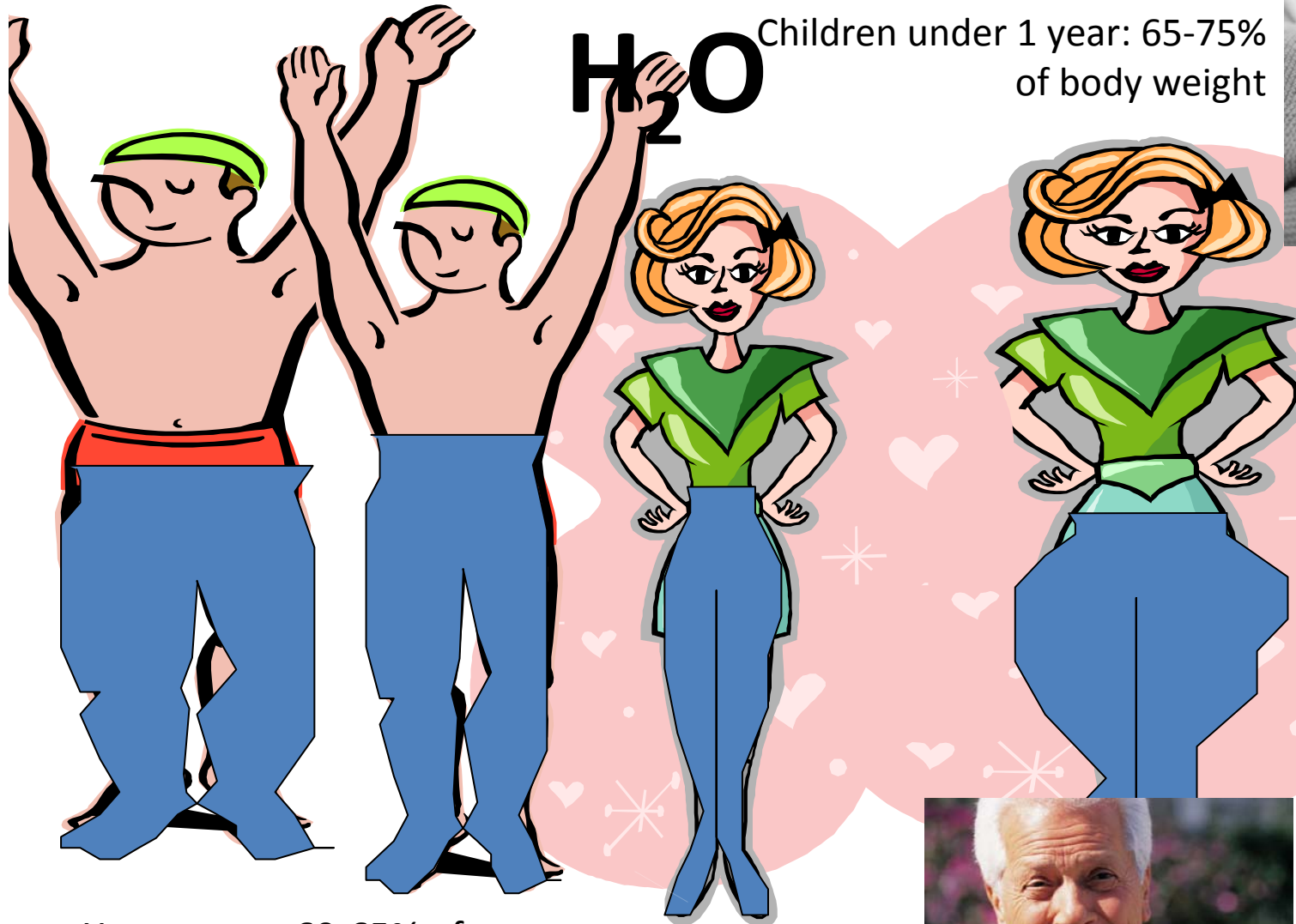


Disturbances of water and mineral balance. Edema

Water Balance

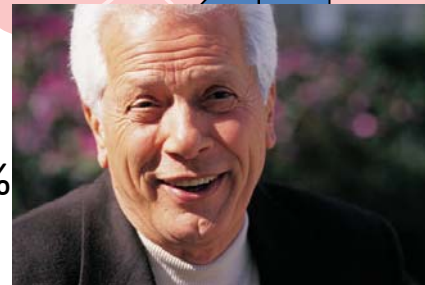


Children under 1 year: 65-75% of body weight



Young men: 60-65% of body weight

Young women: 50-55% of body weight



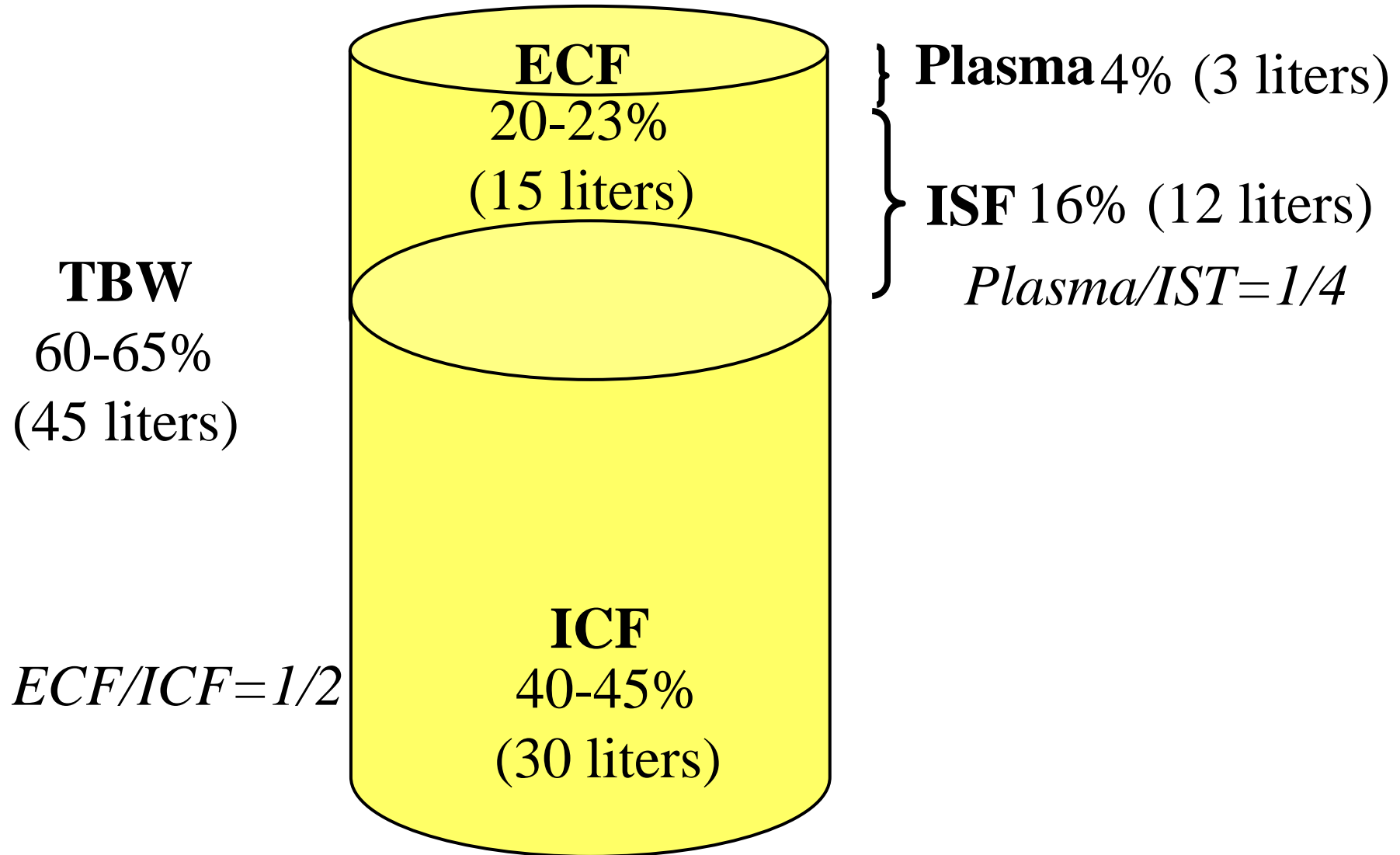
A man over 60 years: 50% of body weight

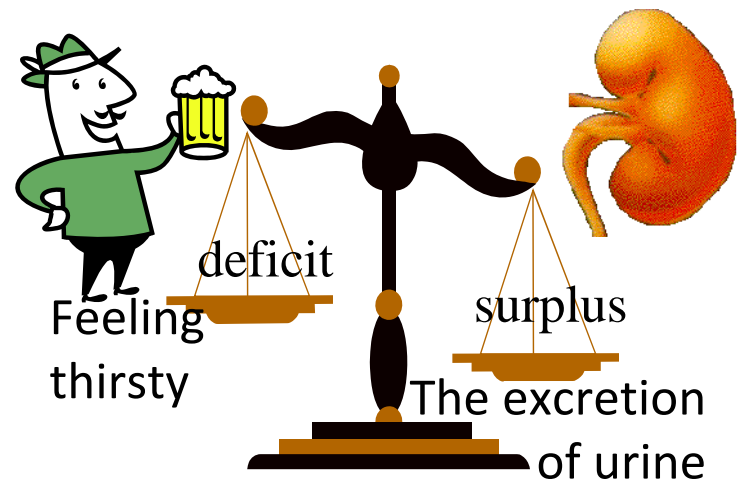
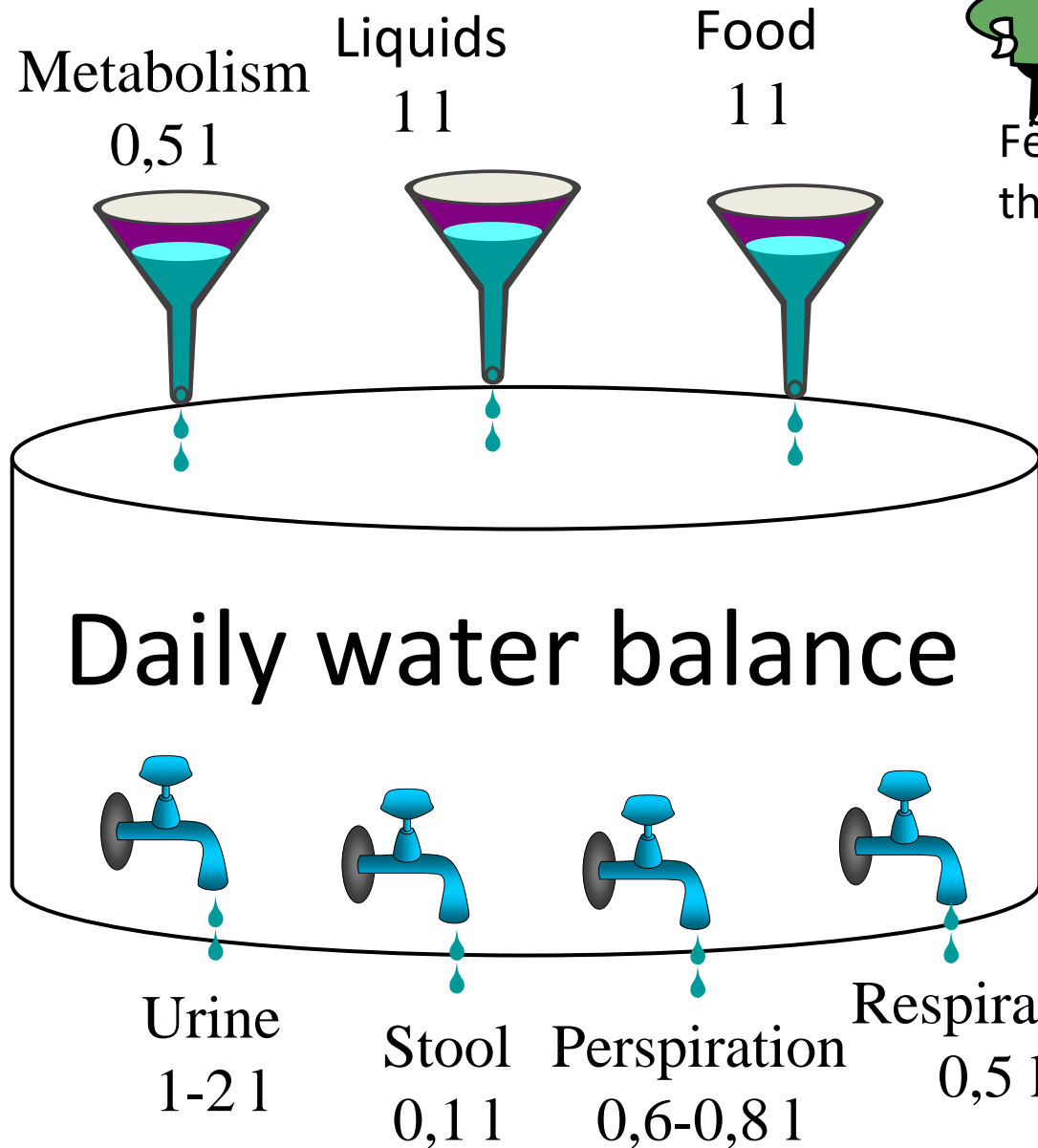


Women over 60 years: 45% of body weight

In obese individuals, the water forms a smaller percentage than in thin

Water content in the body





Major ions

ECF (extracellular fluid)

4,3 mmol/l K^+

140 mmol/l Na^+

104 mmol/l Cl^-

24-27 mmol/l HCO_3^-

ICF (intracellular fluid)

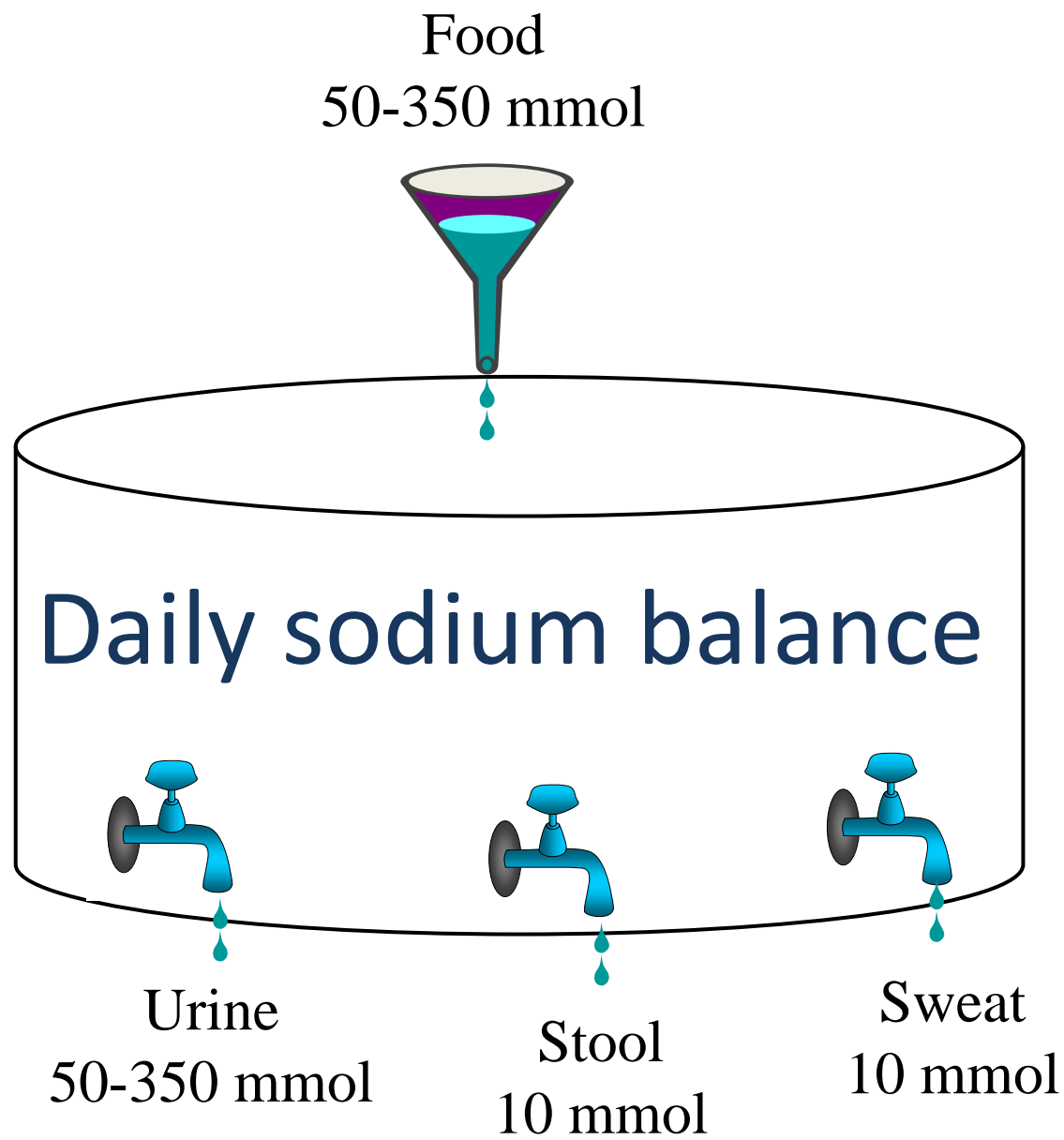
140 mmol/l K^+

12 mmol/l Na^+

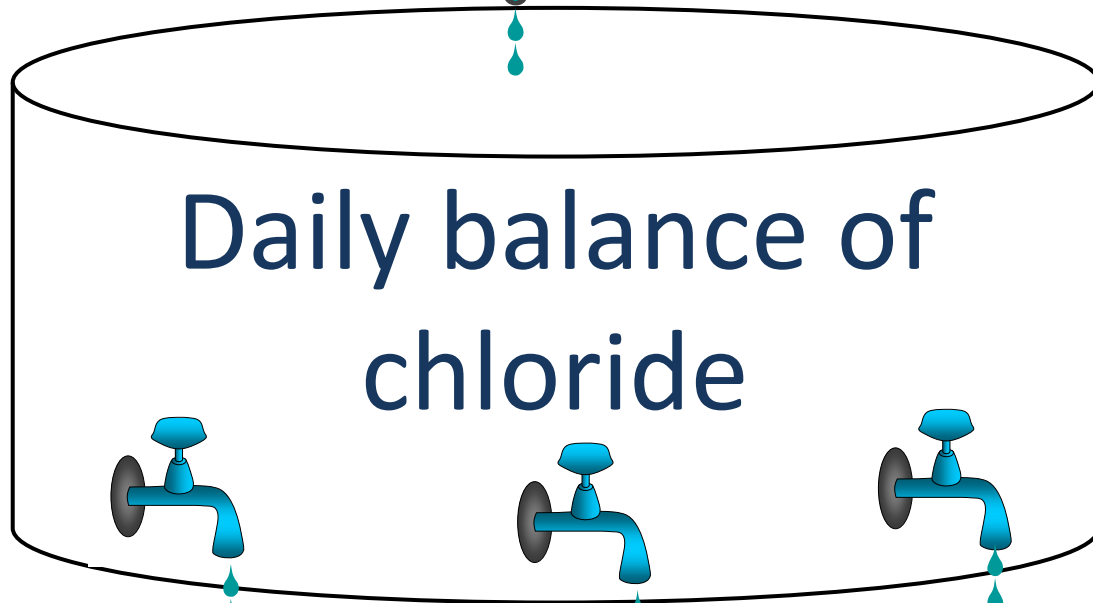
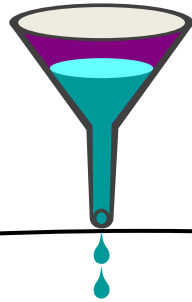
3 mmol/l Cl^-

10 mmol/l HCO_3^-

The balance of sodium and chloride



Food
50-350 mmol



Urine
50-350 mmol

Stool
10 mmol

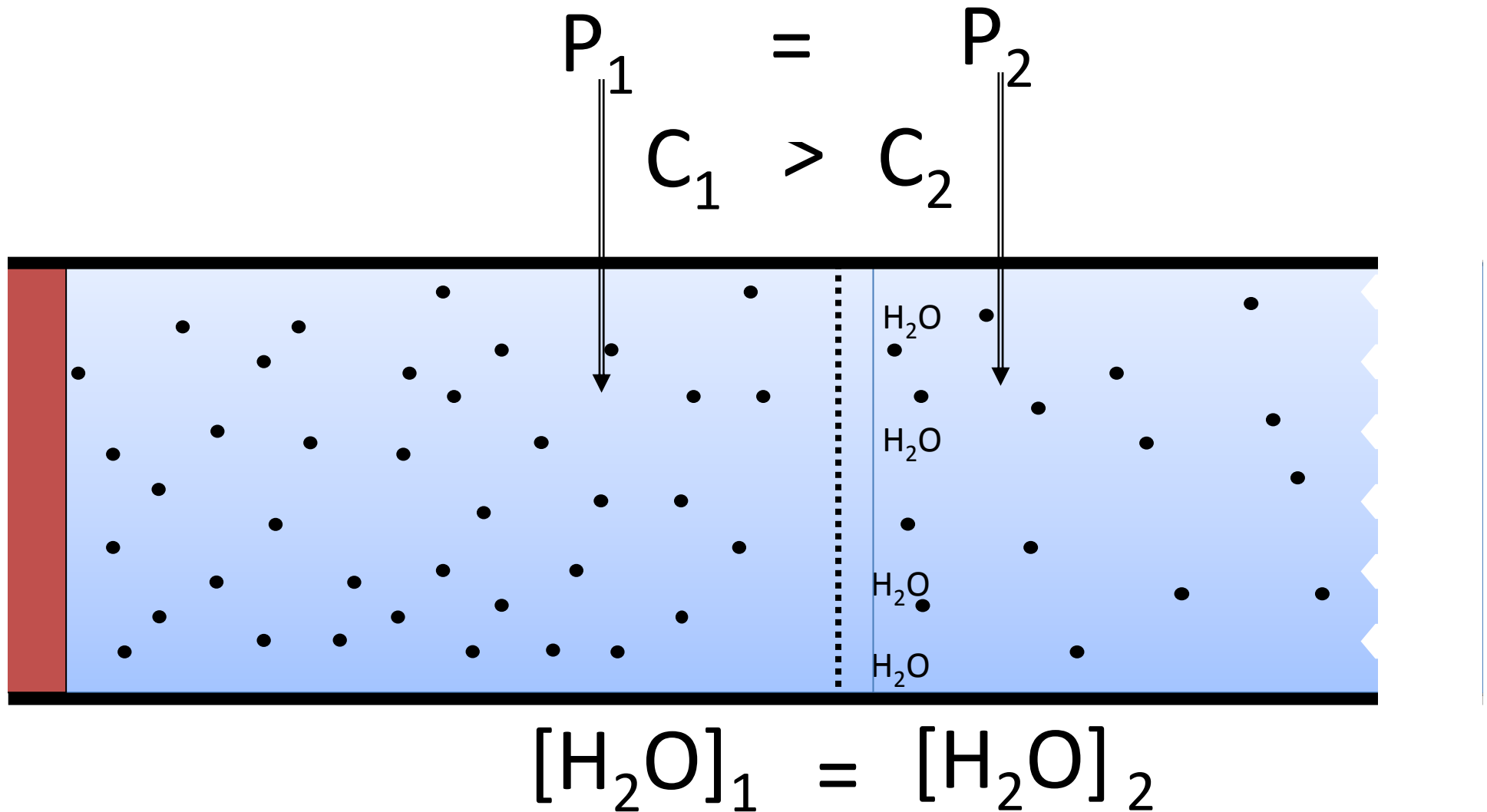
Sweat
10 mmol



Osmotic balance in body fluids

Osmotic pressure

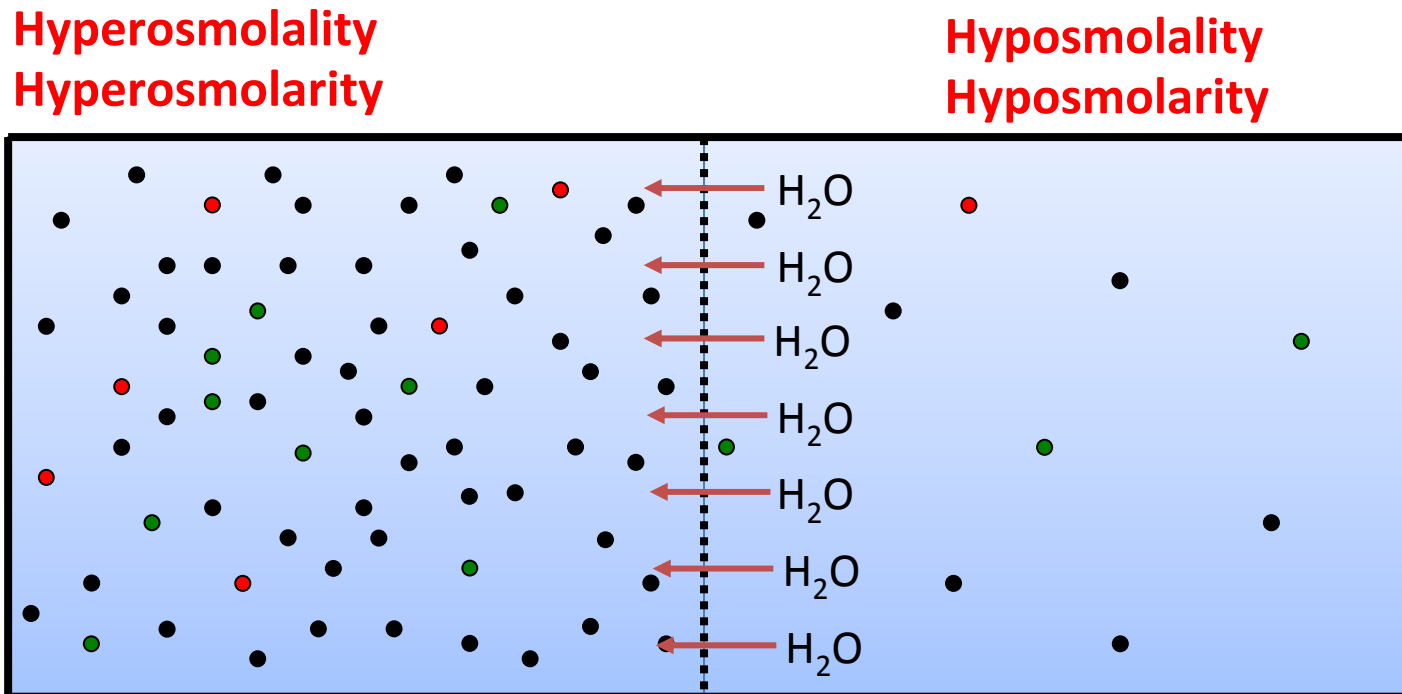
is related to the concentration of all dissolved particles



Osmotic pressure

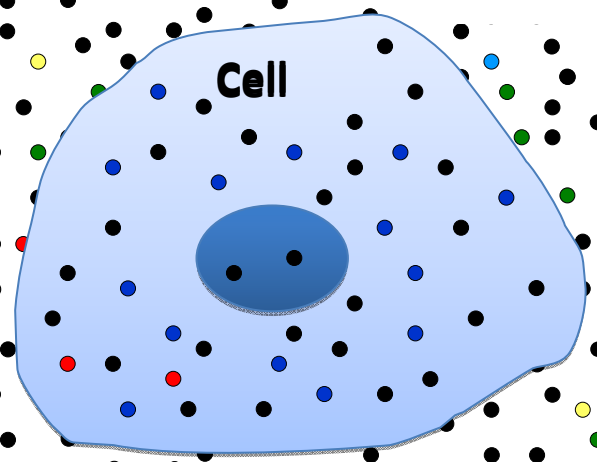
*is related to the concentration of dissolved particles relative to the weight of the solvent: **osmolality** (mmol/kg solvent)*

