Recombination ...

FP I ZS 2024 XE 19. 12.

electron -- ion recombination

atomic ions

molecular ions

ion -ion recombination

Recombination



Radiative recombination

 $Ar^+ + e^- \rightarrow Ar + hv$

Dissociative recombination

 $AB^+ + e \rightarrow A + B$



<1 eV





Capture

AB* resonant state(s)





AB* resonant state(s)

predissociation

To get high recombination rate, we need (a) efficient capture (b) predissociation faster than auto-ionization **Electron**—ion recombination

 $H^+ + e^- \rightarrow products$

- $H_2^+ + e^- \rightarrow products$
- $H_3^+ + e^- \rightarrow products$
- $H_5^+ + e^- \rightarrow products$



$H_2^+ + e^- \rightarrow H + H,$ <u>Dissociative Recombination</u> - DR



Electron collisions with H_2^+ - how to describe ????









THEORETICAL FRAMEWORK

The states involved: exemple for He₂+/He₂ system



2004 DR6 Mosbach

I. Schneider, et al., DR2004 Mosbach

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internuclear separation, R

internuclear separation, R

Direct mechanism

Indirect mechanism

Electron-cold molecular ion reaction: Dissociative Recombination



Recombination of H_{3^+} : No ion-neutral crossing





Concept of recombination rate coefficient (plasma binary reactions) at T



e⁻ + A⁺ \rightarrow products dN_A/dt= - α n_eN_A **RECOMBINATION RATE COEFFICIENT** $\alpha = \alpha(T)$ $\sigma = \sigma(v) = \sigma(z)$ Collision rate coefficient, Recombination rate coefficient $\alpha = < \sigma u_r >$

Direct and indirect process

Recombination processes in plasma

Binary Recombination

$$H^{+} + e \rightarrow H + hv$$

$$RR$$

$$O_{2}^{+} + e \rightarrow O + O$$

$$DR$$

$$\frac{dn_{e}}{dt} = \frac{d[O_{2}^{+}]}{dt} = -\alpha[O_{2}^{+}]n_{e} = -\alpha n_{e}^{2}$$

$$Fe^{7+} + e \rightarrow Fe^{6+}$$

$$DiR$$

Recombination processes in plasma

Binary Recombination

$$H^{+} + e \rightarrow H + h\nu$$

$$PR$$

$$O_{2}^{+} + e \rightarrow O + O$$

$$DR$$

$$\frac{dn_{e}}{dt} = \frac{d[O_{2}^{+}]}{dt} = -\alpha[O_{2}^{+}]n_{e} = -\alpha n_{e}^{2}$$

$$Fe^{7+} + e \rightarrow Fe^{6+}$$

$$DiR$$

$$Ternary electron assisted recombination$$

$$Ar^{+} + e + e \rightarrow Ar + e$$

$$\frac{dn_{e}}{dt} = \frac{d[Ar^{+}]}{dt} = -K_{e}[Ar^{+}]n_{e}^{2} = -\alpha_{eff}[Ar^{+}]n_{e}$$

$$Collisional Radiative Recombination CRR$$

$$\alpha_{eff} = K_{e}n_{e}$$

Ternary neutral assisted recombination

 $Ar^+ + e + He \rightarrow Ar + He$

$$\frac{dn_{e}}{dt} = \frac{d[Ar^{+}]}{dt} = -K_{M}[Ar^{+}]n_{e}[He] = -\alpha_{eff}[Ar^{+}]n_{e}[He]$$

$$\alpha_{eff} = K_M[He]$$

 $f + e \rightarrow O + O$

234311-2 Petrignani et al.





Five exothermic channels are available for vibrational ground state O_2^+ ions in zero relative energy collisions with electrons. They are summarized as follows with the associated kinetic energy releases:

$$O_2^+(X^2\Pi_g, v=0) + e^- \rightarrow O(^3P) + O(^3P) + 6.65 \text{ eV}$$
 (1a)

- $\rightarrow O(^{3}P) + O(^{1}D) + 4.99 \text{ eV}$ (1b)
- $\rightarrow O(^{1}D) + O(^{1}D) + 3.02 \text{ eV}$ (1c)
- $\rightarrow O(^{3}P) + O(^{1}S) + 2.77 \text{ eV}$ (1d)
- $\rightarrow O(^{1}D) + O(^{1}S) + 0.80 \text{ eV.}$ (1e)



FIG. 2. DR rate coefficient k as a function of electron collision energy from 1 meV to 5 eV. Statistical errors are shown at the 1σ level. The dotted line shows the threshold $E^{-1/2}$ behavior. Both the rate coefficient and the energy are shown on a logarithmic scale.

