Recombination ...

electron -- ion recombination

atomic ions

molecular ions

ion -ion recombination

Recombination



Radiative recombination

 $Ar^+ + e^- \rightarrow Ar + hv$

Dissociative recombination

 $AB^+ + e \rightarrow A + B$



<1 eV





Capture

AB* resonant state(s)





AB* resonant state(s)

predissociation

To get high recombination rate, we need (a) efficient capture (b) predissociation faster than auto-ionization **Electron**—ion recombination

 $H^+ + e^- \rightarrow products$

- $H_2^+ + e^- \rightarrow products$
- $H_3^+ + e^- \rightarrow products$
- $H_5^+ + e^- \rightarrow products$



$H_2^+ + e^- \rightarrow H + H,$ Dissociative Recombination - DR



Electron collisions with H_2^+ - how to describe ????









THEORETICAL FRAMEWORK

The states involved: exemple for He₂+/He₂ system



2004 DR6 Mosbach

I. Schneider, et al., DR2004 Mosbach

Charles University Prague

12/10/2024 1:21:18PM

12



internuclear separation, R

internuclear separation, R

Direct mechanism

Indirect mechanism

Electron-cold molecular ion reaction: Dissociative Recombination



Recombination of H_{3^+} : No ion-neutral crossing





Concept of recombination rate coefficient (plasma binary reactions) at T



e⁻ + A⁺ \rightarrow products dN_A/dt= - α n_eN_A **RECOMBINATION RATE COEFFICIENT** $\alpha = \alpha(T)$ $\sigma = \sigma(v) = \sigma(z)$ Collision rate coefficient, Recombination rate coefficient $\alpha = < \sigma u_r >$

Direct and indirect process

Recombination processes in plasma

Binary Recombination

$$H^{+} + e \rightarrow H + hv$$

$$RR$$

$$O_{2}^{+} + e \rightarrow O + O$$

$$DR$$

$$\frac{dn_{e}}{dt} = \frac{d[O_{2}^{+}]}{dt} = -\alpha[O_{2}^{+}]n_{e} = -\alpha n_{e}^{2}$$

$$Fe^{7+} + e \rightarrow Fe^{6+}$$

$$DiR$$

Recombination processes in plasma

Binary Recombination

$$H^{+} + e \rightarrow H + h\nu$$

$$PR$$

$$O_{2}^{+} + e \rightarrow O + O$$

$$PR$$

$$\frac{dn_{e}}{dt} = \frac{d[O_{2}^{+}]}{dt} = -\alpha[O_{2}^{+}]n_{e} = -\alpha n_{e}^{2}$$

$$Fe^{7+} + e \rightarrow Fe^{6+}$$

$$DiR$$

$$Ternary electron assisted recombination$$

$$Ar^{+} + e + e \rightarrow Ar + e$$

$$\frac{dn_{e}}{dt} = \frac{d[Ar^{+}]}{dt} = -K_{e}[Ar^{+}]n_{e}^{2} = -\alpha_{eff}[Ar^{+}]n_{e}$$

$$Collisional Radiative Recombination CRR$$

$$\alpha_{eff} = K_{e}n_{e}$$

Ternary neutral assisted recombination

 $Ar^+ + e + He \rightarrow Ar + He$

$$\frac{dn_{e}}{dt} = \frac{d[Ar^{+}]}{dt} = -K_{M}[Ar^{+}]n_{e}[He] = -\alpha_{eff}[Ar^{+}]n_{e}[He]$$

$$\alpha_{eff} = K_M[He]$$

 $f + e \rightarrow O + O$

234311-2 Petrignani et al.





Five exothermic channels are available for vibrational ground state O_2^+ ions in zero relative energy collisions with electrons. They are summarized as follows with the associated kinetic energy releases:

$$O_2^+(X^2\Pi_g, v=0) + e^- \rightarrow O(^3P) + O(^3P) + 6.65 \text{ eV}$$
 (1a)

- $\rightarrow O(^{3}P) + O(^{1}D) + 4.99 \text{ eV}$ (1b)
- $\rightarrow O(^{1}D) + O(^{1}D) + 3.02 \text{ eV}$ (1c)
- $\rightarrow O(^{3}P) + O(^{1}S) + 2.77 \text{ eV}$ (1d)
- $\rightarrow O(^{1}D) + O(^{1}S) + 0.80 \text{ eV.}$ (1e)



FIG. 2. DR rate coefficient k as a function of electron collision energy from 1 meV to 5 eV. Statistical errors are shown at the 1σ level. The dotted line shows the threshold $E^{-1/2}$ behavior. Both the rate coefficient and the energy are shown on a logarithmic scale.

Electron - Ion Recombination



Resonances

Resonances

Autoionizing and pre-dissociating Rydberg states



Dissociative recombination

Resonances



Dissociative recombination



Details of DR of HD+

Dissociative recombination HD+







Scan of electron ion relative energy EElectron temperature kT_{Per} =4meV (30meV for E>0.3eV) kT_{Par} =0.1meVEnergy resolution ~4....8 meV (E< 0.08eV)</td>Absolute accuracy of cross section ca. +- 30%

Recombination H₂⁺



Recombination only one rotational quanta change the whole spectra



Recombination calculation and theory H_2^+vibrational excitation



COMPUTATIONS vs MEASUREMENTS: H₂⁺/H₂

Ngassam et al 2003

Fifirig et al 2003



Different energy region

Recombination NO⁺

Tennyson et al 1996-2000





Comparison with ASTRID

Schneider et al 2000



Theoretical background **Dissociative Recombination without a Curve Crossing** Theory predicted: DR rate coefficient is vary small ~ 10⁻¹¹ cm³s⁻¹



HeH⁺ and HCO⁺ ionsexamples of a non-crossing case. However, experiments gave $\alpha \approx 2 \times 10^{-8}$ and $\alpha \approx 2 \times 10^{-7}$ cm³s⁻¹

A new mechanism has been proposed!

Multi-step indirect dissociative recombination ("tunneling mode" recombination) REPORTS

Cite as: O. Novotný *et al., Science* 10.1126/science.aax5921 (2019).

Quantum-state-selective electron recombination studies suggest enhanced abundance of primordial HeH⁺

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The epoch of first star formation in the early universe was dominated by simple atomic and molecular species consisting mainly of two elements: hydrogen and helium. Gaining insight into this constitutive era requires thorough understanding of molecular reactivity under primordial conditions. We used a cryogenic ion storage ring combined with a merged electron beam to measure state-specific rate coefficients of dissociative recombination, a process by which electrons destroy molecular ions. We found a dramatic decrease of the electron recombination rates for the lowest rotational states of HeH⁺, compared to previous measurements at room temperature. The reduced destruction of cold HeH⁺ translates into an enhanced abundance of this primordial molecule at redshifts of first star and galaxy formation.