*the volume of solution : **osmolarity** (mmol/l solution).*



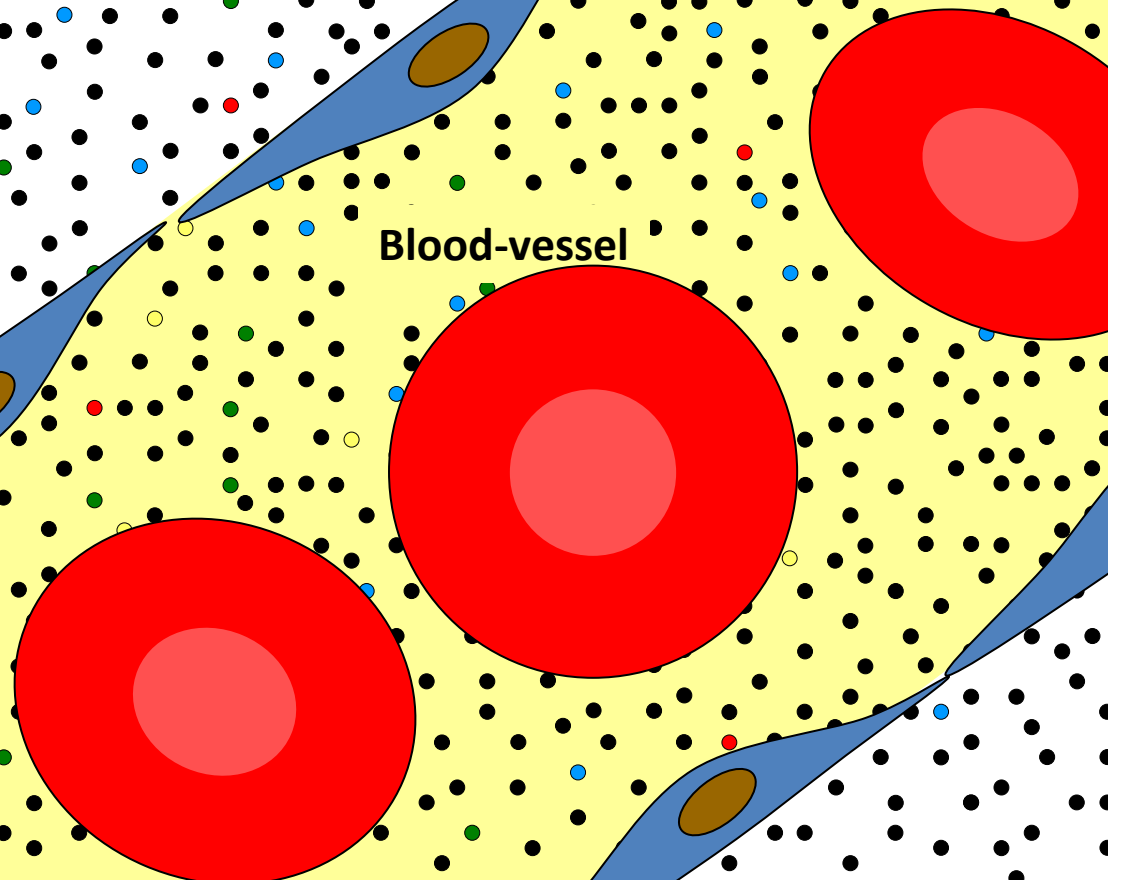
Isotonic environment

290 ± 10 mmol/l

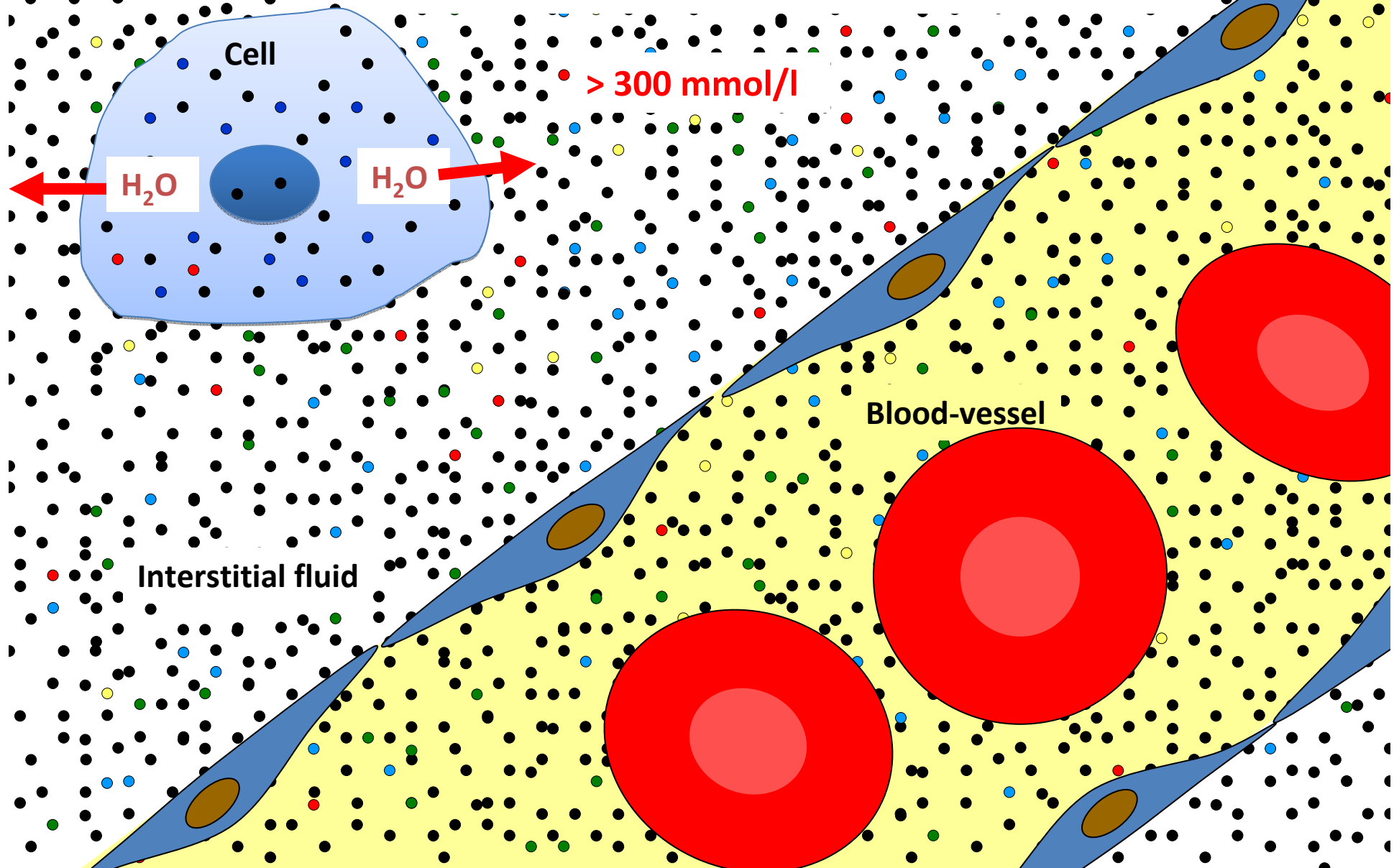


Interstitial fluid

Blood-vessel

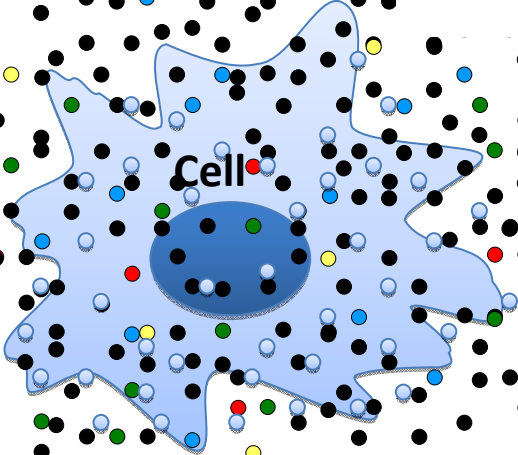


Hypertonic environment



Hypertonic environment

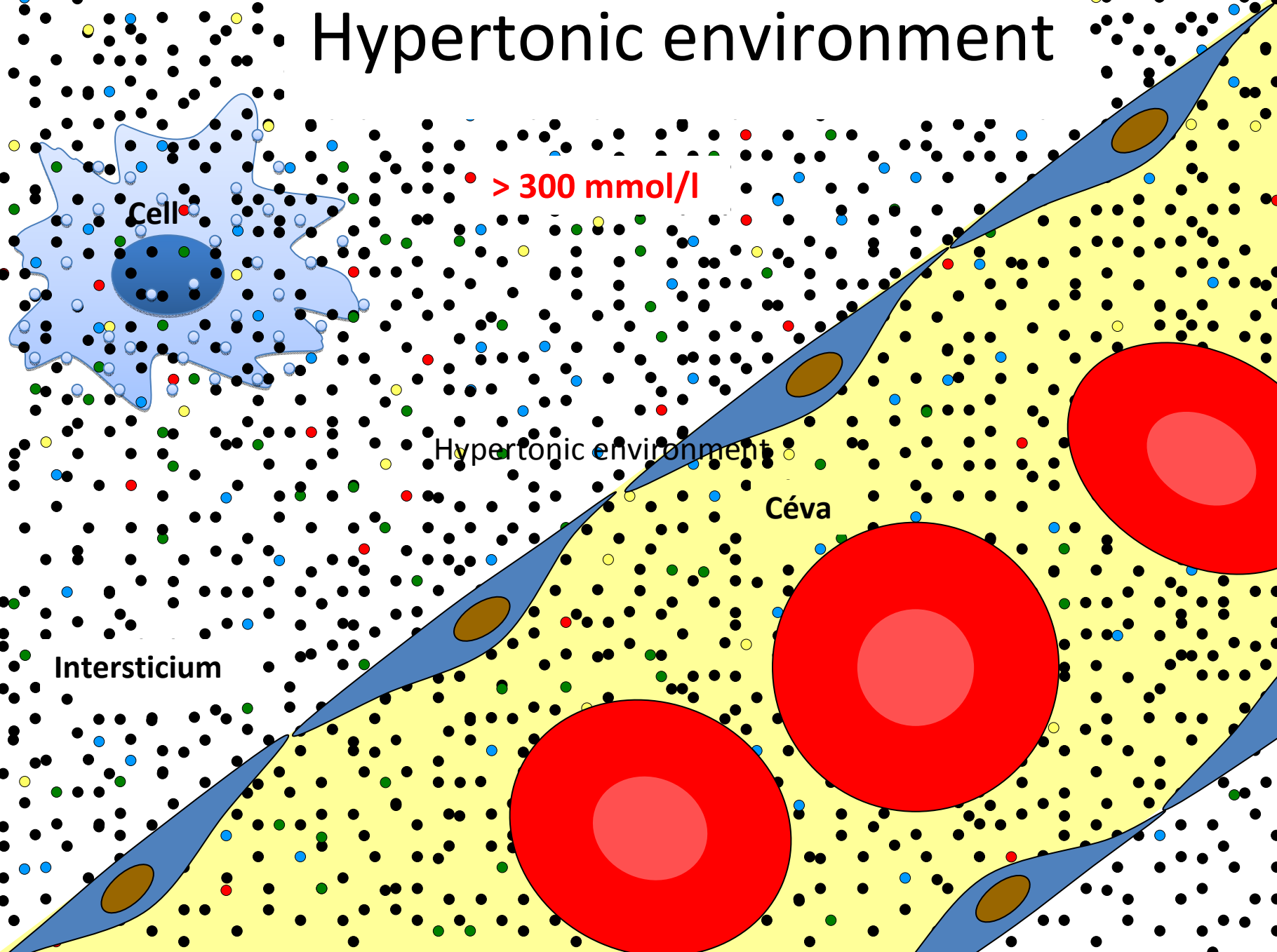
> 300 mmol/l



Hypertonic environment

Céva

Intersticium



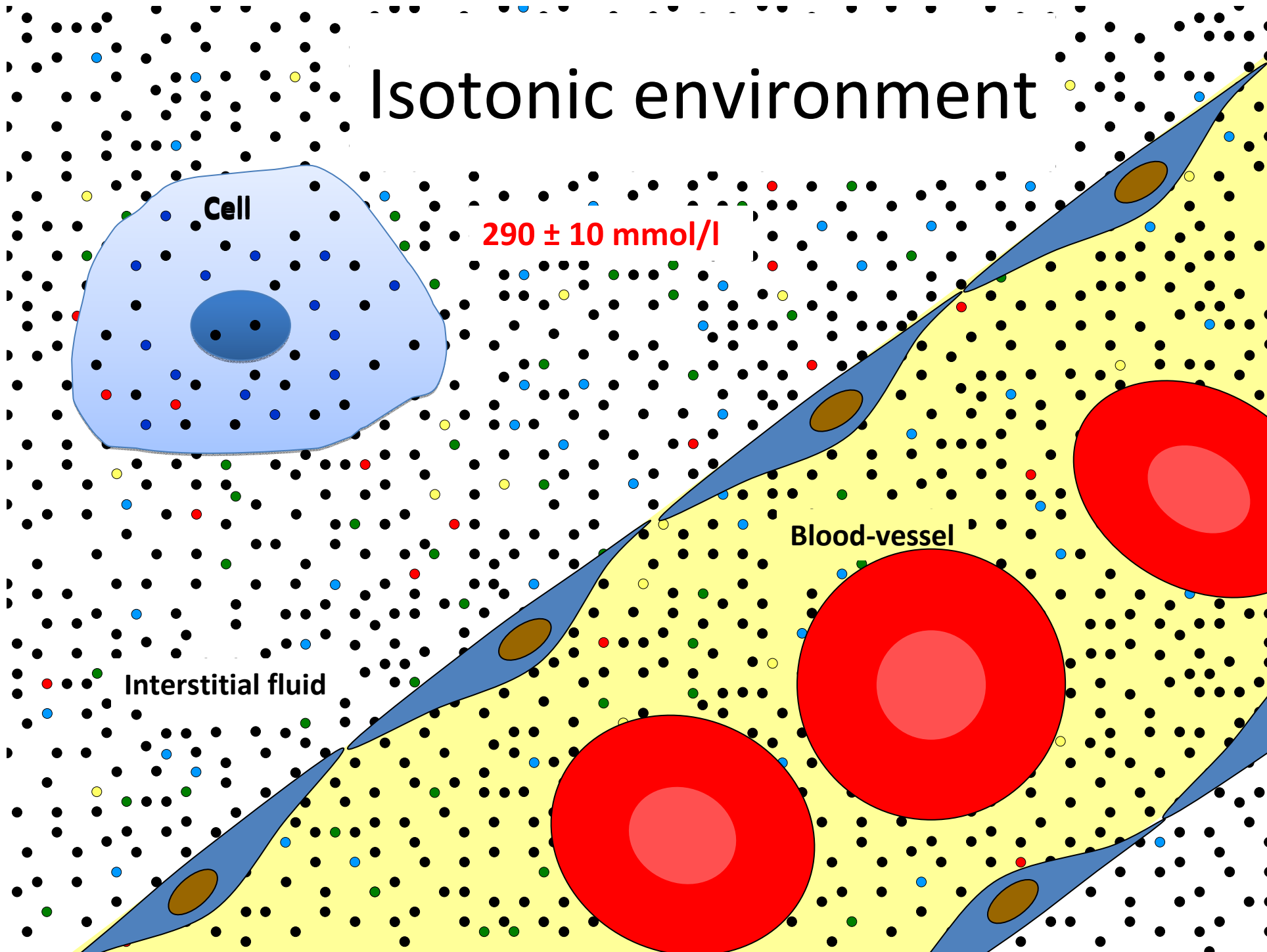
Isotonic environment

$290 \pm 10 \text{ mmol/l}$

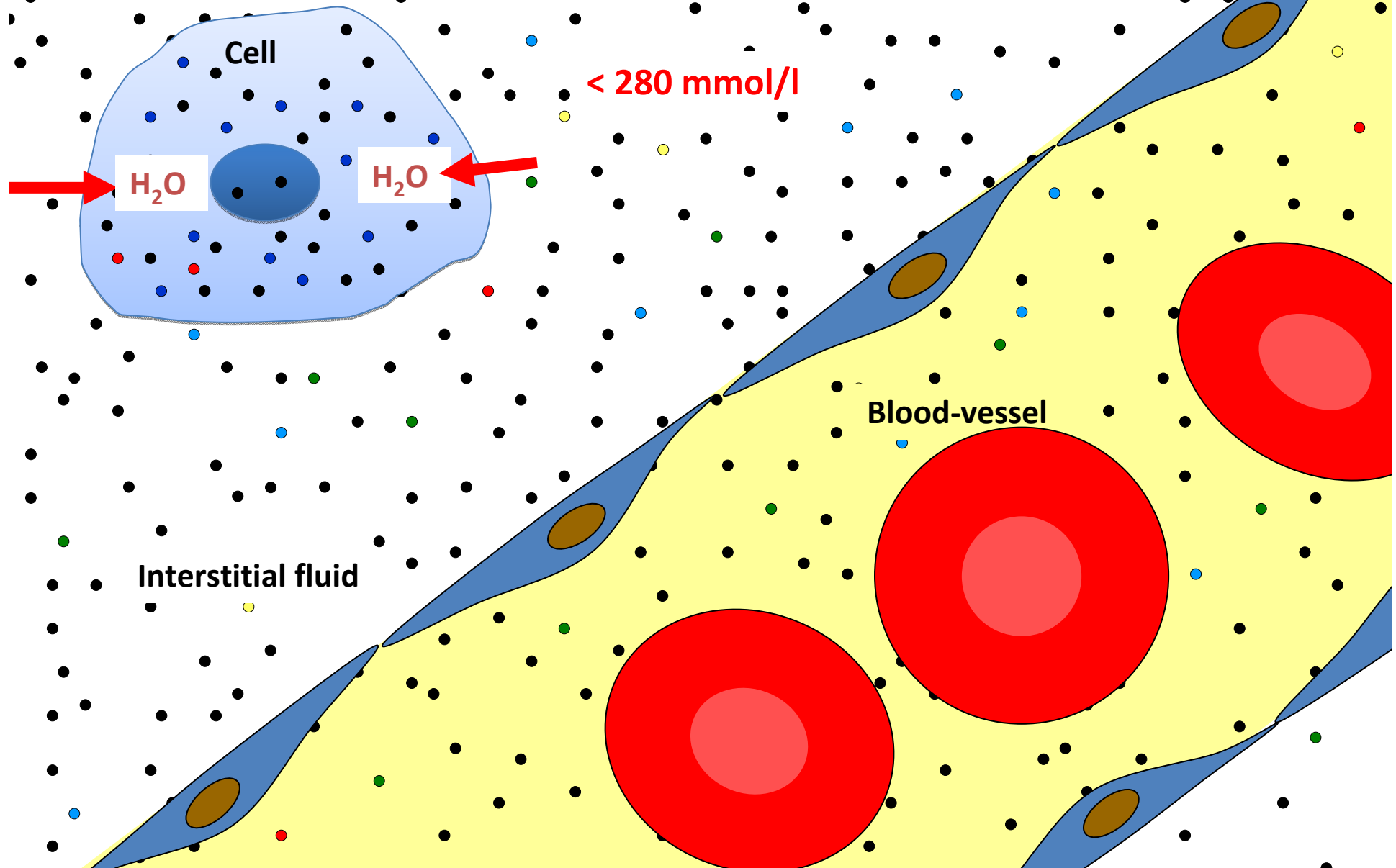
Cell

Interstitial fluid

Blood-vessel

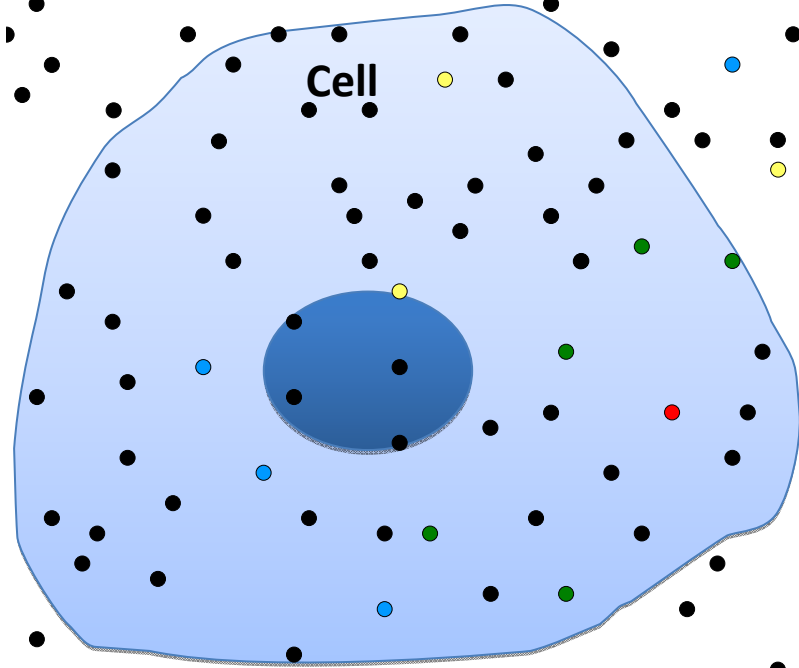


Hypotonic environment



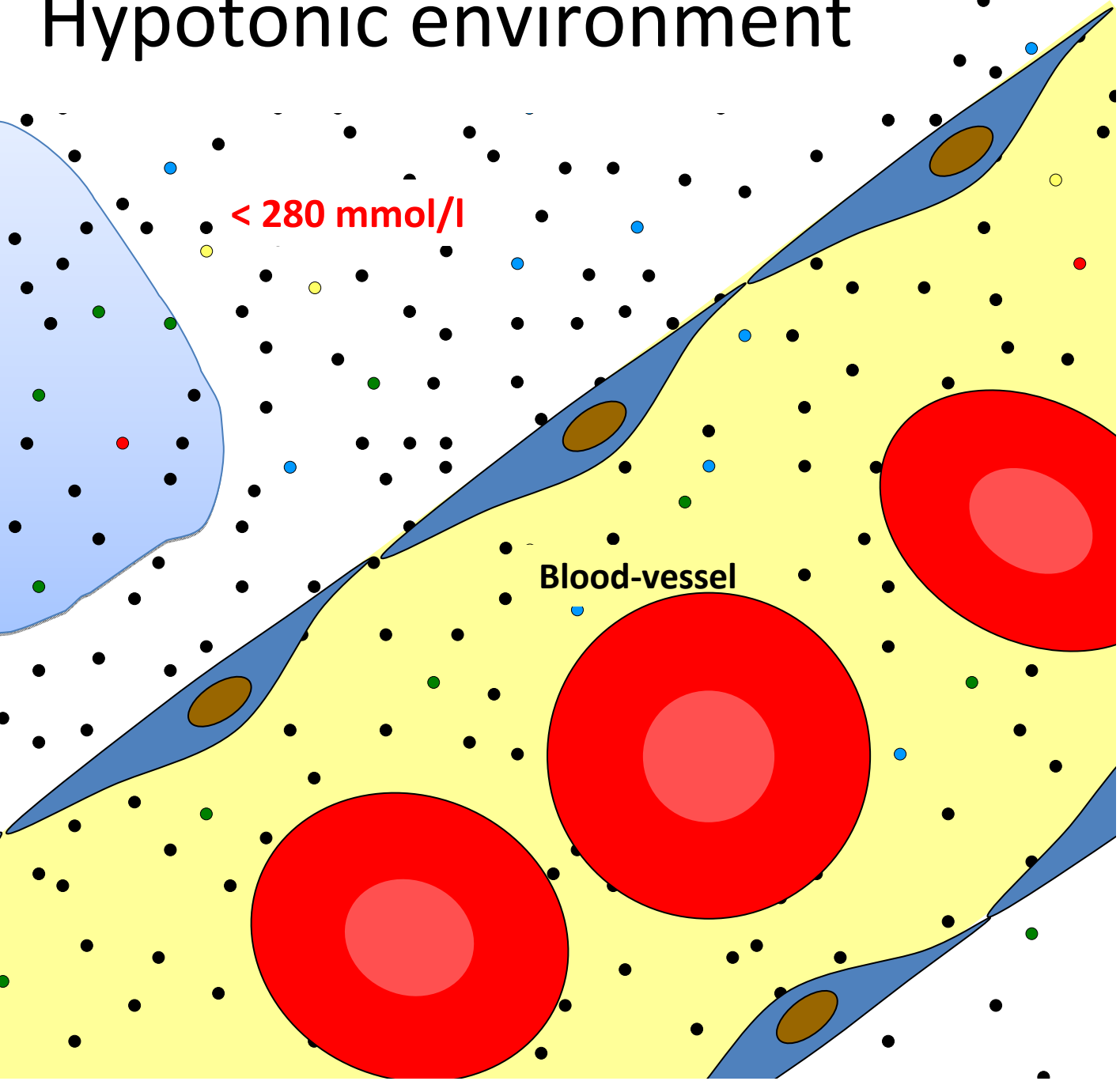
Hypotonic environment

< 280 mmol/l



Cell

Interstitial fluid

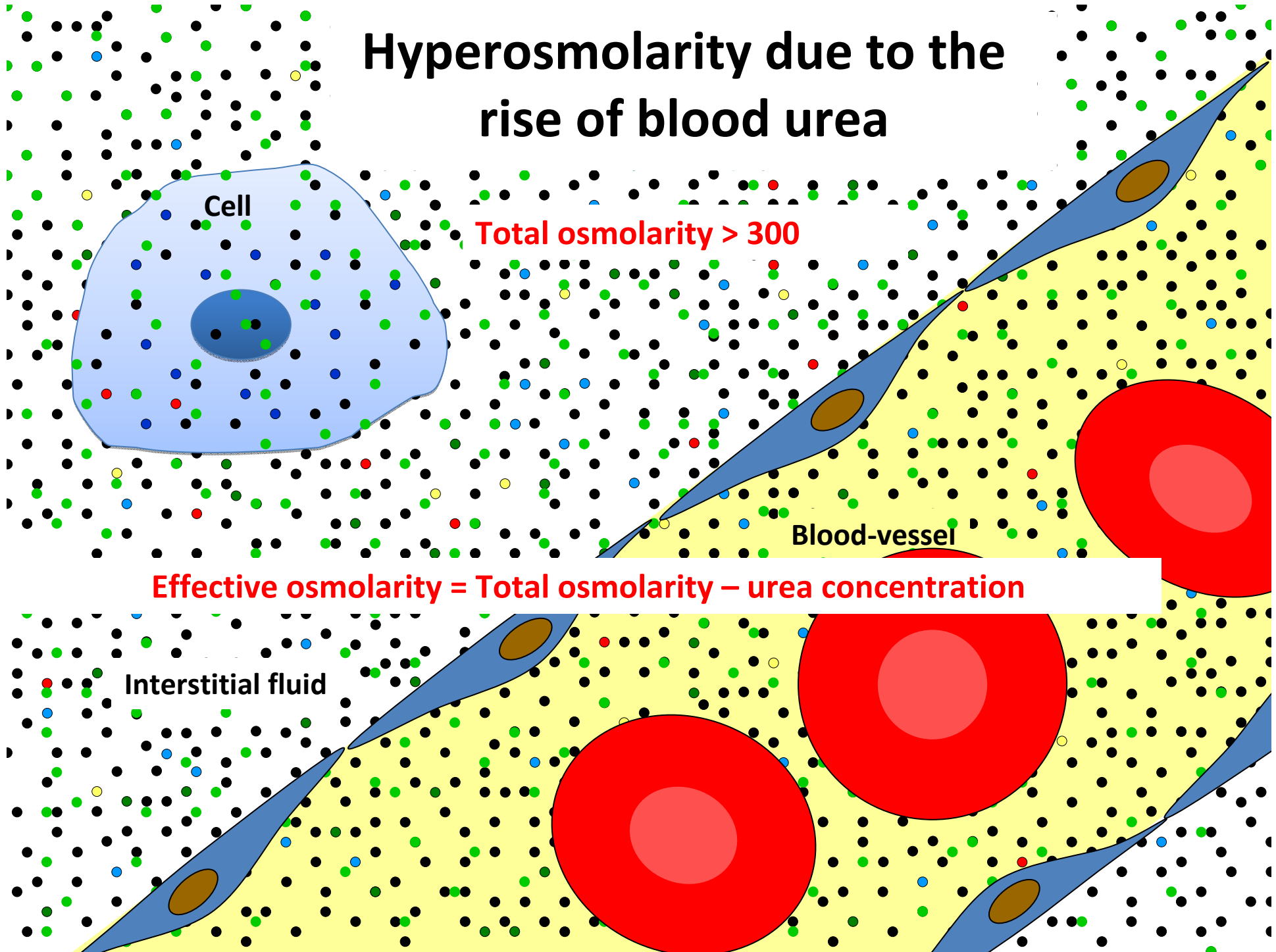


Blood-vessel

Hyperosmolarity due to the rise of blood urea

Total osmolarity > 300

Effective osmolarity = Total osmolarity – urea concentration

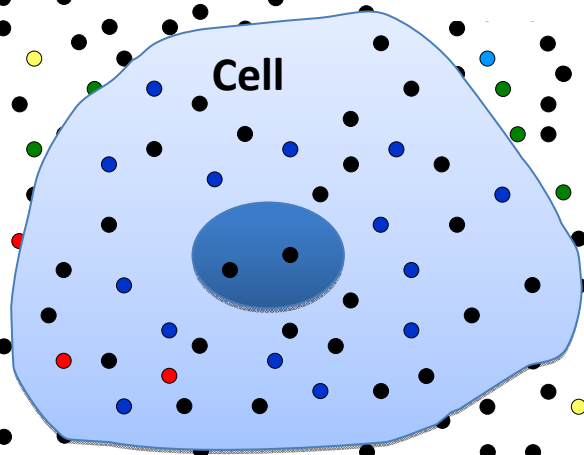


The balance of the capillary and its disorders

(water and soluti movement between plasma and interstitium)