Electron - Ion Recombination



Resonances

Resonances

Autoionizing and pre-dissociating Rydberg states



Dissociative recombination

Resonances



Dissociative recombination



Details of DR of HD+

Dissociative recombination HD+







Scan of electron ion relative energy EElectron temperature kT_{Per} =4meV (30meV for E>0.3eV) kT_{Par} =0.1meVEnergy resolution ~4....8 meV (E< 0.08eV)</td>Absolute accuracy of cross section ca. +- 30%

Recombination H₂⁺



Recombination only one rotational quanta change the whole spectra



Recombination calculation and theory H_2^+vibrational excitation



COMPUTATIONS vs MEASUREMENTS: H₂⁺/H₂

Ngassam et al 2003

Fifirig et al 2003



Different energy region

Recombination NO⁺

Tennyson et al 1996-2000





Comparison with ASTRID

Schneider et al 2000



Theoretical background **Dissociative Recombination without a Curve Crossing** Theory predicted: DR rate coefficient is vary small ~ 10⁻¹¹ cm³s⁻¹



HeH⁺ and HCO⁺ ionsexamples of a non-crossing case. <u>However, experiments gave</u> $\alpha \approx 2 \times 10^{-8}$ and $\alpha \approx 2 \times 10^{-7}$ cm³s⁻¹

A new mechanism has been proposed!

Multi-step indirect dissociative recombination ("tunneling mode" recombination)

Science

REPORTS

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Quantum-state-selective electron recombination studies suggest enhanced abundance of primordial HeH⁺

2019

Oldřich Novotný^{1*}, Patrick Wilhelm¹, Daniel Paul¹, <u>Abel Kálosi^{1,2}</u>, Sunny Saurabh¹, Arno Becker¹, Klaus Blaum¹, Sebastian George^{1,3}, Jürgen Göck¹, Manfred Grieser¹, Florian Grussie¹, Robert von Hahn¹, Claude Krantz¹, Holger Kreckel¹, Christian Meyer¹, Preeti M. Mishra¹, Damian Muell¹, Felix Nuesslein¹, Dmitry A. Orlov¹, Marius Rimmler¹, Viviane C. Schmidt¹, Andrey Shornikov¹, Aleksandr S. Terekhov⁴, Stephen Vogel¹, Daniel Zajfman⁵, Andreas Wolf⁴

¹Max-Planck-Institut für Kernphysik, Saupfercheckweg 1, 69117 Heidelberg, Germany. ²Charles University, 18000 Praha, Czech Republic. ³Universität Greifswald, Institut für Physik, 17487 Greifswald, Germany. ⁴Rzhanov Institute of Semiconductor Physics, Novosibirsk 630090, Russia. ⁵Weizmann Institute of Science, Rehovot 76100, Israel. *Corresponding author. Email: oldrich.novotny@mpi-hd.mpg.de

The epoch of first star formation in the early universe was dominated by simple atomic and molecular species consisting mainly of two elements: hydrogen and helium. Gaining insight into this constitutive era requires thorough understanding of molecular reactivity under primordial conditions. We used a cryogenic ion storage ring combined with a merged electron beam to measure state-specific rate coefficients of dissociative recombination, a process by which electrons destroy molecular ions. We found a dramatic decrease of the electron recombination rates for the lowest rotational states of HeH⁺, compared to previous measurements at room temperature. The reduced destruction of cold HeH⁺ translates into an enhanced abundance of this primordial molecule at redshifts of first star and galaxy formation.





Fig. 1. Dissociative recombination in the cryogenic storage ring CSR. (A) Scheme of the CSR ring structure with injected and stored HeH⁺ ion beam (red), merged electron beam (blue), reaction products (green) and particle detector. (B) Reaction scheme and position-sensitive detection of coincident fragments. (C) Equilibrium rotational state populations of HeH+ for previous studies (300 K) and the estimated radiation field in the CSR.