Fig. 1. Dissociative recombination in the cryogenic storage ring CSR. (A) Scheme of the CSR ring structure with injected and stored HeH+ ion beam (red), merged electron beam (blue), reaction products (green) and particle detector. (B) Reaction scheme and position-sensitive detection of coincident fragments. (C) Equilibrium rotational state populations of HeH+ for previous studies (300 K) and the estimated radiation field in the CSR.



Fig. 2. DR for rotationally cold HeH⁺. (A) Blue dots: merged-beams rate coefficient α_{DR} as a function of detuning energy E_d after the relaxation to >50% J = 0 (this experiment, 10 s < t < 50 s, mean ± SD); absolute scaling uncertainty ±20% (SEM). Red symbols: roomtemperature data from Ref. (11) (squares, absolute scaling uncertainty ±10% SEM) and from Ref. (12) (triangles, scaled to Ref. (11) at 0.03 eV). (B) Fragment distance distribution projected into the detector plane for $E_d = 0$ (blue) with fit (19) for isotropic angular distribution (red). (C) Projected fragment distance distribution for Ed = 0.044 eV (blue) with fit (19) for a $|Y_{10}|^2$ angular distribution of the fragments (red). The angular dependences in (B) and (C) are indicated schematically.

Fig. 4. Rotational-state selective DR rates for HeH⁺. (A) Merged-beams rate coefficients $\alpha_{DR}^J(E_d)$ for $J \leq 2$ and average for $J \geq 3$ (mainly 3 and 4; mean \pm SD). The dashed lines mark the shift of the maximum as J increases. (B) Full lines: single-J plasma rate coefficients $\alpha_{DR,pl}^J(T_{pl})$ for $J \leq 2$ and average for $J \geq 3$ (mainly 3 and 4; mean with shaded areas as \pm SD). Dotted: fully thermal rate coefficient $\alpha_{DR,therm}(T_{rot} = T_{pl})$. Dashed-dotted: values applied in early-universe models (21, 22) and astrochemistry databases (23–25). See (19) for further discussion, numerical fitting functions and parameters.



Dissociative recombination of N_2H^+ ions with electrons in the temperature range of 80–350 K

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 $N_2H^+ + e^- \xrightarrow{\alpha_{bin}} neutral products$

Recombination of N_2H^+ ions with electrons was studied using a stationary afterglow with cavity ring-down spectrometer. We probed in situ the time evolutions of number densities of different rotational and vibrational states of recombining N_2H^+ ions and determined the thermal recombination rate coefficients for N_2H^+ in the temperature range of 80 – 350 K. The newly calculated vibrational transition moments of N_2H^+ are used to explain the different values of recombination rate coefficients obtained in some of the previous studies. No statistically significant dependence of the measured recombination rate coefficient on the buffer gas number density was observed.

I. INTRODUCTION

 N_2H^+ , an important interstellar ion, has been observed in different interstellar environments such as dark and translucent clouds^{1,2}, protostellar cores³, protoplanetary disks⁴ and is considered to play a role in the atmospheric chemistry of Titan⁵. N_2H^+ serves as an important tracer for N_2 in dark clouds, therefore detailed information on production and destruction processes of N_2H^+ could help with the prediction of N_2 abundance in this environment. N_2H^+ in the interstellar medium is mainly produced in proton transfer from H_3^+ to N_2 and its main destruction mechanisms are proton transfer to CO and dissociative recombination with electrons⁶.

$N_2H^+ + e^- \xrightarrow{\alpha_{bin}} neutral products$

II. EXPERIMENT

The recombination rate coefficients^{7,32} are measured in a stationary afterglow (SA) in conjunction with cavity ring-down spectroscopy (CRDS) to monitor the decay of the densities of different rotational and vibrational states of N₂H⁺ ions. The plasma is generated in a pulsed microwave discharge in a fused silica tube (inner diameter ≈ 1.3 cm). The microwave generator is equipped with an external fast high-voltage switch to cut off the power to the magnetron within a fall time of less than 30 µs. A low microwave power in the range of 10 – 25 W, with $\approx 40\%$ duty cycle, is used to avoid excessive heating of the gas during the discharge. The discharge tube temperature (*T*_{tec}) is measured by a thermocouple outside of the discharge and can be varied between 80 and 350 K.






FIG. 2. The dependence of the rotational temperature measured during the discharge on kinetic temperature. All the displayed data were obtained in helium buffer gas. Insert: An example of the Boltzmann plots used for determination of the rotational temperature obtained at T = 200 K and T = 78 K.



FIG. 3. An example of measured time evolutions of number densities of N_2H^+ ions in the ground and the first excited vibrational state. The lower panel shows the relative fractions of the vibrational states and the dashed lines denote the corresponding fraction in thermal equilibrium at temperature of 321 K. The particular vibrational states number densities were calculated from the measured number densities of the *J* = 6 rotational state of the ground vibrational state and of *J* = 9 state of the (01¹0) vibrational state under the assumption of the same rotational temperature in both vibrational states.

FIG. 4. Dependence of the measured vibrational temperature $T_{\rm vib}$ on kinetic temperature $T_{\rm kin}$ of the N₂H⁺ ions measured in the discharge. The kinetic temperature was obtained from the Doppler broadening of the P(6) line of the (200) \leftarrow (000) vibrational band of N₂H⁺. It was assumed that the rotational temperature of (000) and (010) states is equal to $T_{\rm kin}$. The vibrational temperature was then evaluated from the P(6) line of the (200) \leftarrow (000) vibrational band and from the R(9)^f line of the (210) \leftarrow (010) vibrational band of the N₂H⁺ ion. The full line denotes equality of $T_{\rm vib} = T_{\rm kin}$ and the dashed line indicates $T_{\rm vib} = T_{\rm kin} + 30$ K. The displayed errors are statistical errors of the fits.





FIG. 8. The temperature dependence of the measured recombination rate coefficients of N₂H⁺ (full circles, the value of α_{bin} at 350 K was obtained in H₂ buffer gas, for the rest of the data points, helium buffer gas was used) compared to values obtained in previous experiments. Rhomboids: FALP¹¹, squares: FALP¹³, stars: FALP¹⁴, triangles: stationary afterglow with absorption spectroscopy¹⁷, full line: ion storage ring⁶, double-dot-dashed line: merged beams⁸ and to recent theoretical calculations by Fonseca dos Santos²¹ (dot-dashed line, the rate coefficient was calculated from the cross sections for the direct and indirect recombination process in ref.²¹). The dashed line denotes fit to the data: $\alpha_{N_2H^+} = (2.81 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.81\pm0.10)} \text{ cm}^3 \text{s}^{-1}$ for T > 240 K and $\alpha_{N_2H^+} = (3.29 \pm 0.04) \times 10^{-7} (T/300)^{-(0.06\pm0.02)} \text{ cm}^3 \text{s}^{-1}$ otherwise. The dotted lines show 15% deviation from the fitted value (estimated systematic error of the measurement reflects mainly the uncertainty in the effective discharge column length and in the calculated vibrational transition moments).