Isotonic environment

290 ± 10 mmol/l



Interstitial fluid

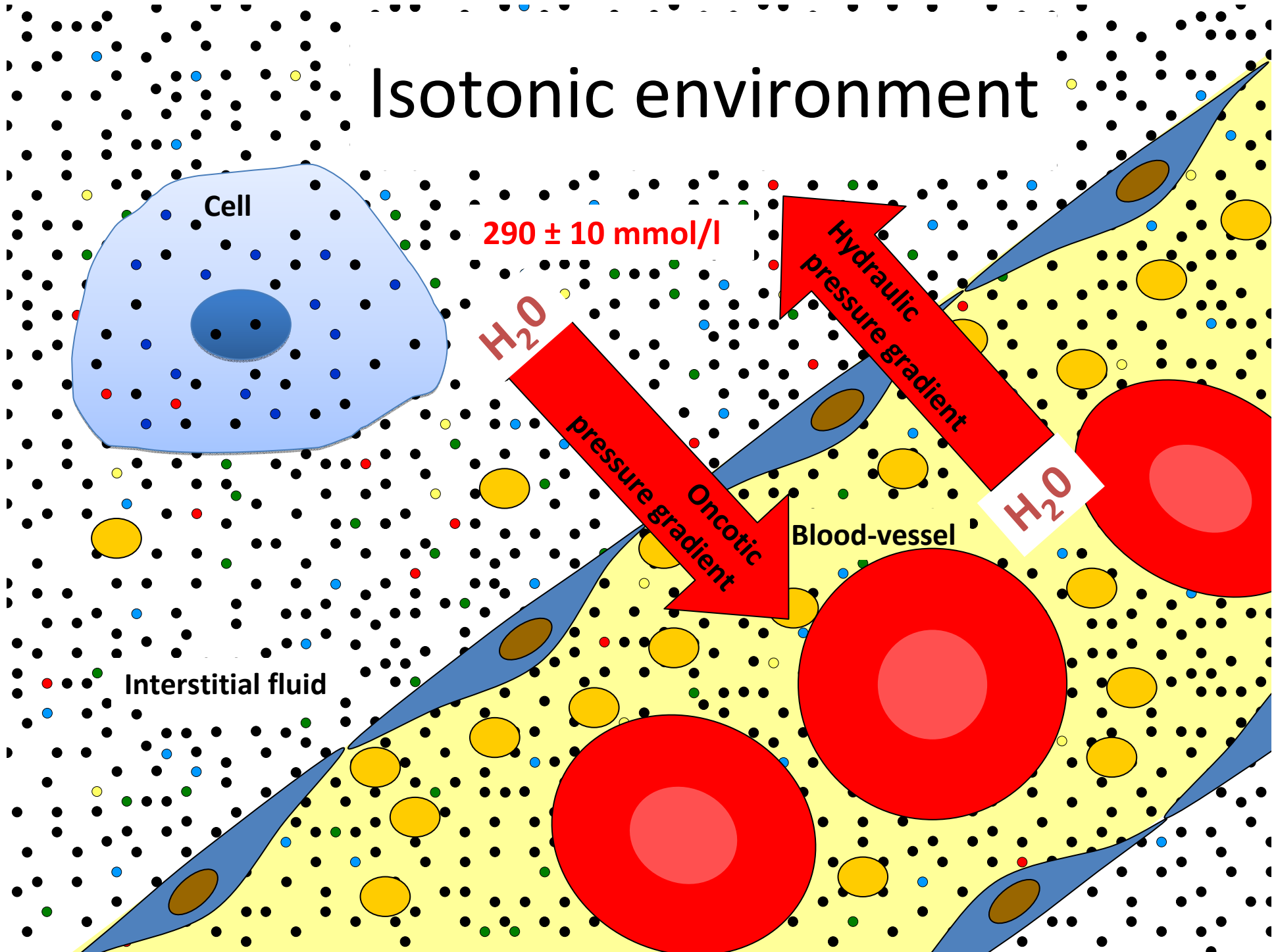
H₂O

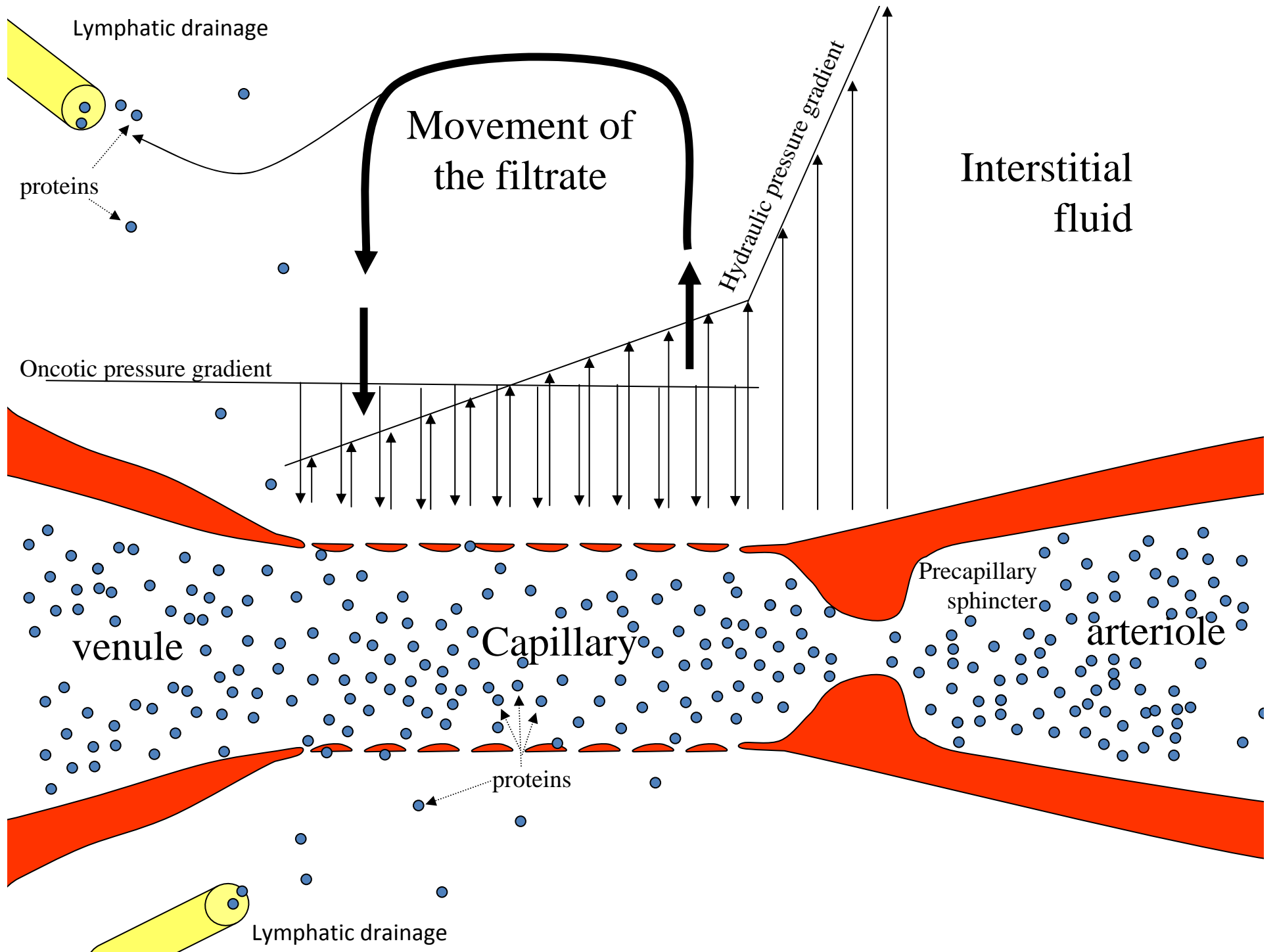
Oncotic pressure gradient

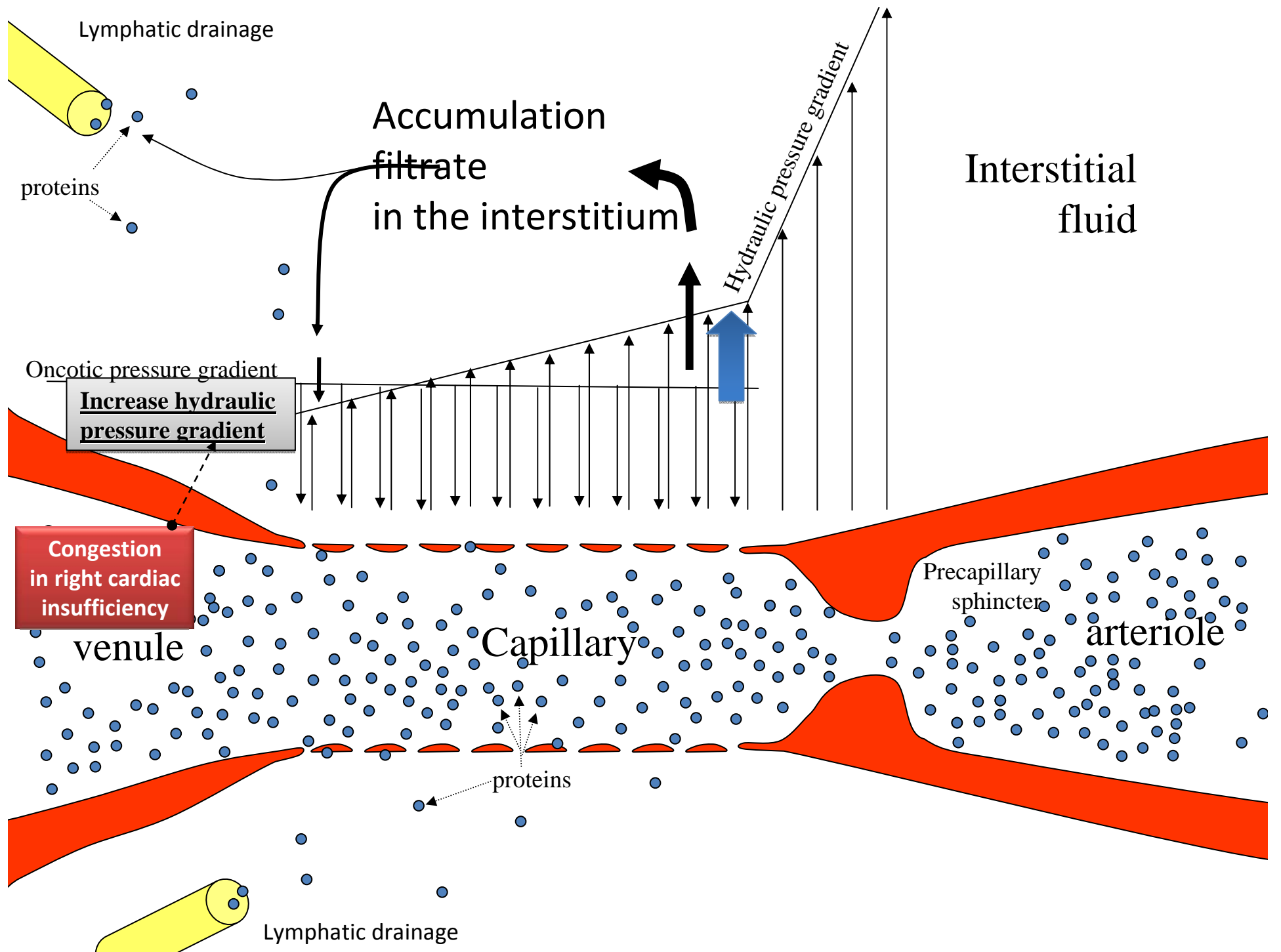
Blood-vessel

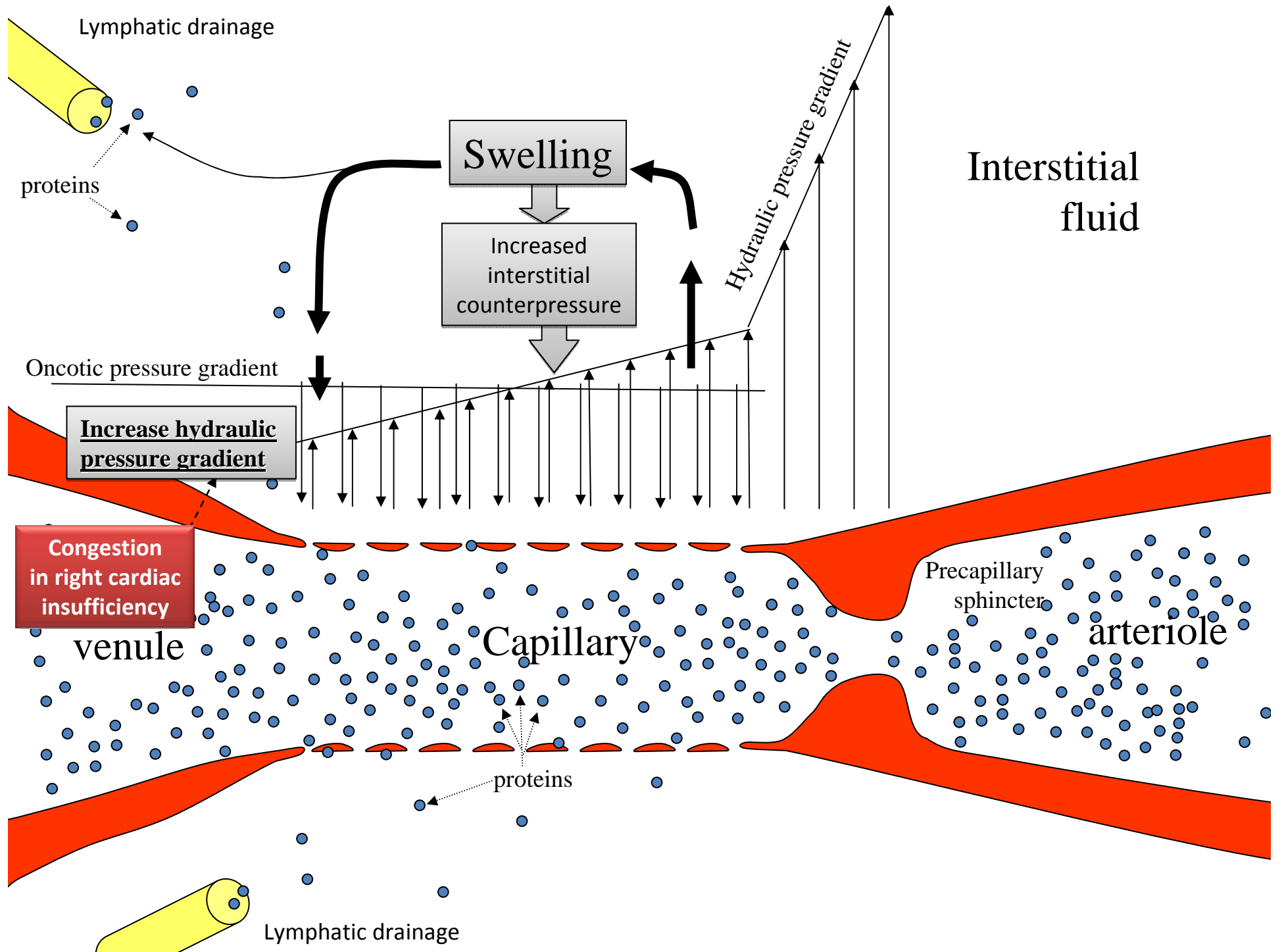
Hydraulic pressure gradient

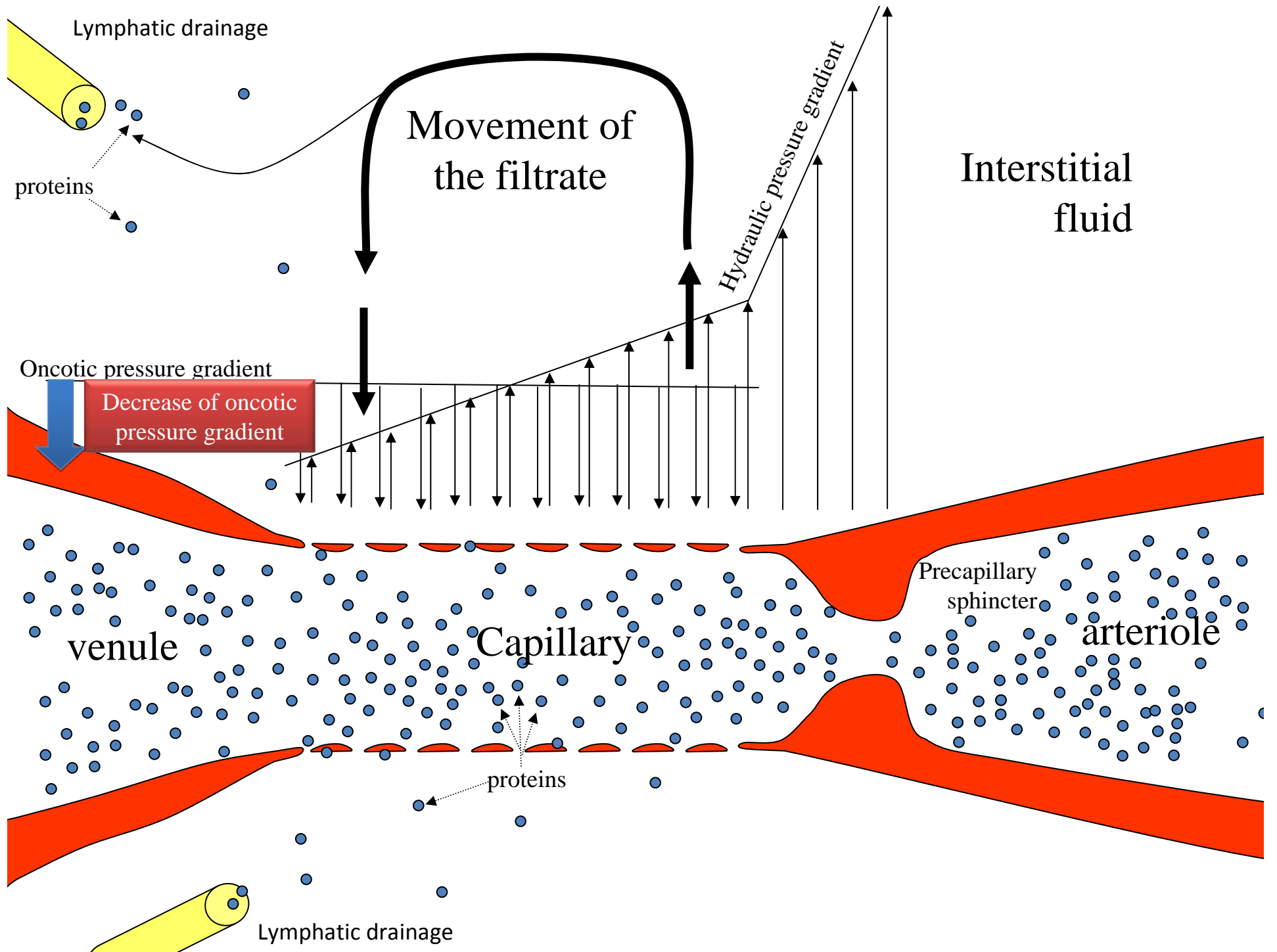
H₂O

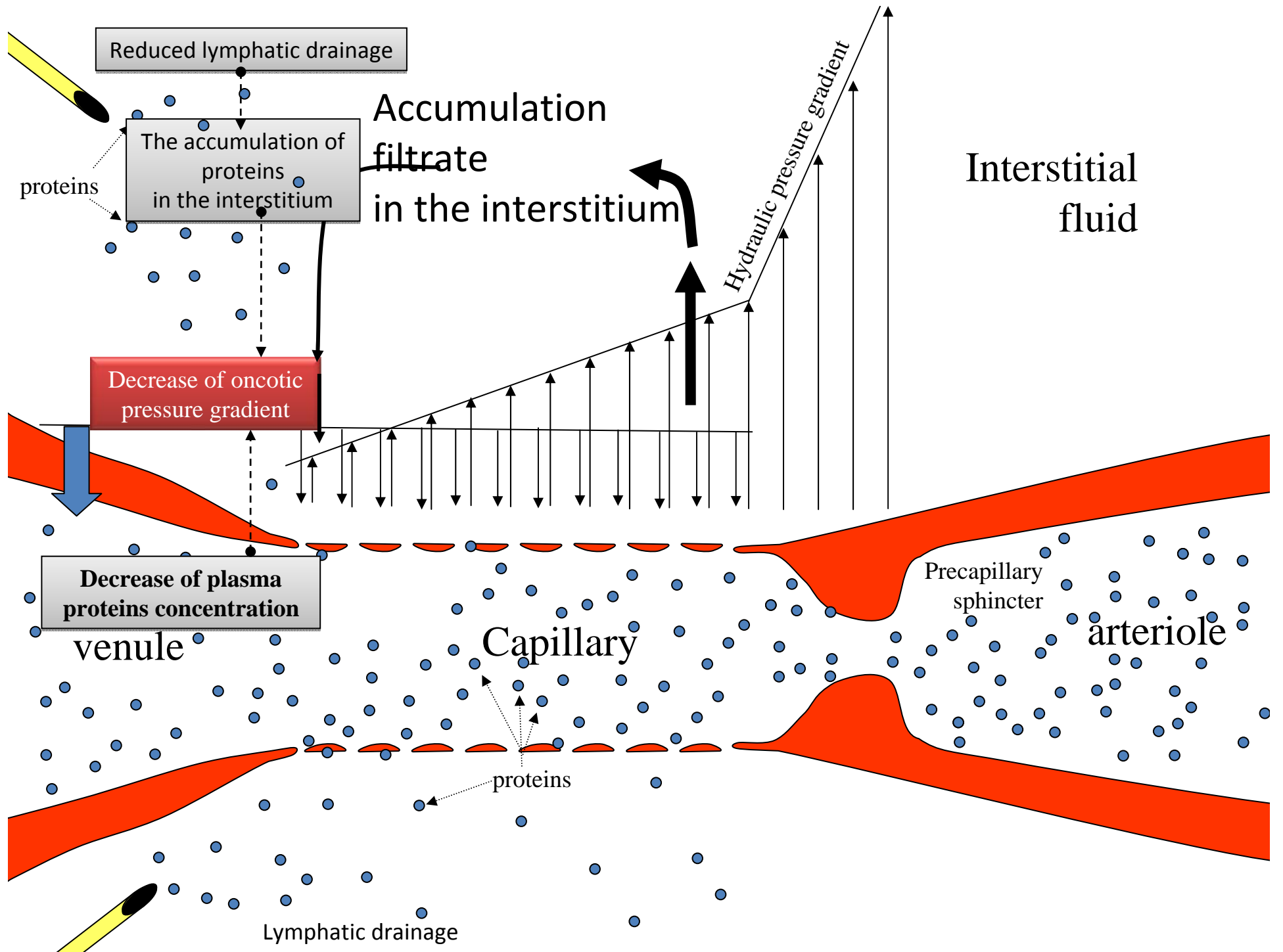


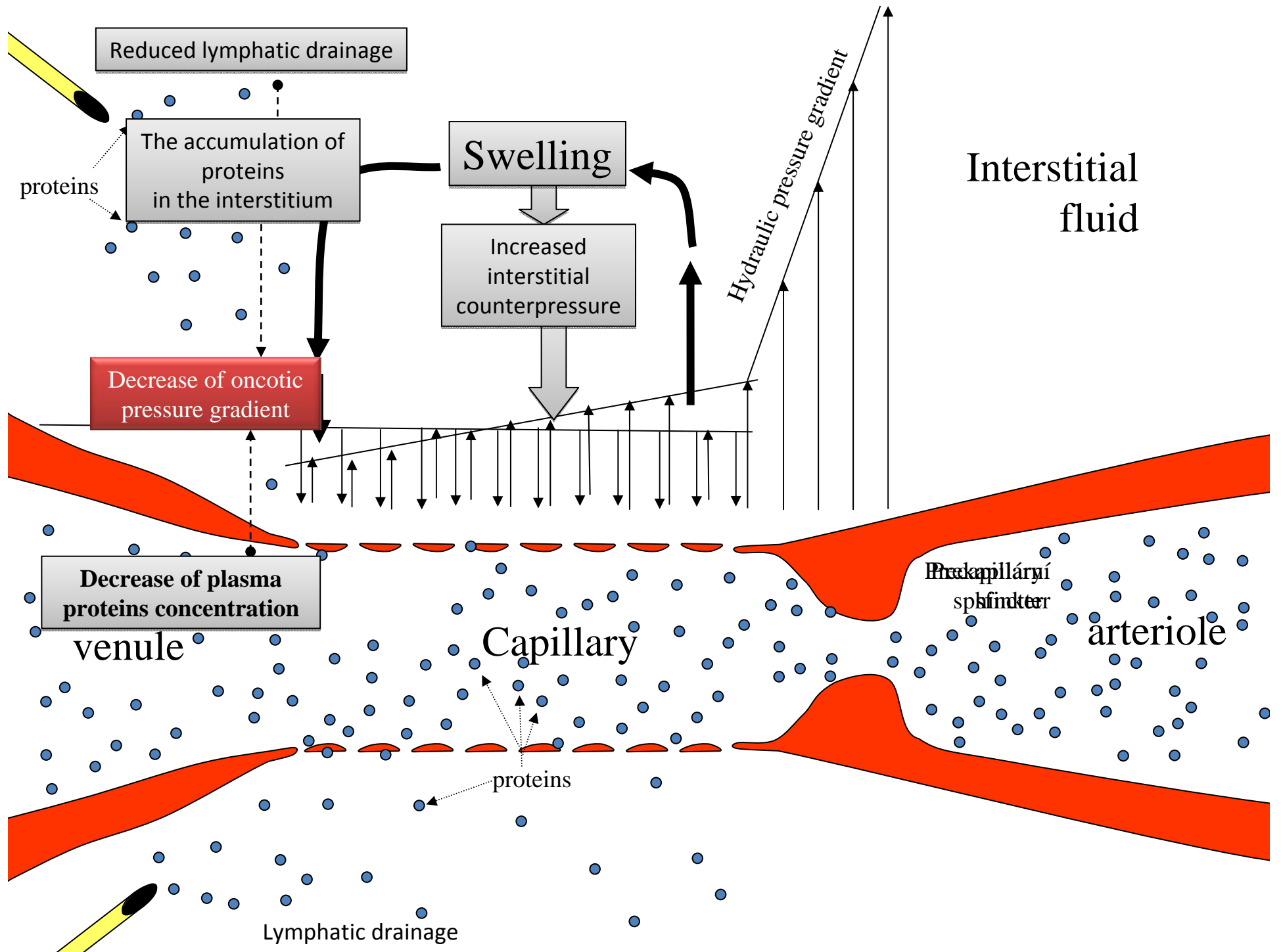


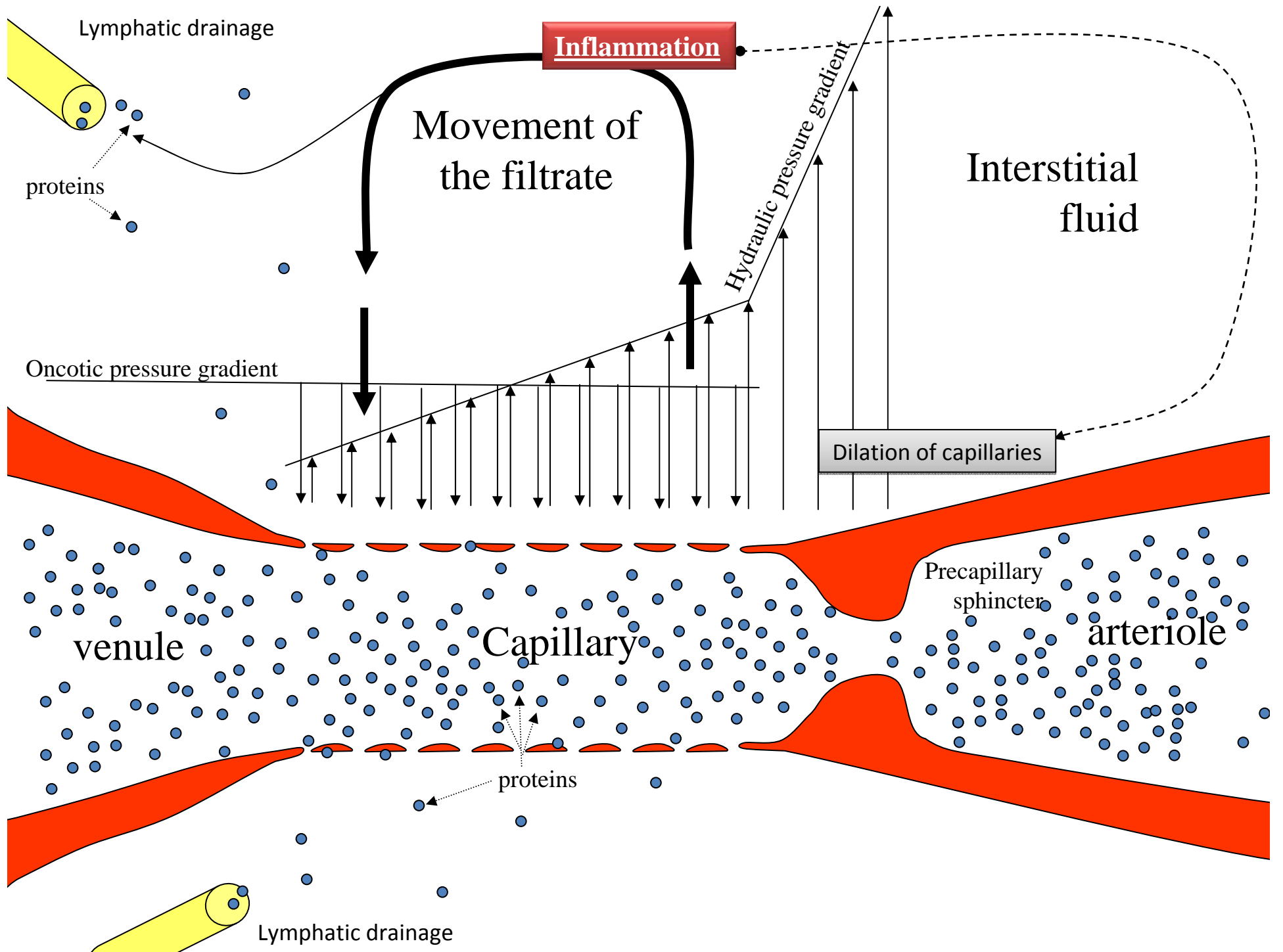


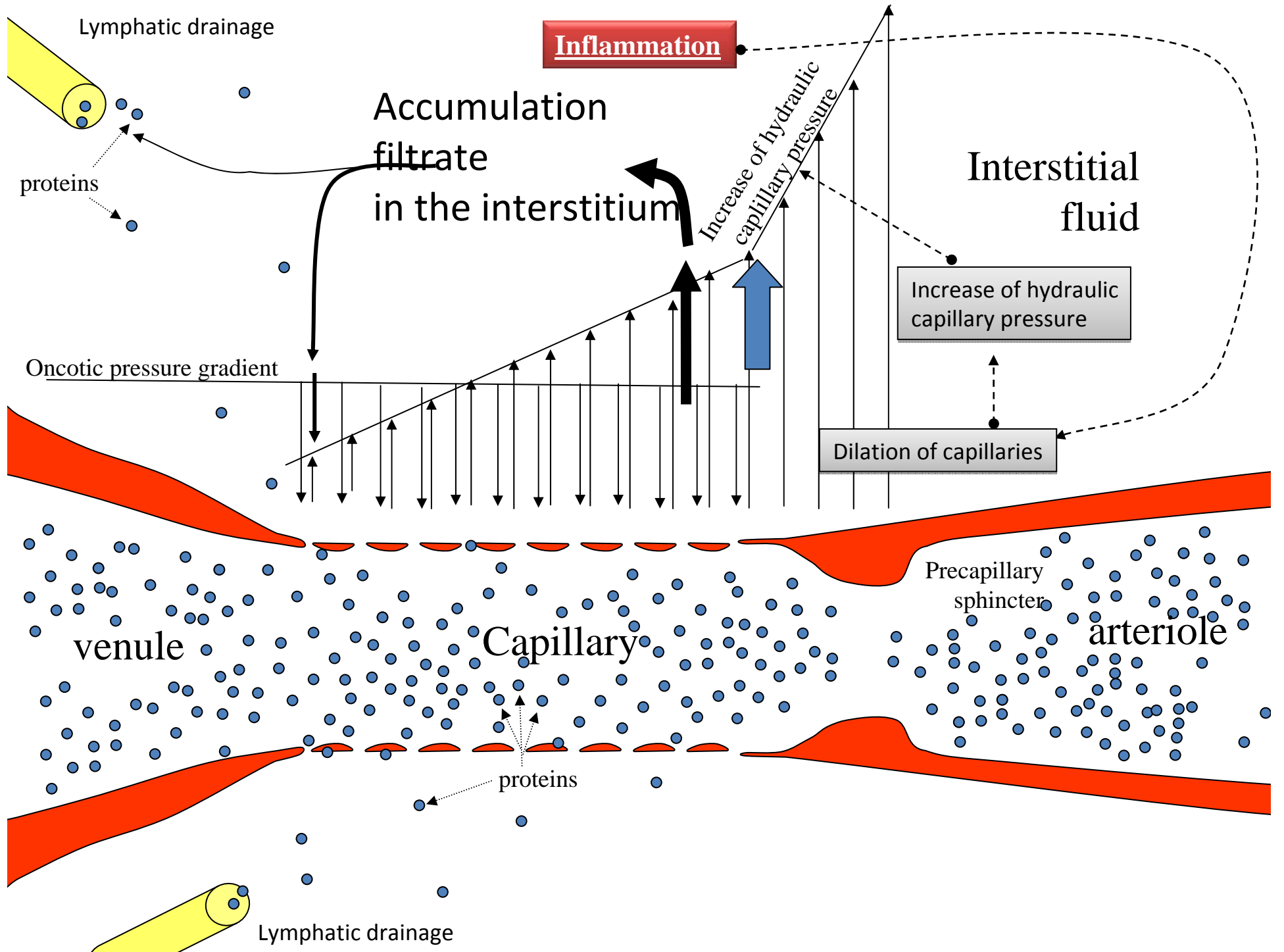


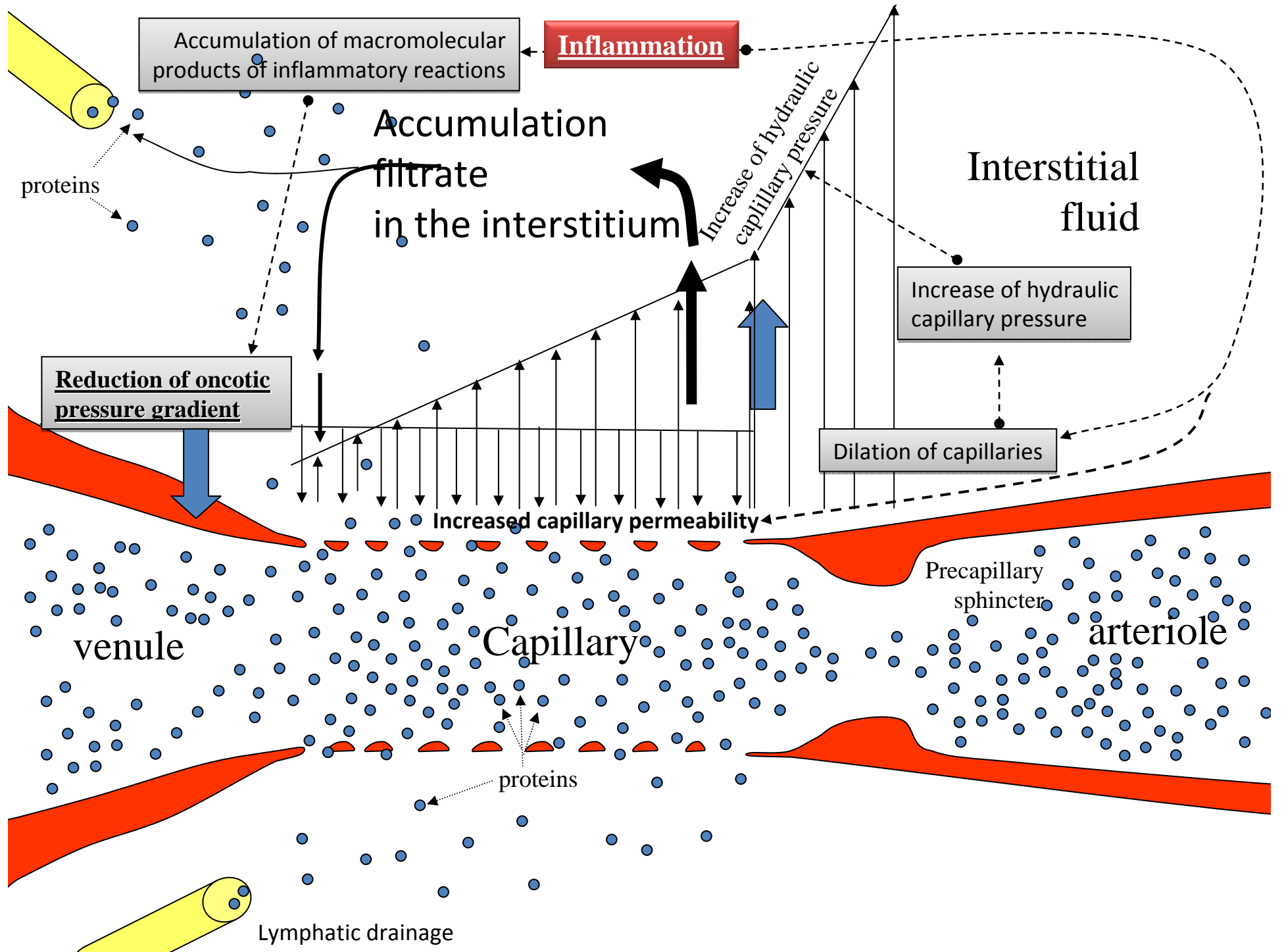


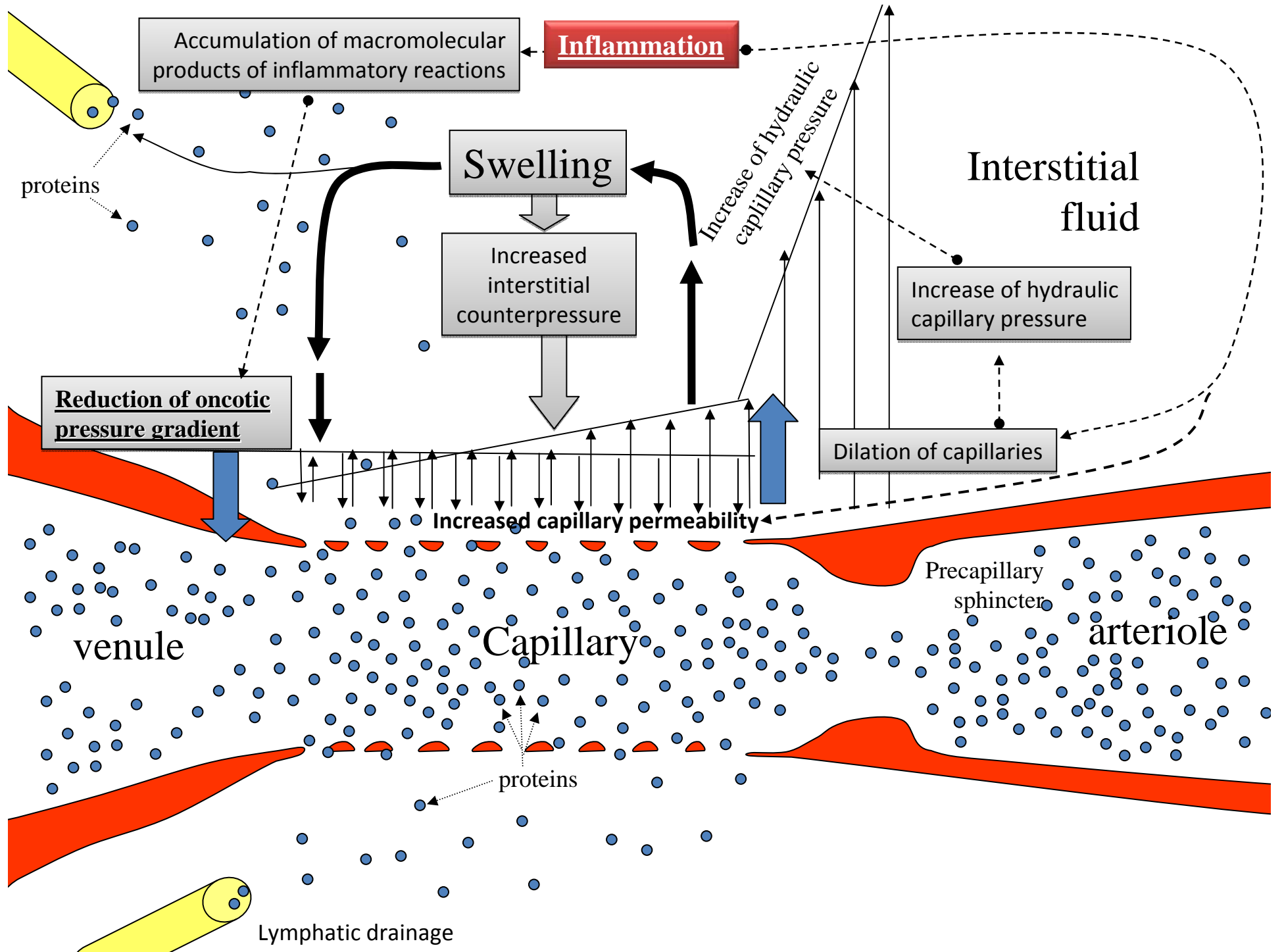














Blood volume [l]



Blood volume [l]: 5
 Resist. blood vol.: 0.6 l

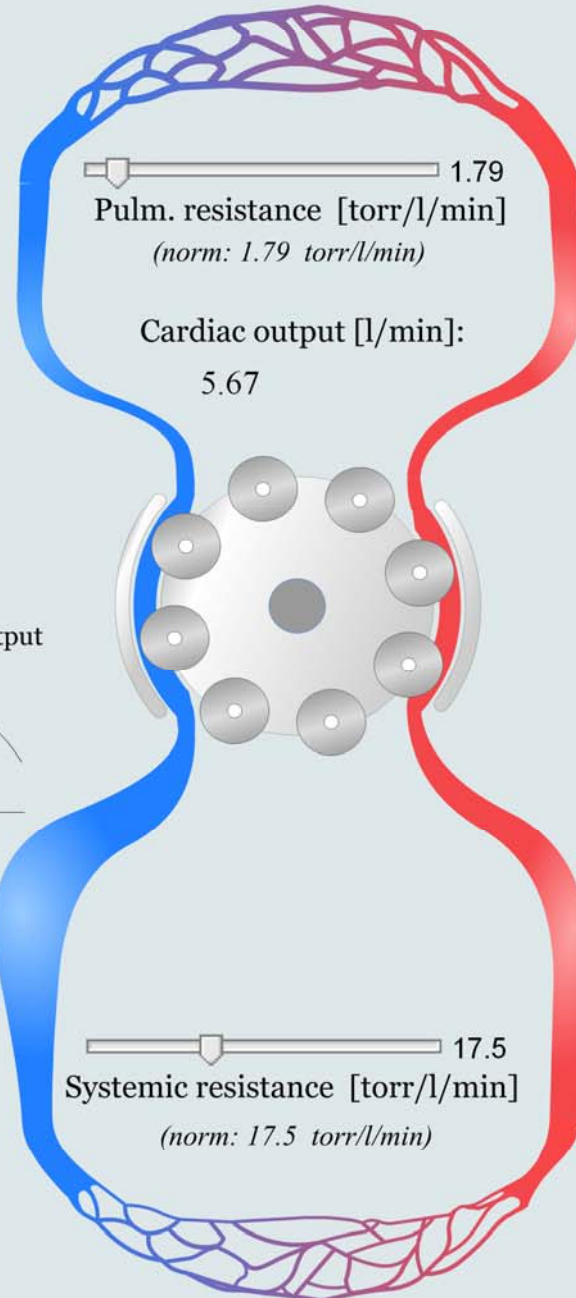


(norm All)

Pulmonary veins

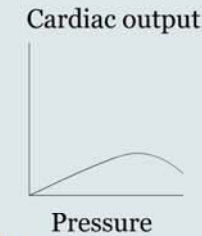


Press. [torr] 5.06
 Volume [l] 0.4



Slider: 1.79
 Pulm. resistance [torr/l/min]
 (norm: 1.79 torr/l/min)

Cardiac output [l/min]:
 5.67



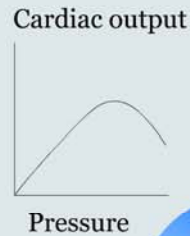
Left heart

Slider: 1.12
 Slope of Starling curve [l/min/torr]
 (norm: 1.12 l/min/torr)

Pulmonary art.

Slider: 6.67
 Compliance [ml/torr]
 (norm: 6.67 ml/torr)

Press. [torr] 15.21
 Volume [l] 0.1



Right heart

Slider: 2.85
 Slope of Starling curve [l/min/torr]
 (norm: 2.8 l/min/torr)

Systemic art.

Slider: 10
 Compliance [ml/torr]
 (norm: 10 ml/torr)

Press. [torr] 101.21
 Volume [l] 1.01

Slider: 17.5
 Systemic resistance [torr/l/min]
 (norm: 17.5 torr/l/min)

Systemic veins


Slider: 1750
 Compliance [ml/torr]
 (norm: 1750 ml/torr)

Press. [torr] 1.99
 Volume [l] 3.48

Control volume and osmolarity

Control volume related to regulation of **circulating blood volume** and thus with hemodynamics (particularly with the regulation of arterial pressure)

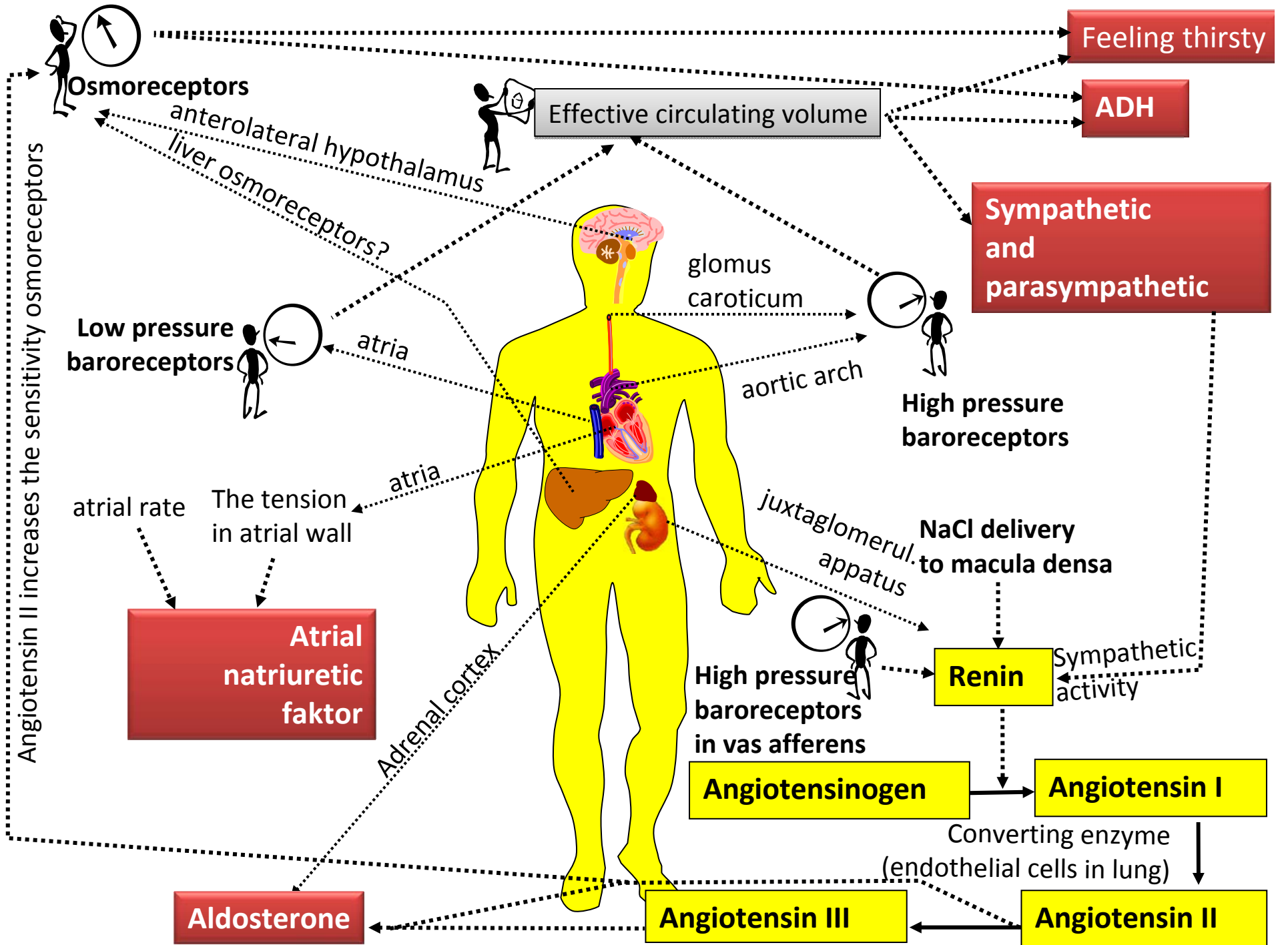
Therefore, we are talking about so-called "**effective circulating volume**"



Volume - is detected by atrial tension in the wall (low-pressure baroreceptors), by tension in the artery wall and in the glomus caroticus the aortic arch (high-receptors)

Osmolarity – is detected by osmoreceptors in the hypothalamus and possibly also liver

Control volume and osmolarity closely related **to regulation of sodium excretion** (sympaticus, aldosterone, ANF) and **control urine osmolarity** (ADH)



Osmoreceptors

anterolateral hypothalamus
liver osmoreceptors?

Effective circulating volume

Feeling thirsty

ADH

Sympathetic and parasympathetic

glomus caroticum

High pressure baroreceptors

aortic arch

Low pressure baroreceptors

atria

atrial rate

The tension in atrial wall

Atrial natriuretic faktor

atria

juxtaglomerular apparatus

NaCl delivery to macula densa

High pressure baroreceptors in vas afferens

Renin

Sympathetic activity

Angiotensinogen

Angiotensin I

Converting enzyme (endothelial cells in lung)

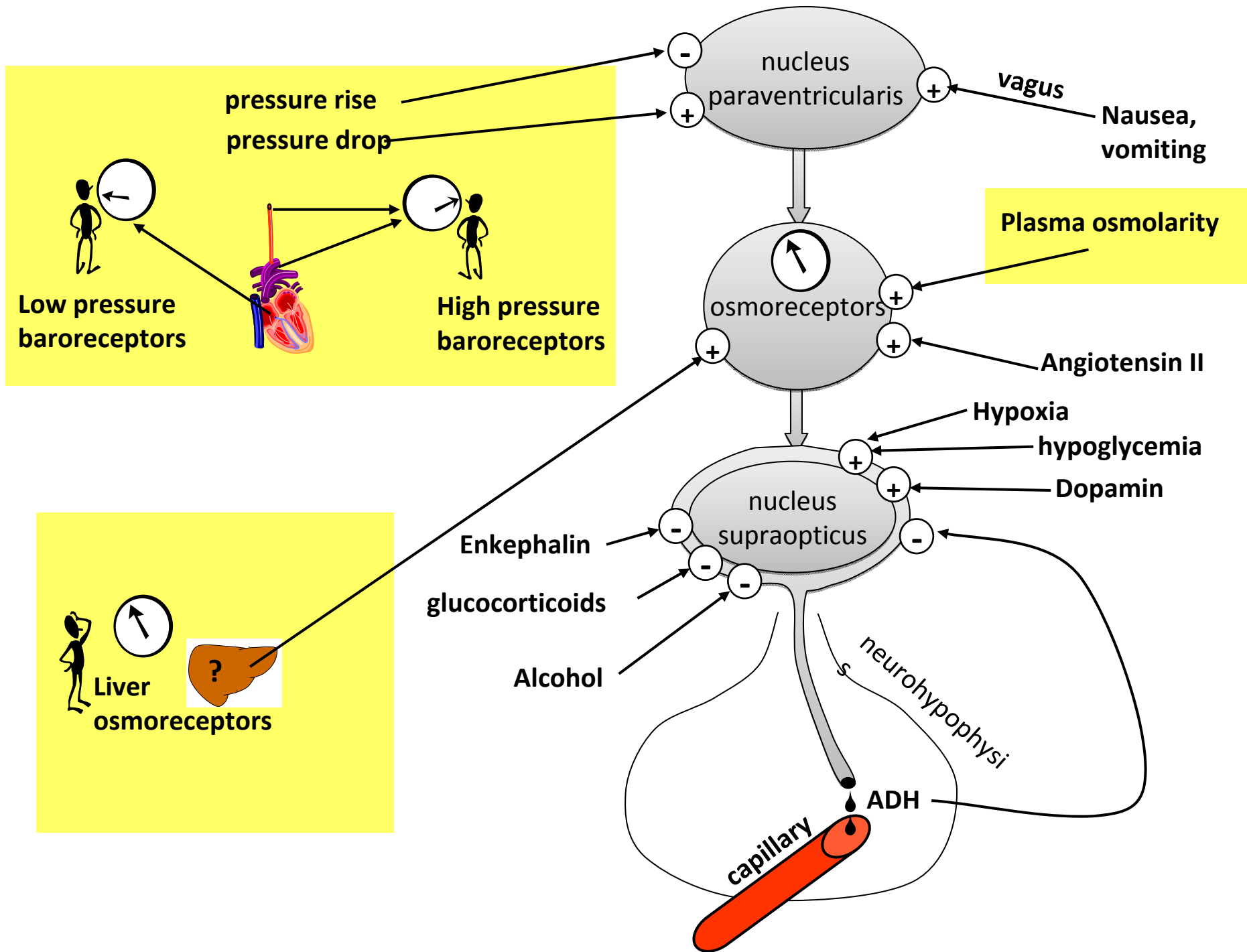
Angiotensin III

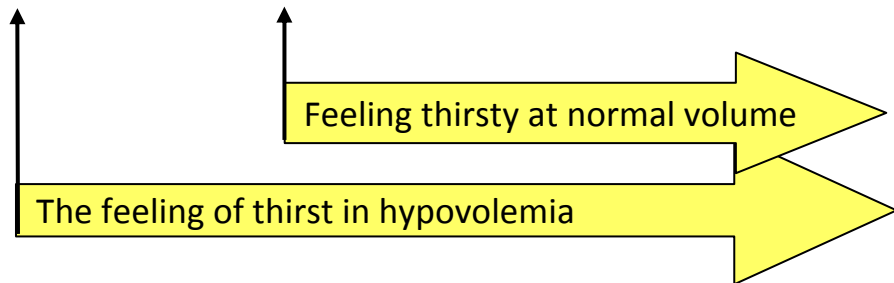
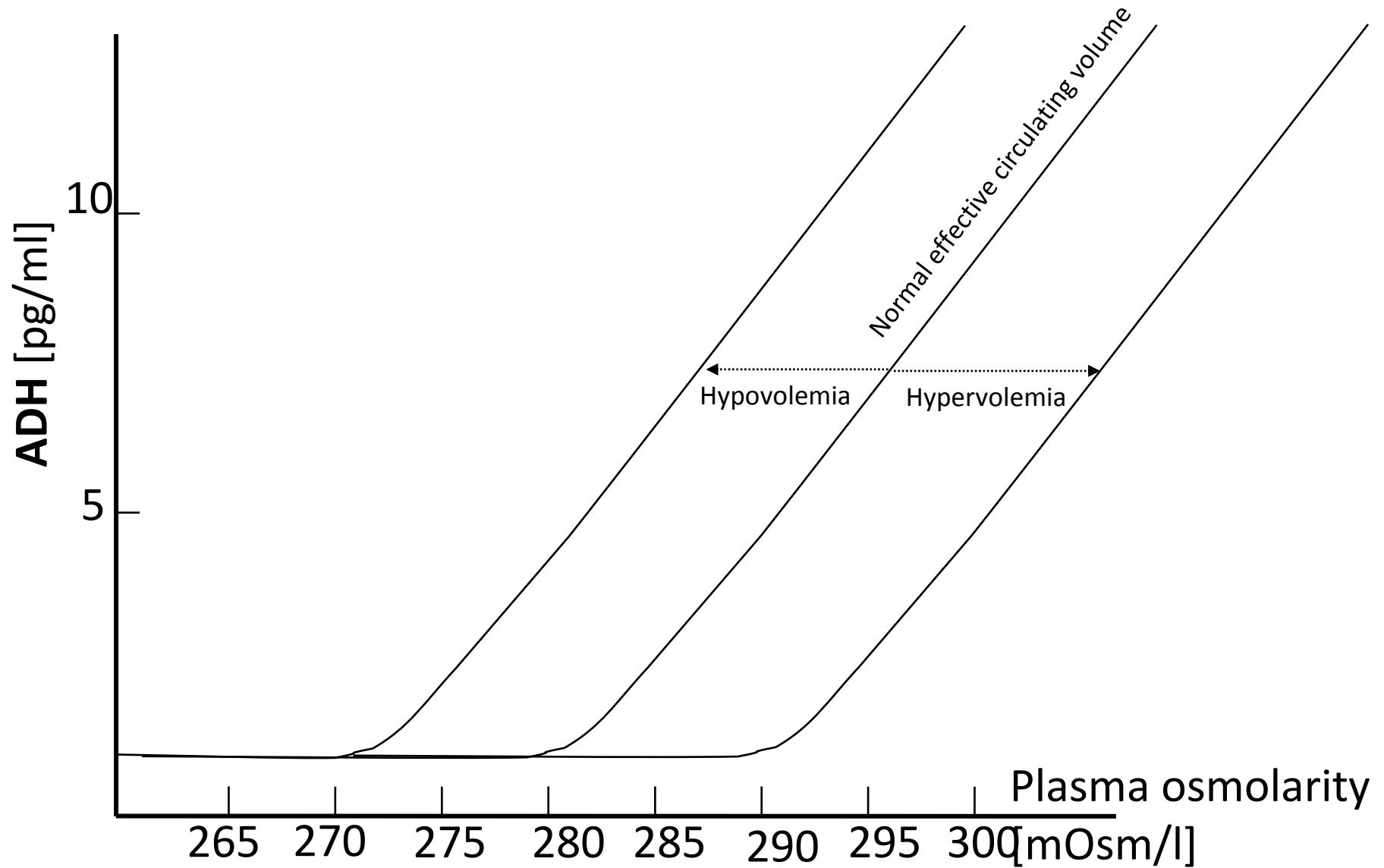
Angiotensin II