Fig. 2. DR for rotationally cold HeH⁺. (A) Blue dots: merged-beams rate coefficient α_{DR} as a function of detuning energy E_d after the relaxation to >50% J = 0 (this experiment, 10 s < t < 50 s, mean ± SD); absolute scaling uncertainty ±20% (SEM). Red symbols: roomtemperature data from Ref. (11) (squares, absolute scaling uncertainty ±10% SEM) and from Ref. (12) (triangles, scaled to Ref. (11) at 0.03 eV). (B) Fragment distance distribution projected into the detector plane for $E_d = 0$ (blue) with fit (19) for isotropic angular distribution (red). (C) Projected fragment distance distribution for Ed = 0.044 eV (blue) with fit (19) for a $|Y_{10}|^2$ angular distribution of the fragments (red). The angular dependences in (B) and (C) are indicated schematically.

Fig. 4. Rotational-state selective DR rates for HeH⁺. (A) Merged-beams rate coefficients $\alpha_{DR}^J(E_d)$ for $J \leq 2$ and average for $J \geq 3$ (mainly 3 and 4; mean \pm SD). The dashed lines mark the shift of the maximum as J increases. (B) Full lines: single-J plasma rate coefficients $\alpha_{DR,pl}^J(T_{pl})$ for $J \leq 2$ and average for $J \geq 3$ (mainly 3 and 4; mean with shaded areas as \pm SD). Dotted: fully thermal rate coefficient $\alpha_{DR,therm}(T_{rot} = T_{pl})$. Dashed-dotted: values applied in early-universe models (21, 22) and astrochemistry databases (23–25). See (19) for further discussion, numerical fitting functions and parameters.



Dissociative recombination of N₂H⁺ ions with electrons in the temperature range of 80–350 K

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Dmytro Shapko, Petr Dohnal ⁽¹⁾, Miroslava Kassayová, Ábel Kálosi ⁽¹⁾, Serhiy Rednyk ⁽¹⁾, _{Štěpán Roučka}, Radek _{Plašil} ⁽¹⁾, Lucie D. _{Augustovičová}, Rainer Johnsen ⁽¹⁾, Vladimír _{Špirko} ⁽¹⁾, and Juraj Glosík

Recombination of N_2H^+ ions with electrons was studied using a stationary afterglow with cavity ring-down spectrometer. We probed in situ the time evolutions of number densities of different rotational and vibrational states of recombining N_2H^+ ions and determined the thermal recombination rate coefficients for N_2H^+ in the temperature range of 80 - 350 K. The newly calculated vibrational transition moments of N_2H^+ are used to explain the different values of recombination rate coefficients obtained in some of the previous studies. No statistically significant dependence of the measured recombination rate coefficient on the buffer gas number density was observed.

$N_2H^+ + e^- \xrightarrow{\alpha_{bin}}$ neutral products

I. INTRODUCTION

 N_2H^+ , an important interstellar ion, has been observed in different interstellar environments such as dark and translucent clouds^{1,2}, protostellar cores³, protoplanetary disks⁴ and is considered to play a role in the atmospheric chemistry of Titan⁵. N_2H^+ serves as an important tracer for N_2 in dark clouds, therefore detailed information on production and destruction processes of N_2H^+ could help with the prediction of N_2 abundance in this environment. N_2H^+ in the interstellar medium is mainly produced in proton transfer from H_3^+ to N_2 and its main destruction mechanisms are proton transfer to CO and dissociative recombination with electrons⁶.

$N_2H^+ + e^- \xrightarrow{\alpha_{bin}} neutral products$

II. EXPERIMENT

The recombination rate coefficients^{7,32} are measured in a stationary afterglow (SA) in conjunction with cavity ring-down spectroscopy (CRDS) to monitor the decay of the densities of different rotational and vibrational states of N₂H⁺ ions. The plasma is generated in a pulsed microwave discharge in a fused silica tube (inner diameter ≈ 1.3 cm). The microwave generator is equipped with an external fast high-voltage switch to cut off the power to the magnetron within a fall time of less than 30 µs. A low microwave power in the range of 10 – 25 W, with $\approx 40\%$ duty cycle, is used to avoid excessive heating of the gas during the discharge. The discharge tube temperature (*T*_{tec}) is measured by a thermocouple outside of the discharge and can be varied between 80 and 350 K.