FIG. 9. Time evolution of the measured overall number density of N₂H⁺ ions (full line) compared to the results obtained from the model of chemical kinetics. The data were obtained at T = 140 K, [He] = 1.5×10^{17} cm⁻³, [Ar] = 2.5×10^{14} cm⁻³, [H₂] = 5×10^{14} cm⁻³ and [N₂] = 4×10^{13} cm⁻³ (same as in Figure 1) and [NH₃] = 5×10^{11} cm⁻³. [He^m](t = 0) = 1/3 $n_e(t = 0)$.

Recombination of H₃⁺



$\begin{array}{rcl} H_{3}^{+}\!\!\!\!&+ e^{-} & \rightarrow & H\!+\!H\!+\!H \\ & \longrightarrow & H_{2}\!\!+\!H \\ & \longrightarrow & H_{3}^{*} & (?) \end{array}$

Tunneling dissociative recombination





Tunneling dissociative recombination







Dissociative recombination of H_3^+ . Relevant potential curves



Dissociative recombination of H₃+



Remote curve crossing

Electron capture via Jahn-Teller coupling of electronic and ro-vibrational motion

Prototype system for electron capture and dissociation mechanisms in polyatomic species

Symmetric deformation

Three atomic ions

Dissociative recombination of H₃+

Prototype system for electron capture and dissociation mechanisms in polyatomic species





FIG. 4. The figure demonstrates how the Jahn-Teller effect produces a high rate of dissociative recombination. One $2p\sigma$ potential surface and two $2p\pi$ potential surfaces [47] of the neutral molecule are shown. The conical intersection is produced by Jahn-Teller coupling. When an electron arrives, it scatters first into a low-lying vibrationally excited Rydberg state $\{01^1\}$. Then, after the nuclei vibrate, the system finds its way with high probability into a $2p\pi$ state having high vibrational excitation, near the point of conical intersection. The contour plot at the bottom of the figure represents the lowest $2p\pi_1$ surface. All three potential surfaces are shown in the reduced 2D space of dimensionless normal coordinates. The coordinates used here are the normal asymmetric Q_x , Q_y coordinate, with ρ and ϕ their polar components [17,18]. The third vibrational coordinate—the symmetric stretch coordinate Q_1 —is kept constant for this graph.

2004-7

PHYSICAL REVIEW A 68, 012703 (2003) 0.2Energy (a.u.) -0.2 1.5 2 2.5 3 Hyperradius (a.u.)

FIG. 1. The problem of DR of H_3^+ in the hyperspherical adiabatic approximation. The lowest hyperspherical adiabatic potential (thick full line) of the H_3^+ and number of hyperspherical adiabatic potentials of the neutral molecule (thin lines). Lower family of lines (darker lines) dissociate to the H_2 +H channel; the upper family (lighter lines) dissociate to the H+H+H channel. To calculate hyperspherical adiabatic curves we used the three-dimensional H_3^+ potential from Ref. [48] and the H_3 potential from Refs. [35–37]. Since the density of hyperspherical states is high, only every tenth H_3 potential curve is shown in the figure. The dashed line shows the position of the ground vibrational level of the ion, which is the only one populated in the relevant experiments.



FIG. 4. The figure demonstrates how the Jahn-Teller effect produces a high rate of dissociative recombination. One $2p\sigma$ potential surface and two $2p\pi$ potential surfaces [47] of the neutral molecule are shown. The conical intersection is produced by Jahn-Teller coupling. When an electron arrives, it scatters first into a low-lying vibrationally excited Rydberg state {01¹}. Then, after the nuclei vibrate, the system finds its way with high probability into a $2p\pi$ state having high vibrational excitation, near the point of conical intersection. The contour plot at the bottom of the figure represents the lowest $2p\pi_1$ surface. All three potential surfaces are shown in the reduced 2D space of dimensionless normal coordinates. The coordinates used here are the normal asymmetric Q_x , Q_y coordinate, with ρ and ϕ their polar components [17,18]. The third vibrational coordinate—the symmetric stretch coordinate Q_1 —is kept constant for this graph.



(b) predissociation faster than auto-ionization





... history is repeating itself

... One remaining problem is to understand the plasma afterglow experiments.



Plasma in TDE P(A) $(U_{j,2}E_{j}) \leq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{j}) \leq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{j}) \leq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{i}) \geq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{i}) \geq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i,2}E_{i}) \geq \sum_{i=1}^{N} P(E_{i}),$ $(U_{i$

.... many times it was concluded, that the task was finished....

... and the caravan is on its way

Calculated life time from Slava



$$H_3^+ + e \rightarrow H_3^*$$



Dear Juraj and Chris, I'm sending you the figure with the DR probabilities for two different symmetries (red and black curves). The red curve corresponds to the rotational autoionization region. Fro this figure you can have an idea about the widths of the resonances. With best wishes, Slava

$$H_3^+ + e \Leftrightarrow H_3^* \rightarrow$$



Recombination rate coefficients



FIG. 5. (Color online) The present theoretical thermal rate coefficient for dissociative recombination of H_3^+ is compared with the experimental rate coefficient deduced from the storage ring experiment of McCall and co-workers (Refs. 9 and 10).



FIG. 3. (Color online) This figure compares the theoretical DR rate coefficient to the high-resolution storage ring experiment of Kreckel *et al.*¹² carried out at TSR. The experimental resolution parameters are ΔE_{\parallel} and ΔE_{\perp} are 25 μ eV and 0.5 meV, respectively. The theoretical curve shown has been calculated with these parameters and rotational temperature T_{rv} = 1000 K. The figure also shows the theoretical DR rate coefficients calculated separately for ortho- and paraconfigurations of H₃⁺ with the same parameters ΔE_{\parallel} , ΔE_{\perp} , and T_{rv} .