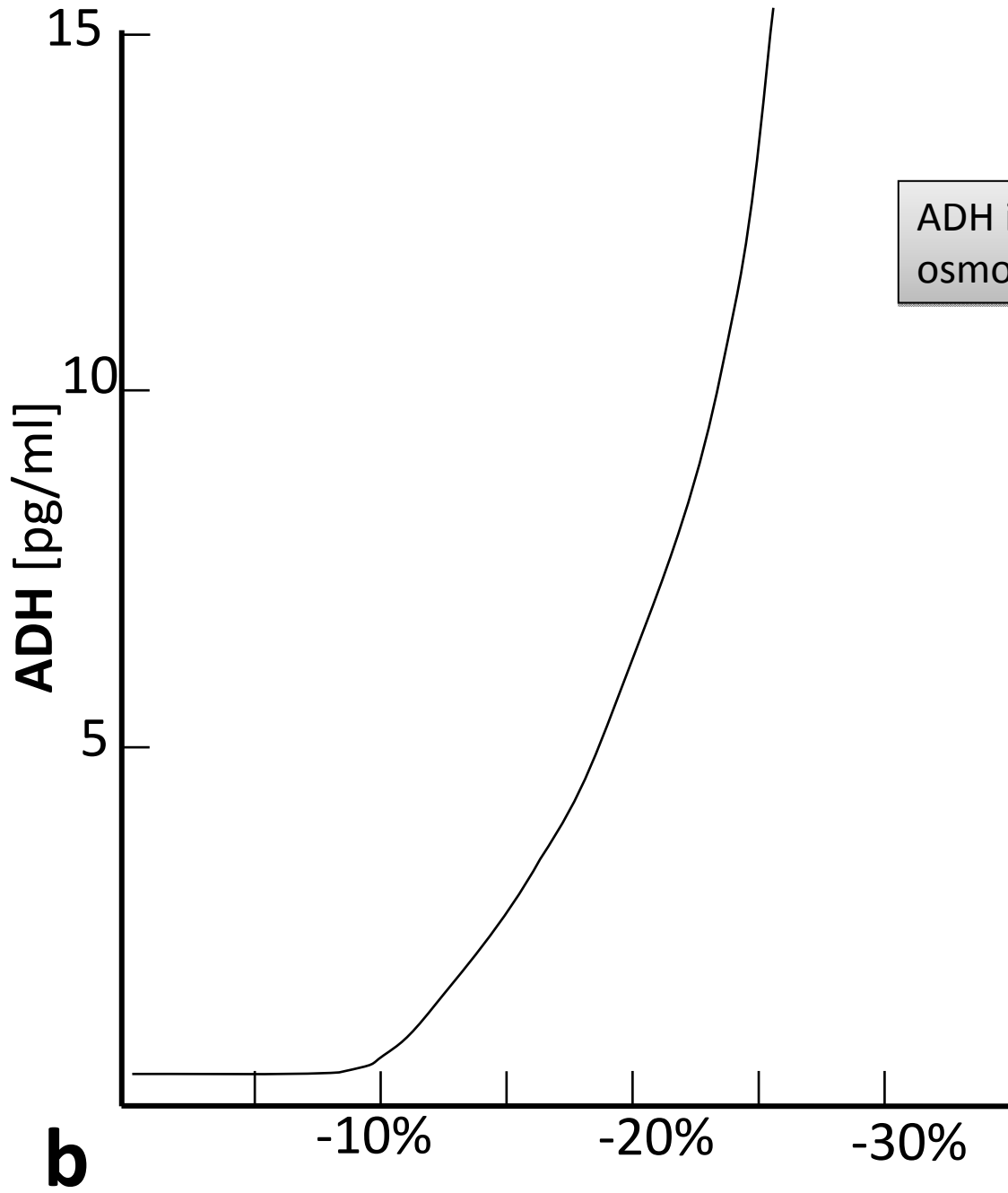
Aldosterone

Adrenal cortex

Angiotensin II increases the sensitivity osmoreceptors





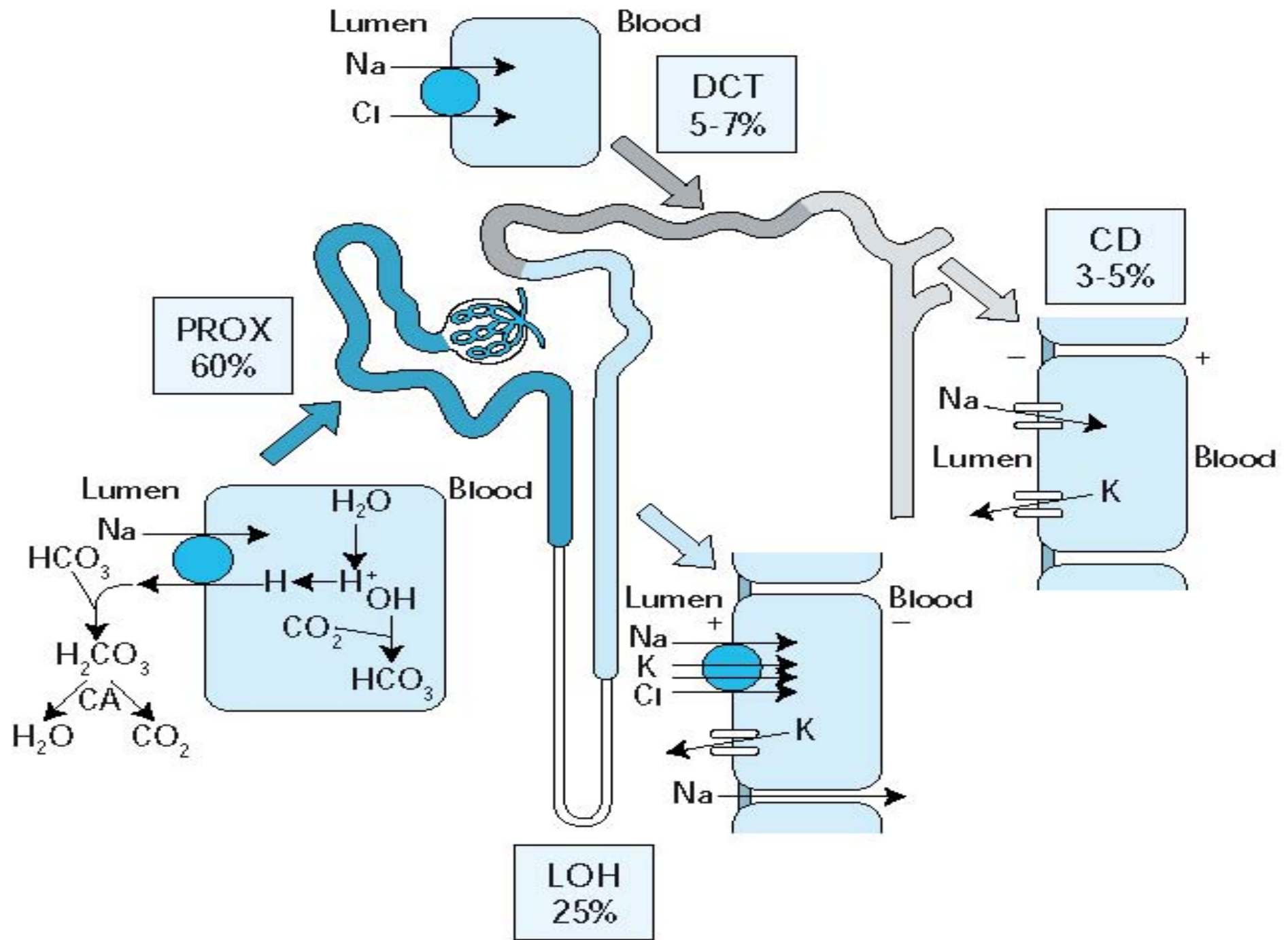


ADH is mainly regulated by osmolarity, less by volume

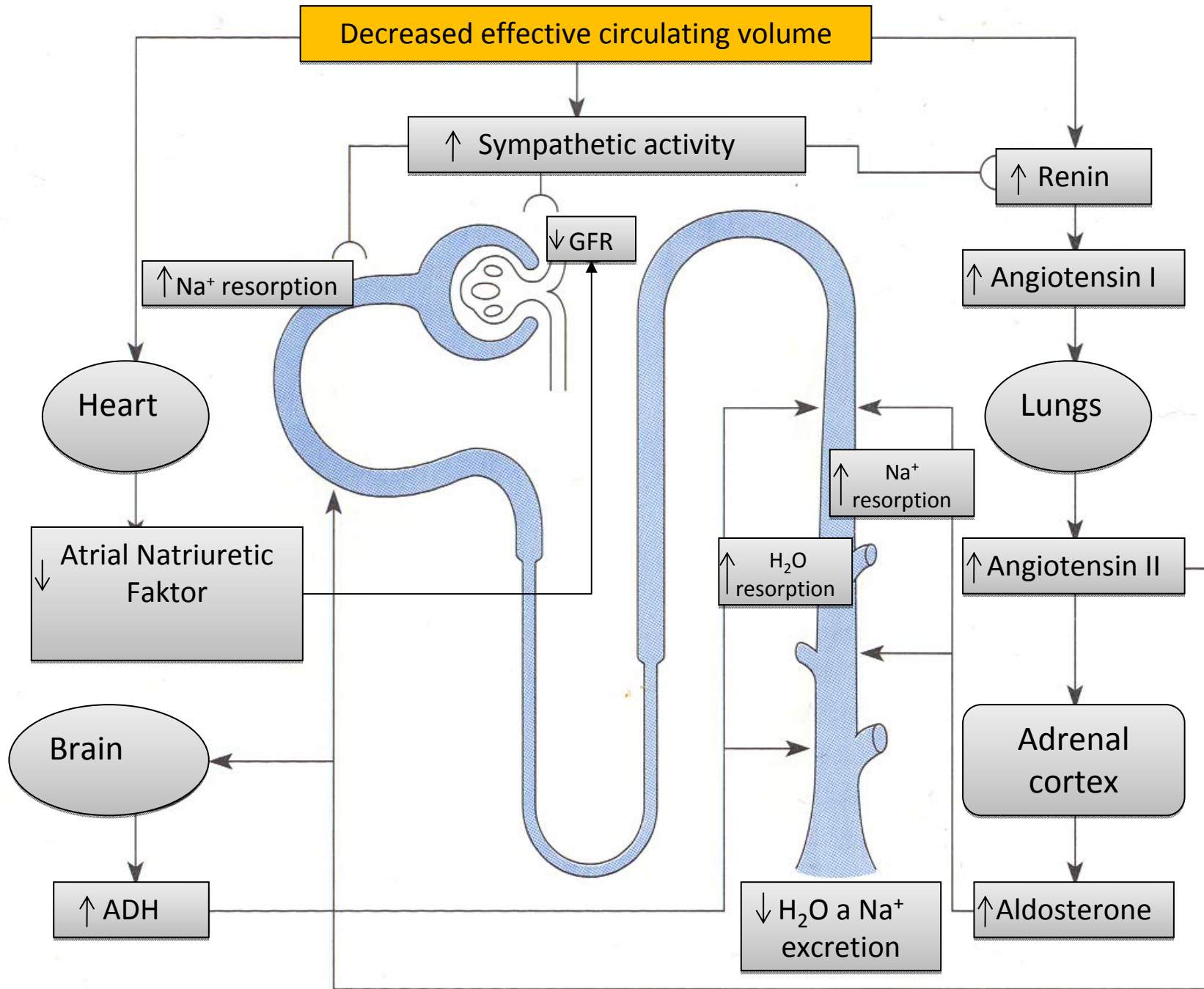
Decrease of the effective circulating volume at normal osmolarity

b

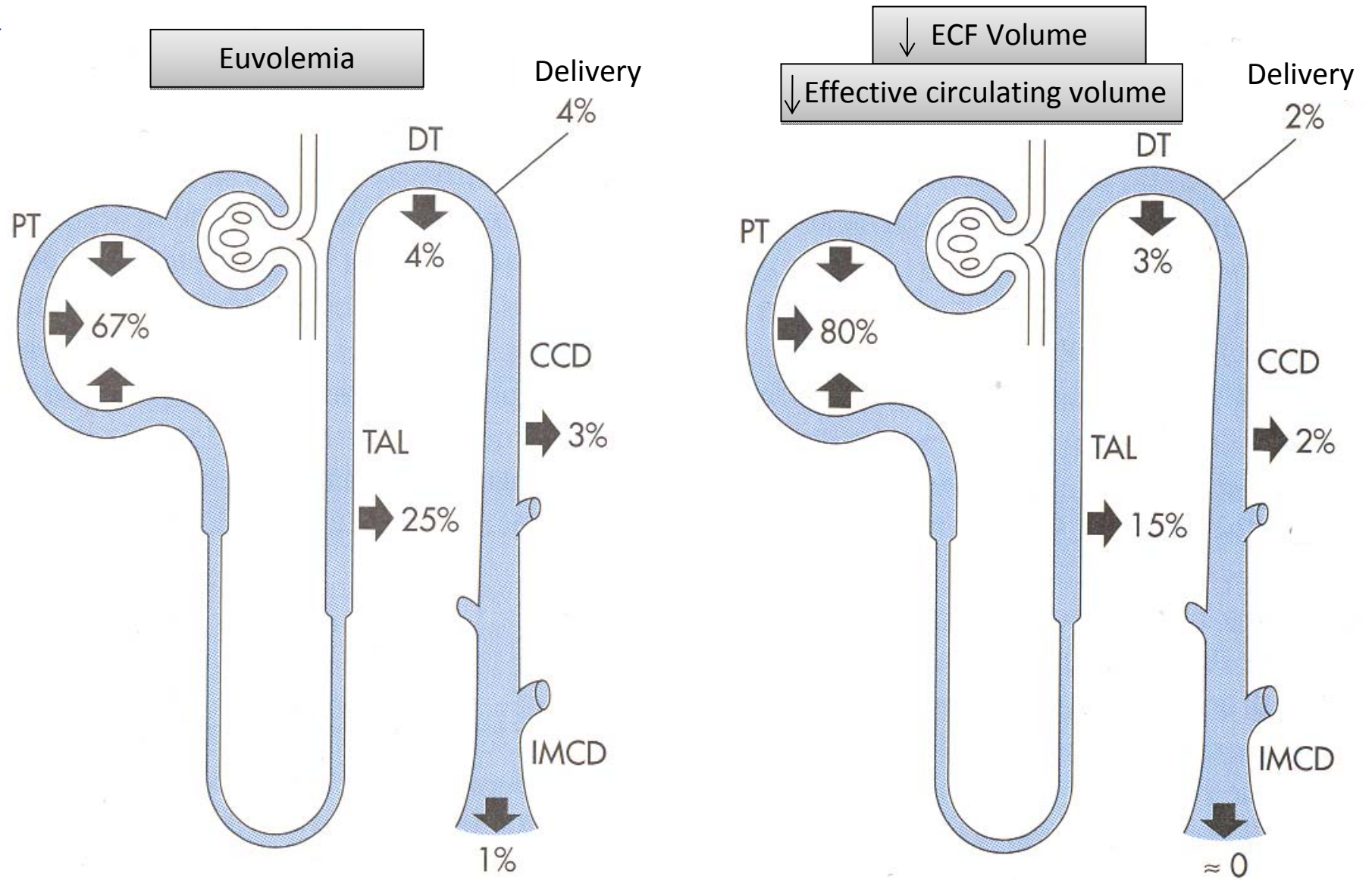
Na⁺ excretion in the kidney



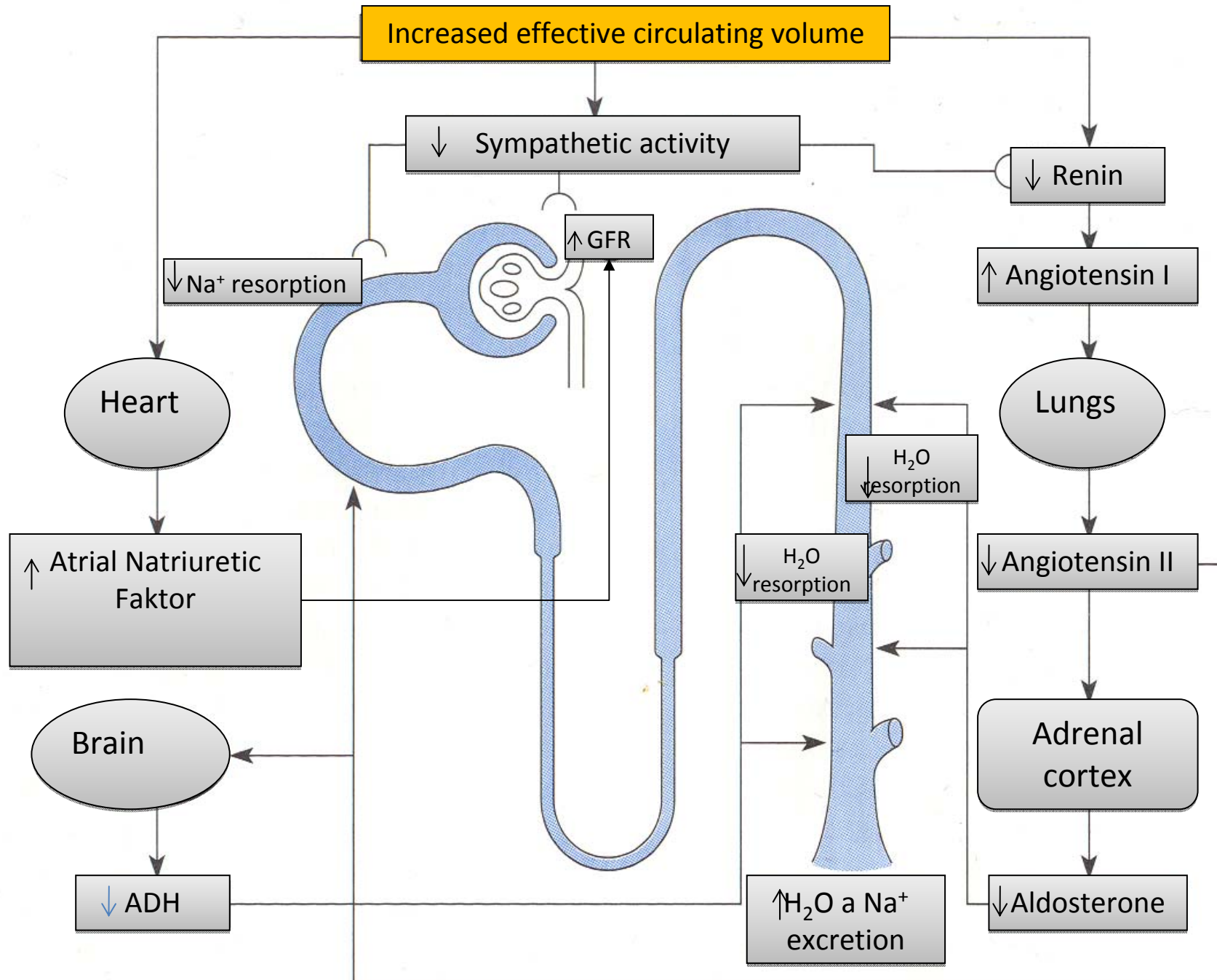
Na⁺



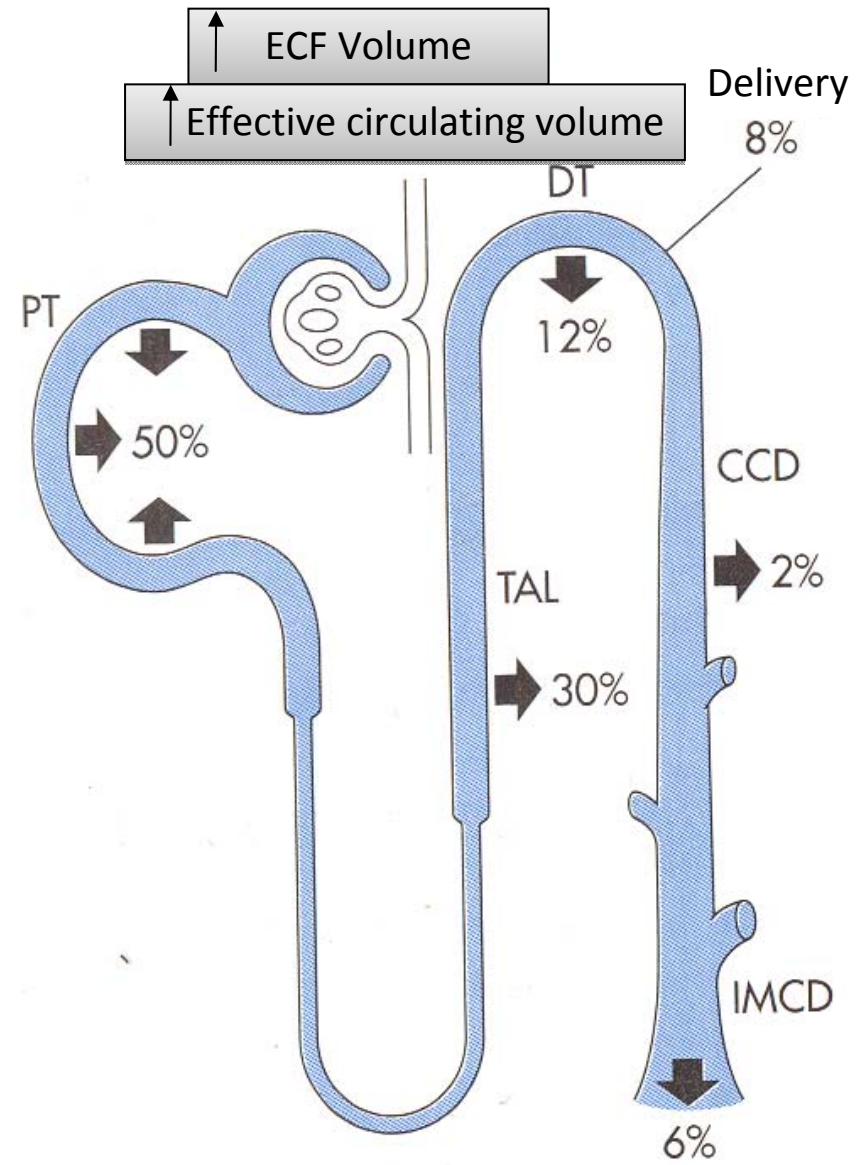
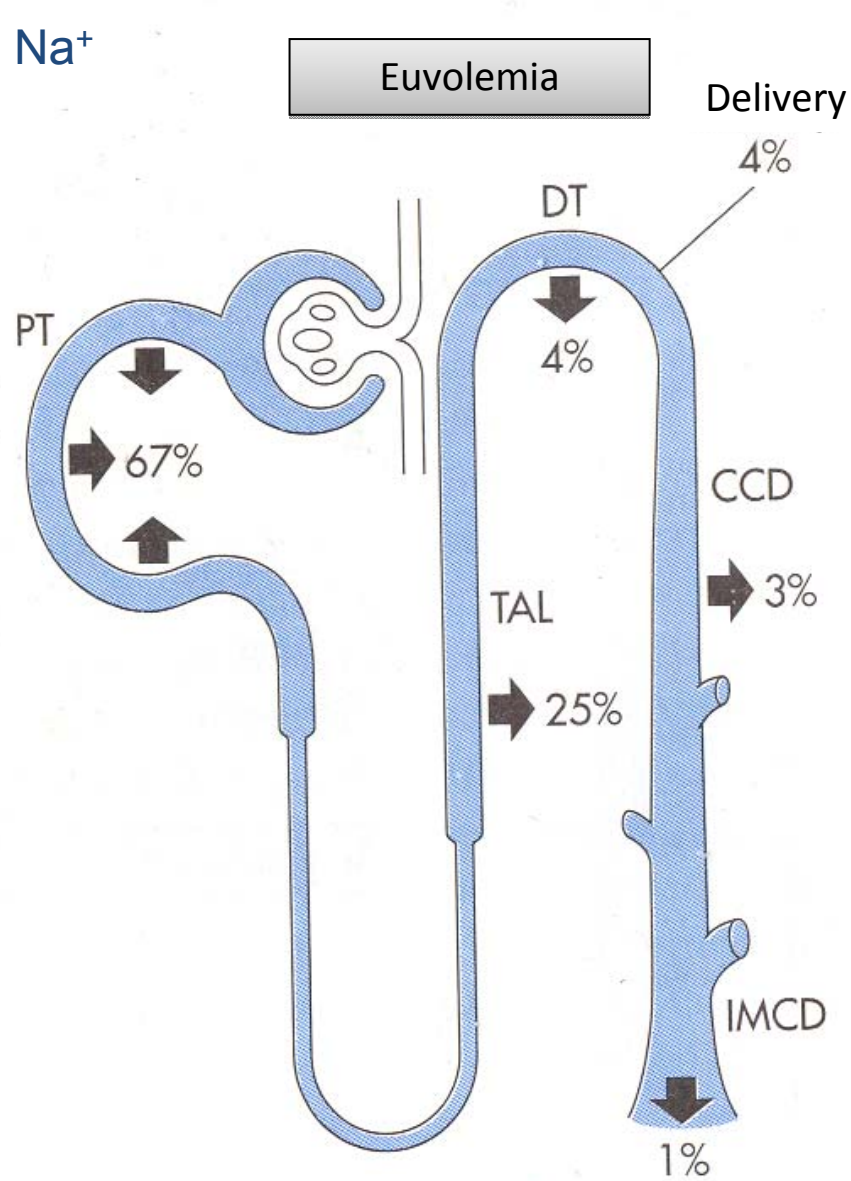
Na⁺

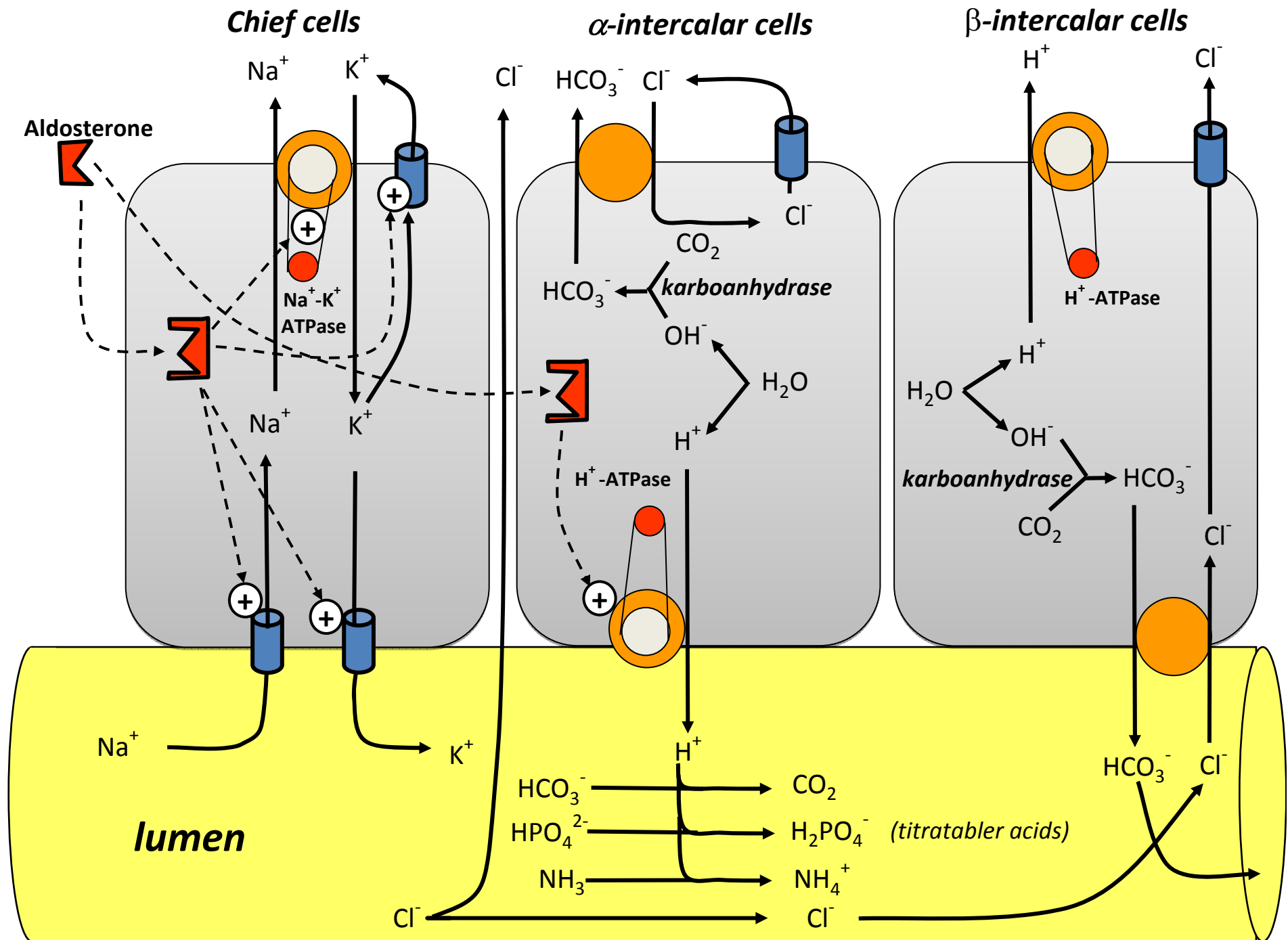


Na⁺



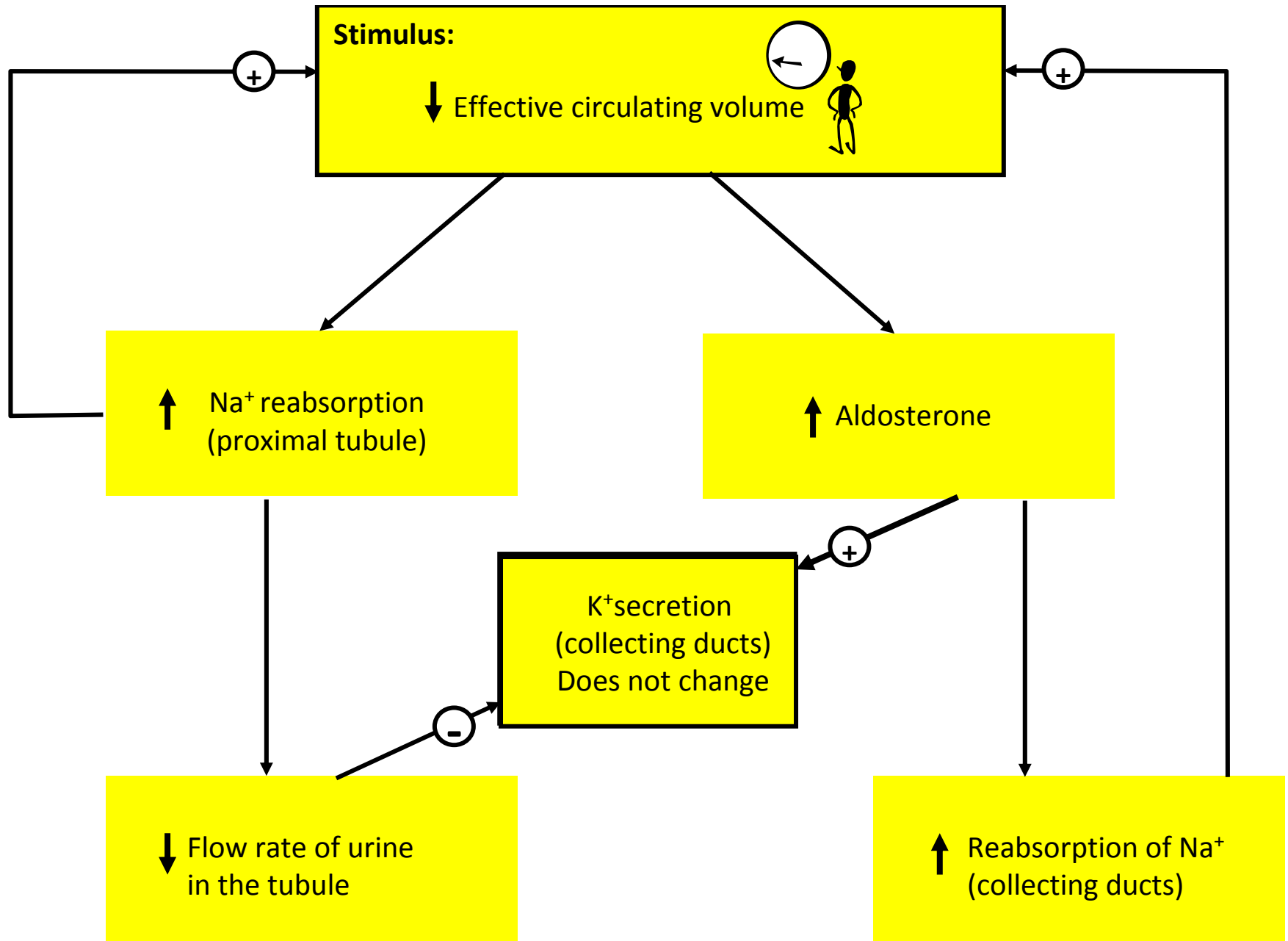
Na⁺

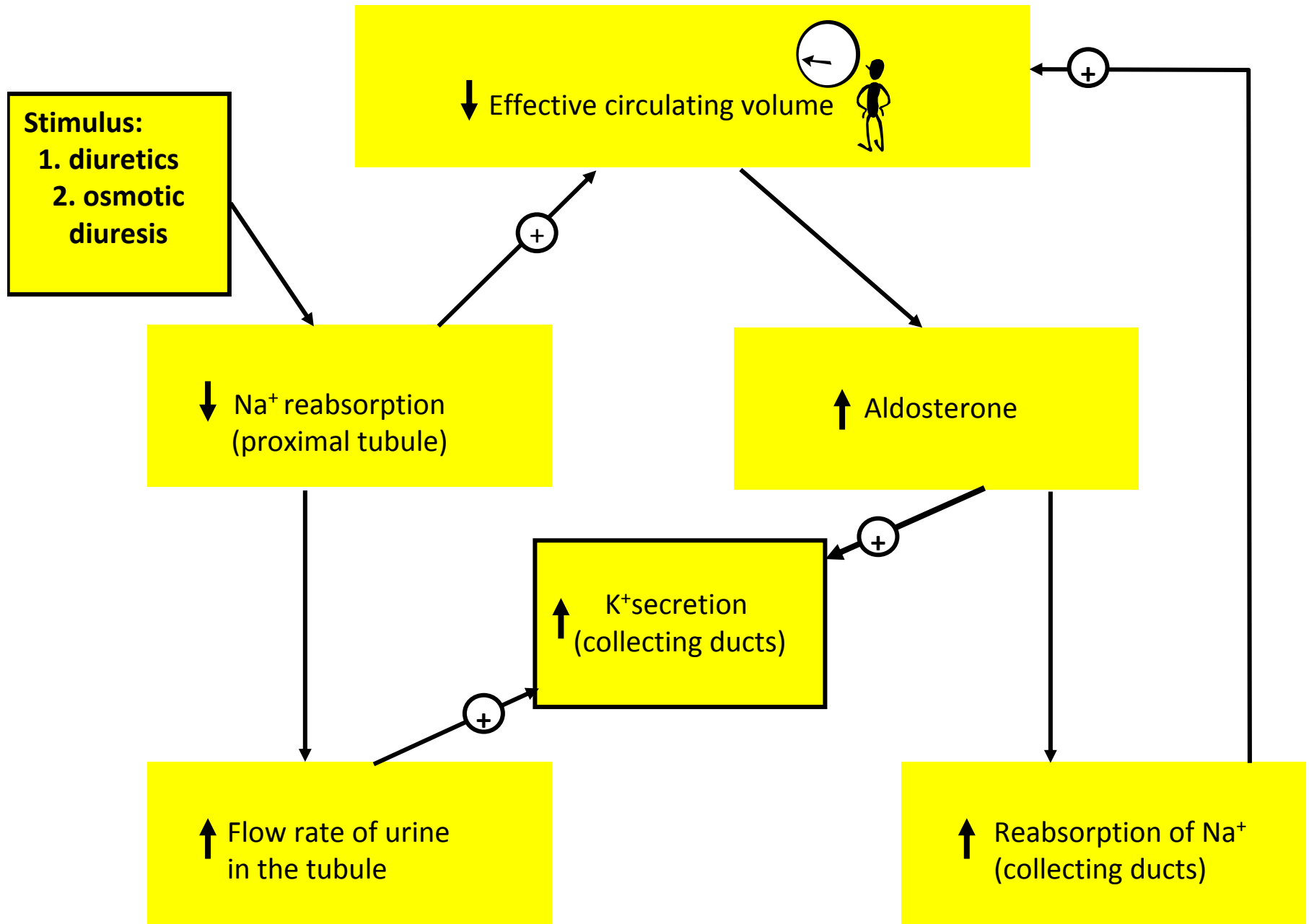




Question

Why in healthy people with reducing the volume of extracellular fluid and subsequent activation of renin-angiotensin-aldosterone system are not lost potassium?



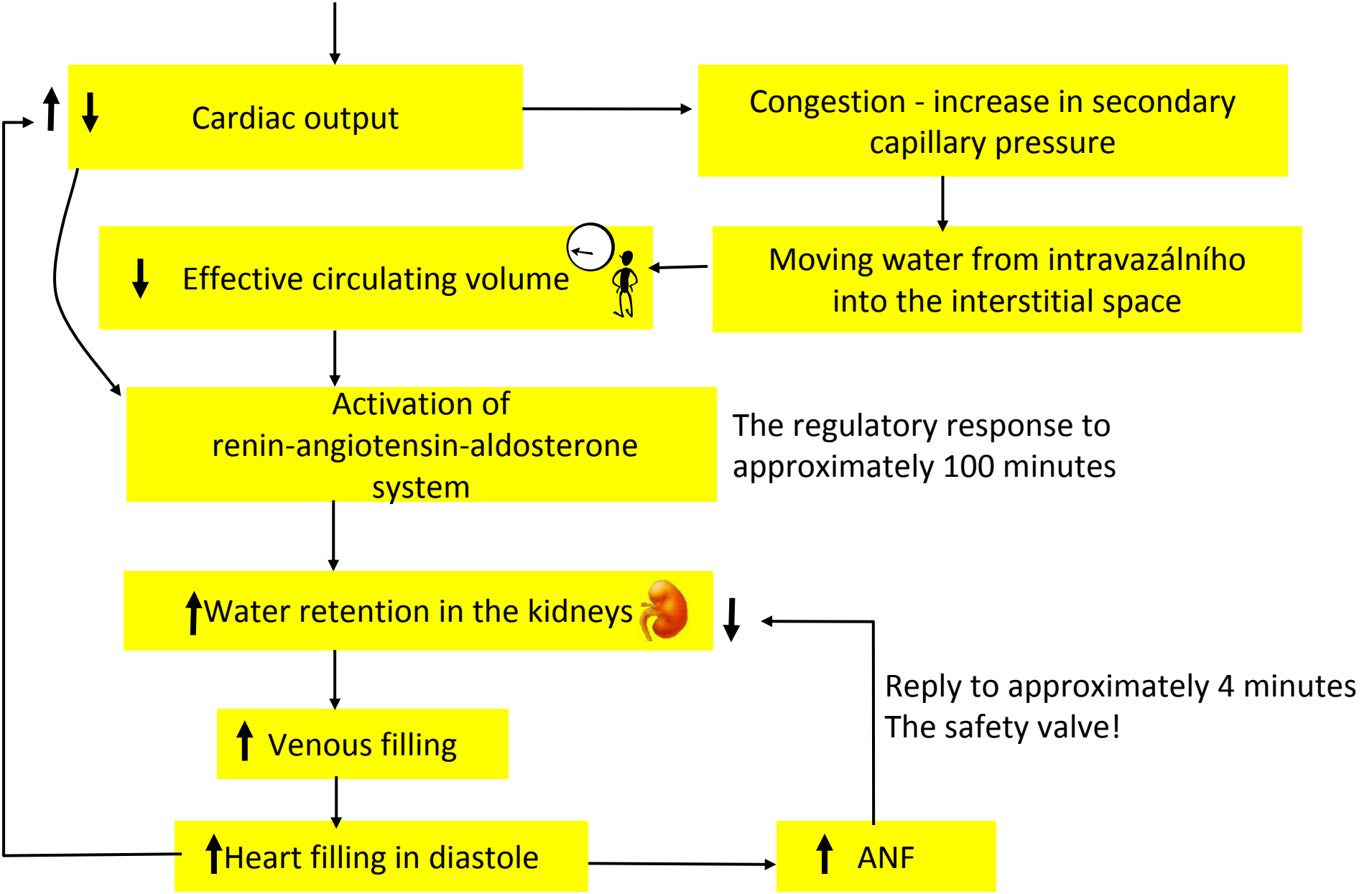


Question

What is the regulatory importance of atrial natriuretic factor?