FIG. 2. The dependence of the rotational temperature measured during the discharge on kinetic temperature. All the displayed data were obtained in helium buffer gas. Insert: An example of the Boltzmann plots used for determination of the rotational temperature obtained at T = 200 K and T = 78 K.



FIG. 3. An example of measured time evolutions of number densities of N_2H^+ ions in the ground and the first excited vibrational state. The lower panel shows the relative fractions of the vibrational states and the dashed lines denote the corresponding fraction in thermal equilibrium at temperature of 321 K. The particular vibrational states number densities were calculated from the measured number densities of the *J* = 6 rotational state of the ground vibrational state and of *J* = 9 state of the (01¹0) vibrational state under the assumption of the same rotational temperature in both vibrational states.

FIG. 4. Dependence of the measured vibrational temperature $T_{\rm vib}$ on kinetic temperature $T_{\rm kin}$ of the N₂H⁺ ions measured in the discharge. The kinetic temperature was obtained from the Doppler broadening of the P(6) line of the (200) \leftarrow (000) vibrational band of N₂H⁺. It was assumed that the rotational temperature of (000) and (010) states is equal to $T_{\rm kin}$. The vibrational temperature was then evaluated from the P(6) line of the (200) \leftarrow (000) vibrational band and from the R(9)^f line of the (210) \leftarrow (010) vibrational band of the N₂H⁺ ion. The full line denotes equality of $T_{\rm vib} = T_{\rm kin}$ and the dashed line indicates $T_{\rm vib} = T_{\rm kin} + 30$ K. The displayed errors are statistical errors of the fits.





FIG. 8. The temperature dependence of the measured recombination rate coefficients of N₂H⁺ (full circles, the value of α_{bin} at 350 K was obtained in H₂ buffer gas, for the rest of the data points, helium buffer gas was used) compared to values obtained in previous experiments. Rhomboids: FALP¹¹, squares: FALP¹³, stars: FALP¹⁴, triangles: stationary afterglow with absorption spectroscopy¹⁷, full line: ion storage ring⁶, double-dot-dashed line: merged beams⁸ and to recent theoretical calculations by Fonseca dos Santos²¹ (dot-dashed line, the rate coefficient was calculated from the cross sections for the direct and indirect recombination process in ref.²¹). The dashed line denotes fit to the data: $\alpha_{N_2H^+} = (2.81 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.06\pm0.02)} \text{ cm}^3 \text{s}^{-1}$ otherwise. The dotted lines show 15% deviation from the fitted value (estimated systematic error of the measurement reflects mainly the uncertainty in the effective discharge column length and in the calculated vibrational transition moments).



FIG. 9. Time evolution of the measured overall number density of N₂H⁺ ions (full line) compared to the results obtained from the model of chemical kinetics. The data were obtained at T = 140 K, [He] = 1.5×10^{17} cm⁻³, [Ar] = 2.5×10^{14} cm⁻³, [H₂] = 5×10^{14} cm⁻³ and [N₂] = 4×10^{13} cm⁻³ (same as in Figure 1) and [NH₃] = 5×10^{11} cm⁻³. [He^m](t = 0) = 1/3 $n_e(t = 0)$.

Ternary recombination

Ternary electron assisted recombination

$$Ar^+ + e + e \rightarrow Ar + e$$

$$\frac{dn_{e}}{dt} = \frac{d[Ar^{+}]}{dt} = -K_{e}[Ar^{+}]n_{e}^{2} = -\alpha_{eff}[Ar^{+}]n_{e}$$

Collisional Radiative Recombination CRR

$$\alpha_{eff} = K_e n_e$$

Ternary neutral assisted recombination

$$\frac{dn_{e}}{dt} = \frac{d[Ar^{+}]}{dt} = -K_{M}[Ar^{+}]n_{e}[He] = -\alpha_{eff}[Ar^{+}]n_{e}$$

$$Ar^+ + e + He \rightarrow Ar + He$$

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 $\alpha_{eff} = K_M[He]$

$Ar^{+} + e^{-} + e^{-}$

Colisional Radiative Recombination -CRR

$$\frac{dn_{e}}{dt} = -K_{CRR} \ [Ar^{+}]n_{e}^{2} - \frac{n_{e}}{\tau_{D}} = -K_{CRR} \ n_{e}^{3} - \frac{n_{e}}{\tau_{D}}$$

$$\alpha_{CRR} = K_{CRR} n_e$$

$$H^+ + e^- + e^-$$

Anti hydrogen formation

Kvantovka na každý deň

 \mathbf{r}_1

PRL 98, 133201 (2007)

