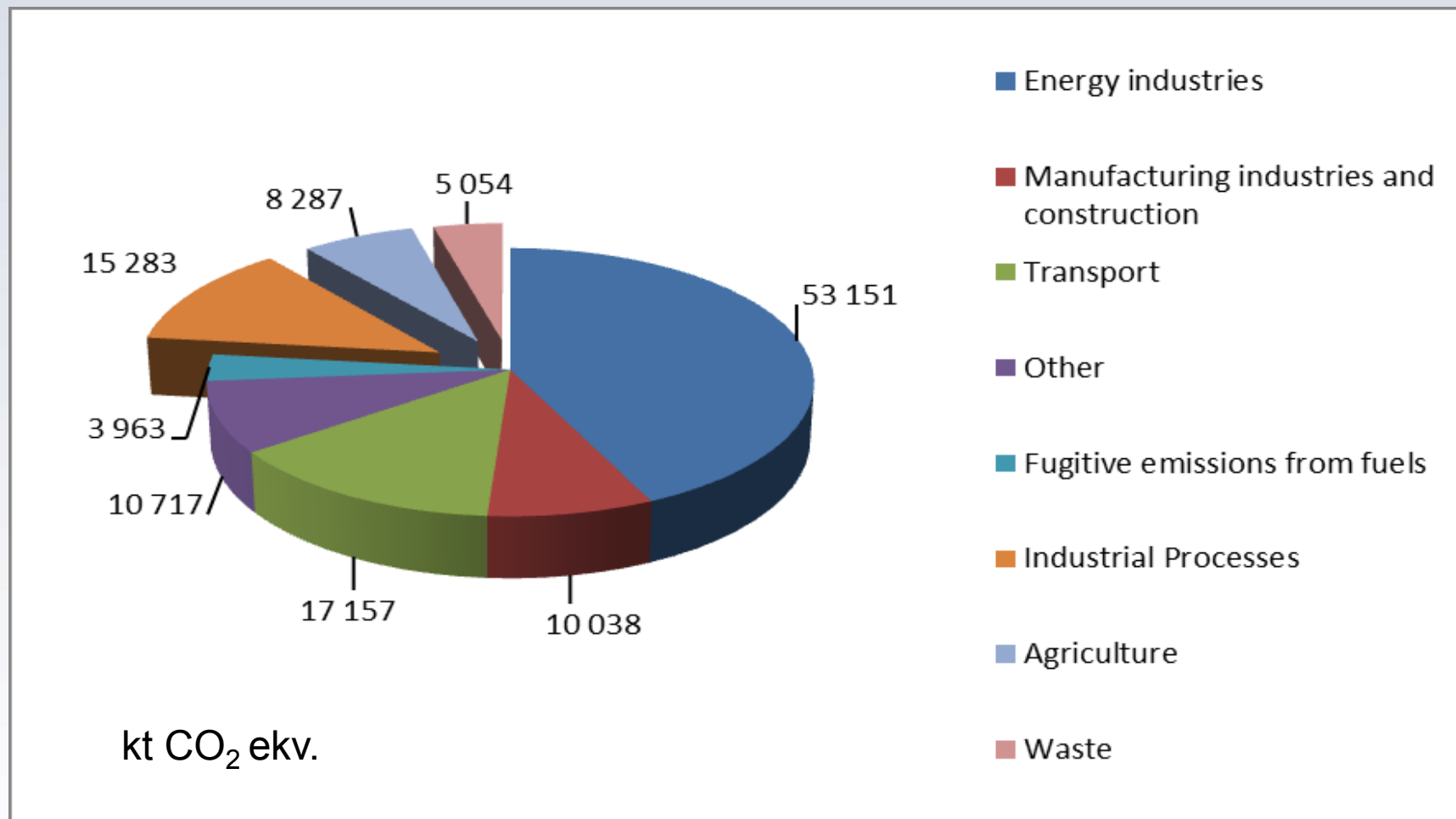


# Emise skleníkových plynů v ČR 2014 podle sektorů

**123,65 mil. t**

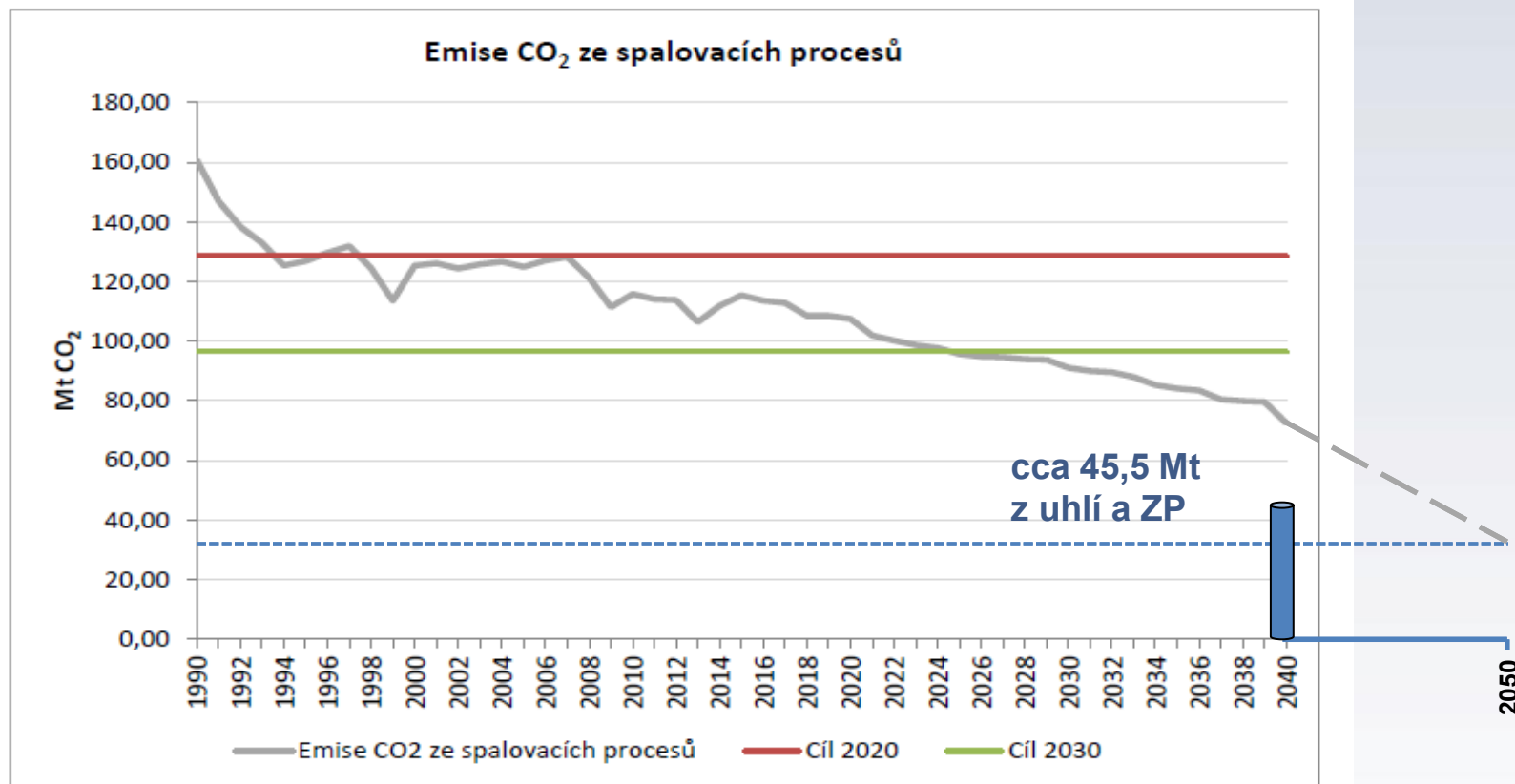