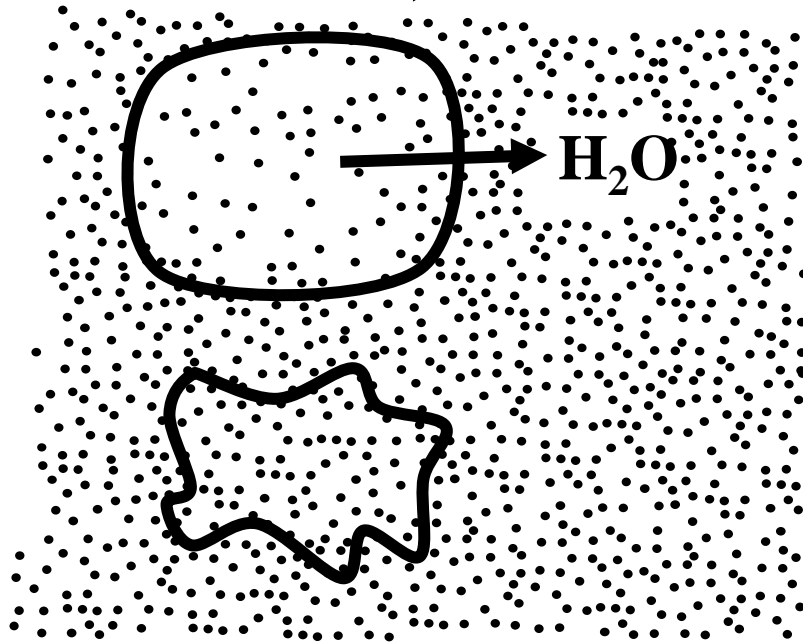
ANF acts as a **safety valve** when the compensatory increase in circulating blood volume

Heart failure



Transfers of water between the ICF and ECF

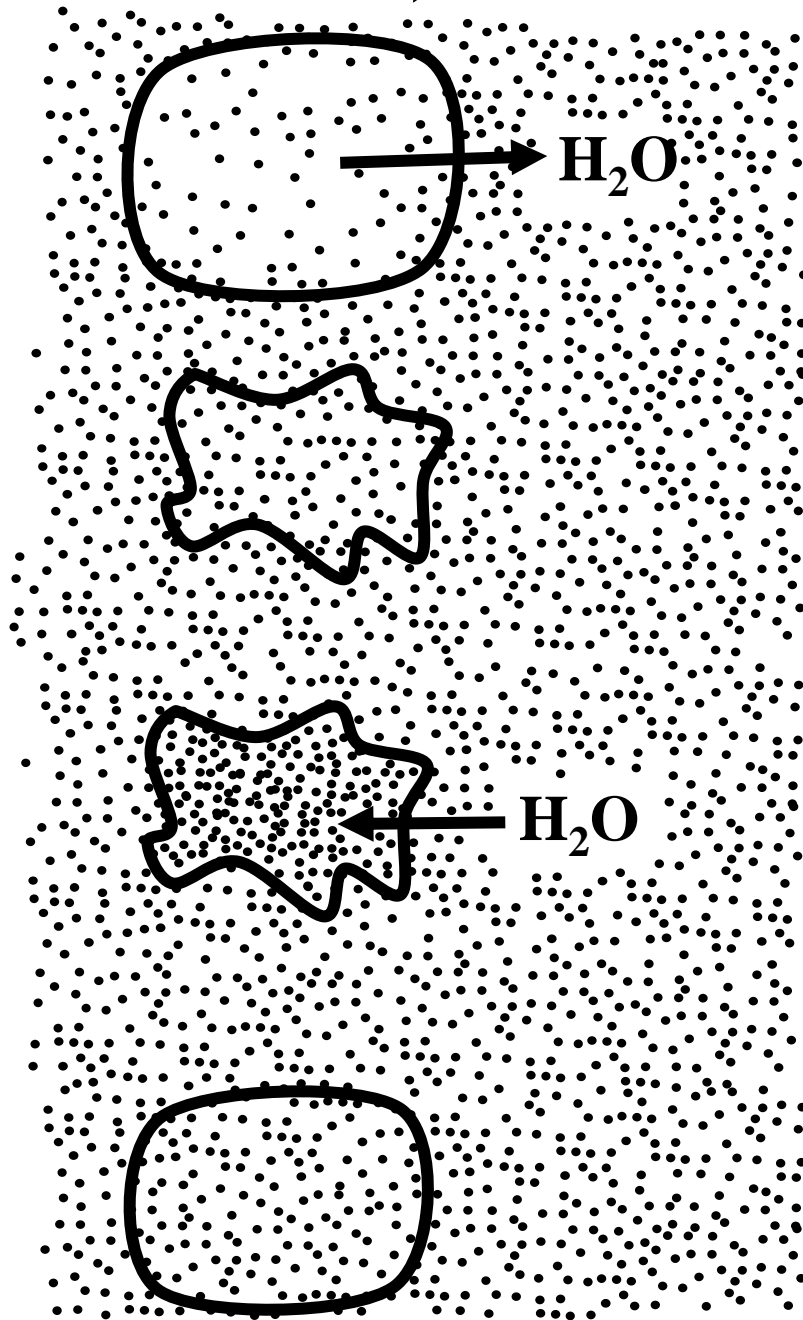
a) Cells in the hypertonic environment



Osmotic shift of water from cells

Reducing the volume of cells

a) Cells in the hypertonic environment

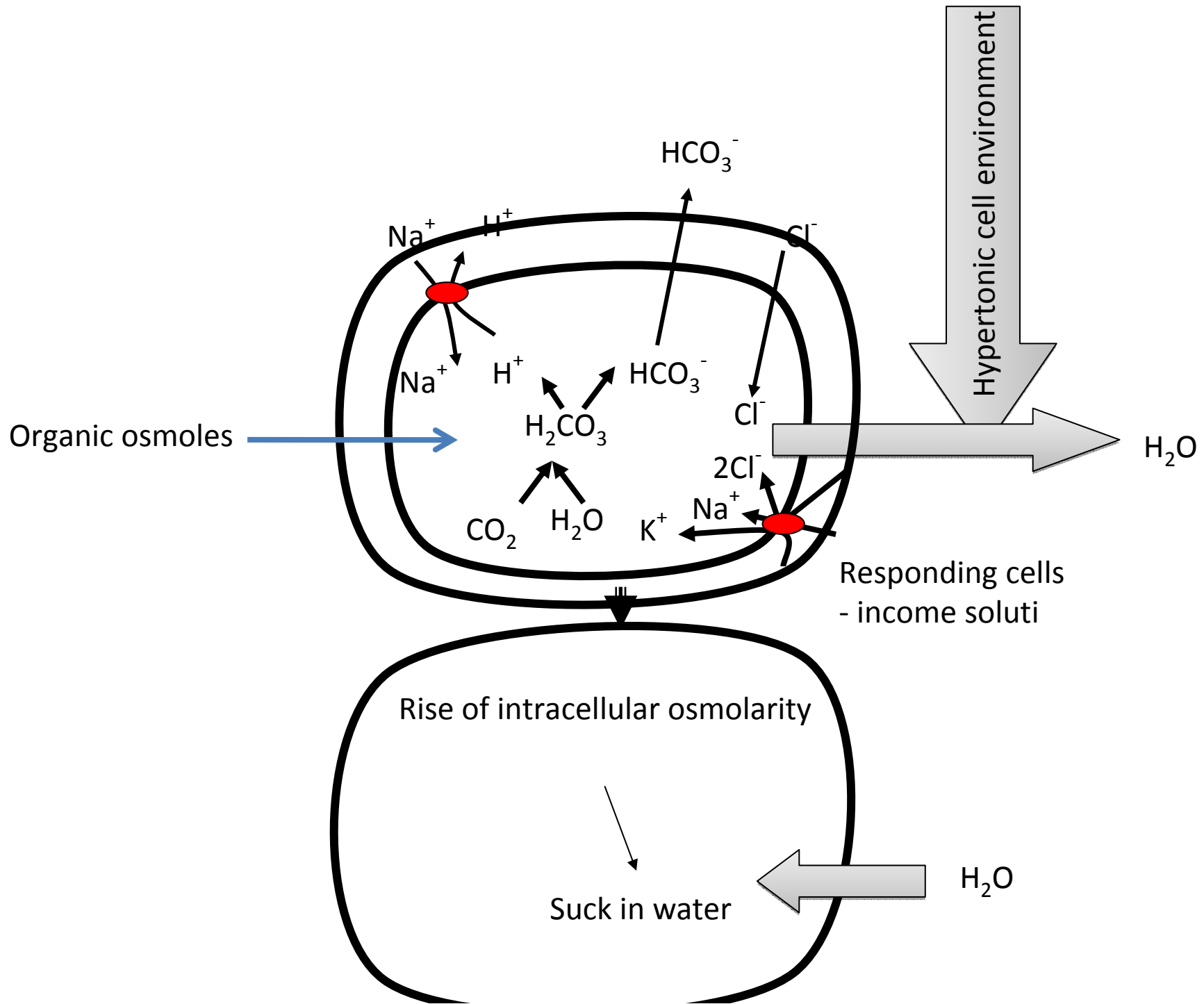


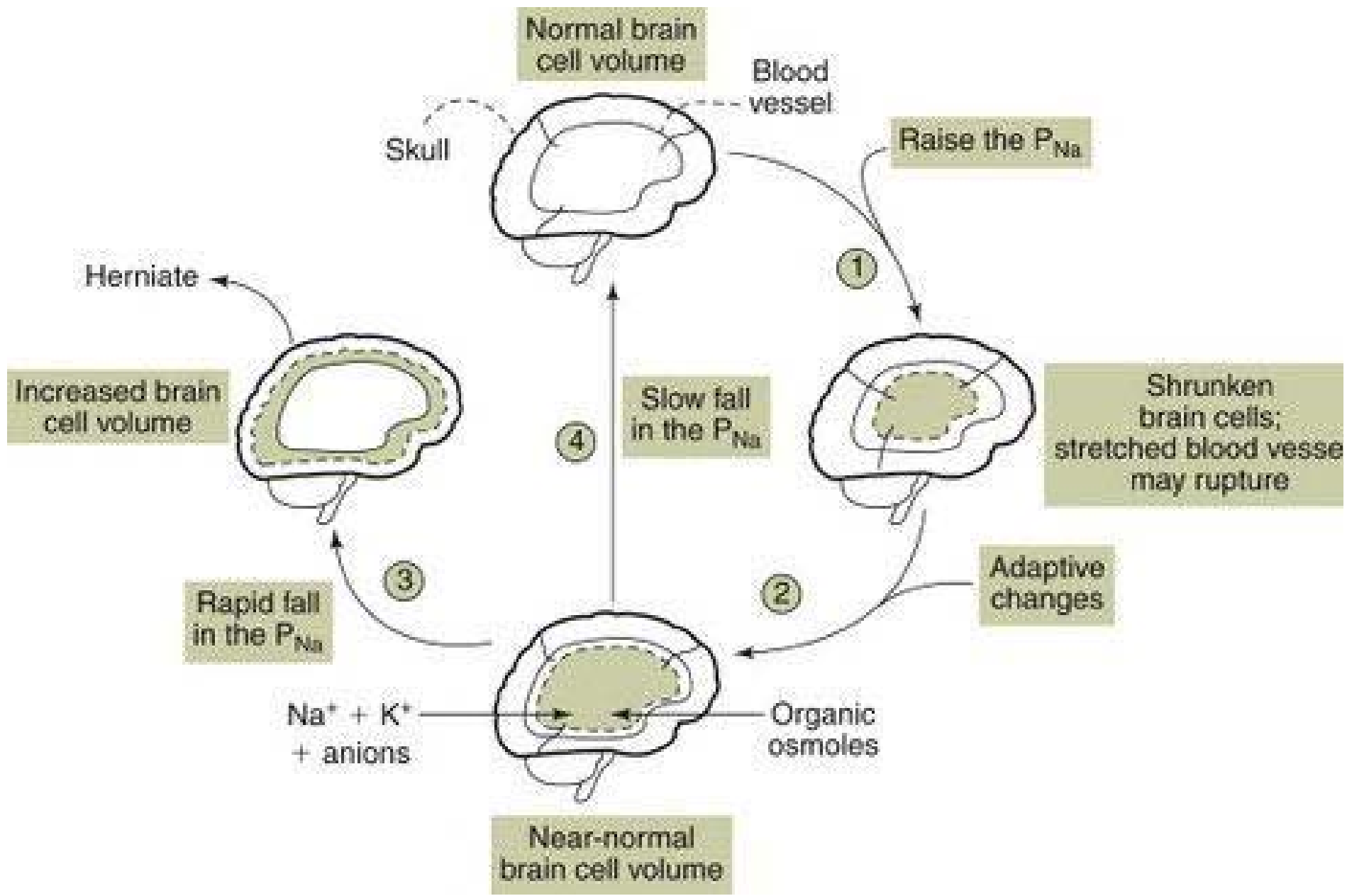
Osmotic shift of water from cells

Reducing the volume of cells

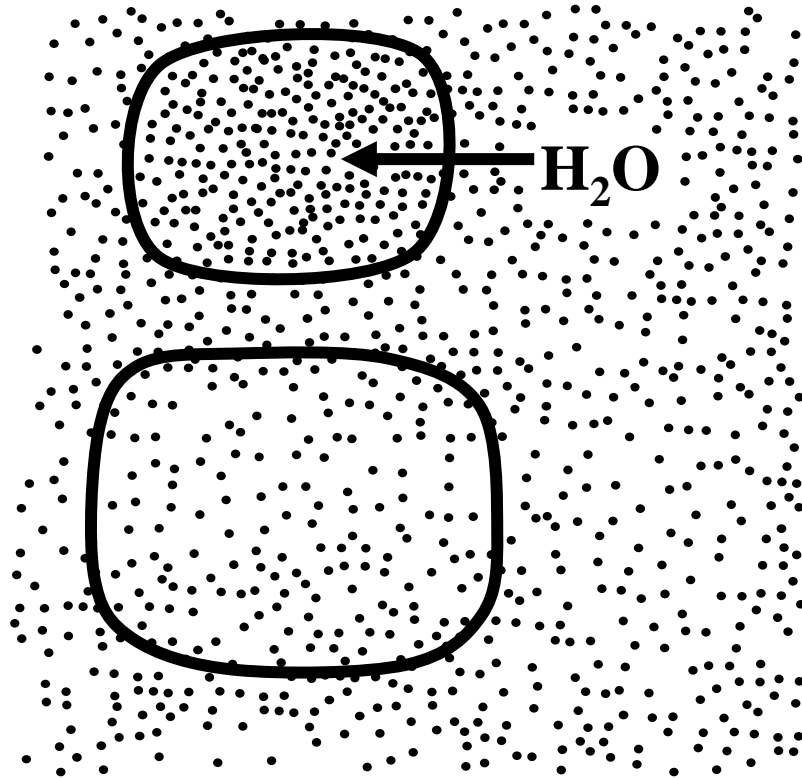
Active increase in osmotic pressure in the cell and subsequent transfer of water

Cell volume increases slightly





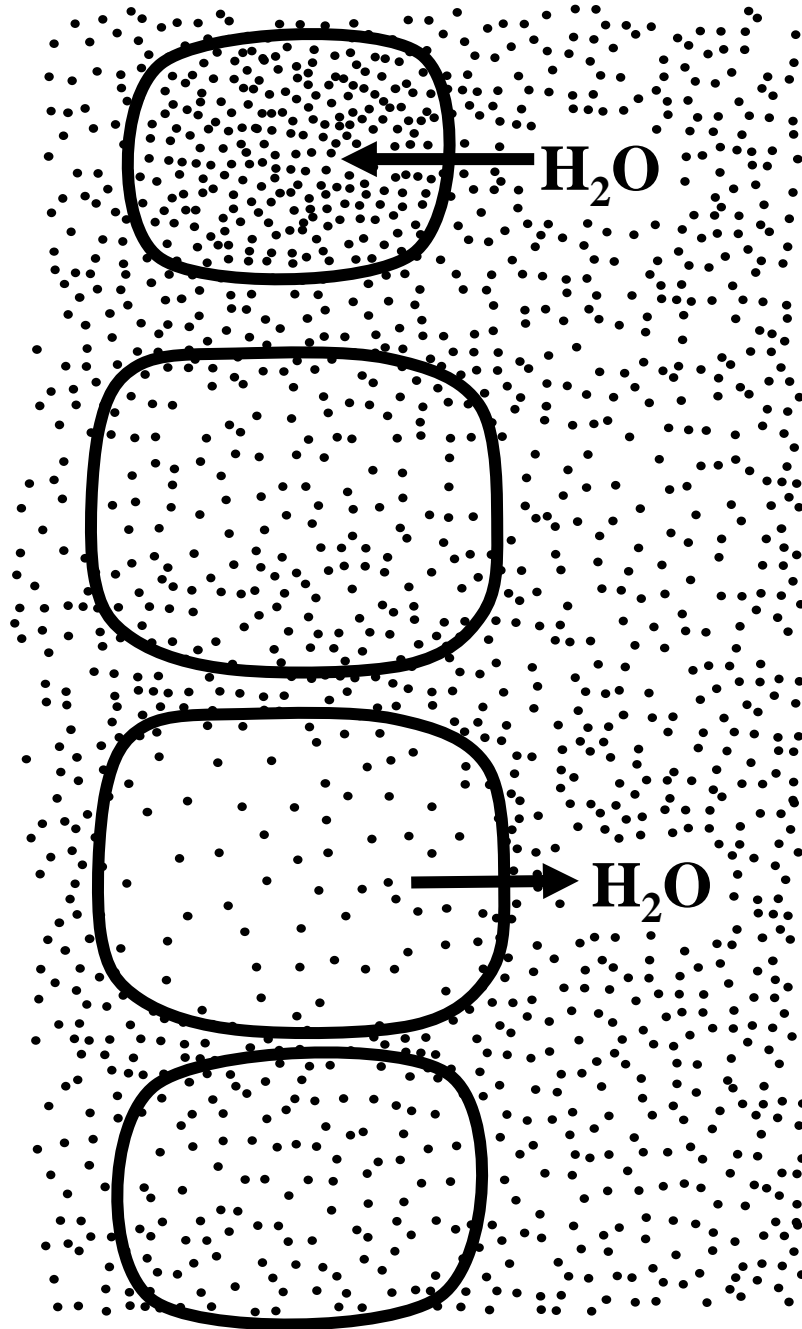
b) Cell in hypotonic environment



Osmotic movement of water into the cell

Increasing the volume of cells

b) Cell in hypotonic environment

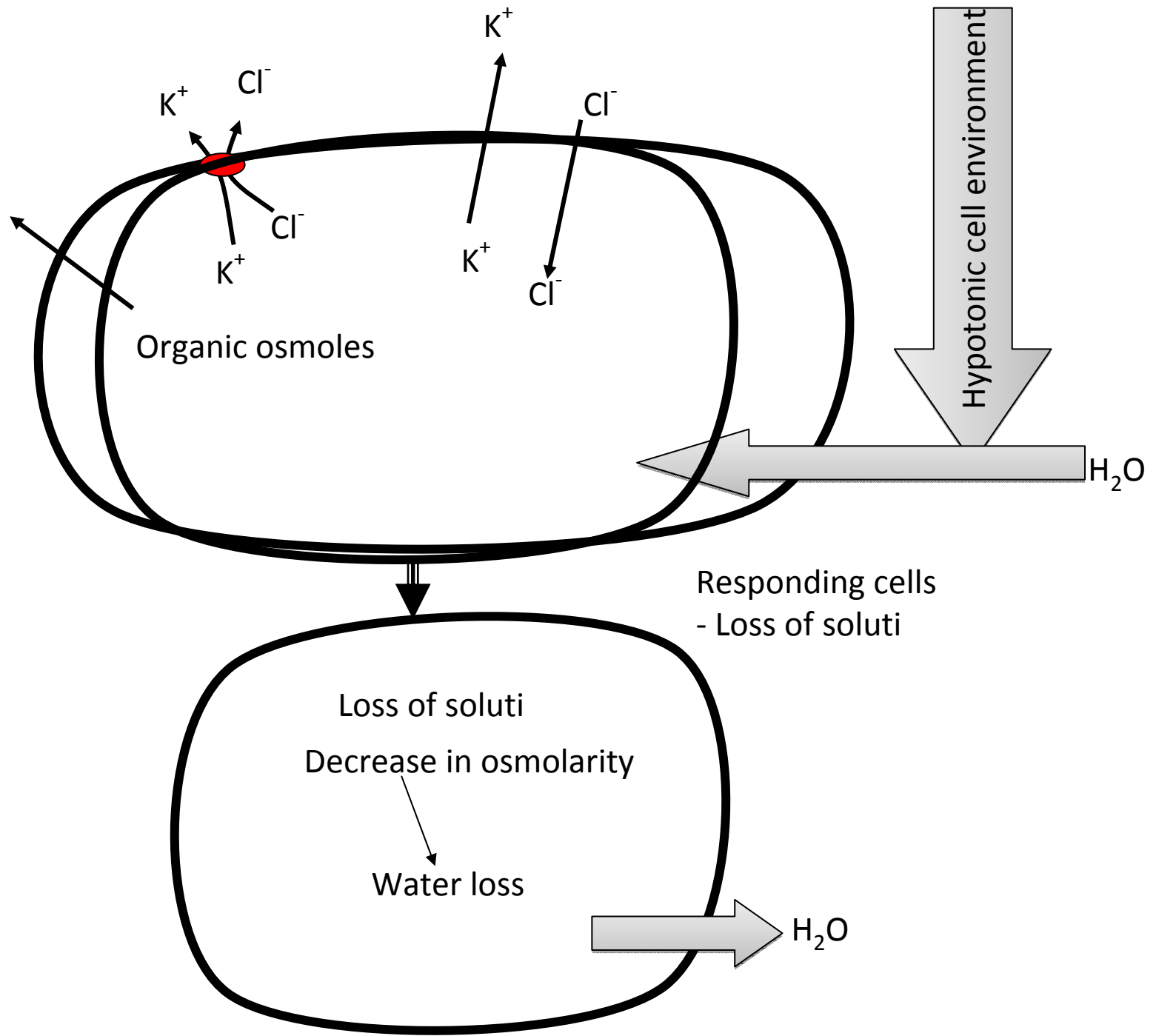


Osmotic movement of water into the cell

Increasing the volume of cells

Active reduction of osmotic pressure in the cell and subsequent transfer of water

Cell volume is somewhat reduced



K^+ Cl^-
 K^+ Cl^-

Organic osmoles

K^+ Cl^-
 K^+ Cl^-

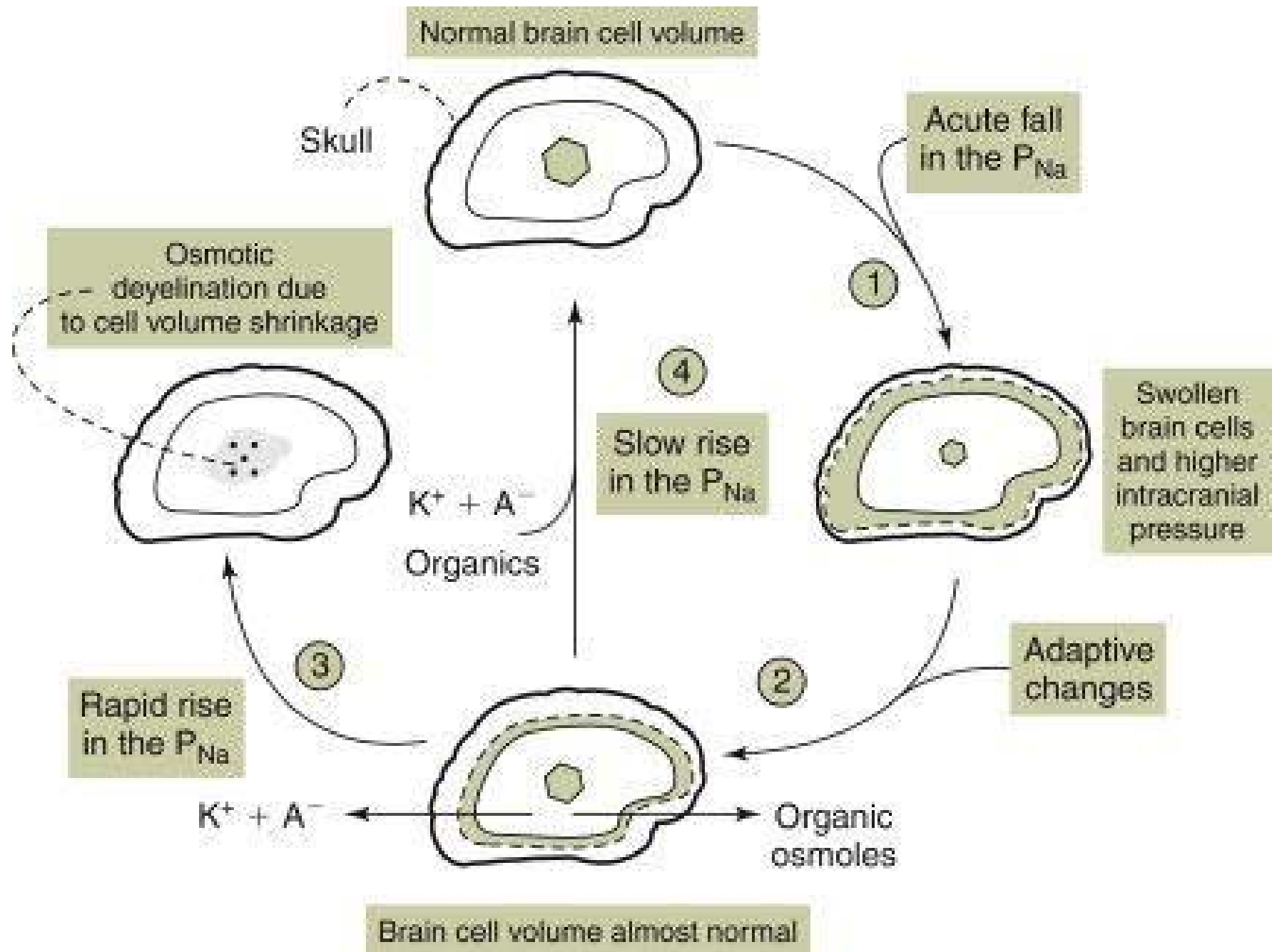
Hypotonic cell environment

H_2O

Responding cells
- Loss of soluti

Loss of soluti
Decrease in osmolarity
Water loss

H_2O



Disorders of the volume and osmolarity


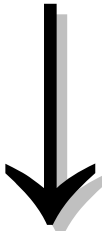
Intravascular fluid

- hypovolemia
shock
- hypervolemia
Renal failure and
water intake

Extravascular fluid

- Dehydration
 - hypertonic
 - isotonic
 - hypotonic
- Hyperhydration
 - hypotonic
 - isotonic
 - (hypertonic)

Clinical signs

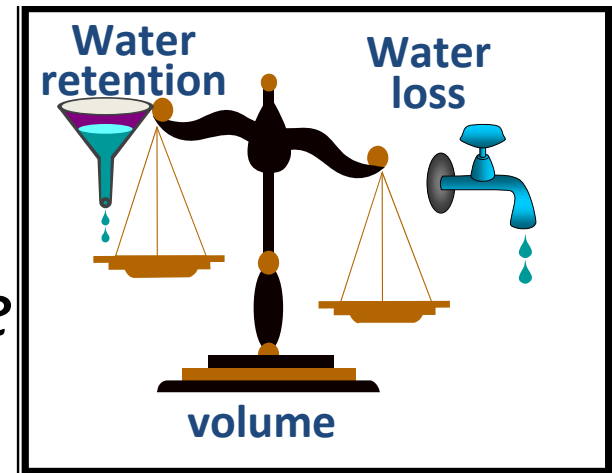
	Hct, Hb, Tot.pl.prot.	Plasma Na ⁺ PlasmOsm.	MCV
Dehydration hyperosmotic isoosmotic normoosmotic		↑ normal ↓	↓ normal ↑
Hyperhydratation hypoosmotic isoosmotic hyperosmotic		↓ normal ↑	↑ normal ↓

Dehydration



Euvolemia

- *Hypotonic (hypoosmolar)*
- *Isotonic – normal physiology state*
- *Hypertonic (hyperosmolar)*

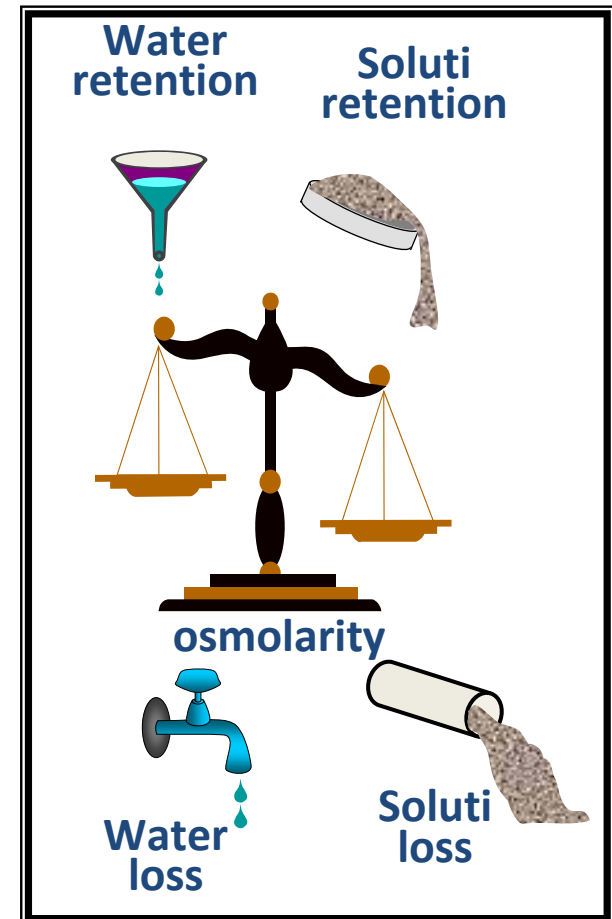


Dehydration

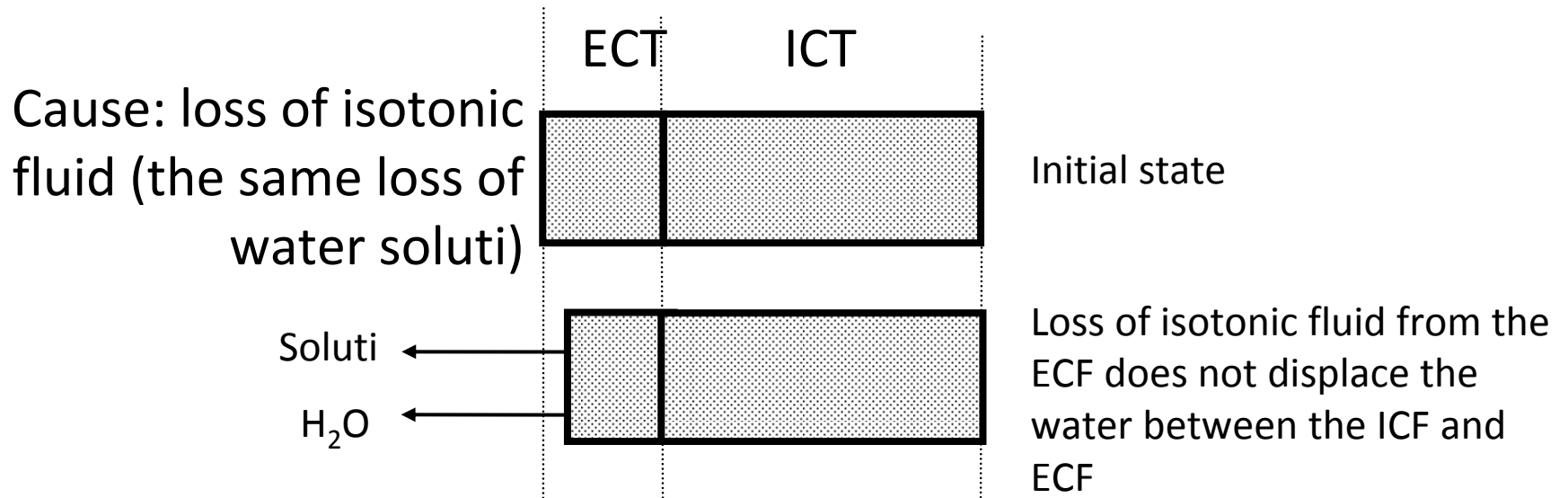
- *Hypotonic (hypoosmolar)*
- *Isotonic (isoosmolar)*
- *Hypertonic (hyperosmolar)*

Hyperhydration

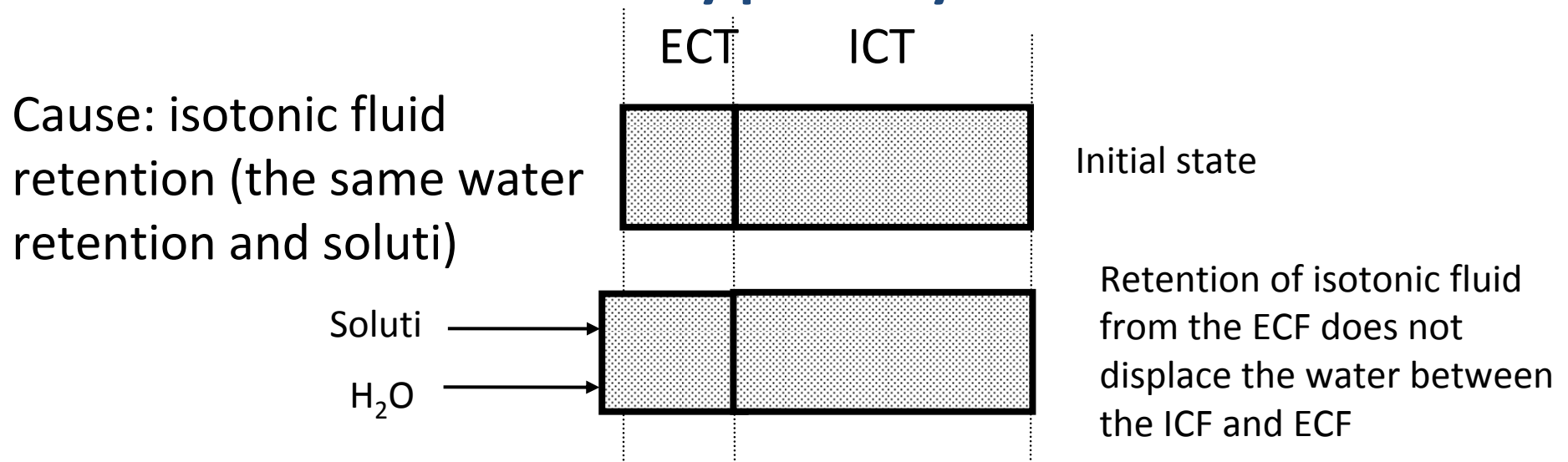
- *Hypotonic (hypoosmolar)*
- *Isotonic (isoosmolar)*
- *Hypertonic (hyperosmolar)*



Isotonic dehydration

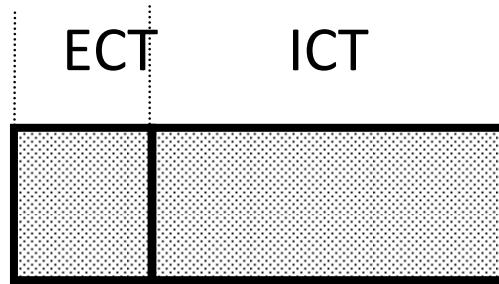


Isotonic hyperhydration

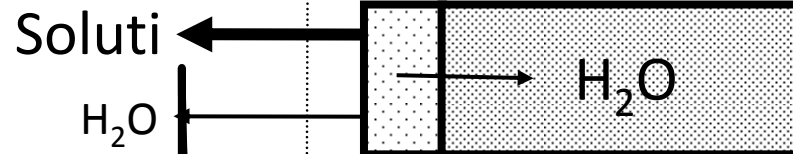


Hypotonic dehydration

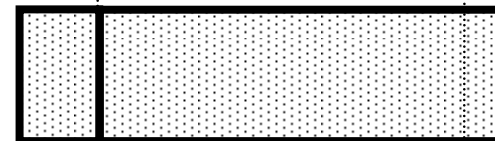
Cause: The greater soluti loss than water loss



Initial state



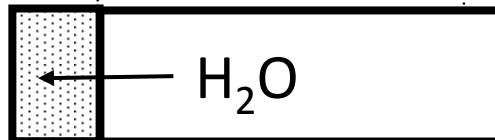
Start failure



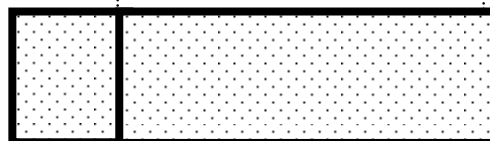
Osmotic equilibrium-
increasing volume of ICF



Compensatory reduction in
osmolarity in ICF ...



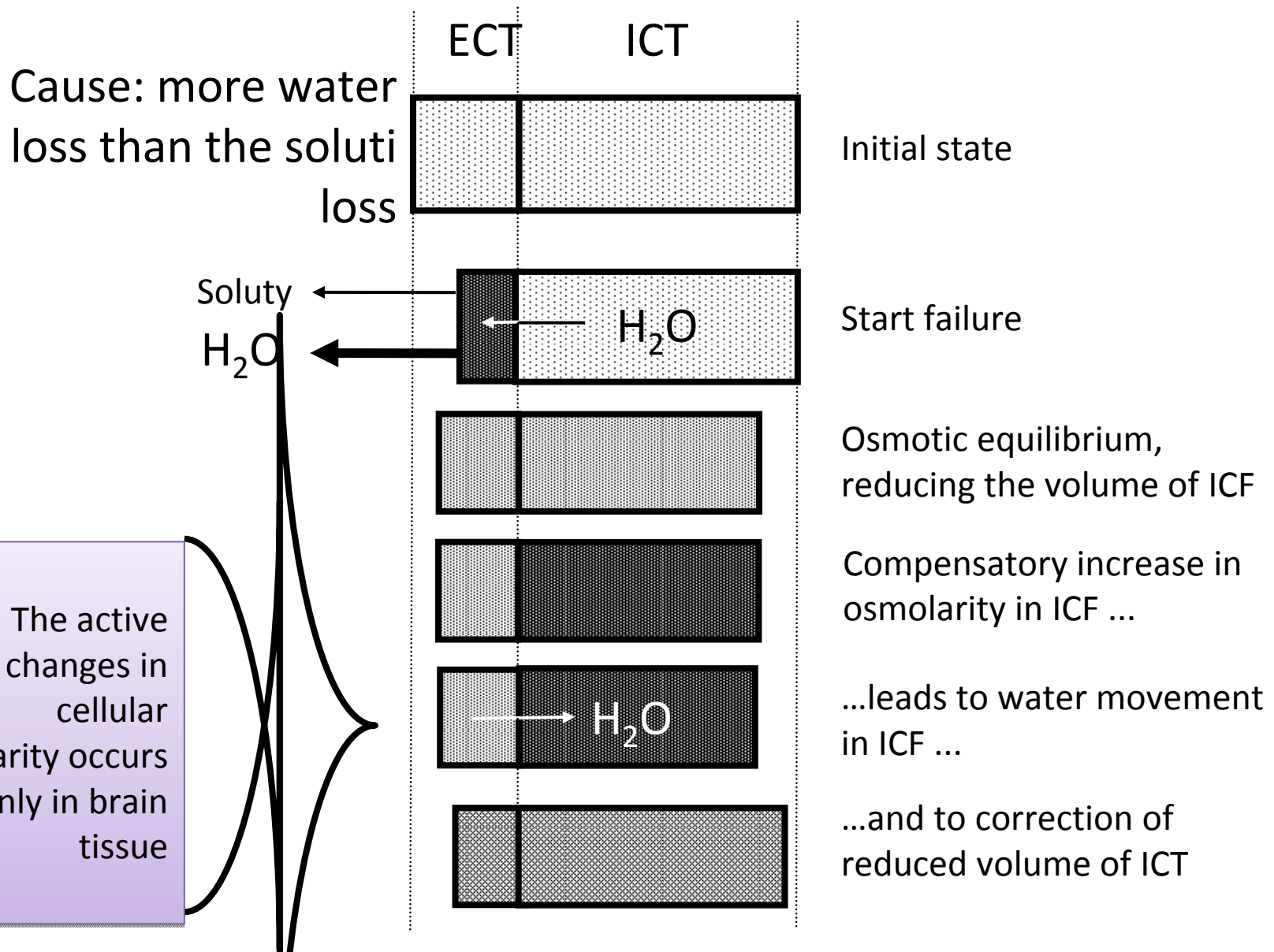
leads to a shift of water from
ICF ...



...and to correction of
increased volume of ICT

The active changes in cellular osmolarity occurs mainly in brain tissue

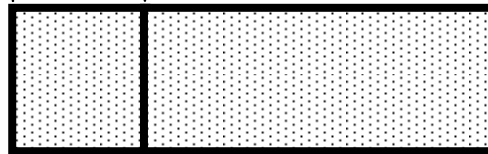
Hypertonic dehydration



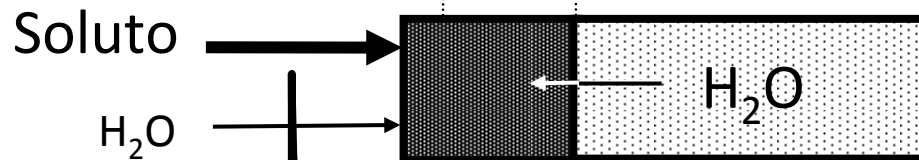
Hypertonic hyperhydratation

Cause: more solute retention than water retention

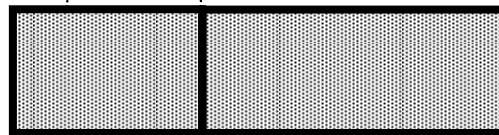
ECT ICT



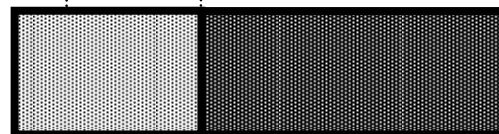
Initial state



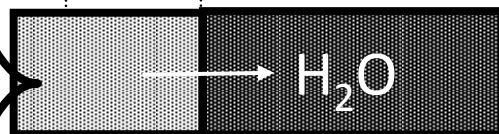
Start failure



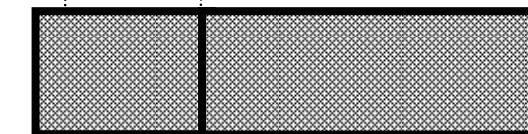
Osmotic equilibrium,
reducing volume of ICF



Compensatory increase in
osmolarity in ICF...