The battle ship enters the stage

FAL



Πλασμα





10-8





Pressure dependence

Pulsed (stationary) afterglow



+ molecules

We measure effective – apparent binary recombination rate coefficient



Quasineutral H₃⁺ dominated plasma

 $\frac{\mathrm{d}n_{\mathrm{e}}}{\mathrm{d}t}$ $n_{\rm e}$ $-\alpha_{\rm eff}n_{\rm e}$ $au_{
m L}$

$$\frac{1}{[H_3^+]} = \frac{1}{[H_3^+]_0} + \alpha t$$

We measure effective – apparent binary recombination rate coefficient



Quasineutral H₃⁺ dominated plasma

$$\frac{\mathrm{d}n_{\mathrm{e}}}{\mathrm{d}t} = -\alpha_{\mathrm{eff}}n_{\mathrm{e}}^2 - \frac{n_{\mathrm{e}}}{\tau_{\mathrm{L}}}$$

$$\frac{1}{[H_3^+]} = \frac{1}{[H_3^+]_0} + \alpha t$$

processes at high densities at low T

Decay in <u>diffusion</u> and <u>recombination</u> governed plasma

$$\frac{dn_e}{dt} = -\alpha n_e^2 - \frac{D_a}{\Lambda^2} n_e$$

$$\frac{1}{n_e} = \alpha \frac{\exp(\nu t) - 1}{\nu_D} + \frac{1}{n_0} \exp(\nu_D t)$$



Decay in <u>diffusion</u> and <u>recombination</u> governed plasma

$$\frac{dn_e}{dt} = -\alpha \ n_e^2 - \frac{D_a}{\Lambda^2} n_e$$

$$\frac{1}{n_e} = \alpha \ \frac{\exp(\nu t) - 1}{\nu_b} + \frac{1}{n_0} \exp(\nu_b t)$$
Limit for t -->0
$$\frac{1}{n_e} = \alpha \ \frac{(1 + \nu_b t) - 1}{\nu_b} + \frac{1}{n_0} (1 + \nu_b t)$$

$$\frac{1}{n_e} = \alpha t + \frac{1}{n_0} (1)$$

$$\frac{1}{n_e} = \alpha t + \frac{1}{n_0} (1)$$



Srovnani modelu a experimentu. Pocatecni podminka: $[H_3^+] = n_e = 2 \times 10^{11} \text{ cm}^{-3}$.



$$\frac{dn_e}{dt} = -[\alpha_1 n_1(t) + \alpha_2 n_2(t)]n_e$$
$$\Rightarrow \alpha_{eff}(t) = [\alpha_1 f_1(t) + \alpha_2 f_2(t)]$$
$$f_1 + f_2 = 1$$

Model + data. Pocatecni podminka: $n_e = He^m = [H_3^+]$.

Poznamka. Namerene τ difuznich ztrat 1.6 ms. Teoreticke τ pri danem tlaku je 1.8 ms (odpovida cca 4×10¹⁰ cm⁻³ koncentraci necistot (pri 2×10⁻⁹ cm³s⁻¹ rychlosti reakce H₃⁺ s necistotami). Namerena koncentrace vody [H₂O] = 5×10¹⁰ cm⁻³ ([He] = 8×10¹⁷ cm⁻³).

If there are 2 or more ion species, the fast recombining species disappears first

$$\frac{dn_e}{dt} = -[\alpha_1 n_1(t) + \alpha_2 n_2(t)]n_e$$
$$\Rightarrow \alpha_{eff}(t) = [\alpha_1 f_1(t) + \alpha_2 f_2(t)]$$
$$f_1 + f_2 = 1$$

Diffusion and recombination



 $\frac{dn_e}{L} = -\alpha n_e^2 - \frac{D_a}{\Lambda^2} n_e$

dt

 $\frac{1}{n_e} = \alpha \frac{\exp(\nu t)}{\nu}$

 $\frac{1}{n_0} + \frac{1}{n_0} \exp(vt)$

J. Glosík, G. Bánó, R. Plašil, A. Luca, P. Zakouril,

Study of the electron ion recombination in high pressure flowing afterglow. Recombination of NH_4^+ .(NH_3)₂, International J. Mass Spectrom., 189, 103-113 (1999)

Advanced analyze O_2^+

$$dn_{e} / dt = -\alpha [O_{2}^{+}]n_{e} - v_{D}n_{e} - v_{R}n_{e} = -\alpha n_{e}^{2} - v_{D}n_{e} - v_{R}n_{e}$$

$$\frac{1}{n_e} - \frac{1}{n_0} = \alpha(t_e - t_0)$$

$$-n_e'/n_e^2 = \alpha + (\nu_D + \nu_R)/n_e$$





Stationary afterglow (M. Biondy, R. Johnsen)



M.T.Leu, M.A. Biondy, R. Johnsen recombination of H_3^+ and H_5^+





R. Johnsen N4+ recombination at 300-800 Torr a= 2.6x10⁻⁶cm³s⁻¹

High pressure SA

RF probe, spark discharge



ADVANCED INTEGRATED STATIONARY AFTERGLOW

AISA



40 cm diameter UHV - 10⁻⁹ Torr External magnetron 2 Torr of He/Ar/H₂ PULSED STATIONARY AFTERGLOW 20-100ms decay



AISA - ADVANCED INTEGRATED STATIONARY AFTERGLOW



CALCULATION OF PLASMA DECAY IN CYLINDER



DIFFUSION AND RECOMBINATION τ_{D} =60 ms, α =1x10⁻⁷cm³s⁻¹ and α =5x10⁻⁹cm³s⁻¹



₂₋₂ Diffusion in Cylinder Time= 8.10e-3 40 max 1.18 1.15 1.10 1.05 1.00 0.95 35 30. 0.90 0.85 0.80 25. 0.70 0.65 20 0.55 0.50 0.45 0.40 0.35 0.30 0.25 0.20 0.15 0.10 0.05 0.00 0.00 15. 10. 5. c: b: a: min 0 0.05 0.1 0.15 0.2 0.25 0.3 -0.1 -0.05 0. R

Time evolution

Spatial distribution after 8 ms





FORMATION:

Ion molecule reactions during the early afterglow

VT - AISA

$dn_i/dt = -\alpha n_i n_e$

He/Ar/H₂





40 cm diameter UHV - 10⁻⁹ Torr External magnetron 2 Torr of He/Ar/H₂

PULSED STATIONARY AFTERGLOW 20-100ms decay $n_e(\tau), n_i(\tau)$



Time resolved mass spectra

time [ms]



From over 30 years of systematic studies of kinetics of IMR follows:

Kinetics of formation of H₃+ is clear IT IS NOT A MYSTERY !!
Table of reactions

R. No	Reaction	Rate coef. [cm ³ s ⁻¹] or [cm ⁶ s ⁻¹]	Reactant number density [cm ⁻³]	Reaction time [ms]	Ref.
1	$He^+ + 2He \rightarrow He_2^+ + He$	1E-31	6,5E16	2	
2	$He^M + He^M \rightarrow He_2^+ + e$	5E-9	5E10 (assumption)	4	
3	$He^M + Ar \rightarrow Ar^+ + He + e$	7E-11	1E14	0,14	
4	$He_2^+ + Ar \rightarrow Ar^+ + 2He$	2E-10	1E14	0,05	
5	$Ar^{+}+H_{2} \rightarrow ArH^{+}+H$ $\rightarrow H_{2}^{+}+Ar$	8E-10 <i>ArH</i> + is dominat	2E11	6,3	
6	$H_2^+ + Ar \rightarrow ArH^+ + H$	2,3E-9	1E14	<0,01	
7	$H_2^+ + H_2 \rightarrow H_3^+ + H$	2,1E-9	2E11	2,4	
8	$ArH^++H_2 \rightarrow H_3^++Ar$	1.5E-9	2E11	3,3	
9	$H_3^+ + H_2 + He \rightarrow H_5^+ + He$	<1E-29	2E11; 6,5E16	>8000	
10	$H_3^+ + H_2^- + H_2^- \rightarrow H_5^+ + H_2^-$	4,6E-30; 210K	2E11	>1E9	
11	$H^+ + He + He \rightarrow HHe^+ + He$	0,9E-31	6,5E16	2,6	







Measured time evolution

Time resolved mass spectra



Rates of the decays are dependent on [H₂]



Time resolved mass spectra

$$\frac{dn_e}{dt} = -[\alpha_1 n_1(t) + \alpha_2 n_2(t)]n_e$$
$$\Rightarrow \alpha_{eff}(t) = [\alpha_1 f_1(t) + \alpha_2 f_2(t)]$$
$$f_1 + f_2 = 1$$



It is only qualitative information, not sufficient to obtain **CL**



FLOWING AFTERGLOW

Diffusion in FA

 $[A^{+}] = [A^{+}]_{0} \exp(-Dt / \Lambda^{2}) = [A^{+}]_{0} \exp(-Dpt / p\Lambda^{2}) = [A^{+}]_{0} \exp(-D_{0}p_{0}L / vp\Lambda^{2})$ ~ $[A^{+}]_{0} \exp(-D_{0}p_{0}L / vp\Lambda^{2}) \sim [A^{+}]_{0} \exp(-const.L / Q)$



FLOWING AFTERGLOW

Ion-molecule reactions



Flowing Afterglow Langmuir Probe - FALP





FALP - RECOMBINATION OF H₃



Flowing afterglow/Langmuir probe (FALP) D. Smith, N. G. Adams and P. Spanel

• A schematic diagram of the **FALP apparatus** showing the relative positions of the microwave discharge that generates the afterglow plasma and the three reactant gas entry ports **P1**, **P2** and **P3**. The distance (*z*) scale is referenced to the downstream mass spectrometer sampling orifice. The complete flow tube is surrounded by a **vacuum jacket** to facilitate high and low temperature operation (ranging from 80 to 600K).



Positive ion/electron dissociative recombination • e.g. $NO^+ + e \rightarrow N + O$

$$\frac{dn_e}{dt} = -\alpha n_e^2 + D_a \nabla^2 n_e \quad \text{diffusion}$$

$$\frac{dn_e}{dt} = -\alpha n_e^2 - \frac{D_a}{\Lambda^2} n_e$$

$$\frac{1}{n_e} - \frac{1}{n_0} = \alpha(t_e - t_0)$$

$$\frac{1}{n_e} = \alpha \frac{\exp(\nu t) - 1}{\nu} + \frac{1}{n_0} \exp(\nu t) ; \nu = D_a / \Lambda^2$$

experiments (CRDS)

FALP studies of the dissociative recombination coefficients for O_2^+ and NO^+ within the electron temperature range $300-2000 \text{ K}^1$

Patrik Španěl, Libuše Dittrichová[†], David Smith*





Fig. 1. A line diagram (approximately to scale) of the FALP apparatus, indicating the positions of the microwave discharge, the entry port of helium coolant gas (for the argon afterglow plasmas), the entry ports P1 for argon (for the helium afterglow plasmas) and P2 for O₂ and NO, and the mass spectrometer. The Langmuir probe can be positioned at any point on the axis of the flow tube. Also indicated is the outline of the vacuum jacket which facilitates the heating and cooling of the complete flow tube.

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P. Španěl et al./Int. J. Mass Spectrom. Ion Processes 129 (1993) 183-191



Fig. 2. Representative plots of d^2i/dV^2 against V obtained from the Langmuir probe current (i)-voltage (V) characteristics from which electron temperatures T_e are obtained. In every case the carrier gas temperature is 300 K. The good linearity of a plot is indicative of a Maxwellian electron energy distribution function (EEDF). The z values are the positions of the probe along the axis of the flow tube (referenced to the mass spectrometer sampling orifice; see Fig. 1). The entry port P2 via which the O_2 and NO ion source gases were introduced into the afterglows is located at z = 56 cm. (a) Obtained in helium afterglows (at a pressure of 1 Torr) with a small admixture of argon to destroy helium metastable atoms. Note the excellent linearity of the plots and the small scatter of T_e about T_g (= 300 K) along z. (b) Obtained in pure argon afterglows (at a pressure of 0.7 Torr). Note the much higher T_e values (compared with those in (a)) and the small (but obvious) T_e gradient along z. (c) Obtained in argon afterglows into which O_2 (partial pressure of $\approx 5 \text{ mTorr}$) has been added to generate an O_2^+ (electron plasma. Note the somewhat greater fractional decrease in T_e compared with that in the pure argon afterglows into which $frest = 10^{-1}$, has been added to cool the electron gas (see Fig. 1), and then into which O_2 (partial pressure 50 mTorr) has been added to cool the electron gas (see Fig. 1), and then into which O_2 (partial pressure 50 mTorr) has been added to cool the electron gas (see Fig. 1), and then into which O_2 (partial pressure 50 mTorr) has been added to cool the electron gas (see Fig. 1), and then into which O_2 (partial pressure 50 mTorr) has been added to cool the electron gas (see Fig. 1), and then into which O_2 (partial pressure 50 mTorr) has been added to cool the electron gas (see Fig. 1), when into which O_2 (partial pressure 50 mTorr) has been added to cool the electron gas (see Fig. 1), and then into which O_2 (partial pressure 50

$$v_{\rm p}\frac{{\rm d}n_{\rm e}}{{\rm d}z} = -\alpha n_{\rm e}^2 + D_{\rm a}\nabla^2 n_{\rm e} \tag{4}$$