# Emise skleníkových plynů v ČR 2014 podle sektorů



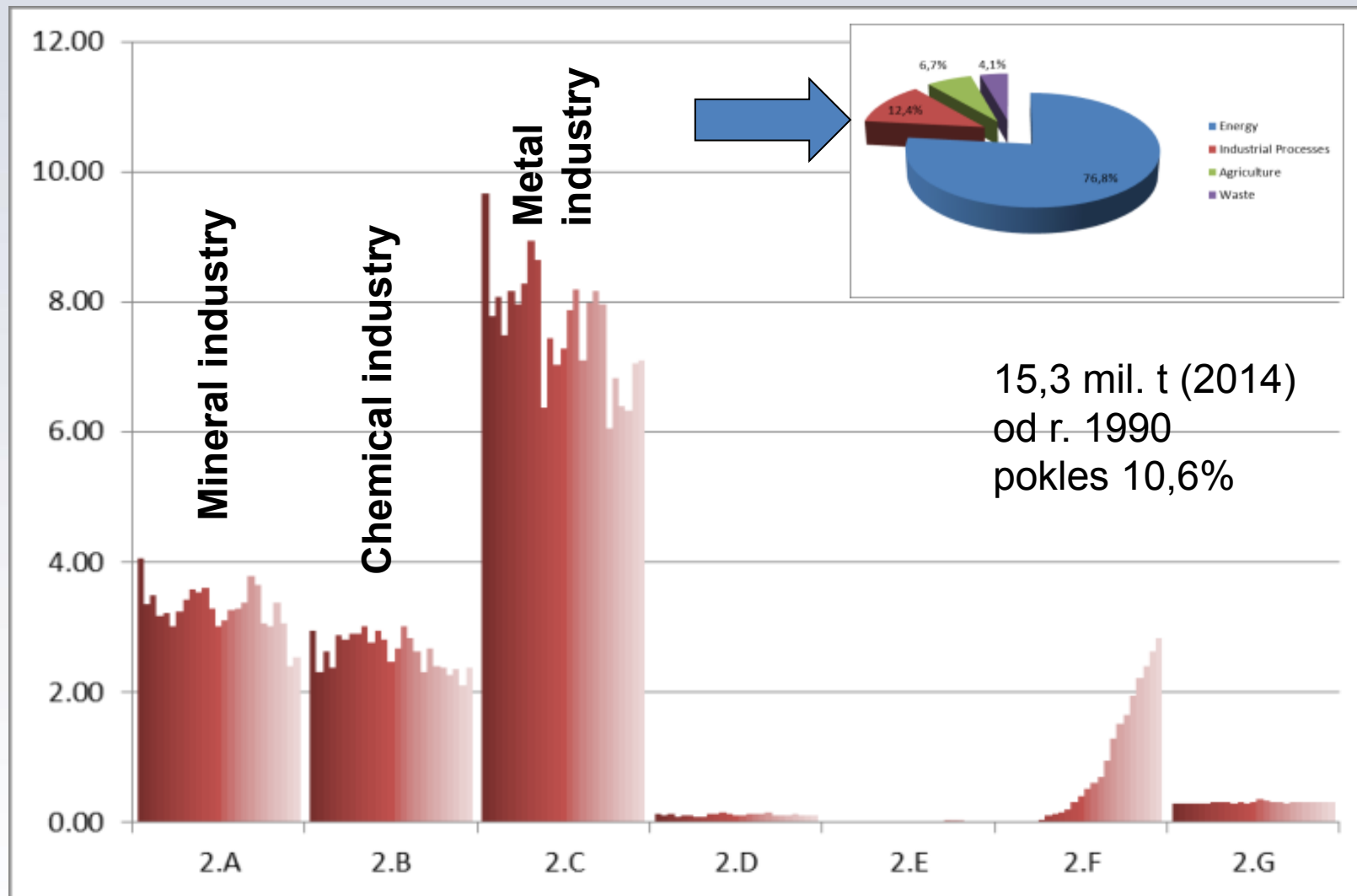


**Graf č. 33:** Emise CO<sub>2</sub> ze spalovacích procesů

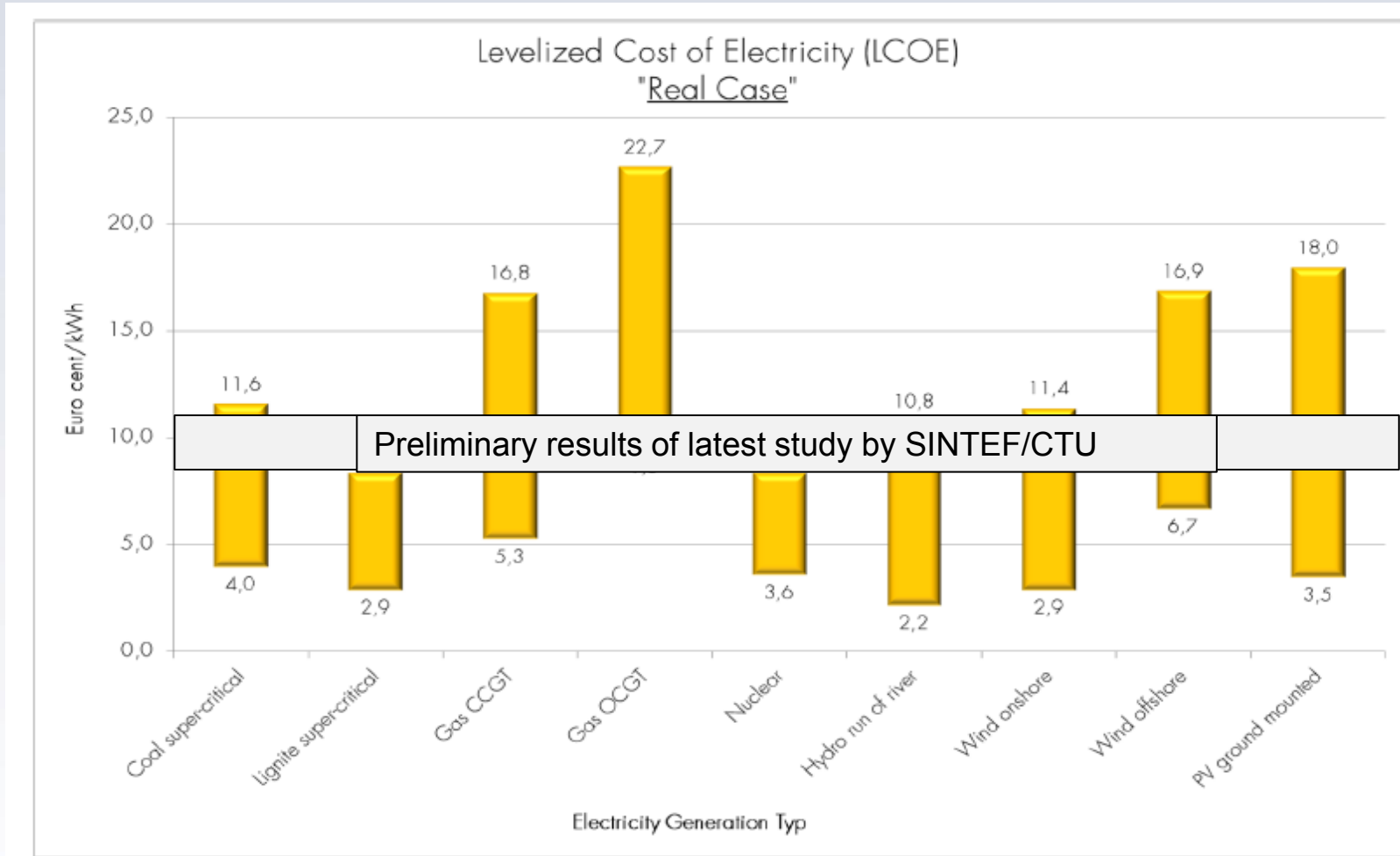


Pozn.: Pro emise CO<sub>2</sub> nejsou stanoveny cíle pro jednotlivé země EU, ale pouze cíl pro EU jako celek. Uvedené linie jsou tedy vypočteny z hodnot cíle EU snížení emisí do roku 2020 o 20 % oproti roku 1990 a cíle EU snížení emisí do roku 2030 o 40 % oproti roku 1990 vztahených k hodnotě emisí ze spalovacích procesů na území České republiky.