...leads to water movement
in ICF ...

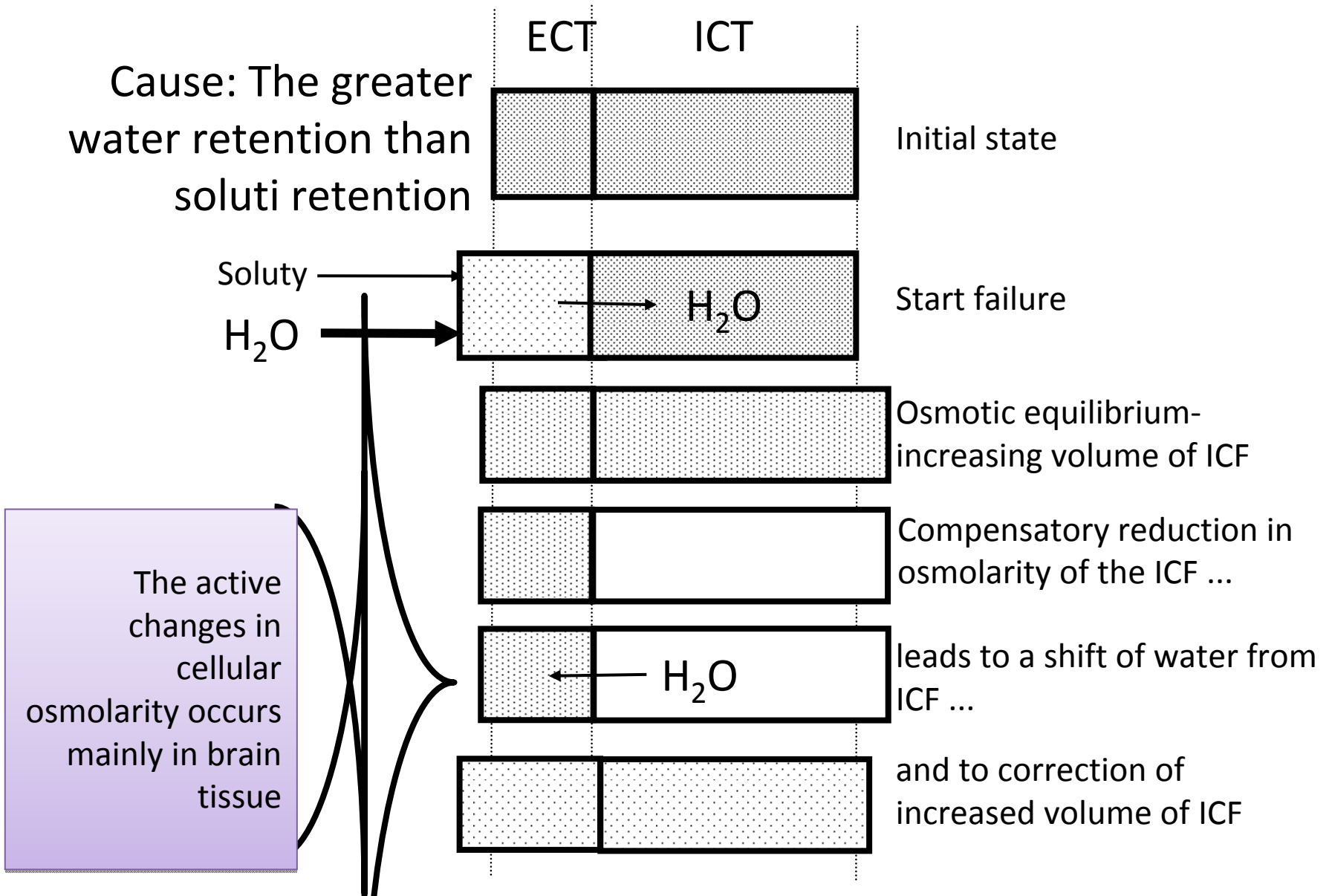


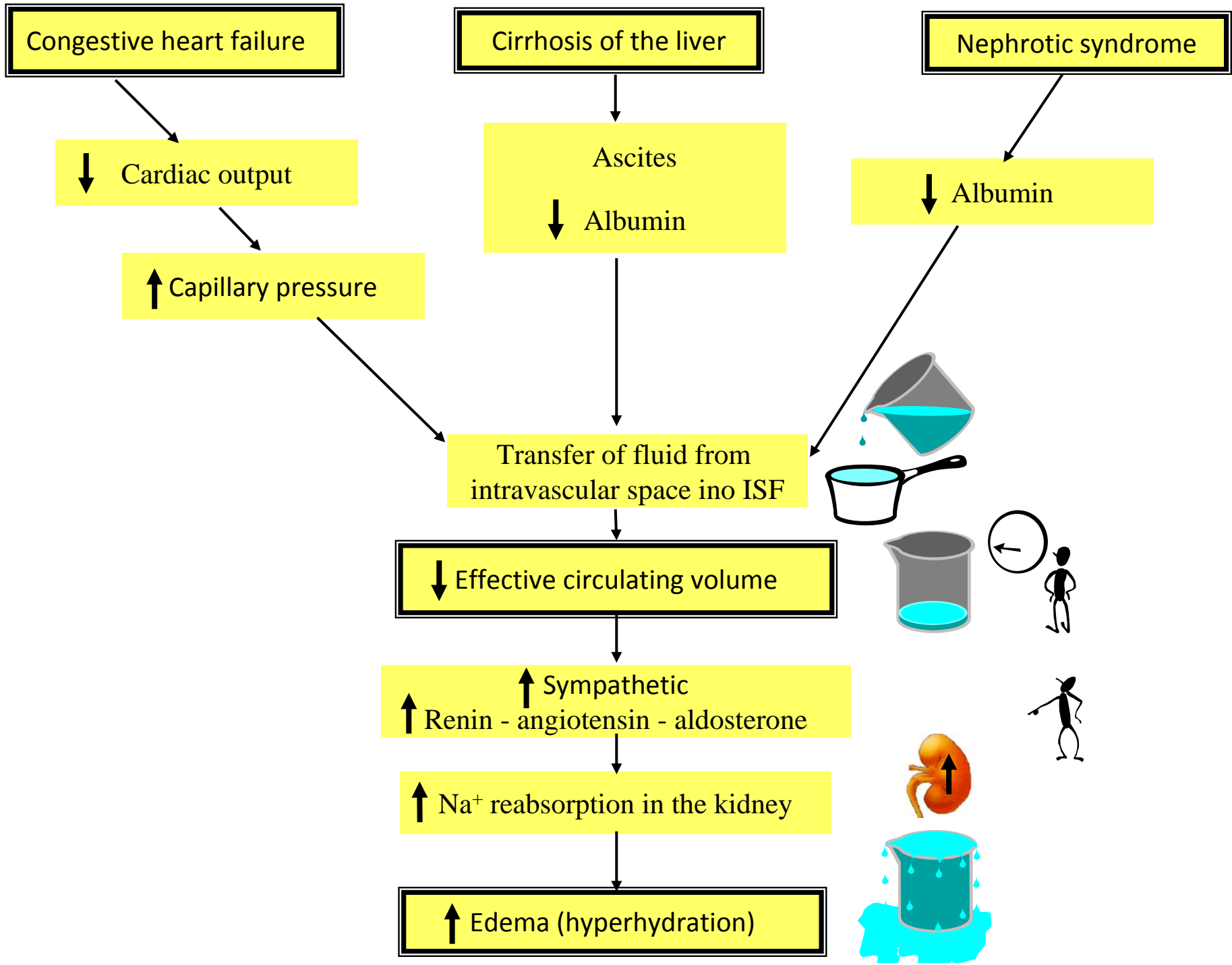
... and to correction of
reduced volume of ICF

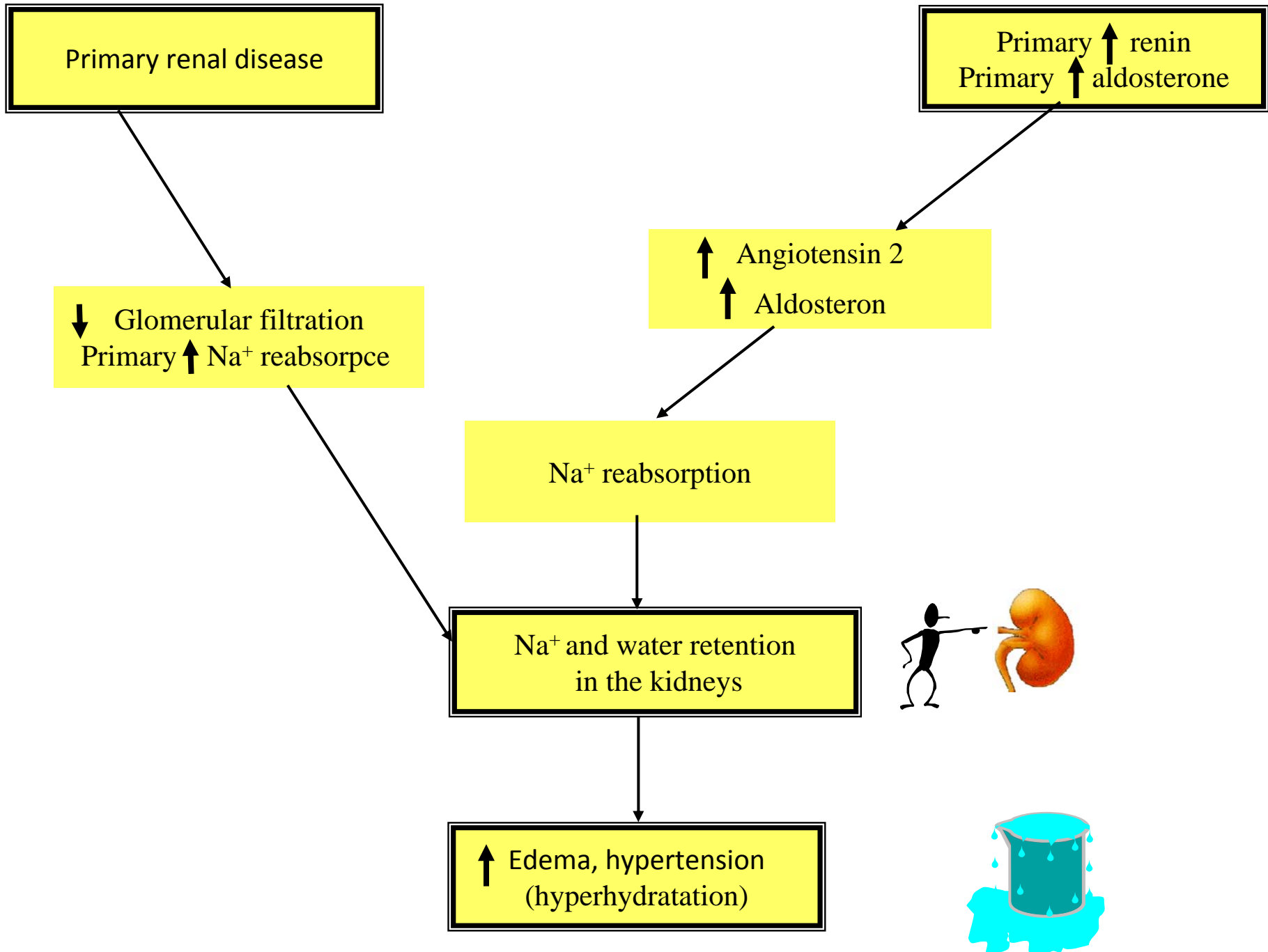
The active changes in cellular osmolarity occurs mainly in brain tissue

Hypotonic hyperhydration

Cause: The greater water retention than soluti retention







Primary renal disease

Primary ↑ renin
Primary ↑ aldosterone

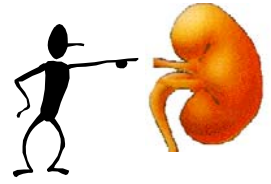
↓ Glomerular filtration
Primary ↑ Na⁺ reabsorpce

↑ Angiotensin 2
↑ Aldosterone

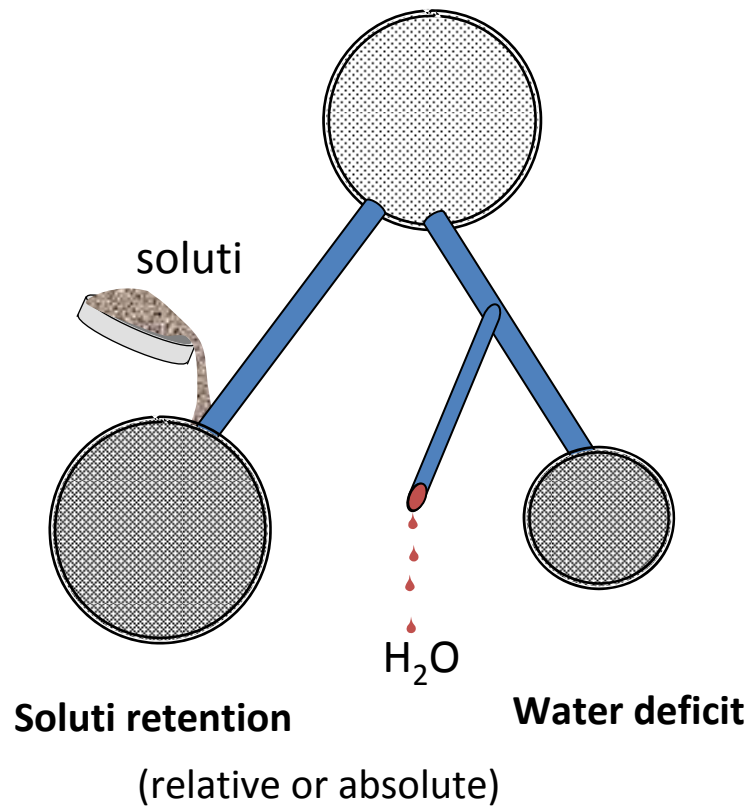
Na⁺ reabsorption

Na⁺ and water retention
in the kidneys

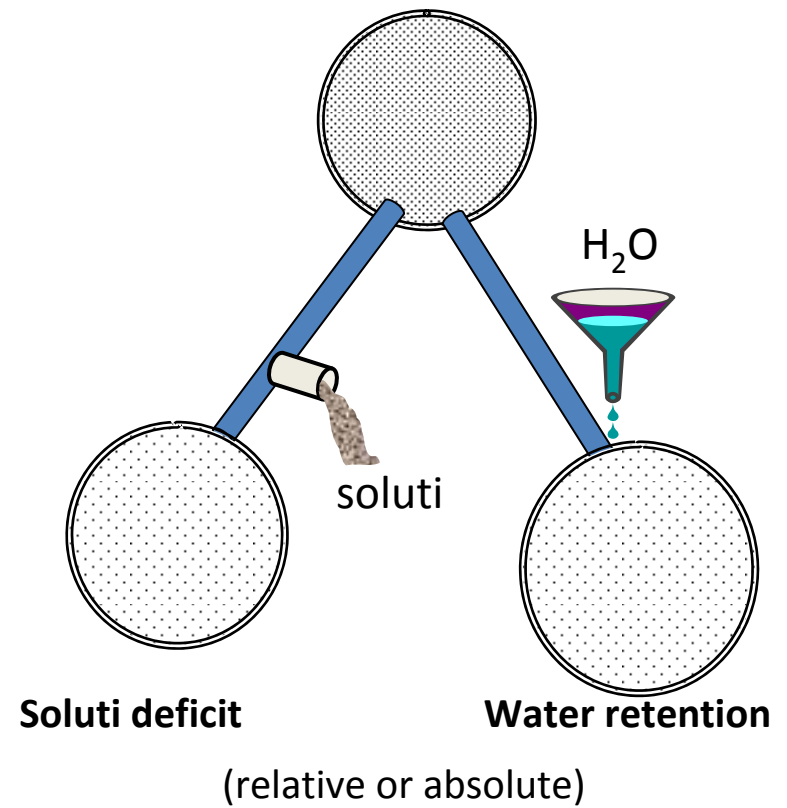
↑ Edema, hypertension
(hyperhydratation)

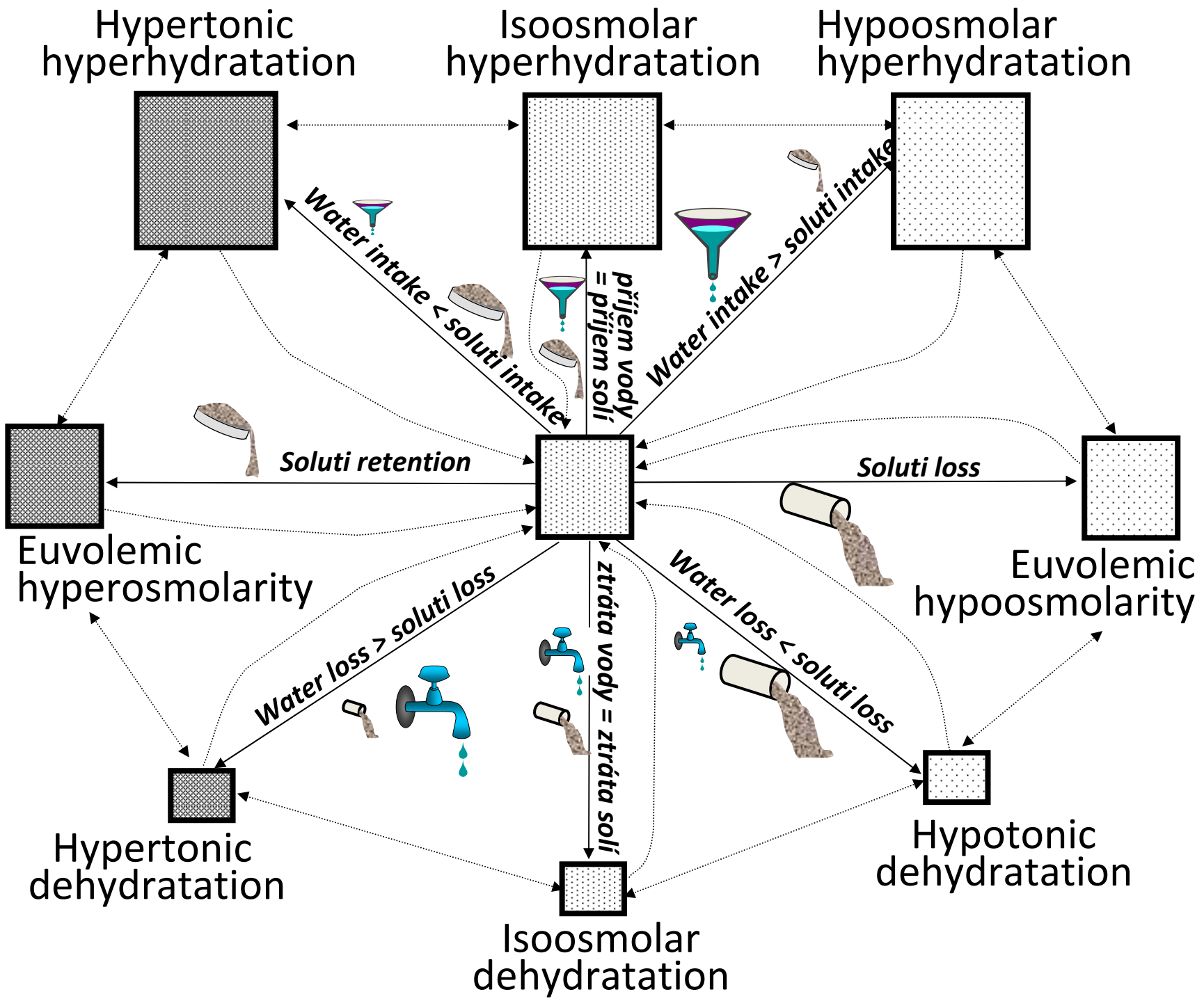


Causes of hyperosmolarity

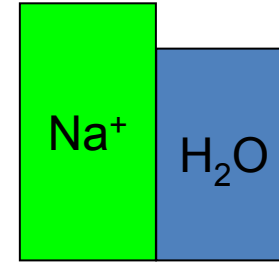
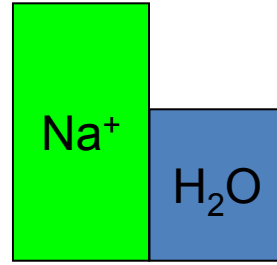
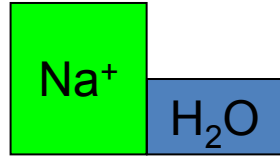


Causes hyposmolarity

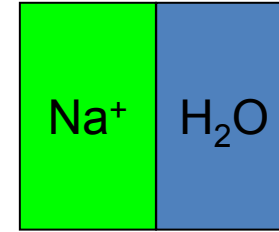
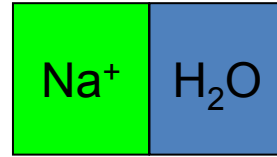
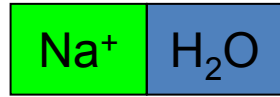




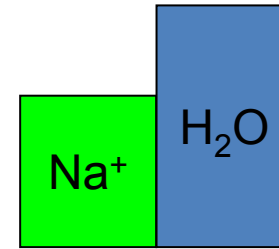
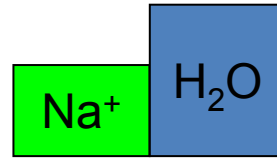
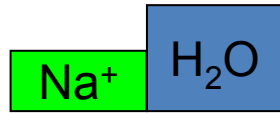
hypernatremia



normonatremia
(140 mmol/l)



hyponatremia

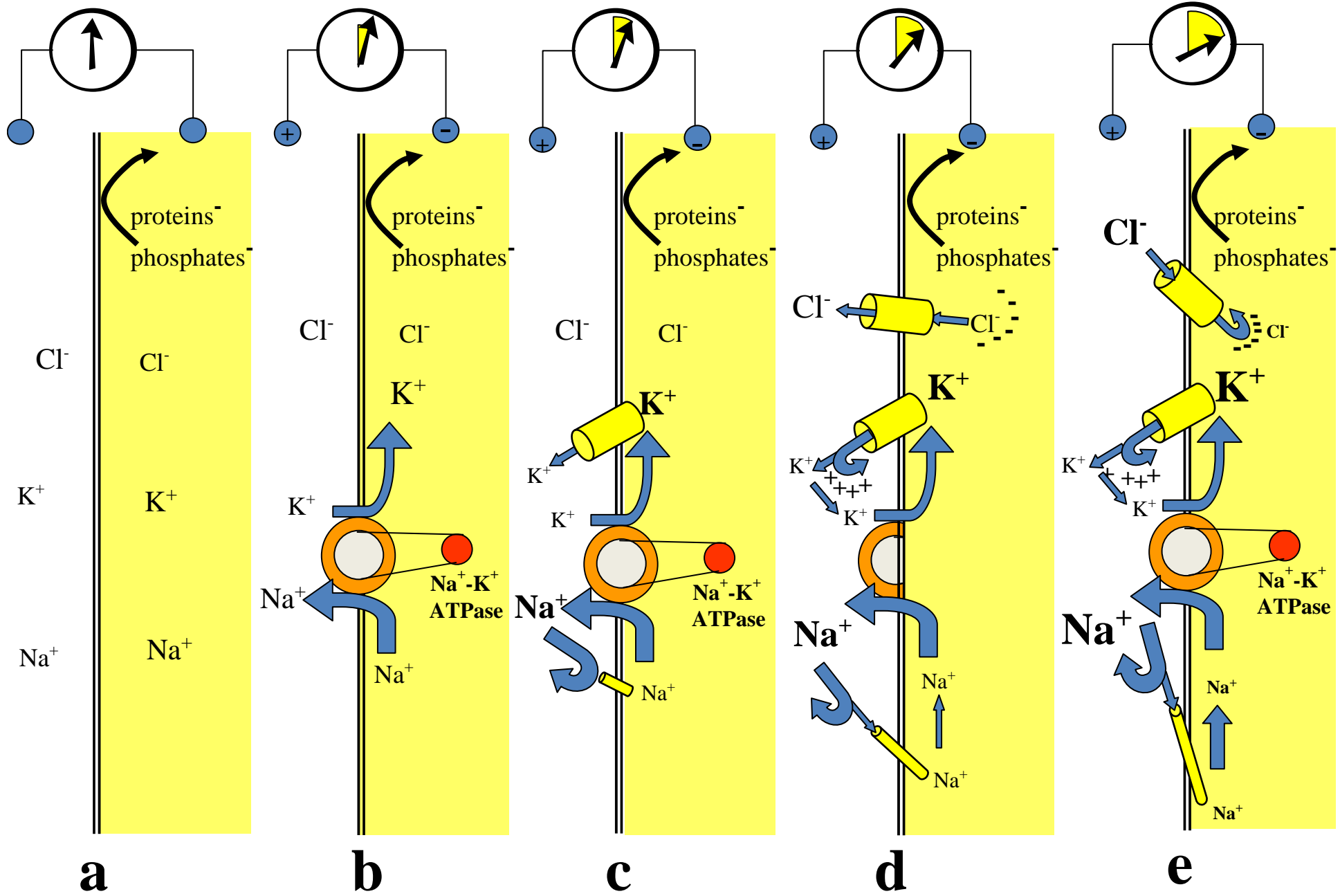


hypovolemia
(dehydration)

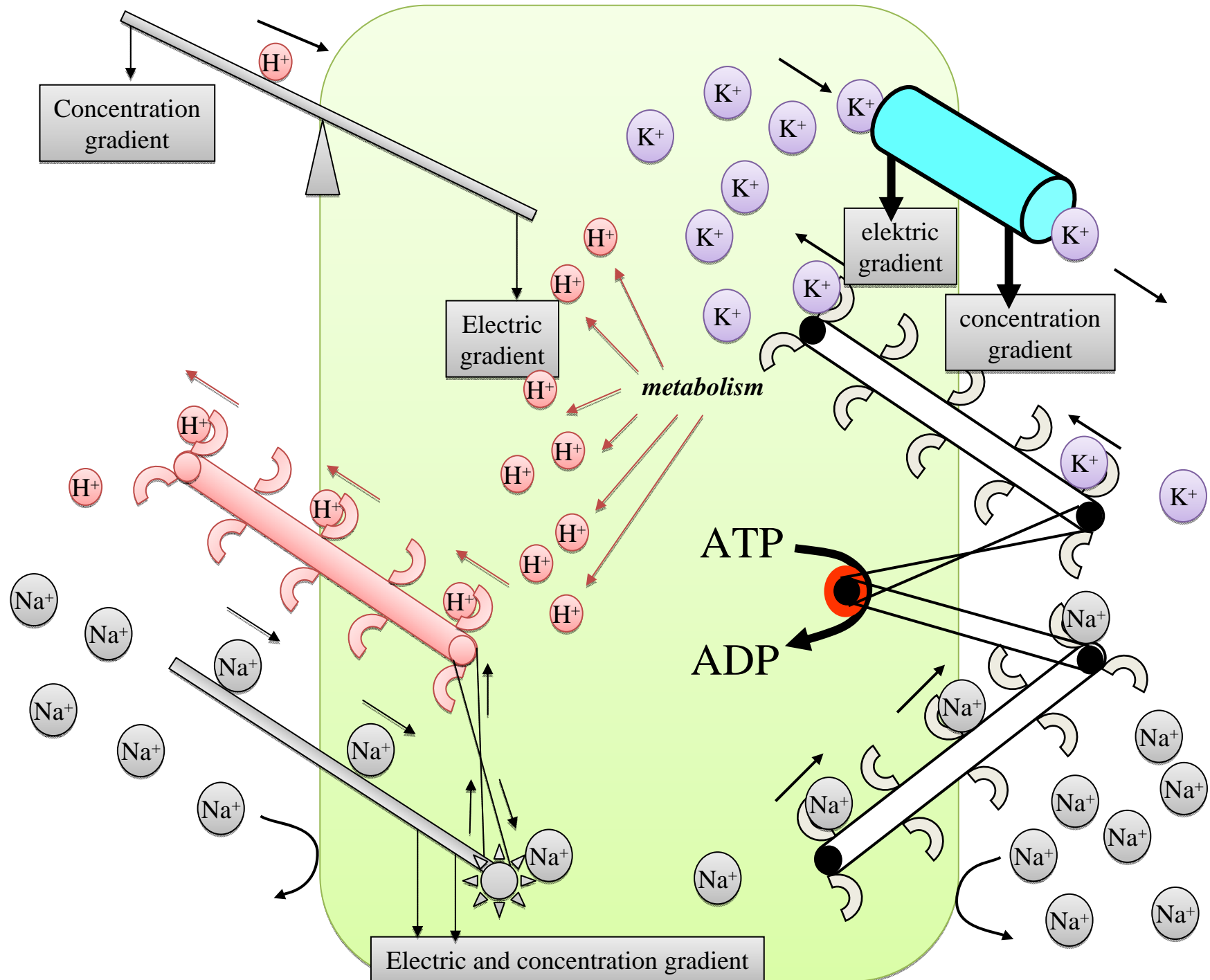
isovolemia

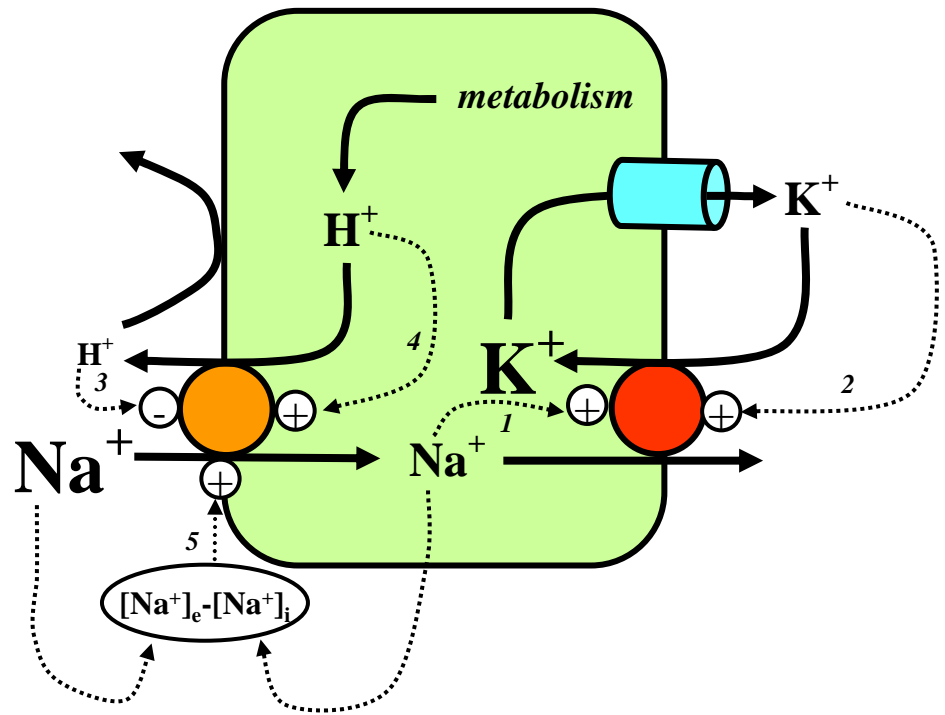
hypervolemia

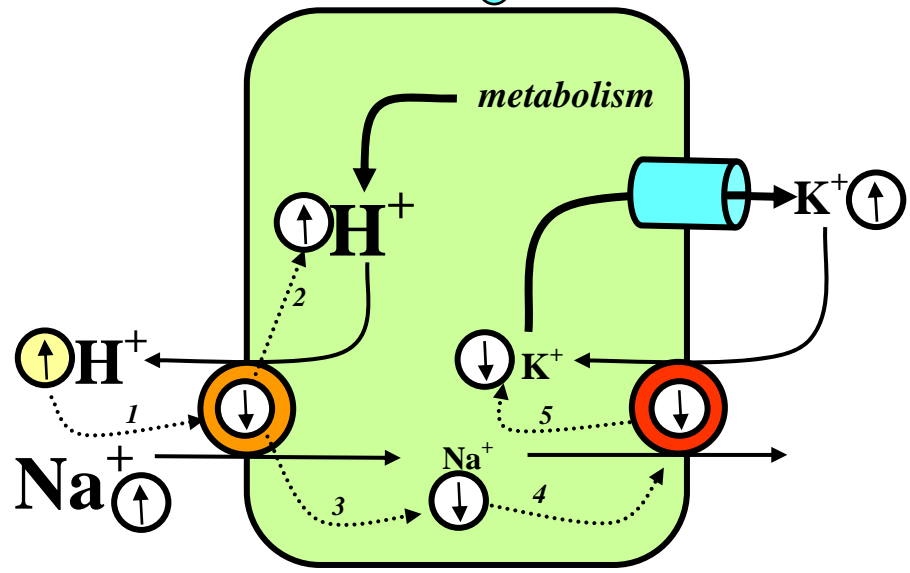
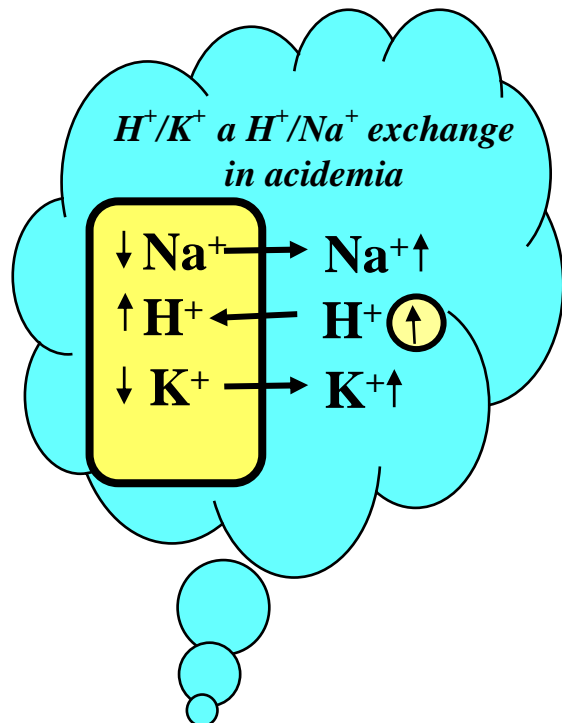
K^+ - main intracellular cation



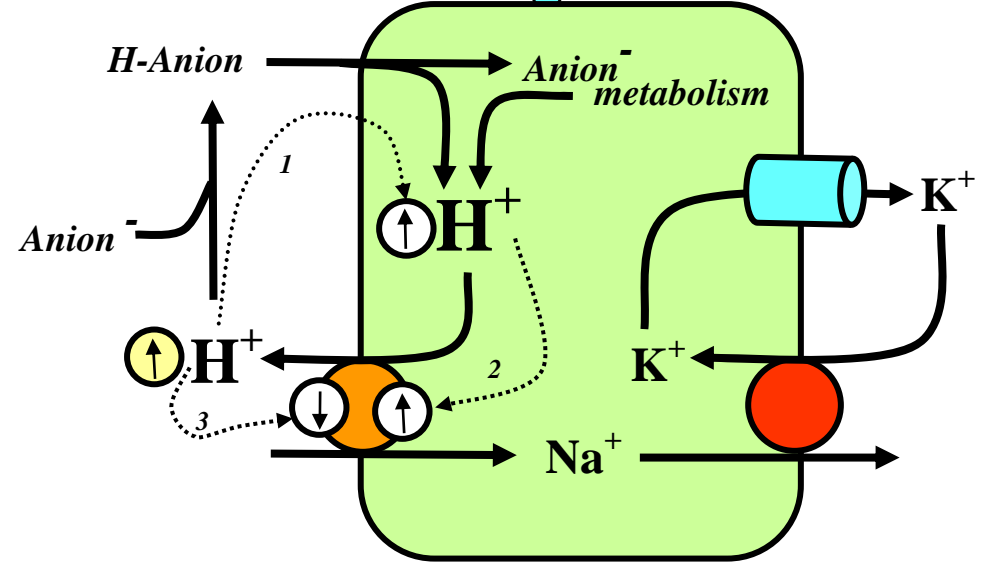
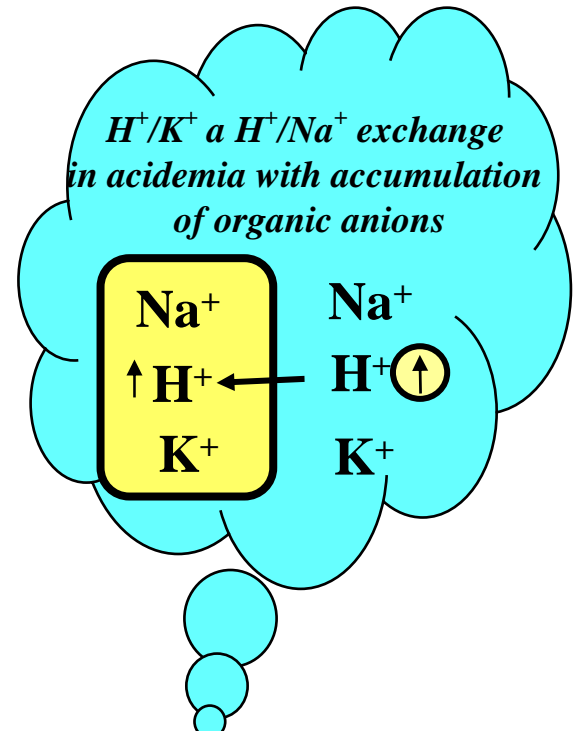
Potassium and Acid-Base