As a reasonable approximation, we assume that diffusive loss is via the fundamental mode only and then

$$v_{\rm p}\frac{{\rm d}n_{\rm e}}{{\rm d}z} = -\alpha n_{\rm e}^2 - \frac{D_{\rm a}}{\Lambda^2} n_{\rm e} \tag{5}$$

where D_a is the ambipolar diffusion coefficient and Λ is the characteristic diffusion length (for the flow tube used here, $\Lambda^2 = 2.76 \text{ cm}^2$), and v_p is the plasma flow velocity $(1.1 \times 10^4 \text{ cm s}^{-1})$. When recombination is the dominant loss process, such as is the case in these studies of both $\alpha(O_2^+)$ and $\alpha(NO^+)$ at the lower T_e , (and certainly at 300 K in the helium carrier gas), then the diffusion term in Eq. (5) can be neglected, and the solution to Eq. (5) is then

$$\frac{1}{n_{\rm t}} - \frac{1}{n_0} = \frac{\alpha(z_{\rm t} - z_0)}{v_{\rm p}} \tag{6}$$

Electron temperature measurement

(a) Obtained in <u>helium</u> afterglow (at a pressure of 1 Torr) with a small admixture of argon to destroy helium metastable atoms.

(b) Obtained in pure argon afterglow (at a pressure of 0.7 Torr).

(c) Obtained in **argon** afterglow into which O_2 (partial pressure of ~5mTorr) has been added to generate an O_2^+ /electron plasma.



D.Smith and P.Spanel

Electron temperature dependence

Plots of $1/n_e$ against the distance z along the flow tube obtained in O_2^+ /electron afterglow plasmas from which values for $\alpha(O_2^+)$ are obtained.

The data indicated by filled circles were obtained in helium carrier gas when $T_e = T_i = T_g = 300$ K; the linearity of the plot over a factor of about ten indicates that dissociative recombination is the dominant loss process for electrons and ions.

The data represented by open circles (T_e =650 K) were obtained in argon carrier gas at T_g = 300 K, and at elevated T_e .



FALP and in-situ data



Dissociative recombination of different ions



RENNES MS-FALP



Figure 1. Sketch of the FALP apparatus.

RENNES absorption studies



Distance z (cm)

Pittsburg Rainer Johnsen FALP

Emission spectroscopy for identification products of recombination collisional radiative recombination of argon ions $Ar^+ + e^- + e^- \rightarrow Ar + e^-$







The Pittsburgh flow tube



N.G.ADAMS University of Georgia, Viktoria Poterya



Figure 1. A schematic of the University of Georgia flowing afterglow. Illustrated are the axially movable Langmuir probe, the downstream mass spectrometer, a 0.66 m monochromator with red sensitive photomultiplier for emission studies, a vuv light source and 1 m vacuum monochromator with uv enhanced photomultiplier for detection of atoms and a YAG pumped dye laser with doubling and mixing capabilities for detection of radical species by LIF and REMPI. All photomultipliers are cooled to reduce the background noise and photon counting is used throughout. Further details of this apparatus are described in a separate review.²⁷

FALP High pressure UHV version - PRAGUE



FALP – Ion detection system





FALP - Pumping units and gas handling system



FALP high pressure version To demonstrante how simple it is in reality

2 men experiment



Variation of the probe characteristics along the flow tube



The decay time is correlated with the position in the flow tube. He pressure 8.8 Torr, temperature 190 K, [Ar] 1.4 mTorr), $[H_2] = 9.6 \times 10^{14}$ cm⁻³.

Evolution of the probe characteristics along the flow tube

Study of H₃⁺ and H₅⁺recombination





PLASMA DECAY H_3^+ and H_5^+ in thermodynamic equilibrium





Note very different time scale!!!

Plasma parameters along the flow tube



EEDF measurements



The time evolution of the EEDF in the recombination dominated FA plasma In He (p=9 Torr) with small admixture of HCOH (0.05 %). EEDF is normalized to the electron number density.



Recombination of $H^+(HCOH)_2 + e^-$


Recombination aldehyde protonated dimmers with electrons

Conditions for measuring of therecombination



 $H^{+} \cdot (HCOH)_{2}$ $H^{+} \cdot (CH_{3}COH)_{2}$ $H^{+} \cdot (CH_{3}COCH_{3})_{2}$ $\alpha = (3.3 \pm 1) \cdot 10^{-6} \text{ [cm}^3\text{s}^{-1}\text{], at } \text{T}_e \sim 400 \text{ K}$ $\alpha = (1.4 \pm 1) \cdot 10^{-6} \text{ [cm}^3\text{s}^{-1}\text{], at } \text{T}_e \sim 500 \text{ K}$ $\alpha = (8 \pm 2) \cdot 10^{-6} \text{ [cm}^3\text{s}^{-1}\text{], at } \text{T}_e \sim 400 \text{ K}$

Dissociative electron attachment CF₃Br +e⁻



Fig. 4. Plots of the rate coefficient β for dissociative electron attachment to CF₃Br as a function of the attaching gas temperature T_g (= the carrier gas temperature) and the electron temperature T_e . The data points indicated (×), i.e. the β vs. $T_g(=T_e)$ data, are from a previous FALP study carried out in helium afterglows [11]. The points indicated (Δ) are for $T_g = 300$ K and those indicated (\diamond) are for $T_g = 520$ K; these data were obtained largely in argon afterglows (see text); (----) is described by Eq. (2) for the condition that $T_e = T_g$; (- -) are described by the same equation but for the fixed T_g values of 300 and 520 K and for variable T_e .

buffer He/Ar mixture



The CRESU technique at Rennes

Carrier gas (He, Ar or N₂) + reactants



Kinetics of anion-molecule reactions at low temperature











Experiments

PLASMA experiments SA and FA

Crossed beam experiments Marched beam, Storage rings - TSR, Cryring, Astrid

- multi collisions
- single collisions
- single collisions

 $\begin{array}{l} \{\alpha(T)\} \\ \{\sigma(v_r)\} \\ \{\sigma(v_r)\} \end{array}$





Electron-cold molecular ion reaction: Dissociative Recombination





Center of mass resolution:

$$\Delta E_{cm} = \left\{ \left[\left(1 - \frac{v_e}{v_i}\right) \frac{m_e}{m_i} \Delta E_i \right]^2 + \left[\left(1 - \frac{v_i}{v_e}\right) \Delta E_e \right]^2 \right\}^{1/2}$$

meV resolution for zero relative kinetic energy!