$\frac{H^+ + e^- + e^-}{H^+ + e^-}$





 \mathbf{r}_{2}

We consider the simplest TBR in the case of hydrogen formation, in which two free electrons interact with a proton. To investigate the three-body interaction dynamics, we numerically solve the six-dimensional (6D) timedependent Schrödinger equation, which has the following form (atomic units are used throughout):

$$i\frac{\partial}{\partial t}\Phi(\mathbf{r}_{1},\mathbf{r}_{2},t) = \left[-\frac{1}{2}(\Delta_{\mathbf{r}_{1}}+\Delta_{\mathbf{r}_{2}})-\frac{1}{r_{1}}-\frac{1}{r_{2}} + \frac{1}{|\mathbf{r}_{1}-\mathbf{r}_{2}|}\right]\Phi(\mathbf{r}_{1},\mathbf{r}_{2},t), \quad (1)$$

where \mathbf{r}_1 and \mathbf{r}_2 are the position vectors of each electron, with respect to the proton. We obtain a more tractable with respect to the proton. We obtain a more tractable solution by using the close-coupling recipe [12]: expanding the 6D wave function $\Phi(\mathbf{r}_1, \mathbf{r}_2|t)$ in terms of bipolar spherical harmonics $Y_{l_1 l_2}^{LS}(\Omega_1, \Omega_2)$, $\Phi(\mathbf{r}_1, \mathbf{r}_2|t) =$ $\sum_{LS} \sum_{l_1 l_2} [\Psi_{l_1 l_2}^{(LS)}(r_1, r_2|t)/r_1 r_2] Y_{l_1 l_2}^{LS}(\Omega_1, \Omega_2)$, for a specific symmetry (*LS*). We can also expand the Coulomb repulsion term $1/|\mathbf{r}_1 - \mathbf{r}_2|$ in terms of spherical harmonics. Substituting these expansions into the above Schrödinger Eq. (1) and integrating over the angles Ω_1 and Ω_2 yields a set of coupled partial differential equations with only two radial variables r_1 and r_2 left:

$$\begin{split} i\frac{\partial}{\partial t}\Psi_{j}(r_{1},r_{2}|t) &= [\hat{T}_{1}+\hat{T}_{2}+\hat{V}_{c}]\Psi_{j}(r_{1},r_{2}|t) \\ &+ \sum_{k}\hat{V}_{j,k}^{I}(r_{1},r_{2}|t)\Psi_{k}(r_{1},r_{2}|t), \quad (2) \end{split}$$

where the partial-wave index j runs from 1 to the total number N of partial waves used for expansion. In Eq. (2),

Kvantovka na každý deň

 $i\frac{\partial}{\partial t}\Psi_{j}(r_{1}, r_{2}|t) = [\hat{T}_{1} + \hat{T}_{2} + \hat{V}_{c}]\Psi_{j}(r_{1}, r_{2}|t) + \sum_{k}\hat{V}_{j,k}^{I}(r_{1}, r_{2}|t)\Psi_{k}(r_{1}, r_{2}|t), \quad (2)$

$H^+ + e^- + e^- \rightarrow H + e^-$

$$P_{nl}(E_2) = 2\sum_{LS} \sum_{l_2} \left| \int dr_1 \int dr_2 \phi_{nl}^*(r_1) \phi_{k_2 l_2}^*(r_2) \Psi_{ll_2}^{(LS)}(r_1, r_2, t = t_f) \right|^2,$$

$K_{E} = 0.1 \text{ eV}$



FIG. 1 (color online). Snapshots of electron probability distribution on the plane spanned by the radial coordinates r_1 and r_2 for different times: (a) t = 0.0 fs, (b) t = 60 fs, (c) t = 100 fs, (d) t = 150 fs, (e) t = 194 fs, and (f) (in log scale) t = 260 fs.