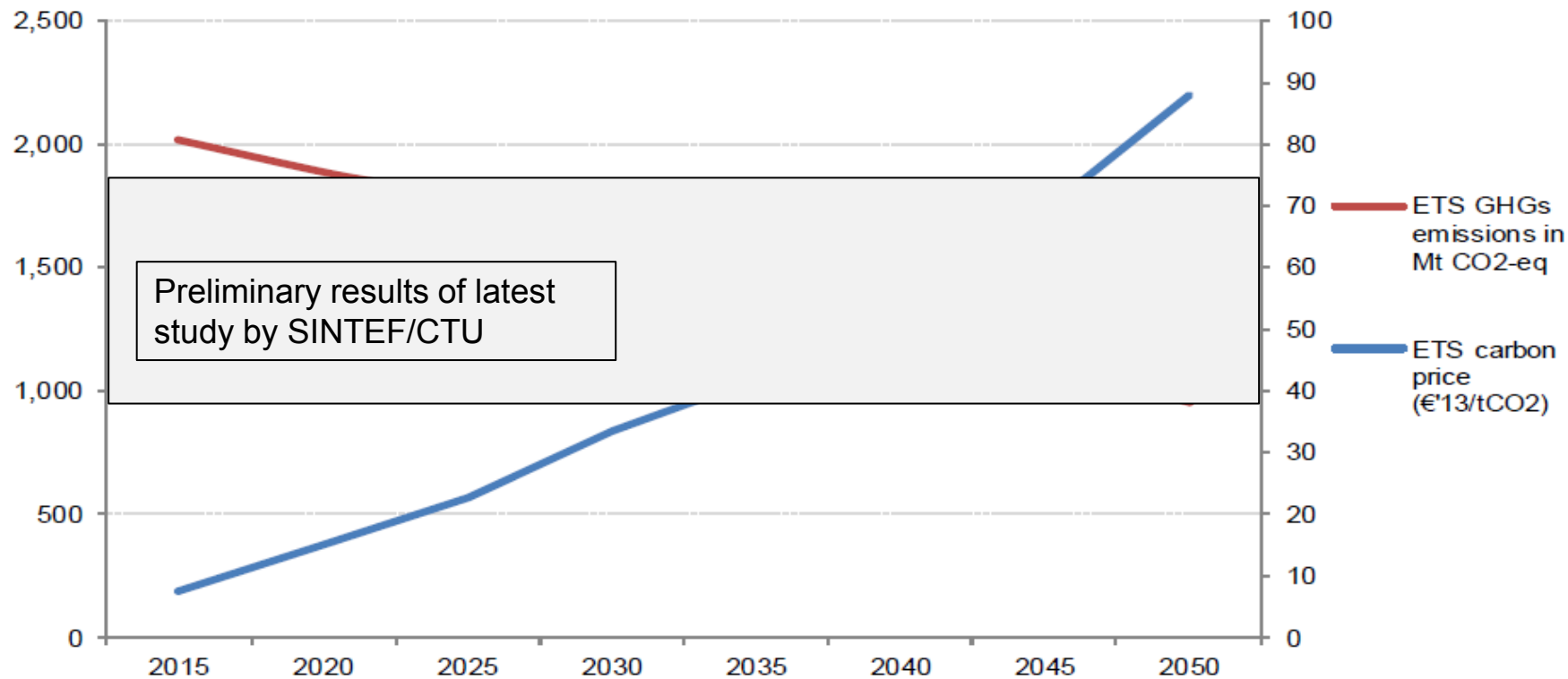
# Emise skleníkových plynů v ČR 1990-2014 – průmyslové procesy