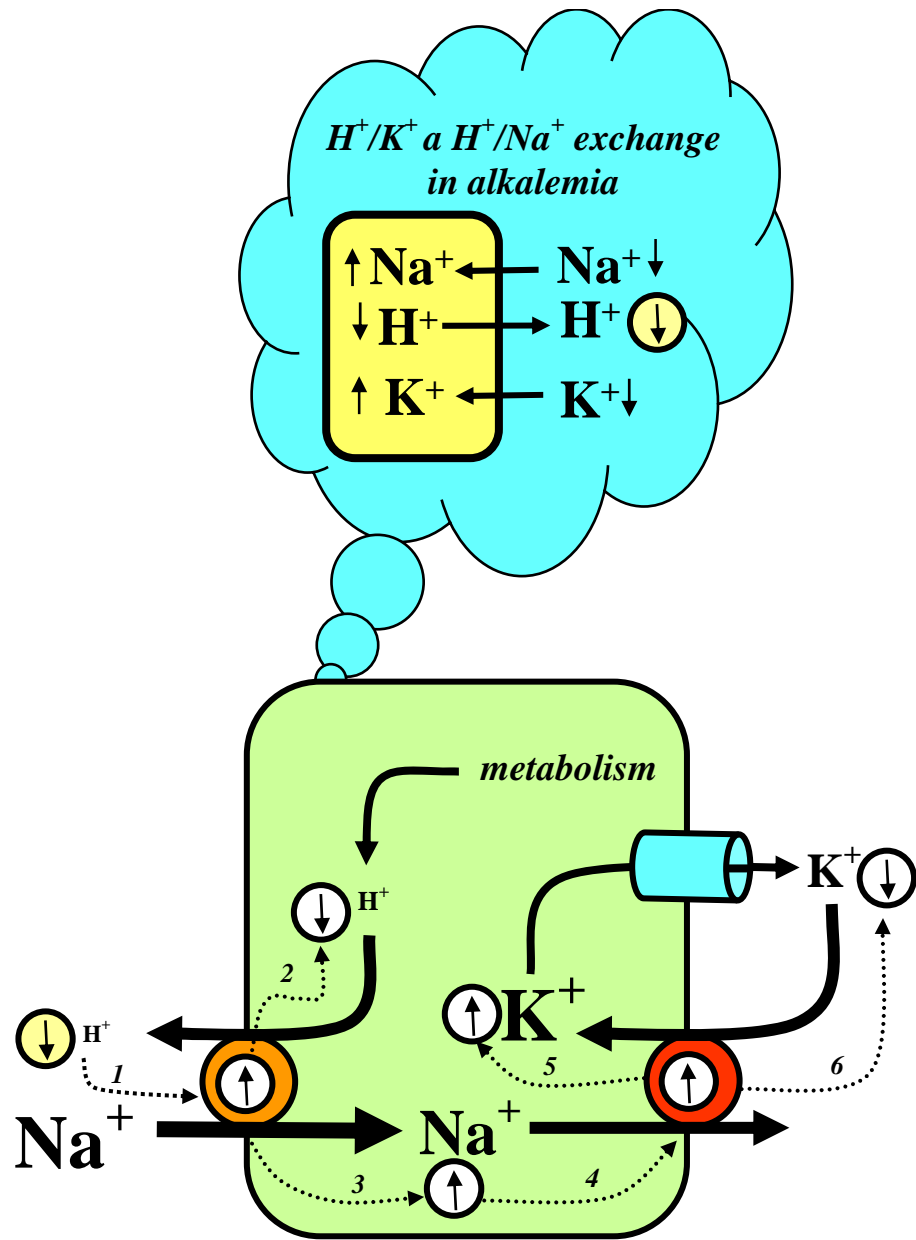




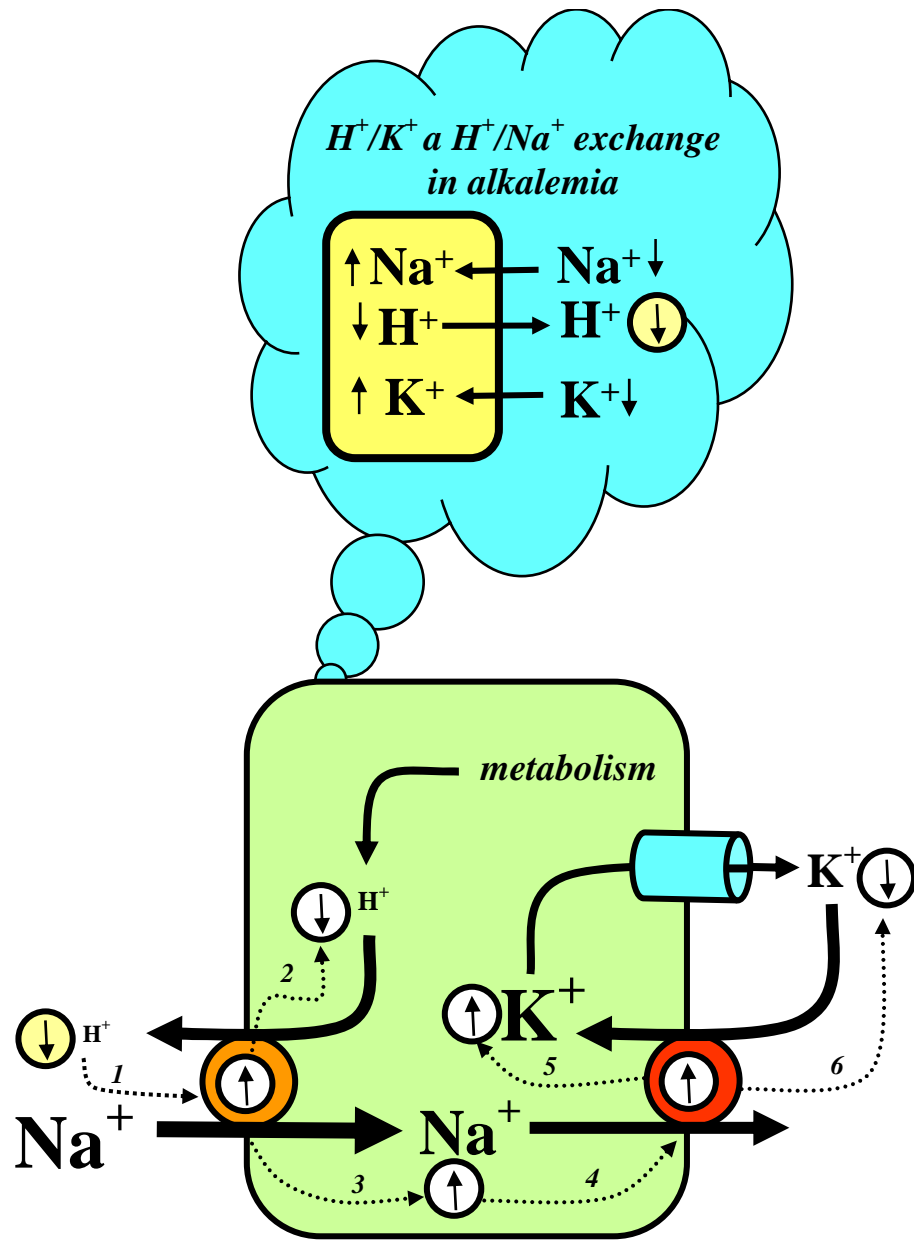
a) acidemia, no accumulation of organic anions



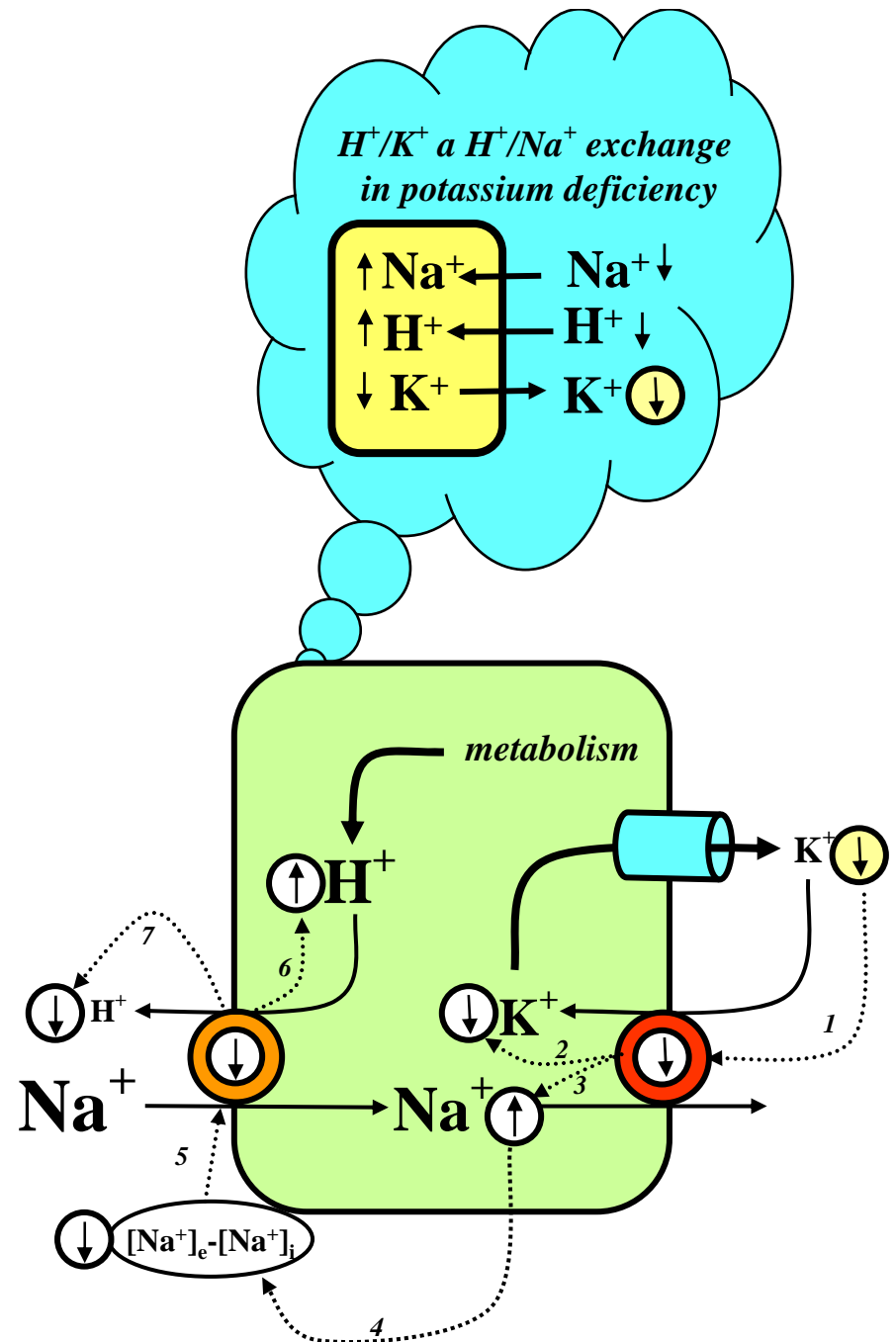
a) acidemia with the accumulation of organic anions



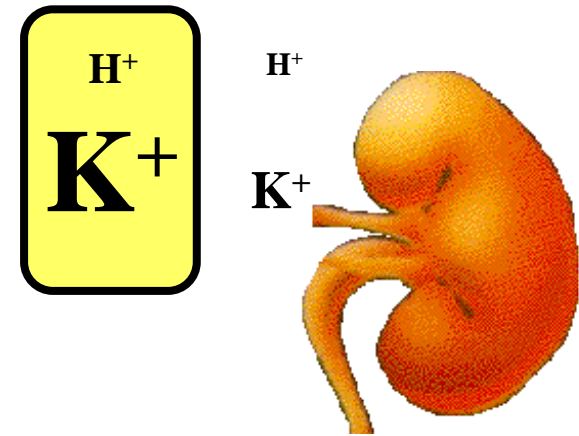
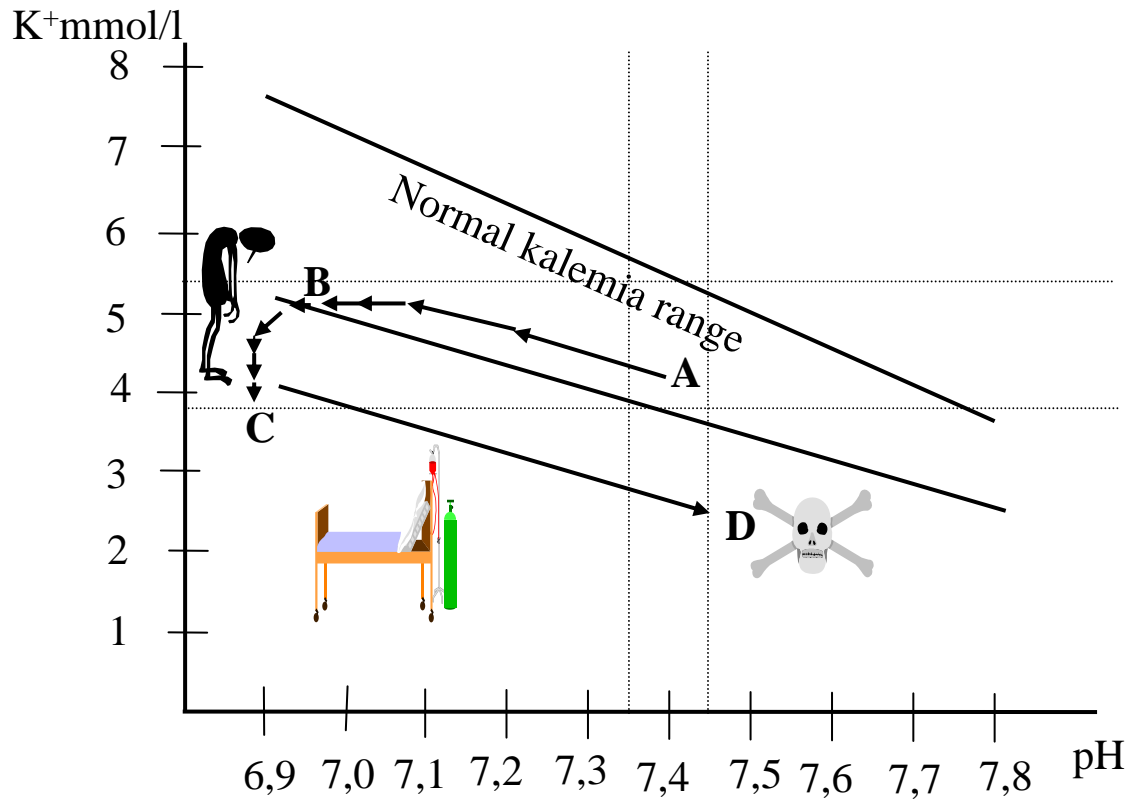
a) Alkalemia without potassium deficiency



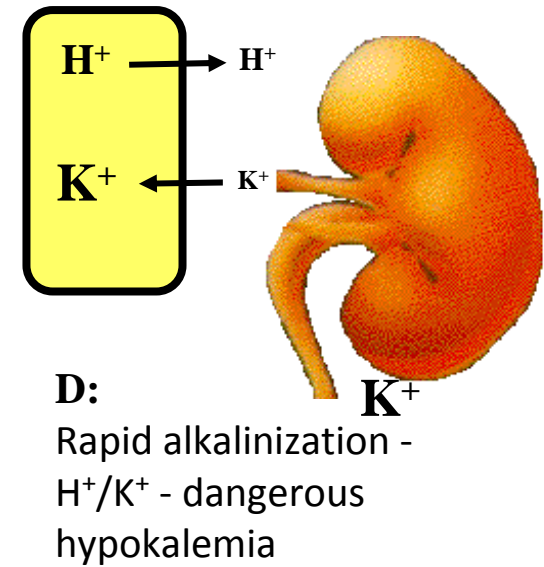
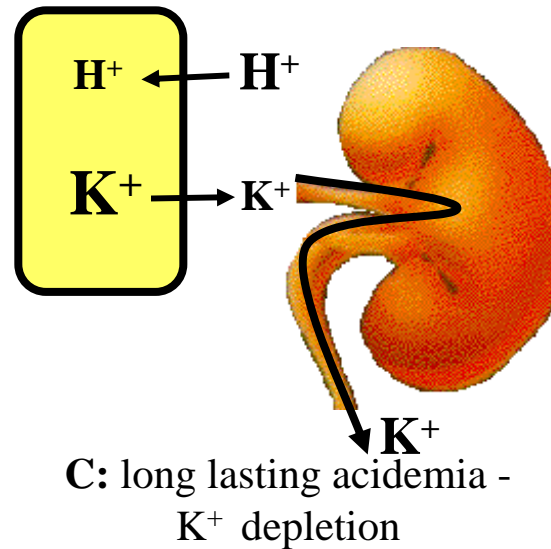
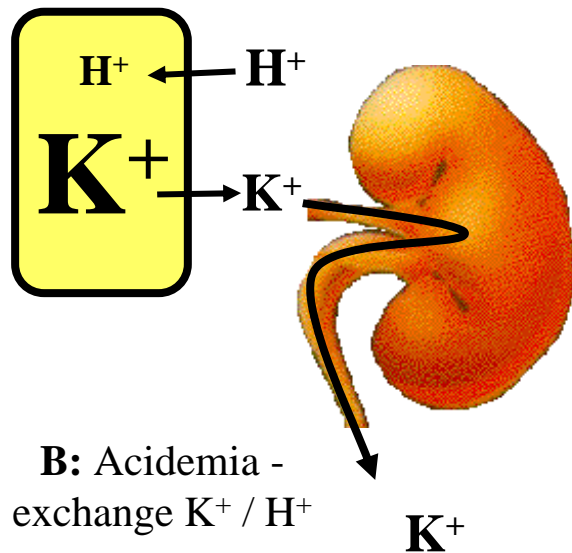
a) Alkalemia without potassium deficiency



a) Alkalemia with potassium deficiency

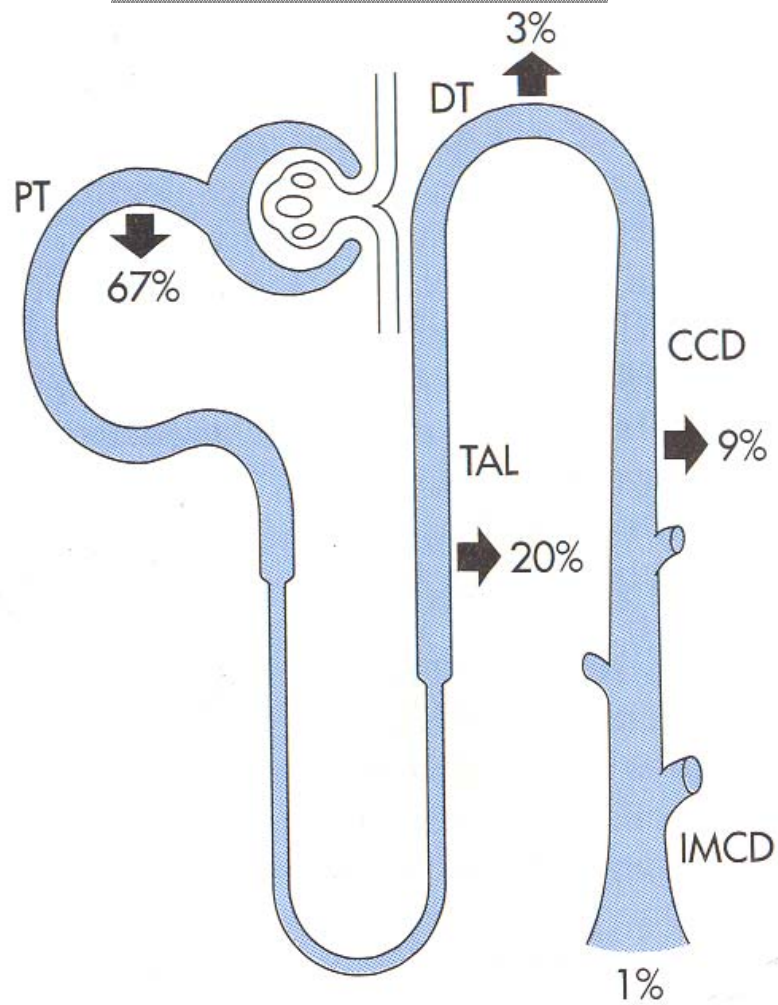


A: Norm

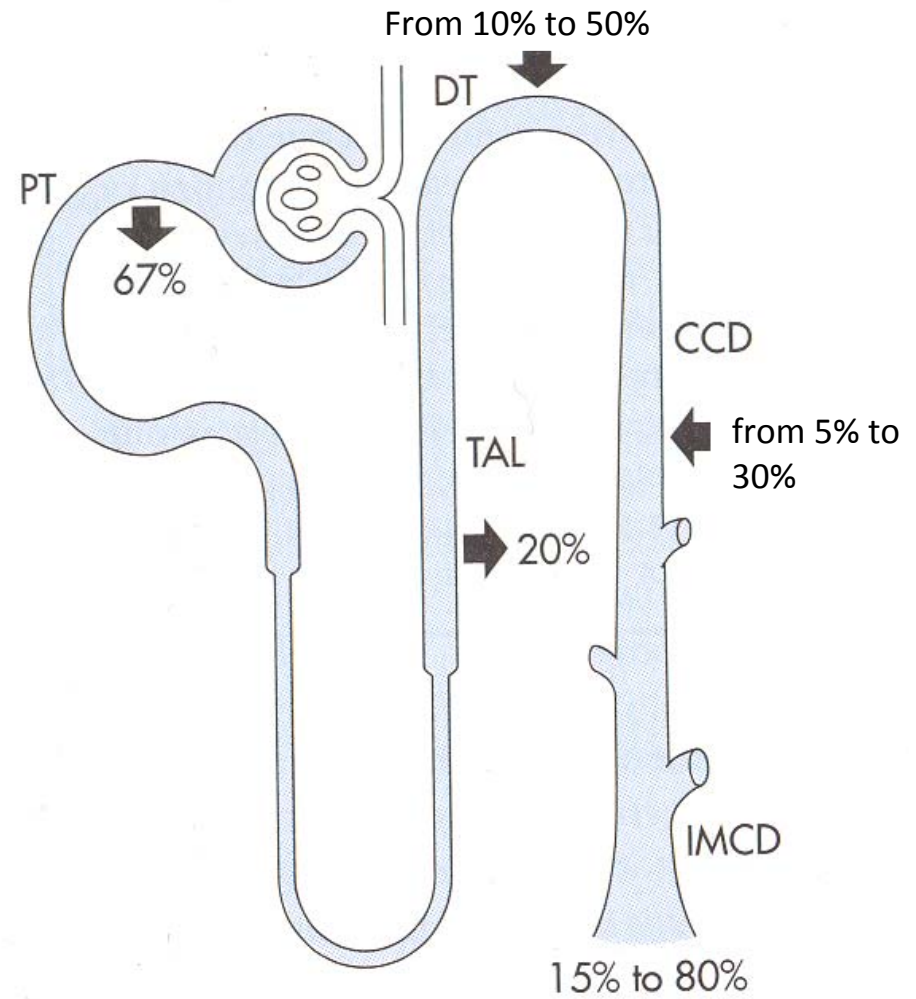


K⁺

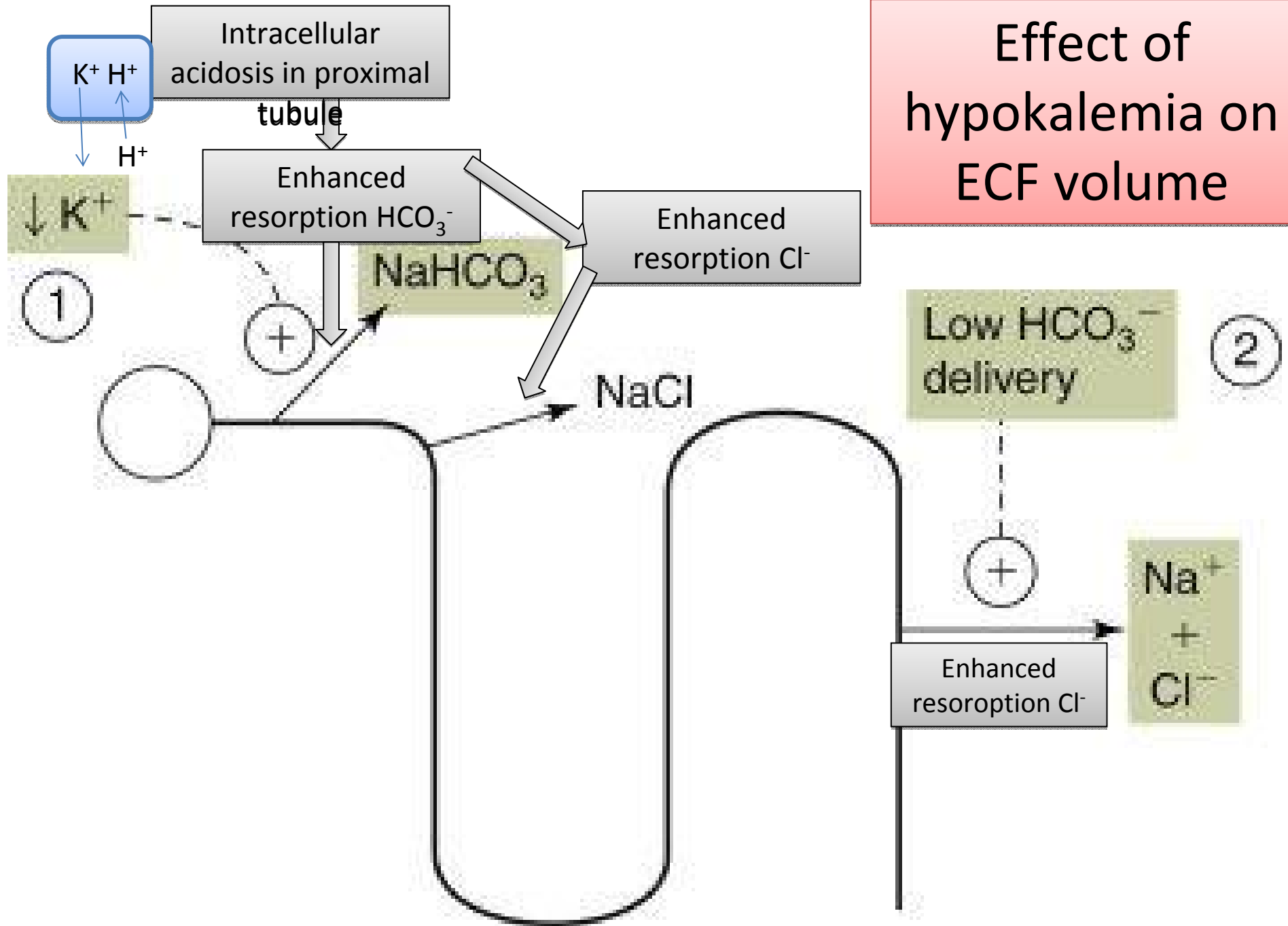
Potassium depletion



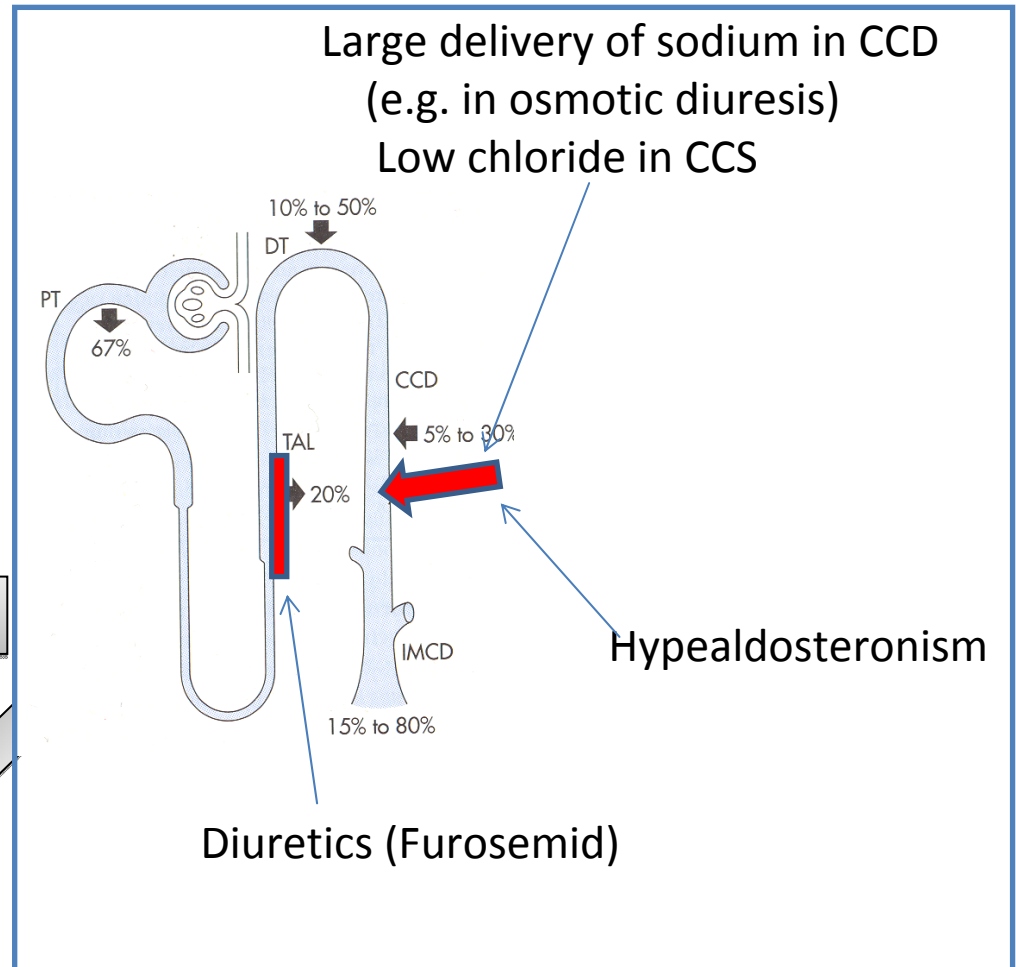
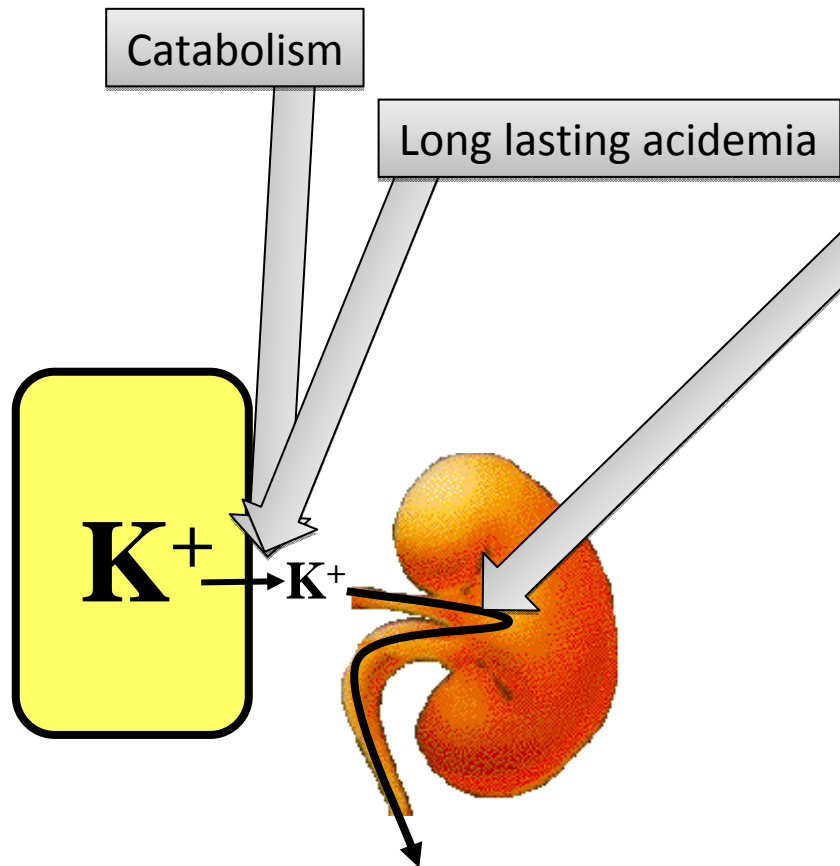
Normal or increased intake of potassium



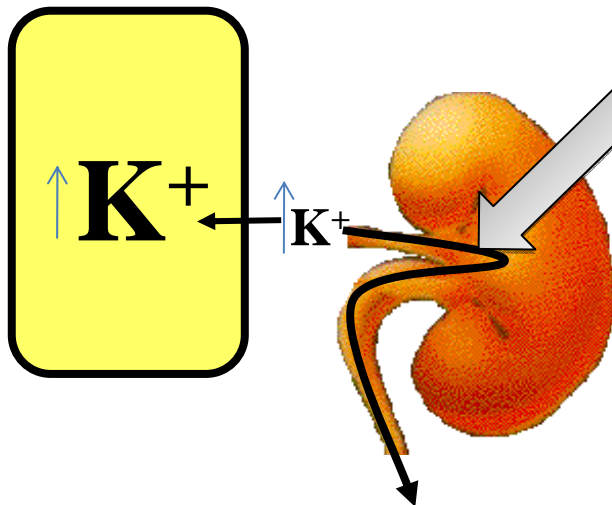
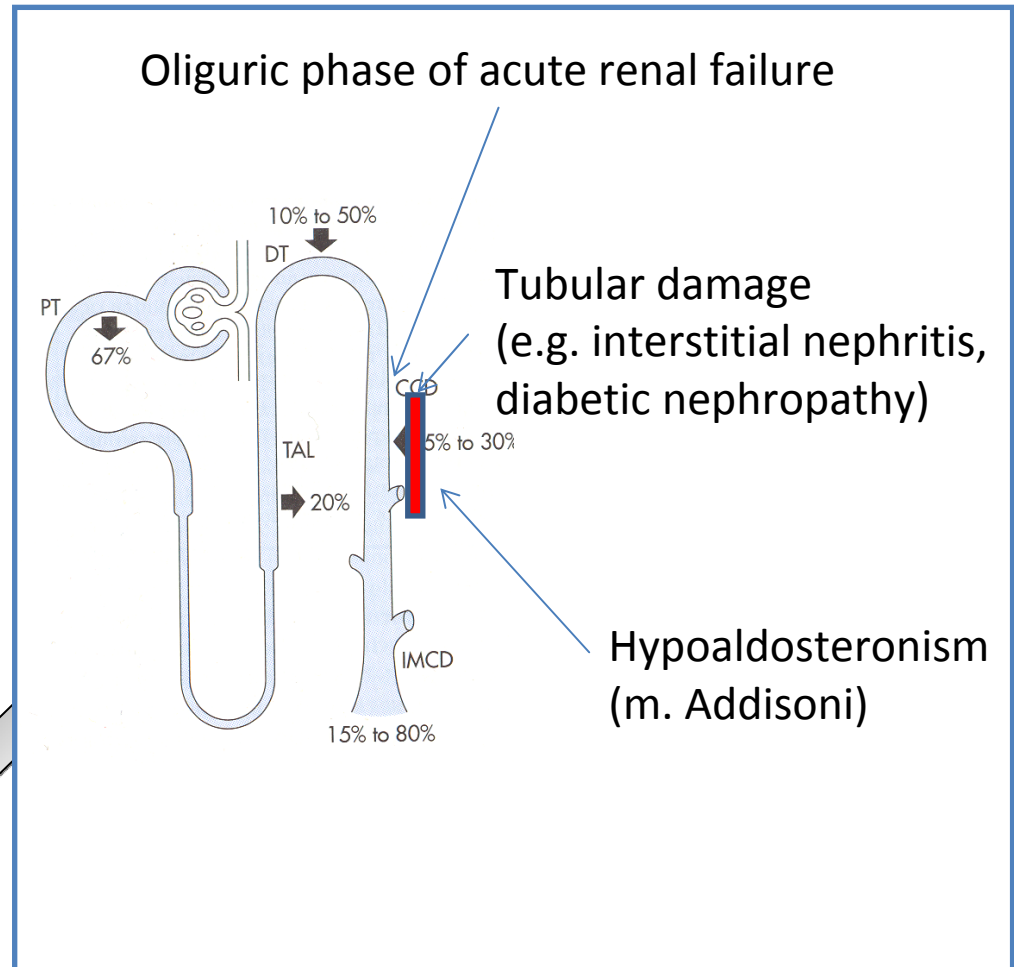
Effect of hypokalemia on ECF volume