Thus, for the case of $K_E = 0.1$ eV considered in Figs. 1 and 2, the total system energy is about $E_{tot} \sim 0.12$ eV instead of $2K_E$. Hence, when one electron recombines to the 10*d* state ($|E_{10d}| \approx 0.136$ eV) of the H atom, the outgoing electron takes an initial total energy of 0.12 eV plus $|E_{10d}|$, thereby $P_{10d}(E_2)$ peaks at $E_2 \sim 0.256$ eV, as shown by the (red) solid line of Fig. 2. Similar energy conservation is also well satisfied for the recombination to the 6*p* state, as is illustrated by the (blue) dash-dotted line in Fig. 2. Our quantum calculations unambiguously reveal the essential feature of a TBR process.

Kvantovka na každý deň

 $\mathbf{H}^{+} + \mathbf{e}^{-} + \mathbf{e}^{-} \rightarrow \mathbf{H} + \mathbf{e}^{-}$

 $K_E = 0.1 \text{ eV}$







FIG. 3 (color online). The recombination probability P_n as a function of the energy level *n*, for different electron kinetic energies K_E marked in each panel.



FIG. 4 (color online). The recombination probability $P_{n=25,l}$ as a function of the angular-momentum quantum number l, for different electron kinetic energies K_E marked in each panel.



$Ar^{+} + e^{-} + e^{-}$

 $\frac{dn_{e}}{dt} = -K_{CRR} \ [Ar^{+}]n_{e}^{2} - \frac{n_{e}}{\tau_{D}} = -K_{CRR} \ n_{e}^{3} - \frac{n_{e}}{\tau_{D}}$

$$\alpha_{\rm CRR} = 3.8 \times 10^{-9} T_{\rm e}^{-4.5} n_{\rm e} + 1.55 \times 10^{-10} T_{\rm e}^{-0.63} + 6 \times 10^{-9} T_{\rm e}^{-2.18} n_{\rm e}^{0.37} {\rm cm}^3 {\rm s}^{-1}$$





Recombination of H₃⁺



$\begin{array}{rcl} \mathrm{H_3^{+\!+\,e^{-}} & \rightarrow \mathrm{H+H+H}} \\ & \rightarrow \mathrm{H_2+H}} \\ & \rightarrow \mathrm{H_3^{*}} & (?) \end{array}$

Tunneling dissociative recombination





Tunneling dissociative recombination







Dissociative recombination of H_3^+ . Relevant potential curves



Dissociative recombination of H₃+



Remote curve crossing

Electron capture via Jahn-Teller coupling of electronic and ro-vibrational motion

Prototype system for electron capture and dissociation mechanisms in polyatomic species

Symmetric deformation

Three atomic ions

Dissociative recombination of H₃+

Prototype system for electron capture and dissociation mechanisms in polyatomic species





FIG. 4. The figure demonstrates how the Jahn-Teller effect produces a high rate of dissociative recombination. One $2p\sigma$ potential surface and two $2p\pi$ potential surfaces [47] of the neutral molecule are shown. The conical intersection is produced by Jahn-Teller coupling. When an electron arrives, it scatters first into a low-lying vibrationally excited Rydberg state {01¹}. Then, after the nuclei vibrate, the system finds its way with high probability into a $2p\pi$ state having high vibrational excitation, near the point of conical intersection. The contour plot at the bottom of the figure represents the lowest $2p\pi_1$ surface. All three potential surfaces are shown in the reduced 2D space of dimensionless normal coordinates. The coordinates used here are the normal asymmetric Q_x , Q_y coordinate, with ρ and ϕ their polar components [17,18]. The third vibrational coordinate—the symmetric stretch coordinate Q_1 —is kept constant for this graph.

2004-7

PHYSICAL REVIEW A 68, 012703 (2003) 0.2Energy (a.u.) -0.2 1.5 2 2.5 3 Hyperradius (a.u.)

FIG. 1. The problem of DR of H_3^+ in the hyperspherical adiabatic approximation. The lowest hyperspherical adiabatic potential (thick full line) of the H_3^+ and number of hyperspherical adiabatic potentials of the neutral molecule (thin lines). Lower family of lines (darker lines) dissociate to the H_2 +H channel; the upper family (lighter lines) dissociate to the H+H+H channel. To calculate hyperspherical adiabatic curves we used the three-dimensional H_3^+ potential from Ref. [48] and the H_3 potential from Refs. [35–37]. Since the density of hyperspherical states is high, only every tenth H_3 potential curve is shown in the figure. The dashed line shows the position of the ground vibrational level of the ion, which is the only one populated in the relevant experiments.