# Je CCS drahé?



## ETS emissions and carbon prices over time

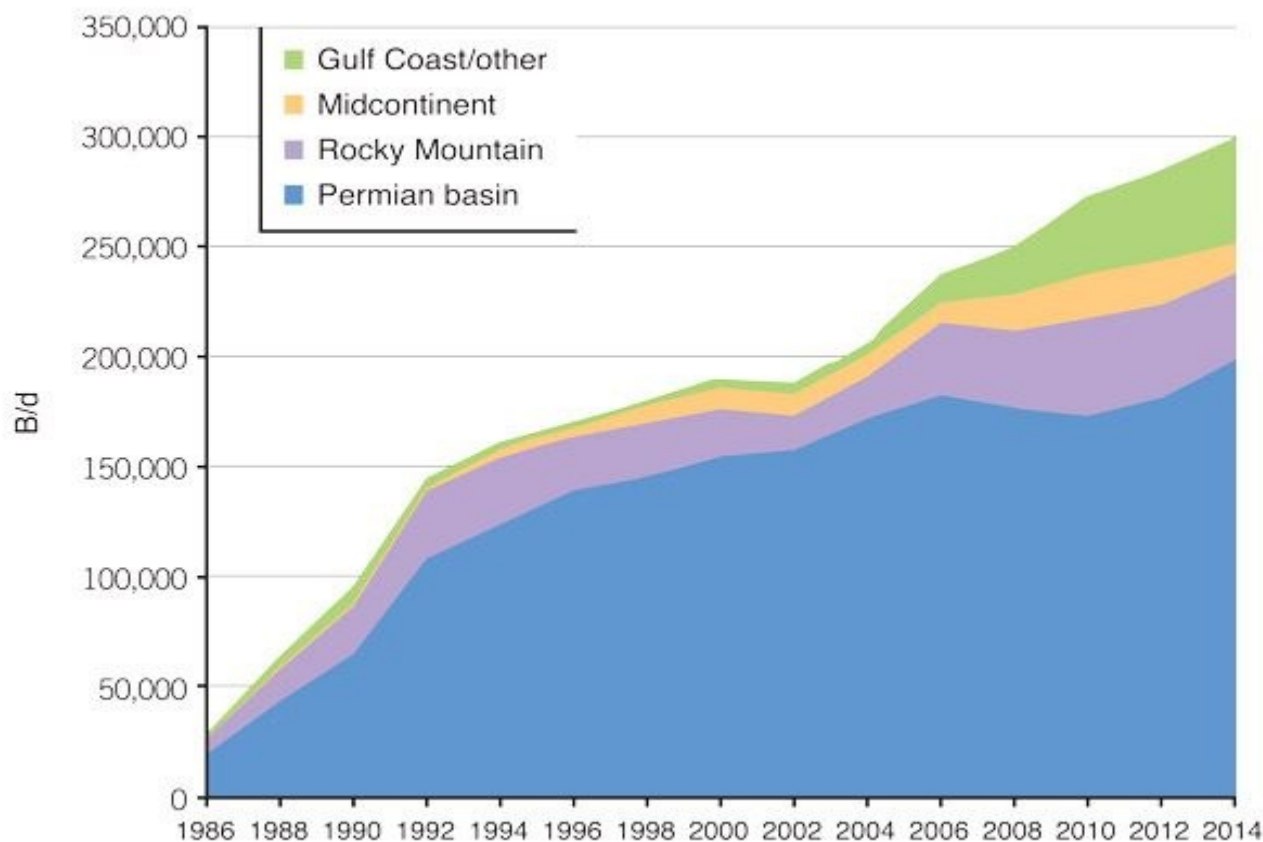


Source: PRIMES, GAINS

- Od 20. let 20. století se separuje CO<sub>2</sub> ze zemního plynu
- 1972 první projekt využití separovaného CO<sub>2</sub> k intenzifikaci těžby ropy v Texasu (spolu s využitím přírodního CO<sub>2</sub> v USA a v Maďarsku)
- 1977 poprvé navrženo využití CCS jako metody pro snížení emisí skleníkových plynů

# Těžba ropy pomocí CO<sub>2</sub>-EOR v USA

**HISTORICAL CO<sub>2</sub>-EOR PRODUCTION** FIG. 2



Source: Advanced Resources International Inc. adjustment to OGI EOR/Heavy Oil Survey 2014