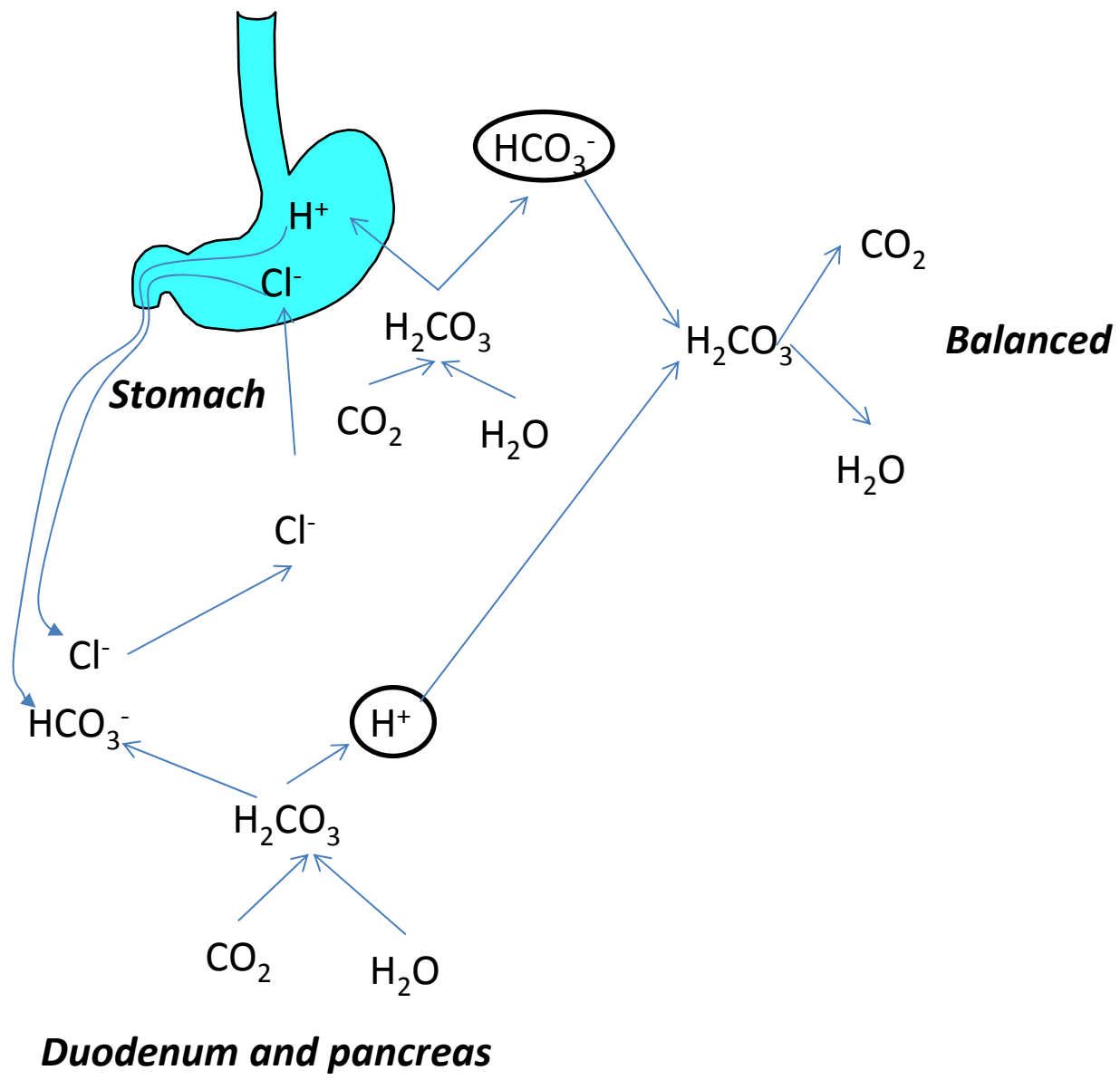
Potassium depletion

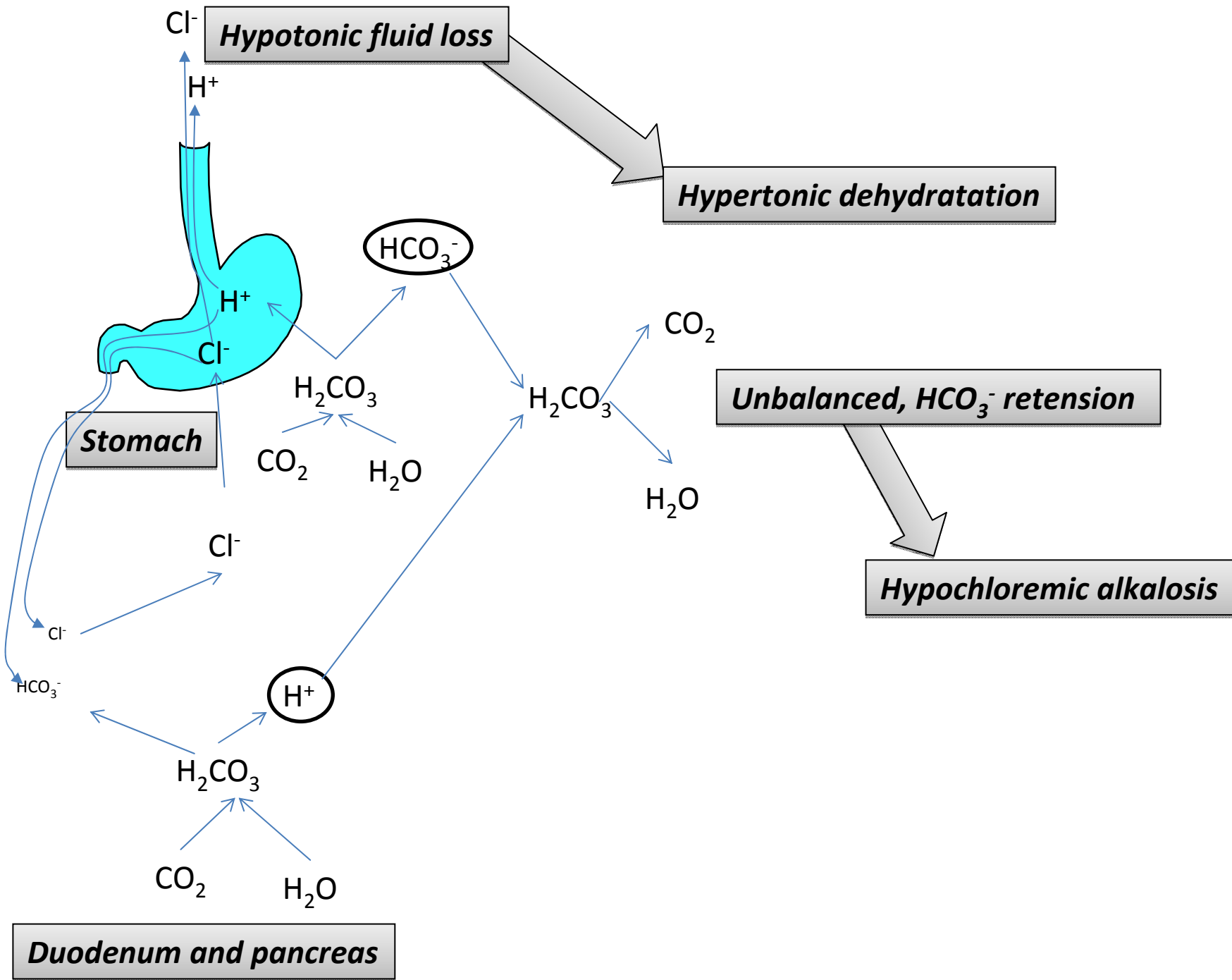


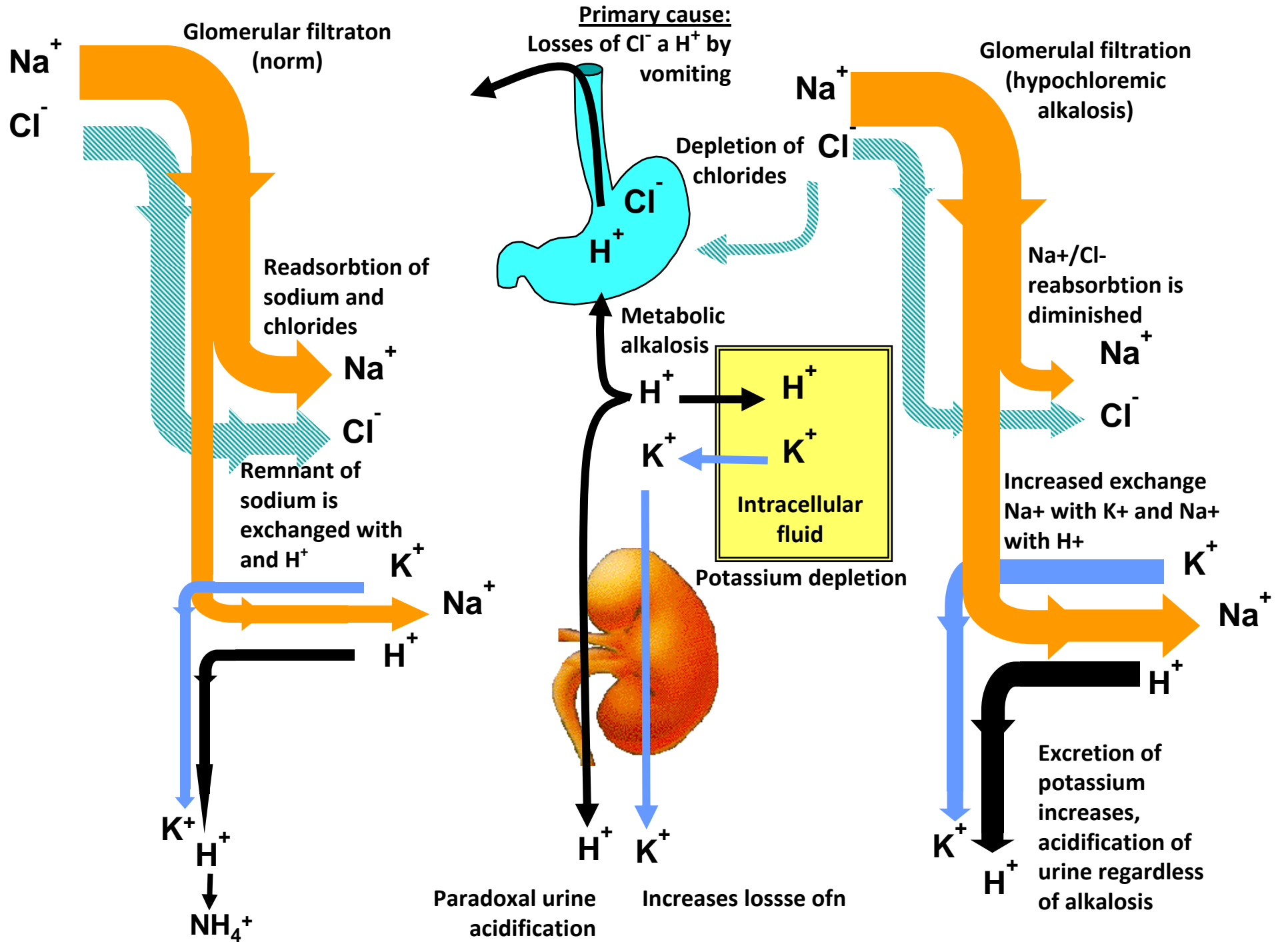
Potassium retention



Vomiting

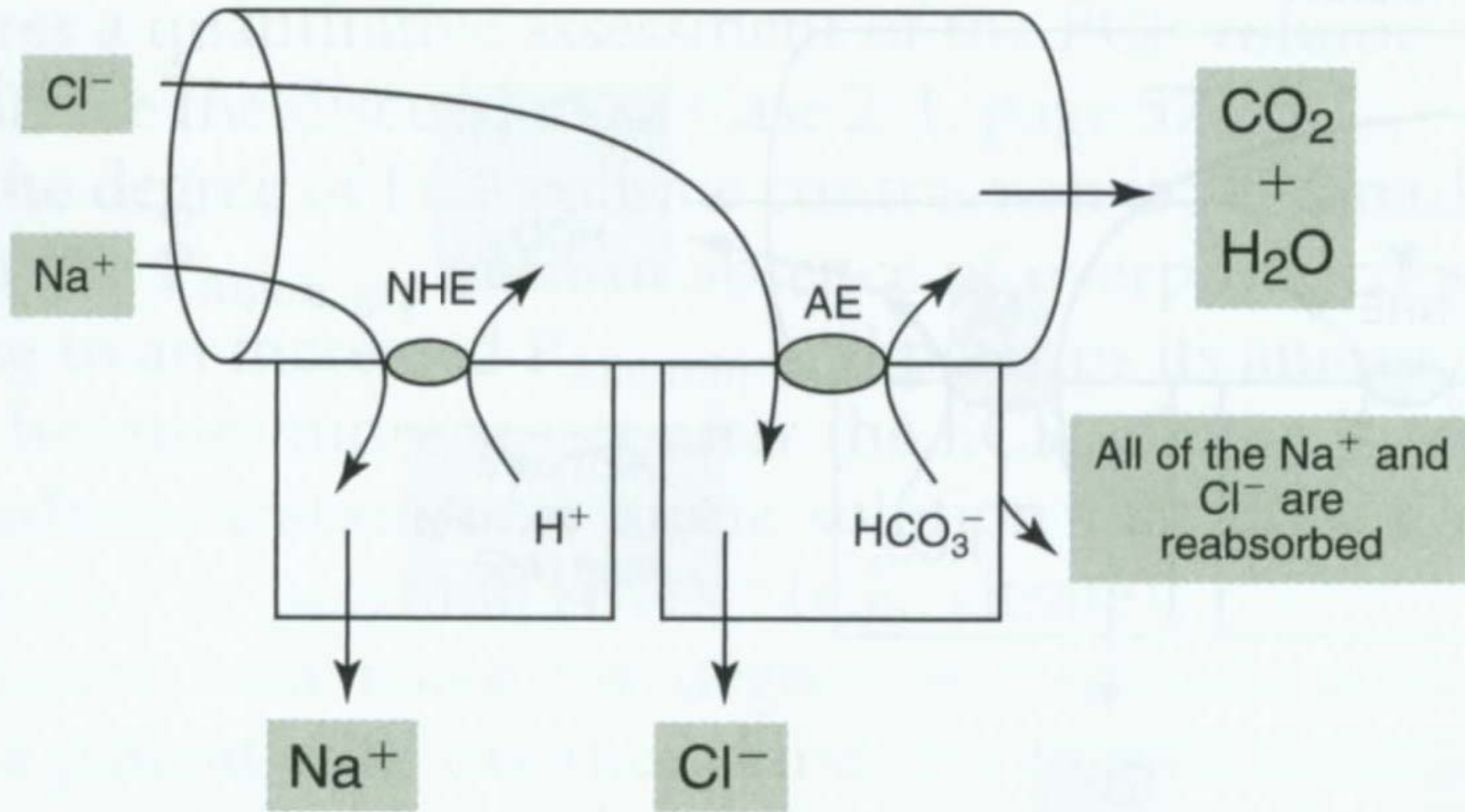


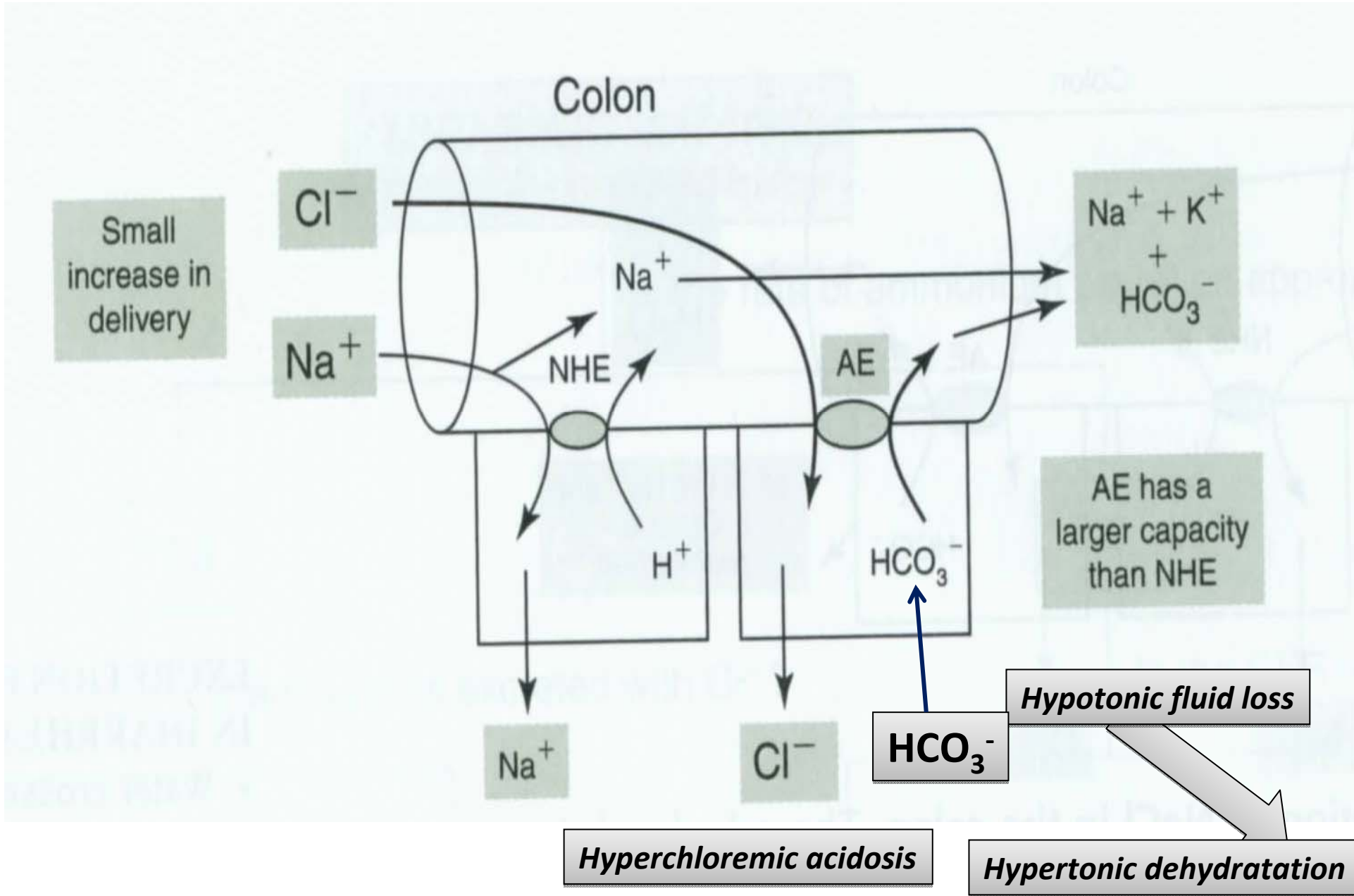




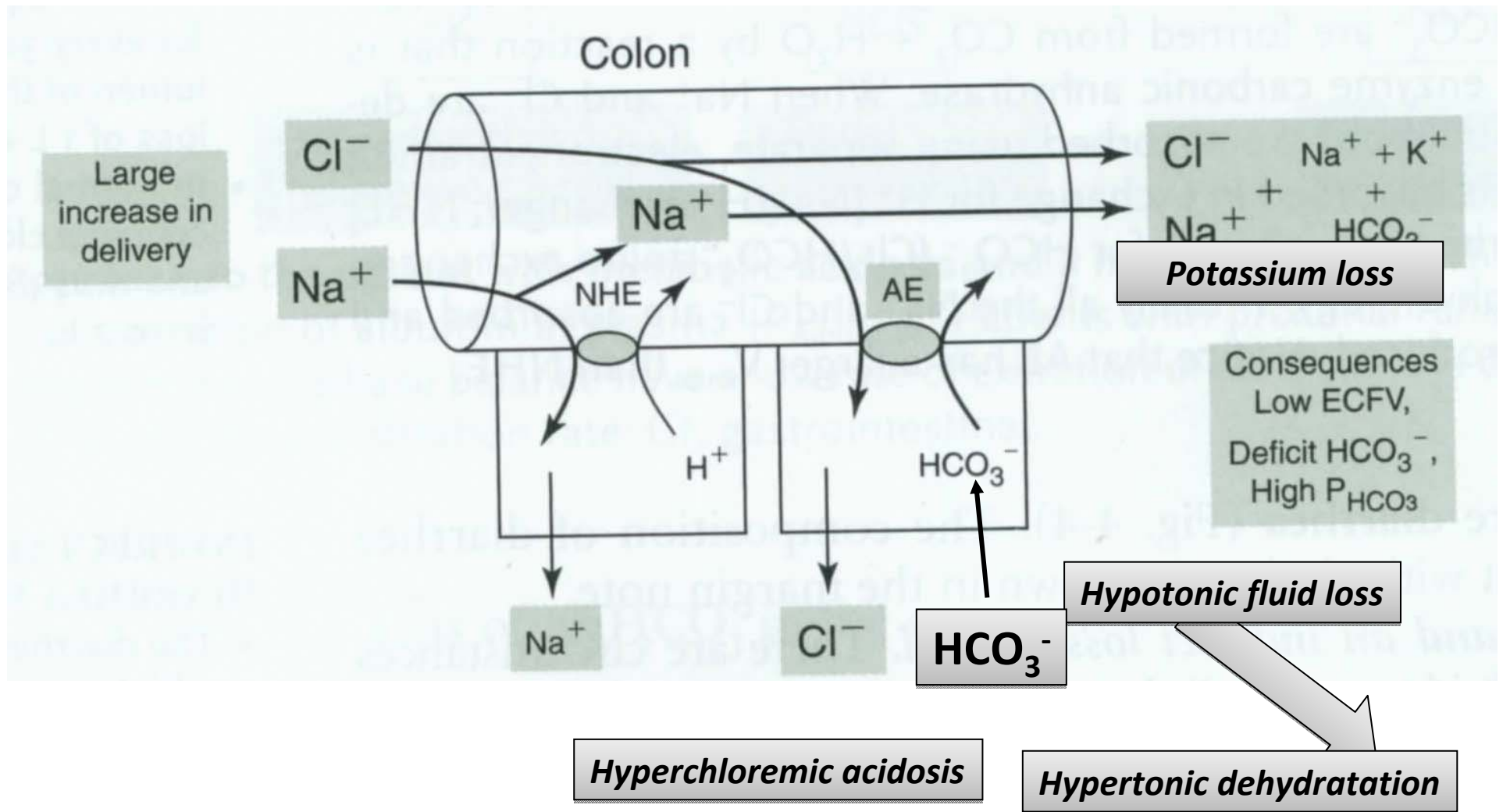
Diarrhoea

Colon

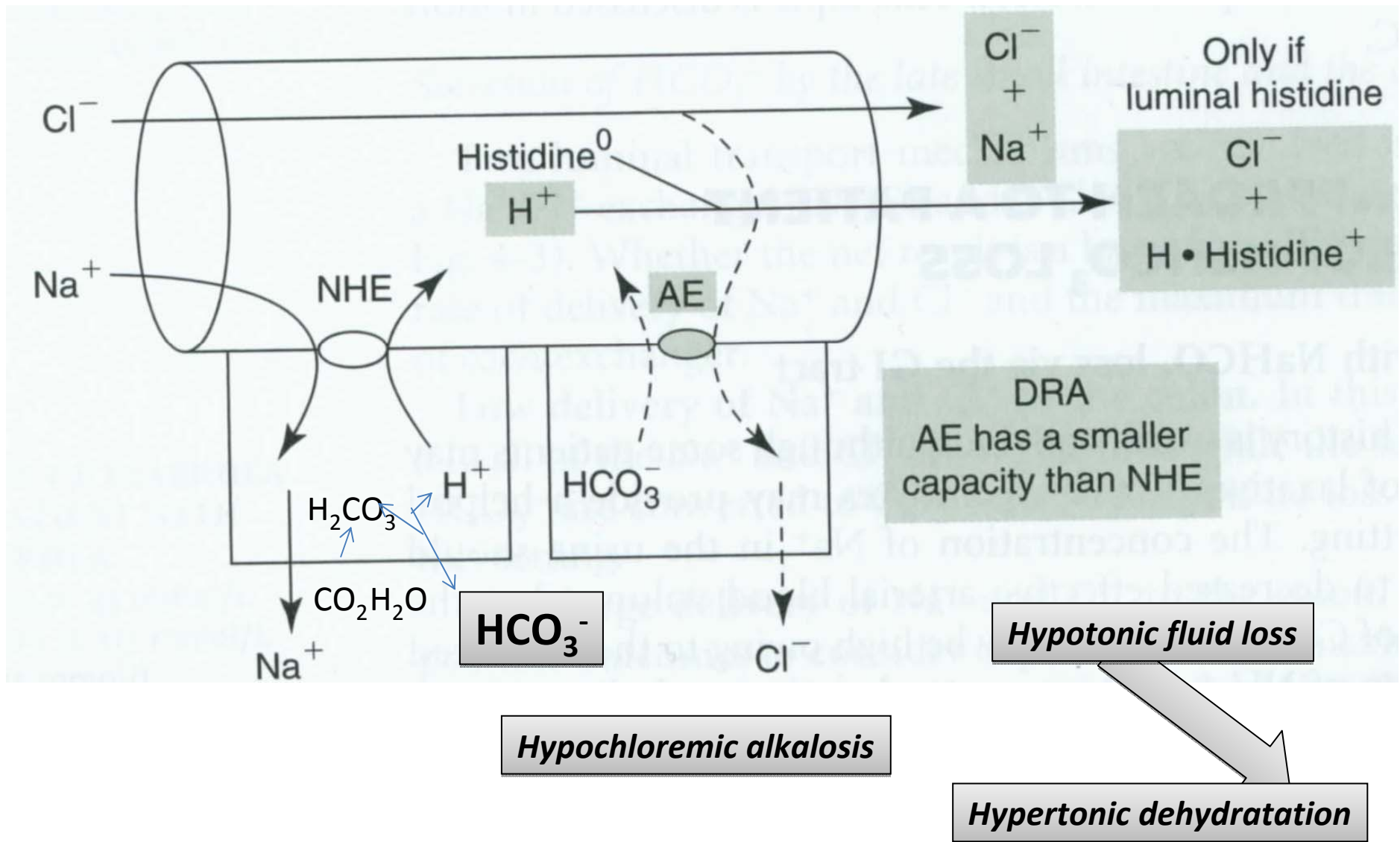




Alkalic diarrhoea



Severe alkalic diarrhoea



Acidic diarrhoea in DRA, down-regulated adenoma

Acute volume changes and acid-base disturbances

Pufrační systém plazmy



Respirační regulace

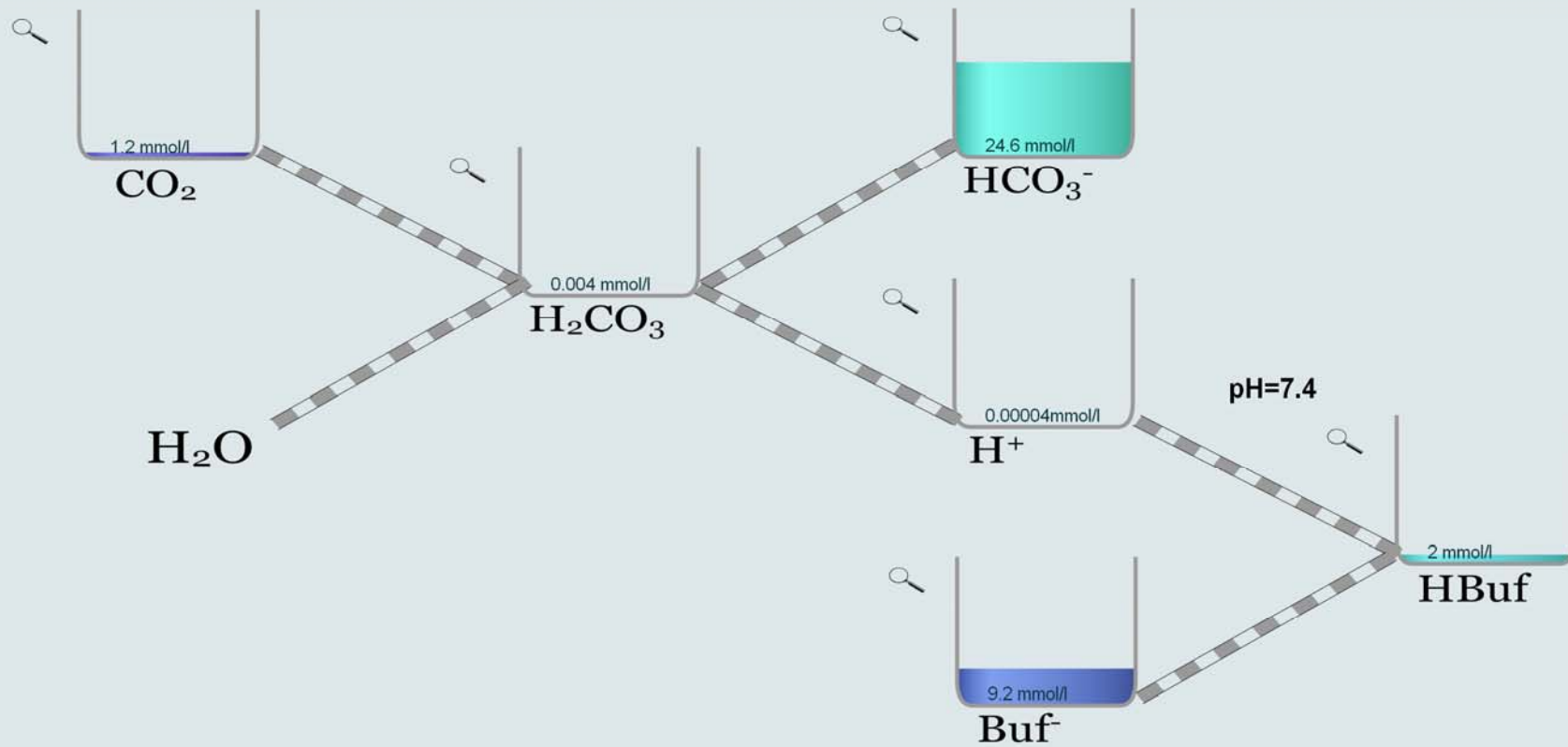
Pufrační ekvilibrace

Pouze bikarbonátový pufr

Renální regulace

Zobraz karbonáty

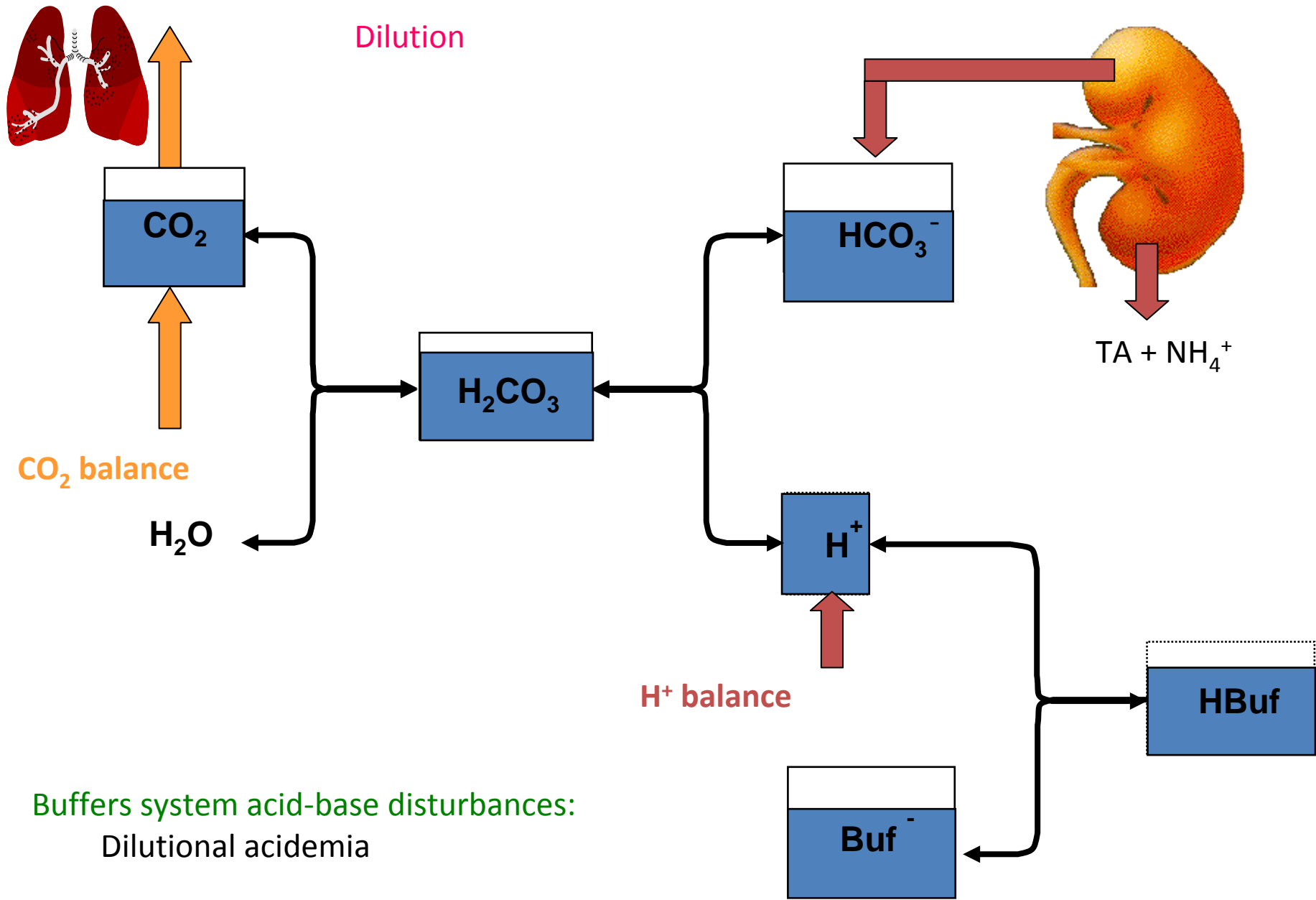
$[\text{HCO}_3^-] + [\text{Buf}^-] = 38.1 \text{ mM/l}$ $\text{BE} = 0 \text{ mM/l}$



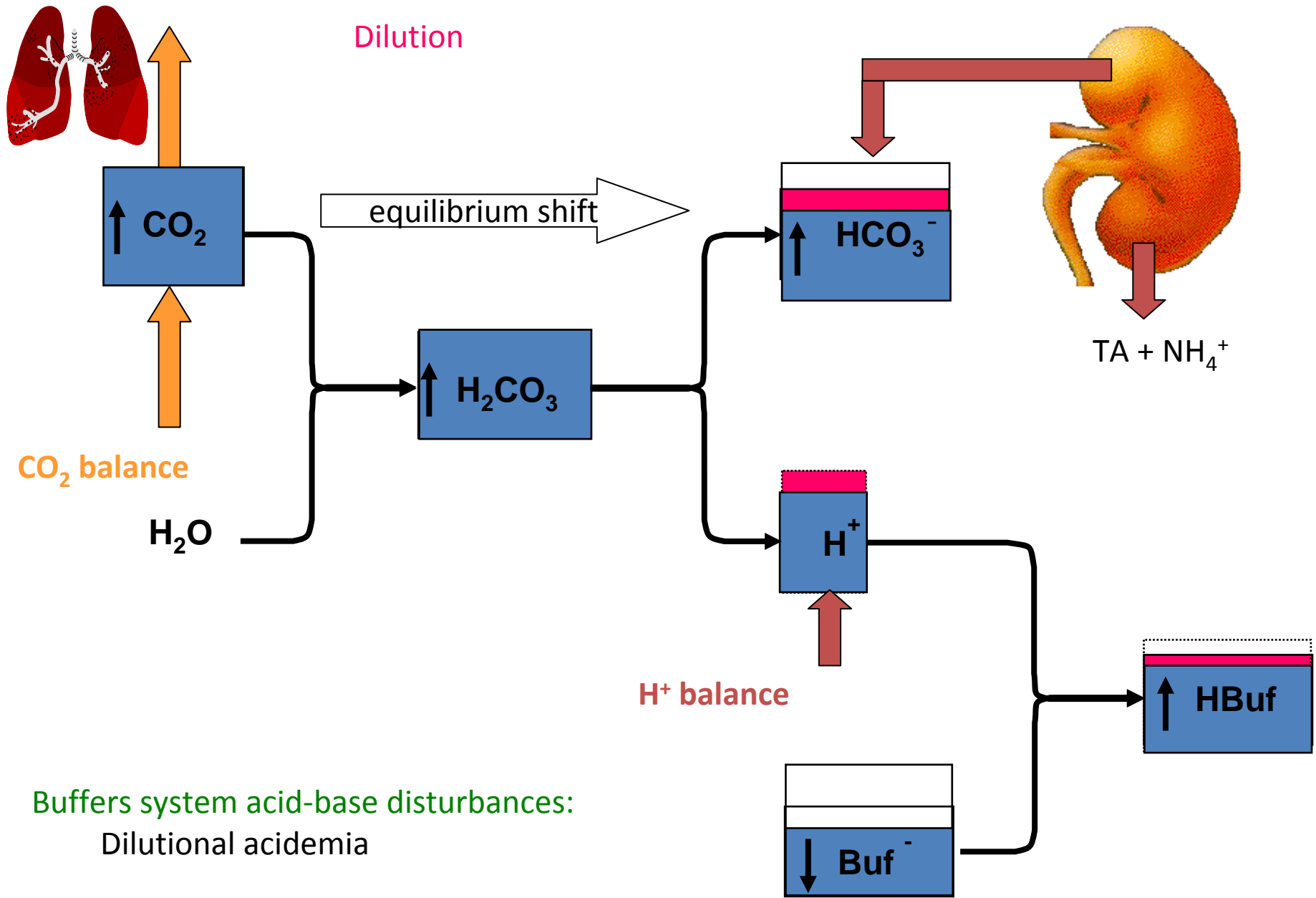
Vše normalizuj!



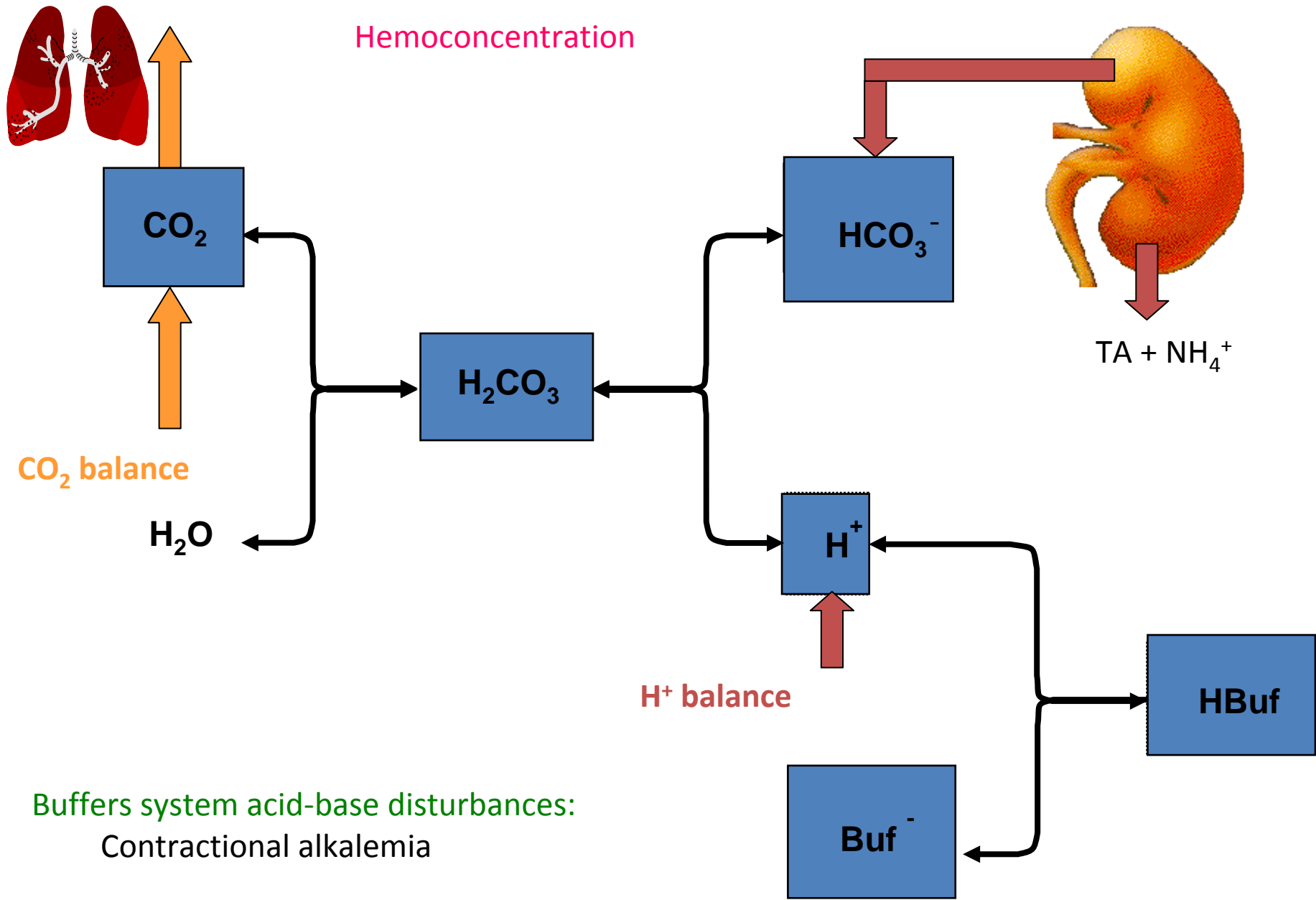
KONCENTRACE - DILUCE



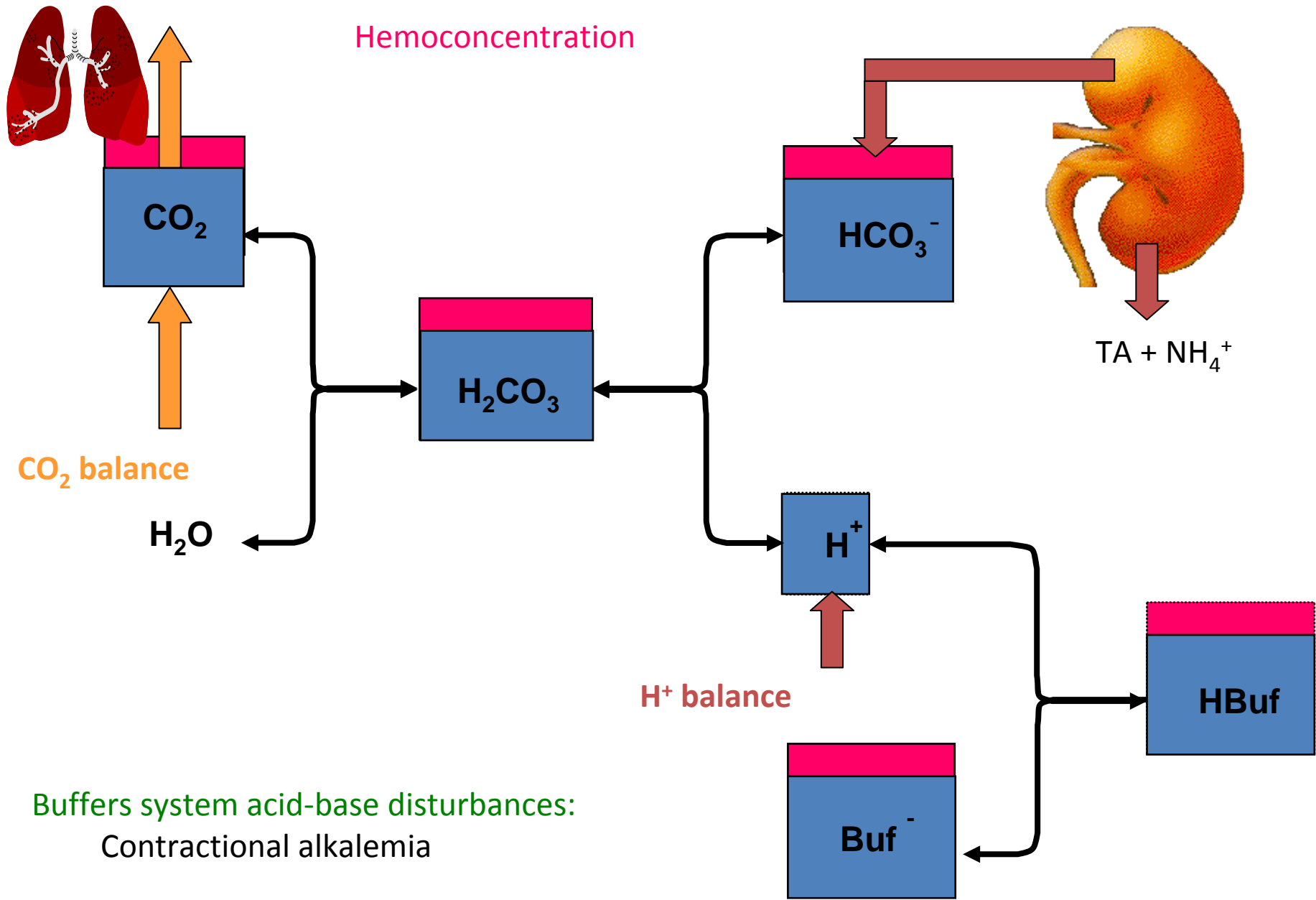
Buffers system acid-base disturbances:
Dilutional acidemia



Buffers system acid-base disturbances:
Dilutional acidemia



Buffers system acid-base disturbances:
 Contractional alkalemia



Buffers system acid-base disturbances:
 Contractural alkalemia