FIG. 4. The figure demonstrates how the Jahn-Teller effect produces a high rate of dissociative recombination. One $2p\sigma$ potential surface and two $2p\pi$ potential surfaces [47] of the neutral molecule are shown. The conical intersection is produced by Jahn-Teller coupling. When an electron arrives, it scatters first into a low-lying vibrationally excited Rydberg state {01¹}. Then, after the nuclei vibrate, the system finds its way with high probability into a $2p\pi$ state having high vibrational excitation, near the point of conical intersection. The contour plot at the bottom of the figure represents the lowest $2p\pi_1$ surface. All three potential surfaces are shown in the reduced 2D space of dimensionless normal coordinates. The coordinates used here are the normal asymmetric Q_x , Q_y coordinate, with ρ and ϕ their polar components [17,18]. The third vibrational coordinate—the symmetric stretch coordinate Q_1 —is kept constant for this graph.



(b) predissociation faster than auto-ionization





... history is repeating itself

... One remaining problem is to understand the plasma afterglow experiments.



Plasma in TDE P(A) $(U_{j,2}E_{j}) \leq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{j}) \leq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{j}) \leq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{i}) \geq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{i}) \geq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{i}) \geq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i$

.... many times it was concluded, that the task was finished....

... and the caravan is on its way

The battle ship enters the stage

FAL



Πλασμα





10-8





Pressure dependence

Pulsed (stationary) afterglow



+ molecules

We measure effective – apparent binary recombination rate coefficient



Quasineutral H₃⁺ dominated plasma

 $\frac{\mathrm{d}n_{\mathrm{e}}}{\mathrm{d}t}$ $n_{\rm e}$ $-\alpha_{\rm eff}n_{\rm e}$ $au_{
m L}$

$$\frac{1}{[H_3^+]} = \frac{1}{[H_3^+]_0} + \alpha t$$

We measure effective – apparent binary recombination rate coefficient



Quasineutral H₃⁺ dominated plasma

$$\frac{\mathrm{d}n_{\mathrm{e}}}{\mathrm{d}t} = -\alpha_{\mathrm{eff}}n_{\mathrm{e}}^2 - \frac{n_{\mathrm{e}}}{\tau_{\mathrm{L}}}$$

$$\frac{1}{[H_3^+]} = \frac{1}{[H_3^+]_0} + \alpha t$$

processes at high densities at low T

Decay in <u>diffusion</u> and <u>recombination</u> governed plasma

$$\frac{dn_e}{dt} = -\alpha n_e^2 - \frac{D_a}{\Lambda^2} n_e$$

$$\frac{1}{n_e} = \alpha \frac{\exp(\nu t) - 1}{\nu_D} + \frac{1}{n_0} \exp(\nu_D t)$$



Decay in <u>diffusion</u> and <u>recombination</u> governed plasma

$$\frac{dn_e}{dt} = -\alpha \ n_e^2 - \frac{D_a}{\Lambda^2} n_e$$

$$\frac{1}{n_e} = \alpha \ \frac{\exp(\nu t) - 1}{\nu_b} + \frac{1}{n_0} \exp(\nu_b t)$$
Limit for t -->0
$$\frac{1}{n_e} = \alpha \ \frac{(1 + \nu_b t) - 1}{\nu_b} + \frac{1}{n_0} (1 + \nu_b t)$$

$$\frac{1}{n_e} = \alpha t + \frac{1}{n_0} (1)$$

$$\frac{1}{n_e} = \alpha t + \frac{1}{n_0} (1)$$



Srovnani modelu a experimentu. Pocatecni podminka: $[H_3^+] = n_e = 2 \times 10^{11} \text{ cm}^{-3}$.



$$\frac{dn_e}{dt} = -[\alpha_1 n_1(t) + \alpha_2 n_2(t)]n_e$$
$$\Rightarrow \alpha_{eff}(t) = [\alpha_1 f_1(t) + \alpha_2 f_2(t)]$$
$$f_1 + f_2 = 1$$

Model + data. Pocatecni podminka: $n_e = He^m = [H_3^+]$.