TSR electron target



Charles University Prague

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Reality - TSR (MPIK Heidelberg)

Injection of <u>INTERNALY COLD</u> H₃⁺ <u>IONS(12-50K)</u> with kinetic energy 1-2 MeV



Detection of neutrals

INTERACTION at meV collision energies

Detection of $H_3^+(v,j)$

PLASMA PHYSICS I/7

Recombination

Dipole-Meutral fragments S = 5720mm INTERACTION Toroid-S = 7210mm Toroid-S = 7200mm Toroid-S = 7200mm Toroid-S = 7200mm Toroid-S = 7200mm Toroid-Toro

Kumulativní prstenec



TSR Heidelberg, ion injection and ion source



Ion storage rings





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Charles University Prague

Electron cooling

Electron cooling



Charles University Prague

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State diagnostics







25

Charles University Prague

Recombination H₂⁺



Thanks for your attention!



The CSR – overview



injection beamline, the neutral beam extraction and some of the corresponding detector units.

Max Planck Institut für Kernphysik, 69117 Heidelberg, Germany

astro

The CSR – overview

R













A set-up for ion—atom collisions at the Cryogenic Storage Ring

 $H_3^+ + C \longrightarrow CH^+ + H_2, \quad H_3^+ + O \longrightarrow OH^+ + H_2$



Figure 4. Overview of the CSR facility and the peripheral infrastructure for the ion-atom collision experiments.



Figure 3. (*a*) Calculated Feshbach resonances for the photodissociation of CH^+ for rotational states J = 0-9. The window of resonances shifts with the rotational quantum number of the initial state. For the lowest rotational states (J = 0-2), some of the resonances overlap. This is the area of interest to monitor the rotational cooling inside the CSR. (*b*) Photodissociation spectrum of CH^+ measured for two different storage time intervals inside the CSR. The resonances can be fitted and assigned to individual rotational states. The resulting state populations are given in the insets.







Figure 5. Detail of the straight section of the CSR that houses the low-energy electron cooler and the corresponding positionsensitive, multi-hit counting detector.







The CSR electron cooler



electrostatic storage ring with circumference≈35 m first beam stored: March 2014 cryogenic operation: since April 2015

High Voltage platforms



Thanks for your attention!



The battle ship enters the stage

FAL



Πλασμα





10-8





Pressure dependence



Line intensity H₃⁺

Energy (cm⁻¹)


Stationary afterglow + Spectroscopic identification of recombining ions





Pulsed discharge – plasma decay



Wavenumber [cm⁻¹]





From Doppler broadening









History of experiments –"time evolution"









$$\frac{dn_e}{dt} = -[\alpha_1 n_1(t) + \alpha_2 n_2(t)]n_e$$
$$\Rightarrow \alpha_{eff}(t) = [\alpha_1 f_1(t) + \alpha_2 f_2(t)]$$
$$f_1 + f_2 = 1$$

Model + data. Pocatecni podminka: $n_e = \text{He}^m = [\text{H}_3^+]$.

Poznamka. Namerene τ difuznich ztrat 1.6 ms. Teoreticke τ pri danem tlaku je 1.8 ms (odpovida cca 4×10¹⁰ cm⁻³ koncentraci necistot (pri 2×10⁻⁹ cm³s⁻¹ rychlosti reakce H₃⁺ s necistotami). Namerena koncentrace vody [H₂O] = 5×10¹⁰ cm⁻³ ([He] = 8×10¹⁷ cm⁻³).

If there are 2 or more ion species, the fast recombining species disappears first

$$\frac{dn_e}{dt} = -[\alpha_1 n_1(t) + \alpha_2 n_2(t)]n_e$$
$$\Rightarrow \alpha_{eff}(t) = [\alpha_1 f_1(t) + \alpha_2 f_2(t)]$$
$$f_1 + f_2 = 1$$

063116-3 Plašil et al.











REVIEW OF SCIENTIFIC INSTRUMENTS 89, 063116 (2018)

Stationary afterglow apparatus with CRDS for study of processes in plasmas from 300 K down to 30 K

R. Plašil,^{1,a)} P. Dohnal,¹ Á. Kálosi,¹ Š. Roučka,¹ D. Shapko,¹ S. Rednyk,¹ R. Johnsen,² and J. Glosík¹

$H_{3^{+}}$ Nuclear spin dependence of $H_{3^{+}}$ recombination

- B. J. McCall, et al. *Physical Review A* (2004)
- H. Kreckel, J. Glosik, et al. Phys. Rev. Lett. 2005,

....2008, new improved calculations

Astronomy & Astrophysics L. Pagani¹, C. Vastel², E. Hugo³, V. Kokoouline⁴, Chris H. Greene⁵, A. Bacmann⁶, E. Bayet⁷, C. Ceccarelli⁶, R. Peng⁸, and S. Schlemmer³

- M. Larsson, B.J. McCall, A.E. Orel (2008)
- J. Glosik, R. Plasil, et al. Phys. Rev. A, 2009.
- H. Kreckel, O. Novotny, et al., Phys. Rev. A (2010).
- K. N. Crabtree, N. Indriolo, et al., Astrophys. J. (2011)
- J. Varju, M. Hejduk, J. Glosik, et al. Phys. Rev. Lett., 2011.
- P. Dohnal, M. Hejduk, J. Glosik, et al. J. Chem. Phys., 2012.



Doubts 2011

"Presently no rate coefficient measurement with a confirmed temperature below 300 K exists".

Petrignani et al. Phys. Rev. A (2011)

FIG. 5. (Color online) The present theoretical thermal rate coefficient for dissociative recombination of H_3^+ is compared with the experimental rate coefficient deduced from the storage ring experiment of McCall and co-workers (Refs. 9 and 10).

. Unfortunately the experiments on storage rings were stopped 😁 😁



State of the art in 2013???



The dissociative recombination of H_{3^+} – a saga coming to an end?

'Yes, the saga is coming to an end; but slowly.'

M. Larsson, B.J. McCall, A.E. Orel (2008)

..... Presently no reliable recombination rate coefficient for H3+ measured with storage rings below 300 K exists.

H. Kreckel, O. Novotny, K. N. Crabtree, et al., Phys. Rev. A (2010). A. Petrignani, S. Altevogt, M. H. Berg, et al., Phys. Rev. A (2011).

The recent observations made towards several diffuse molecular clouds showed large difference between excitation temperatures T10(H2) and T(H3+), for details see ref. [cra11].

These observations lead to conclusion that in reliable chemical models the nuclear spin dependences of the reactions, including recombination of para- and ortho-H3+, have to be considered. The dependences on spin, rotational excitation and temperature have to be measured.

K. N. Crabtree, N. Indriolo, H. Kreckel, B. A. Tom, and B. J. McCall, Astrophys. J. (2011)

Help! Theory for H₃⁺ Recombination Still Needed We still badly need theory and the caravan is on its way





Takeshi Oka, DR2013

.... It is time to present some recent results from afterglow experiments ...