Poznamka. Namerene τ difuznich ztrat 1.6 ms. Teoreticke τ pri danem tlaku je 1.8 ms (odpovida cca 4×10¹⁰ cm⁻³ koncentraci necistot (pri 2×10⁻⁹ cm³s⁻¹ rychlosti reakce H₃⁺ s necistotami). Namerena koncentrace vody [H₂O] = 5×10¹⁰ cm⁻³ ([He] = 8×10¹⁷ cm⁻³).

If there are 2 or more ion species, the fast recombining species disappears first

$$\frac{dn_e}{dt} = -[\alpha_1 n_1(t) + \alpha_2 n_2(t)]n_e$$
$$\Rightarrow \alpha_{eff}(t) = [\alpha_1 f_1(t) + \alpha_2 f_2(t)]$$
$$f_1 + f_2 = 1$$



FORMATION:

Ion molecule reactions during the early afterglow

VT - AISA

$dn_i/dt = -\alpha n_i n_e$

He/Ar/H₂





40 cm diameter UHV - 10⁻⁹ Torr External magnetron 2 Torr of He/Ar/H₂

PULSED STATIONARY AFTERGLOW 20-100ms decay $n_e(\tau), n_i(\tau)$



Time resolved mass spectra

time [ms]



Recombination




























α: $P_{\mu W} = 10$ W, D = 15%100 β: $P_{\mu W} = 25$ W, D = 27%γ: $P_{\mu W} = 10$ W, D = 50%80 $T_{\rm kin}, T_{\rm rot-para}$ (K) 60 X 40 20 short discharge 0 long discharge 20 40 60 80 100 0 $T_{_{\rm H}}$ (K)

Examples of H_{3^+} absorption line profiles

The dependence of the kinetic temperature (T_{kin}) of H_3^+ ions on the temperature T_H (temperature of the discharge tube holder.



FLOWING AFTERGLOW

Diffusion in FA

 $[A^{+}] = [A^{+}]_{0} \exp(-Dt / \Lambda^{2}) = [A^{+}]_{0} \exp(-Dpt / p\Lambda^{2}) = [A^{+}]_{0} \exp(-D_{0}p_{0}L / vp\Lambda^{2})$ ~ $[A^{+}]_{0} \exp(-D_{0}p_{0}L / vp\Lambda^{2}) \sim [A^{+}]_{0} \exp(-const.L / Q)$



FLOWING AFTERGLOW

Ion-molecule reactions



Flowing Afterglow Langmuir Probe - FALP





FALP - RECOMBINATION OF H₃



FALP High pressure UHV version - PRAGUE



FALP – Ion detection system







The CRESU technique at Rennes

Carrier gas (He, Ar or N₂) + reactants



Kinetics of anion-molecule reactions at low temperature













Experiments

PLASMA experiments SA and FA

Crossed beam experiments Marched beam, Storage rings - TSR, Cryring, Astrid

- multi collisions
- single collisions
- single collisions

 $\begin{array}{l} \{\alpha(T)\} \\ \{\sigma(v_r)\} \\ \{\sigma(v_r)\} \end{array}$





Electron-cold molecular ion reaction: Dissociative Recombination





Center of mass resolution:

$$\Delta E_{cm} = \left\{ \left[\left(1 - \frac{v_e}{v_i}\right) \frac{m_e}{m_i} \Delta E_i \right]^2 + \left[\left(1 - \frac{v_i}{v_e}\right) \Delta E_e \right]^2 \right\}^{1/2}$$

meV resolution for zero relative kinetic energy!

TSR electron target



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94

Reality - TSR (MPIK Heidelberg)

Injection of <u>INTERNALY COLD</u> H₃⁺ <u>IONS(12-50K)</u> with kinetic energy 1-2 MeV



Detection of neutrals

INTERACTION at meV collision energies

Detection of $H_3^+(v,j)$

PLASMA PHYSICS I/7

Recombination

Dipole-Meutral fragments S = 5720mm INTERACTION Toroid-S = 7210mm Toroid-S = 7200mm Toroid-Toro

Kumulativní prstenec



TSR Heidelberg, ion injection and ion source



State diagnostics







8

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Thanks for your attention!





The CSR – overview



injection beamline, the neutral beam extraction and some of the corresponding detector units.

Max Planck Institut für Kernphysik, 69117 Heidelberg, Germany

astro

The CSR – overview

R