SA a FALP



DR2007 - Dependence on He and H_2 pressure at 260 K

 $a_{eff} = a_{eff}(T_e, T_i, n_e, [He], [H_2], {}^{o/p}f_2, {}^{o/p}f_3)$

$a_{eff} = a_{eff}(T, [He])$



Afterglow in He/Ar/H₂ mixture



J. Phys. B: At. Mol. Opt. Phys. 41 (2008) 191001 (6pp)

Battle ships

Binary + He assisted ternary recombination



J. Phys. B: At. Mol. Opt. Phys. 41 (2008) 191001 (6pp)

Just before splitting 🙂















Model

$$\alpha_{\text{eff}} = \alpha_{\text{eff}}(T_{e}, T_{i}, n_{e}, [\text{He}], [\text{H}_{2}], {}^{o/p}f_{3})$$

$$H_3^+ + e^- \xrightarrow{\alpha_{Bin}} H_2 + H_{,.}H + H + H$$

$$\xrightarrow{\alpha_{F}} H_{3}^{\#} \xrightarrow{H_{2}...,k_{SH_{2}}} \text{neutrals}$$

$$\xrightarrow{\tau_{a}} H_{3}^{\#} \xrightarrow{H_{2}...,k_{SH_{2}}}$$

By solving the set of balance equations we obtain:

(He/Ar/H₂ mixture)
$$\frac{\partial n_{e}}{\partial t} = -(\alpha_{bin} - \alpha_{F} \frac{k_{SHe}[He] + k_{SH_{2}}[H_{2}]}{\frac{1}{\tau_{a}} + k_{SHe}[He] + k_{SH_{2}}[H_{2}]})[H_{3}^{+}]n_{e}$$

$$K_{\text{He}} = \alpha_{\text{F}} k_{\text{SHe}} \tau_{\text{a}}$$
 $K_{\text{H2}} = \alpha_{\text{F}} k_{\text{SH2}} \tau_{\text{a}}$

$$\alpha_{\text{eff}} = \alpha_{\text{bin}} + \alpha_{\text{F}} \frac{K_{\text{He}}[\text{He}] + K_{\text{H2}}[\text{H}_{2}]}{\alpha_{\text{F}} + K_{\text{He}}[\text{He}] + K_{\text{H2}}[\text{H}_{2}]}$$

In the low density limit ([He] and $[H_2] \rightarrow 0$), linear approximation

$$\alpha_{\rm eff} = \alpha_{\rm bin} + K_{\rm He} [\rm He] + K_{\rm H2} [\rm H_2]$$

Experiments -State of the art 2015

Experiments - State of the art in 2015

$$H_3^+ + e + He \rightarrow \dots + He$$



$H_3^{+} + e + He \rightarrow \dots + He$ $H_3^{+} + e + H_2 \rightarrow \dots + H_2$



$$\alpha_{\rm eff} = \alpha_{\rm bin} + \alpha_{\rm F} \frac{K_{\rm He}[\rm He] + K_{\rm H2}[\rm H_2]}{\alpha_{\rm F} + K_{\rm He}[\rm He] + K_{\rm H2}[\rm H_2]}$$

CRR

$$H_3^+ + e + e \rightarrow \dots + e$$

Rate coefficient binary





para-H₃⁺ and orto-H₃⁺





Rate coefficient of formation



para- H_3^+ and orto- H_3^+





Rate coefficient ternary



$$K_{\rm He} = \alpha_{\rm F} k_{\rm SHe} \tau_{\rm a} \qquad K_{\rm H2} = \alpha_{\rm F} k_{\rm SH2} \tau_{\rm a}$$

$$\alpha_{\rm eff} = \alpha_{\rm bin} + \alpha_{\rm F} \frac{K_{\rm He}[{\rm He}] + K_{\rm H2}[{\rm H}_2]}{\alpha_{\rm F} + K_{\rm He}[{\rm He}] + K_{\rm H2}[{\rm H}_2]}$$







Rate coefficients summary

Plasma Sources Science and Technolog doi:10.1088/0963-0252/24/6/0650

Plasma Sources Sci. Technol. 24 (2015) 065017 (10pp)

Recombination of H_3^+ ions with electrons in He/H₂ ambient gas at temperatures from 240 K to 340 K

J Glosík¹, P Dohnal¹, P Rubovič¹, Á Kálosi¹, R Plašil¹, Š Roučka¹ and R Johnsen²

$$K_{\rm He} = \alpha_{\rm F} k_{\rm SHe} \tau_{\rm a} \qquad K_{\rm H2} = \alpha_{\rm F} k_{\rm SH2} \tau_{\rm a}$$





History and state of the art



Different views & different plasmas

H₃⁺ and its interaction of with e⁻ is FUNDAMENTAL



I JAKO KOMIKS.

J.E.P. Connerney and T. Satoh, Phil. Trans. R. Soc. Lond. A358, 2471 (2000 Ion storage rings

AISA

FALP





New "state selective" study with "cold ion source" observed faster recombination of para H_3^+ in comparison with ortho H_3^+

B. J. McCall, et al. *Physical Review A* (2004)

H. Kreckel, et al. Phys. Rev. Lett. 2005,



In the middle of 2008 M. Larsson et al wrote the Frontier article to Chem. Phys. Let., [Larsson 2008].

This was written in the abstract:

".... Two independent ion storage ring experiments with rovibrationally cold H3+ ions are in excellent agreement, and quantum mechanical calculations agree with the storage ring results quantitatively for the thermal rate constant, <u>if not in all details</u> concerning the cross section. <u>The recombination mechanism is understood</u>. A direct consequence of this progress is that the cosmic-ray ionization rate in diffuse clouds must be shifted upwards to a value larger than $10^{16}s^{-1}$ "

2008, new improved calculations



FIG. 5. (Color online) The present theoretical thermal rate coefficient for dissociative recombination of H_3^+ is compared with the experimental rate coefficient deduced from the storage ring experiment of McCall and co-workers (Refs. 9 and 10).

In the abstract of [Petrignani 2010] it is written:

"... A systematic experimental assessment of <u>heating effects</u> is performed which, together with a survey of other <u>recent storage-ring data</u>, <u>suggests that the present rotationally cool rate-</u> coefficient measurement was performed at 380 (+50 -130) K and that this is the lowest rotational temperature so far realized in storage-ring rate-coefficient measurements on H3+....

.... Unfortunately the experiments on storage rings were stopped $\ldots \otimes \ldots \otimes \ldots \otimes \ldots$













THEORETICAL FRAMEWORK

The states involved: exemple for He_2^+/He_2 system



I. Schneider, et al., DR2004 Mosbach

Electron - Ion Collision

Electron collisions with H₂⁺



